

Supporting Information

Graphene oxide induced multi-layered six-petal flower-shaped rare earth Tb³⁺ hybrid luminescent material: Synthesis, characterization, luminescence and fluorescent anti-counterfeiting properties

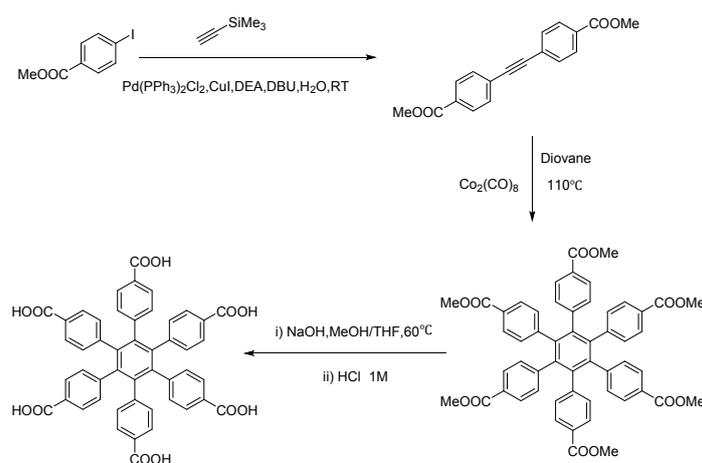


Fig. S1 Schematic illustration of preparation of H₆CPB.

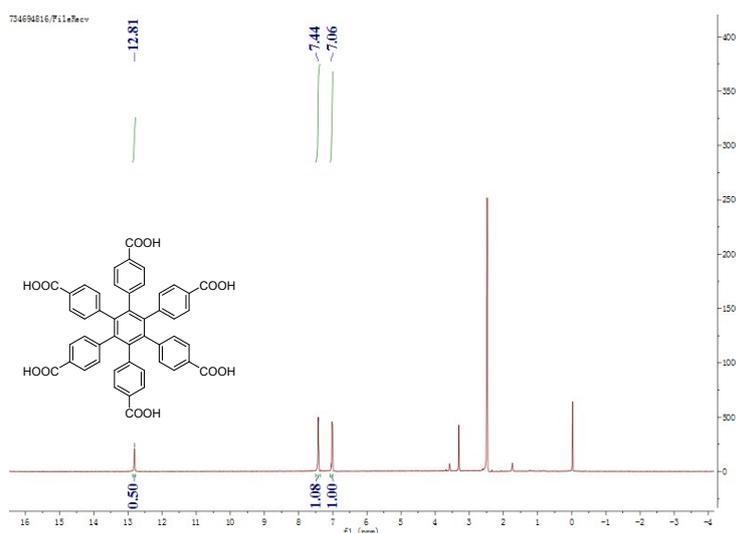


Fig. S2 NMR spectroscopy of H₆CPB (L).

¹H-NMR: σ (DMSO-d₆) = 7.06 (d, 12H), 7.44 (d, 12H), 12.81 (s, 6H) ppm.

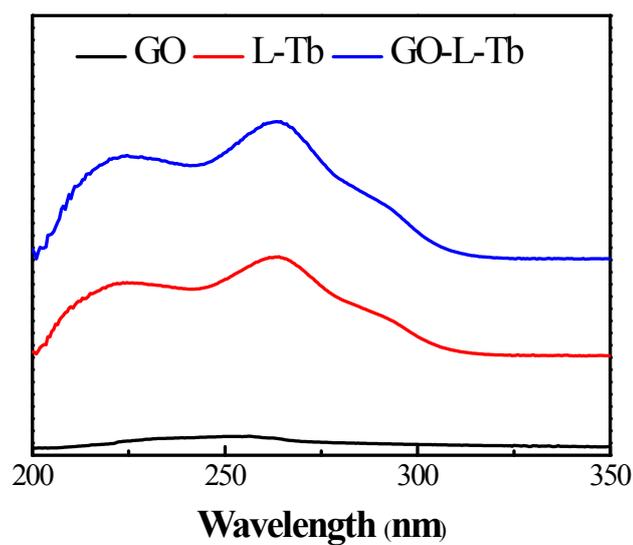


Fig. S4 UV spectrum of GO, L-Tb and GO-L-Tb.

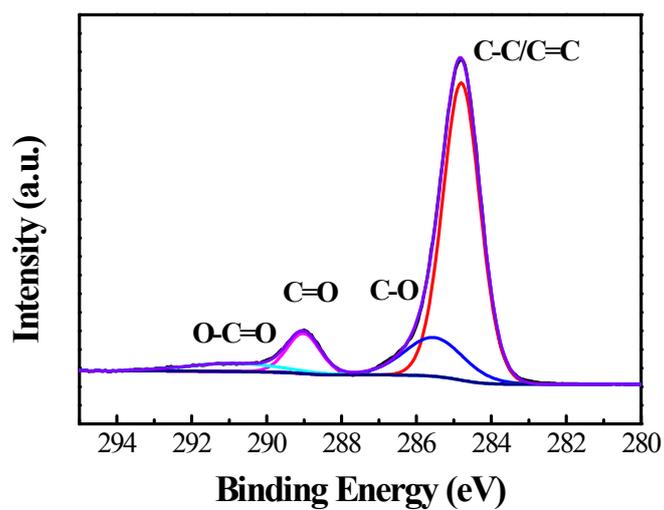


Fig. S5 Resolved C1s spectra of L-Tb.

Table S2. Resolved C1s data of GO.

Position	FWHM	Area	% Conc.
284.8	1.446	85306.21	47.98
286.91	1.148	73268.07	41.21
288.35	1.961	18273.28	10.28
290.98	1.455	950.4	0.53

Table S3. Resolved CIs data of GO-L-Tb.

Position	FWHM	Area	% Conc.
284.8	1.135	64166.84	60.31
285.78	2.821	33183.01	31.19
288.7	1.21	7216.1	6.78
290.72	3.082	1826.52	1.72

Table S4. Resolved CIs data of L-Tb.

Position	FWHM	Area	% Conc.
284.8	1.172	101569.99	71.69
285.54	1.854	20978.63	14.81
289.02	1.041	12171.82	8.59
290.85	2.885	6953.16	4.91

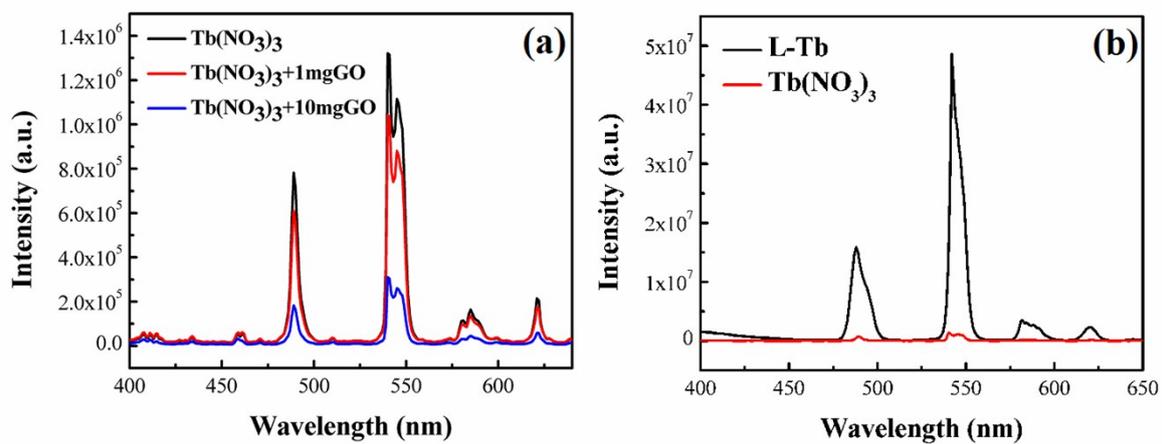


Fig. S6 Fluorescence Spectroscopy of $Tb(NO_3)_3$, $Tb(NO_3)_3+1mg GO$, $Tb(NO_3)_3+10mg GO$ (a), $Tb(NO_3)_3$ and GO-L-Tb (b).

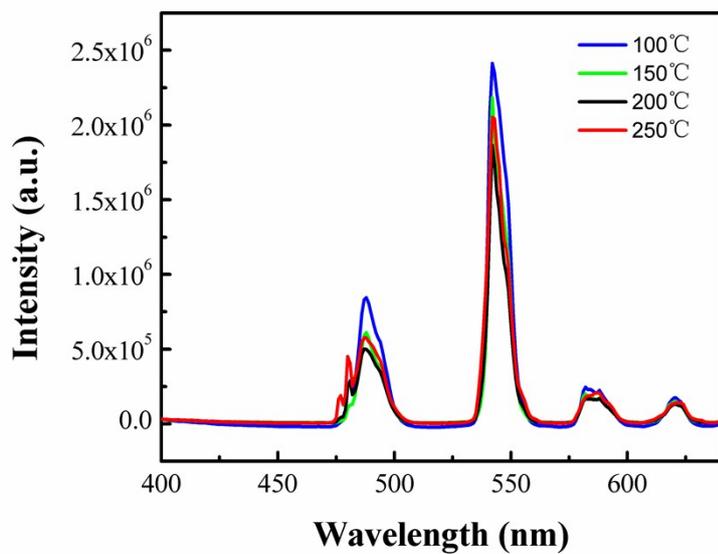


Fig. S7 Fluorescence Spectroscopy of GO-L-Tb after heating at 100°C, 150°C, 200°C, 250°C for half an hour.

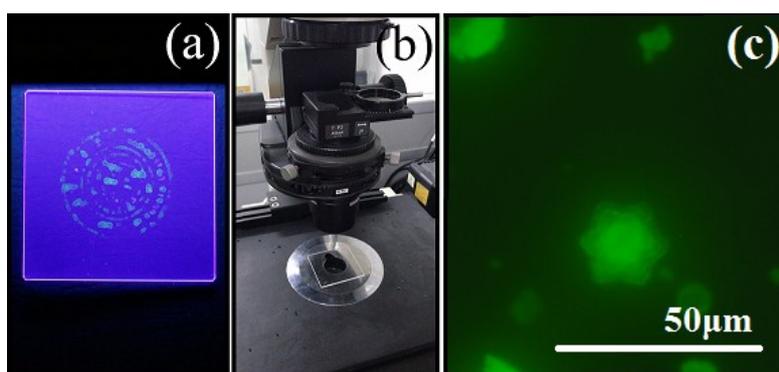


Fig. S8 The pattern made by GO-L-Tb printed on the quartz plate (a), quartz sheet on a fluorescence microscope (b), six-petal shape in GO-L-Tb in the pattern observed under fluorescence microscope.