

Electronic Supporting Information

Nanocrystallization of Lanthanide-Doped KLu_2F_7 - KYb_2F_7 Solid-Solutions in Aluminosilicate Glass for Upconverted Solid-State-Lighting and Photothermal Anti-Counterfeiting

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Table S1 Nominal glass compositions (mol%) for the investigated samples and the corresponding crystallization phases after heat-treatment at 750 °C for 2h. Cub and orth represent cubic and orthorhombic phases, respectively.

Glass compositions (mol%)	Crystallization phase	
	As-quenched	Heat-treatment (750°C/2h)
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9LaF ₃	Cub-K(Lu/La) ₃ F ₁₀	Cub-K(Lu/La) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9CeF ₃	Cub-K(Lu/Ce) ₃ F ₁₀	Cub-K(Lu/Ce) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9PrF ₃	Cub-K(Lu/Pr) ₃ F ₁₀	Cub-K(Lu/Pr) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9NdF ₃	Cub-K(Lu/Nd) ₃ F ₁₀	Cub-K(Lu/Nd) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9SmF ₃	Cub-K(Lu/Sm) ₃ F ₁₀	Cub-K(Lu/Sm) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9EuF ₃	Cub-K(Lu/Eu) ₃ F ₁₀	Cub-K(Lu/Eu) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9GdF ₃	Cub-K(Lu/Gd) ₃ F ₁₀	Cub-K(Lu/Gd) ₃ F ₁₀
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9TbF ₃	Cub-K(Lu/Tb) ₃ F ₁₀ ⁺ Orth-K(Lu/Tb) ₂ F ₇	Cub-K(Lu/Tb) ₃ F ₁₀ ⁺ Orth-K(Lu/Tb) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9DyF ₃	Cub-K(Lu/Dy) ₃ F ₁₀ ⁺ Orth-K(Lu/Dy) ₂ F ₇	Cub-K(Lu/Dy) ₃ F ₁₀ ⁺ Orth-K(Lu/Dy) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9HoF ₃	Cub-K(Lu/Ho) ₃ F ₁₀ ⁺ Orth-K(Lu/Ho) ₂ F ₇	Cub-K(Lu/Ho) ₃ F ₁₀ ⁺ Orth-K(Lu/Ho) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9YF ₃	Cub-K(Lu/Y) ₃ F ₁₀ ⁺ Orth-K(Lu/Y) ₂ F ₇	Cub-K(Lu/Y) ₃ F ₁₀ ⁺ Orth-K(Lu/Y) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9ErF ₃	Orth-K(Lu/Er) ₂ F ₇	Orth-K(Lu/Er) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9TmF ₃	Orth-K(Lu/Tm) ₂ F ₇	Orth-K(Lu/Tm) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9YbF ₃	Orth-K(Lu/Yb) ₂ F ₇	Orth-K(Lu/Yb) ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9LuF ₃	Orth-KLu ₂ F ₇	Orth-KLu ₂ F ₇
100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16LuF ₃ -9ScF ₃	Orth-K(Lu/Sc) ₂ F ₇	Orth-K(Lu/Sc) ₂ F ₇

Table S2 Effective ionic radius of Ln³⁺ in fluorine environment (Z=6)

Lanthanide ions	La ³⁺	Ce ³⁺	Pr ³⁺	Nd ³⁺	Sm ³⁺	Eu ³⁺	Gd ³⁺	Tb ³⁺
Radius (Å)	1.19	1.15	1.14	1.12	1.10	1.109	1.08	1.06
Lanthanide ions	Dy ³⁺	Ho ³⁺	Y ³⁺	Er ³⁺	Tm ³⁺	Yb ³⁺	Lu ³⁺	Sc ³⁺
Radius (Å)	1.05	1.04	1.04	1.03	1.02	1.01	1.00	0.89

Table S3 Nominal glass compositions (mol%) for the Lu-Yb solid-solution NCs@glass samples and the corresponding crystallization phases after heat-treatment at 750 °C for 2h. Orth represents orthorhombic phases.

Samples	Glass compositions (mol%)	Crystallization phase
		GC (750°C/2h)
0%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-24LuF ₃ -0YbF ₃ -0.48Er	Orth-KLu ₂ F ₇
10%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-21.6LuF ₃ -2.4YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
20%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-19.2LuF ₃ -4.8YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
30%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-16.8LuF ₃ -7.2YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
40%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-14.4LuF ₃ -9.6YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
50%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-12LuF ₃ -12YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
60%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-9.6LuF ₃ -14.4YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
70%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-7.2LuF ₃ -16.8YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
80%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-4.8LuF ₃ -19.2YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
90%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-2.4LuF ₃ -21.6YbF ₃ -0.48Er	Orth-K(Lu/Yb) ₂ F ₇
100%Yb	100SiO ₂ -6Al ₂ O ₃ -9K ₂ O-19KF-0LuF ₃ -24YbF ₃ -0.48Er	Orth-KYb ₂ F ₇

Table S4 Color coordinates for laser power dependent UC luminescence yielded from the Er: KYb₂F₇ NCs@glass under irradiation of 980 nm laser.

Laser power density (W/cm ²)	CIE x	CIE y
23	0.5407	0.4437
66	0.5318	0.4558
113	0.5166	0.4630
159	0.4943	0.4860
211	0.4650	0.5110
245	0.4336	0.5313
276	0.4284	0.5411

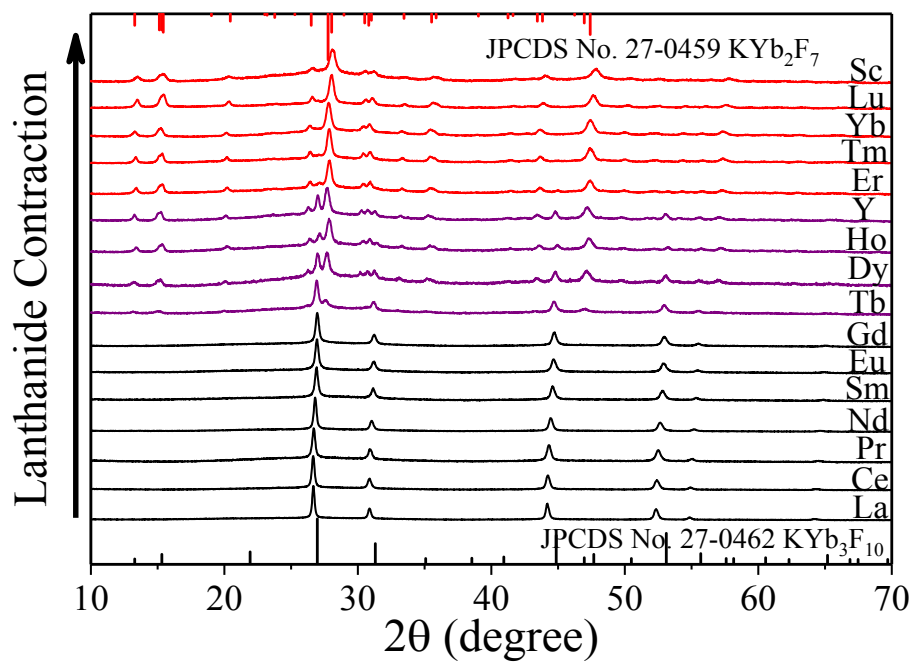


Figure S1 XRD patterns of a series of PGs containing Ln^{3+} dopants ($\text{Ln}=\text{La-Lu, Sc}$). The diffraction bars represent cubic $\text{KYb}_3\text{F}_{10}$ (JPCDS No. 27-0462) and orthorhombic KYb_2F_7 (JPCDS No. 27-0459) crystals. Lu-Ln solid-solution fluorides have precipitated inside glass, indicating that self-crystallization has already occurred in these precursor glasses.

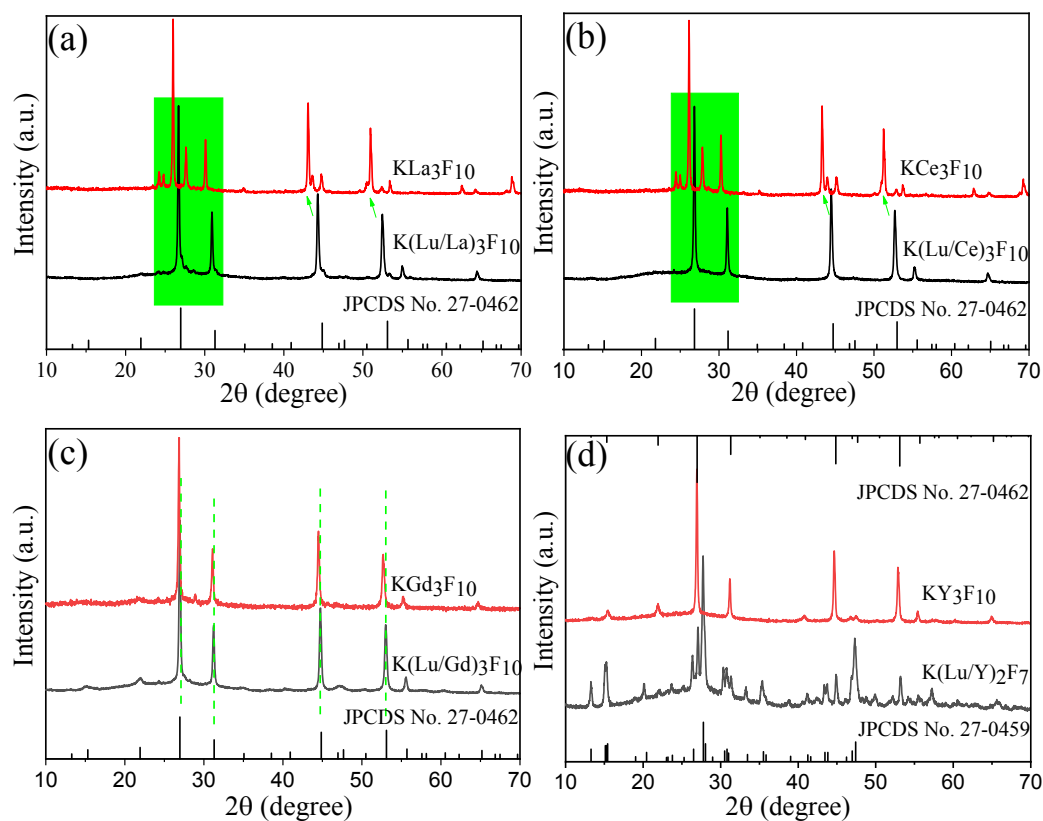


Figure S2 Comparison of XRD patterns of NCs@glass samples with/without the addition of LuF_3 . The glass compositions without LuF_3 are $100\text{SiO}_2\text{-}6\text{Al}_2\text{O}_3\text{-}9\text{K}_2\text{O}\text{-}19\text{KF}\text{-}25\text{LnF}_3$: (a) $\text{Ln}=\text{La}$, (b) $\text{Ln}=\text{Ce}$, (c) $\text{Ln}=\text{Gd}$, and (d) $\text{Ln}=\text{Y}$. The diffraction bars represent cubic $\text{KYb}_3\text{F}_{10}$ (JPCDS No. 27-0462) and orthorhombic KYb_2F_7 (JPCDS No. 27-0459) crystals. With the addition of LuF_3 , the diffraction peaks shift towards large angles, pure cubic phases are obtained for Lu-La, Lu-Ce and Lu-Gd samples and extra orthorhombic phase occurs for Lu-Y sample, verifying that the crystallized phases in glasses are indeed Lu-Ln fluoride solid-solutions and Lu^{3+} ions are beneficial for the growth of pure $\text{K}(\text{Lu}/\text{Ln})_3\text{F}_{10}$ or $\text{K}(\text{Lu}/\text{Ln})_2\text{F}_7$ solid-solutions.

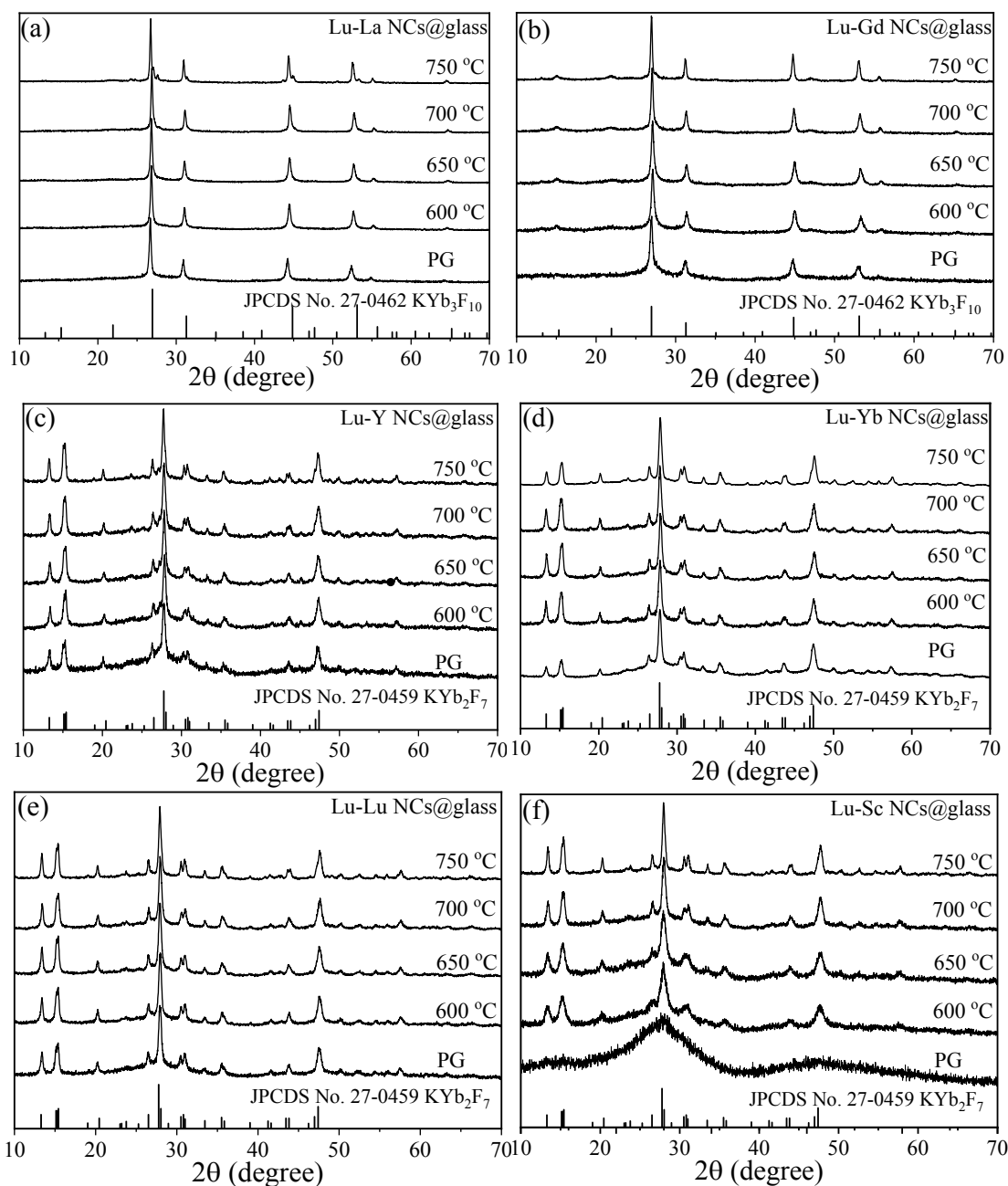


Figure S3 XRD patterns of PGs and Lu-Ln NCs@glass samples prepared by heat-treatment at various temperatures (600, 650, 700, 750 °C) for 2 h: (a) Ln=La, (b) Ln=Gd, (c) Ln=Y, (d) Ln=Yb, (e) Ln=Lu and (f) Ln=Sc.

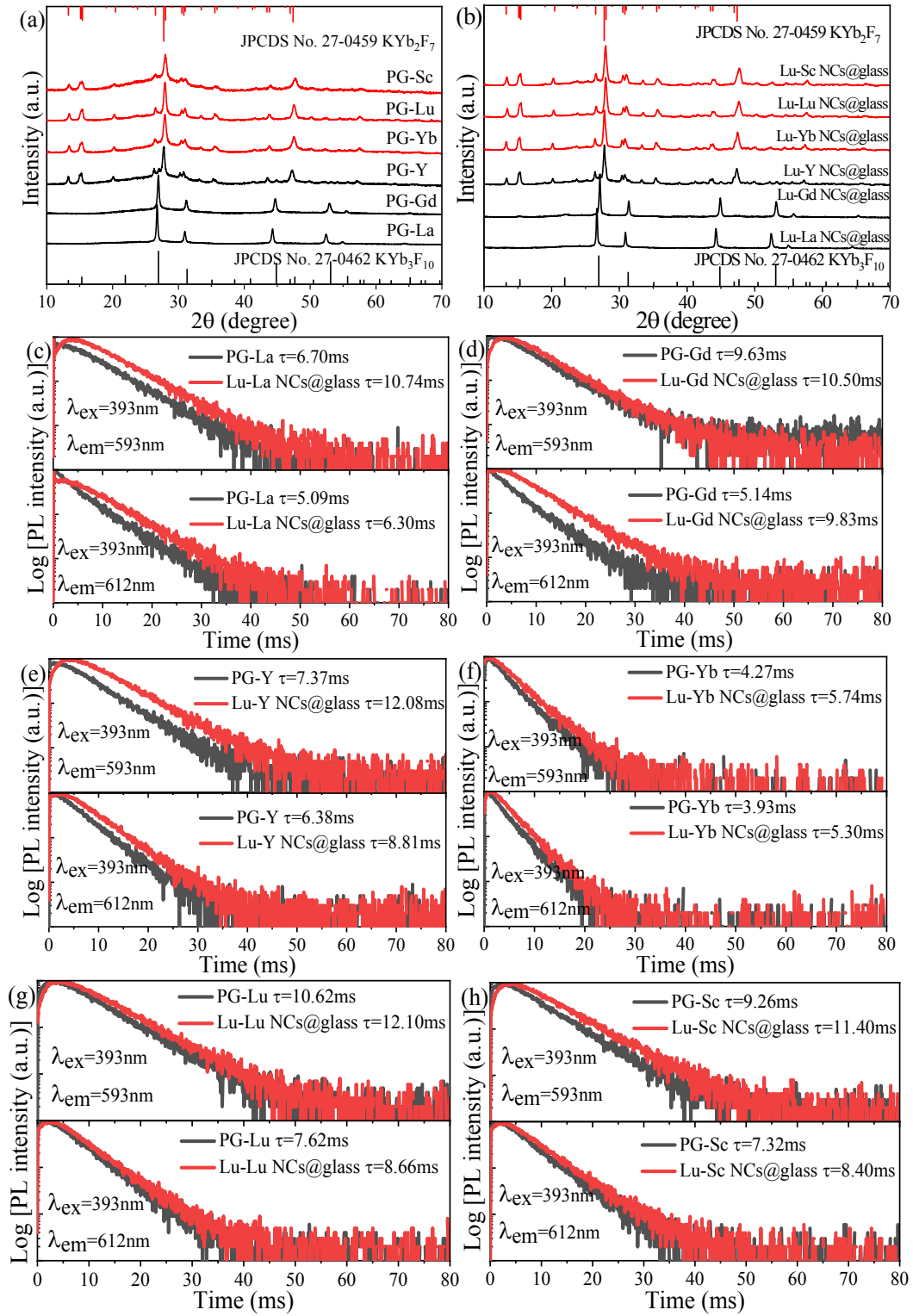


Figure S4 XRD patterns of (a) Eu³⁺ doped PGs and (b) Lu-Ln NCs@glass samples prepared by heating PGs at 750 °C for 2 h. Decay curves (λ_{ex} =393 nm) for the corresponding samples by monitoring Eu³⁺ ⁵D₀→⁷F₁ (λ_{em} =593 nm) and ⁵D₀→⁷F₂ (λ_{em} =612 nm) transitions, respectively: (c) Ln=La, (d) Ln=Gd, (e) Ln=Y, (f) Ln=Yb, (g) Ln=Lu and (h) Ln=Sc.

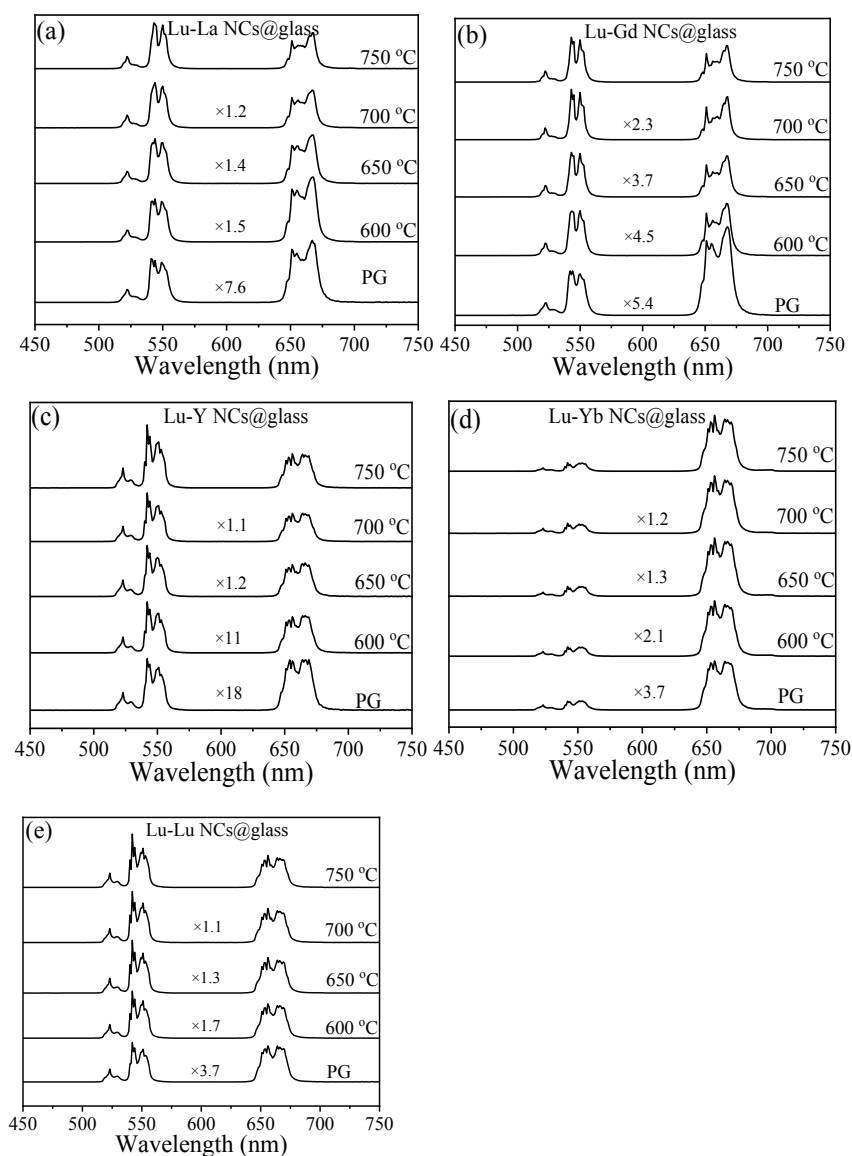


Figure S5 UC emission spectra ($\lambda_{\text{ex}} = 980 \text{ nm}$) of PGs and Lu-Ln NCs@glass samples prepared by heat-treatment at various temperatures (600, 650, 700, 750 °C) for 2 h: (a) Ln=La, (b) Ln=Gd, (c) Ln=Y, (d) Ln=Yb and (e) Ln=Lu. The glass compositions (mol %) are 100SiO₂-6Al₂O₃-9K₂O-19KF-16LuF₃-9LnF₃-1YbF₃-0.32ErF₃ (Ln=La, Gd, Y, Yb, Lu).

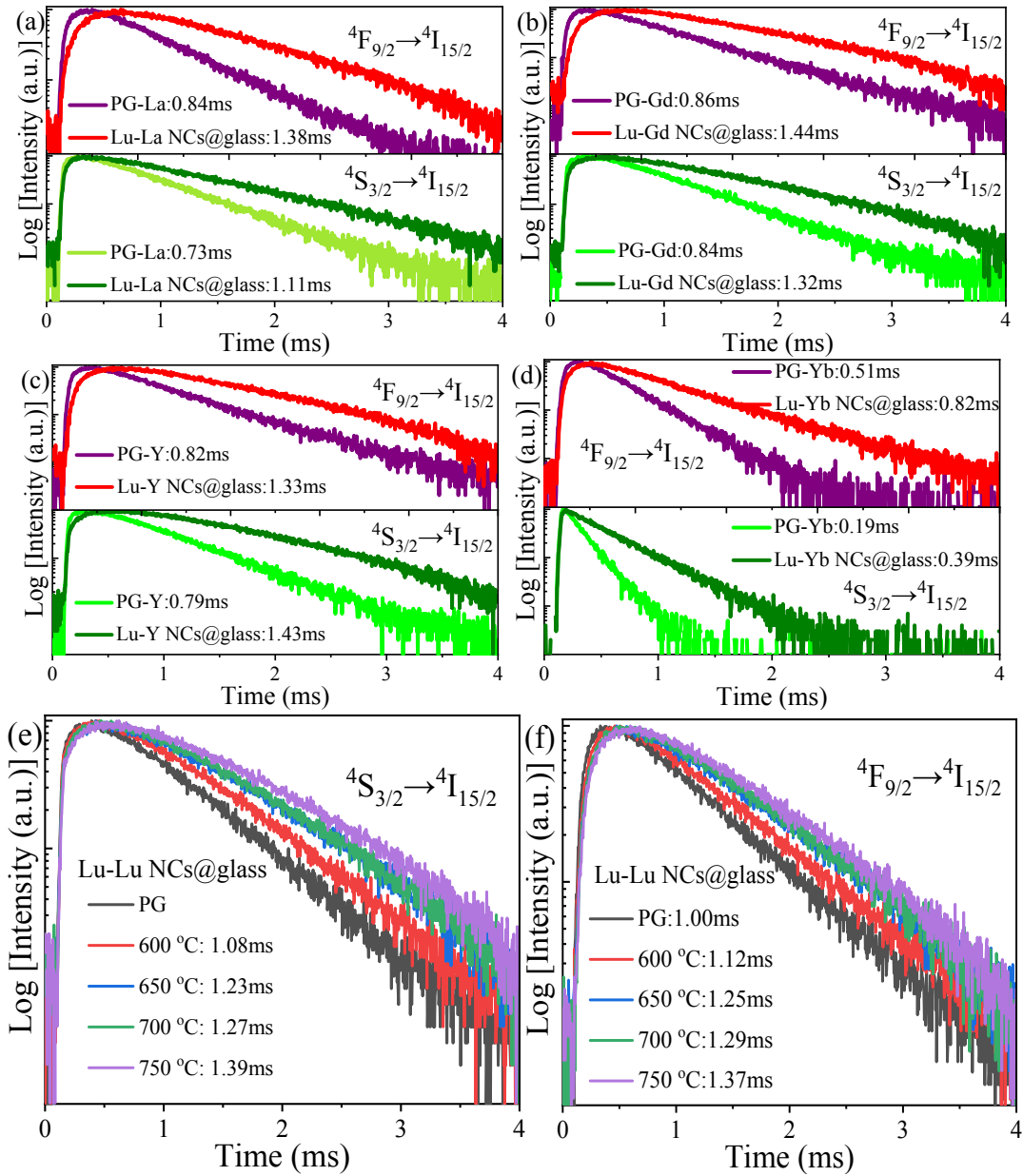


Figure S6 UC decay curves of PGs and Lu-Ln NCs@glass samples by monitoring Er³⁺ green ($^4S_{3/2} \rightarrow ^4I_{15/2}$) and red ($^4F_{9/2} \rightarrow ^4I_{15/2}$) emissions, respectively: (a) Ln=La, (b) Ln=Gd, (c) Ln=Y, (d) Ln=Yb and (e, f) Ln=Lu.

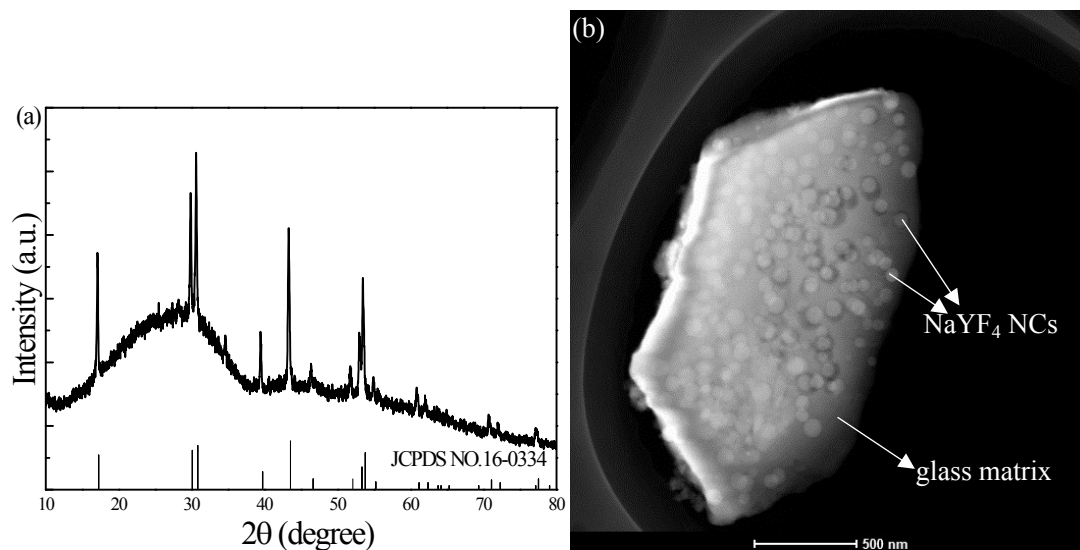


Figure S7 (a) XRD pattern and (b) HAADF-STEM image of Yb/Er: NaYF₄@glass sample. Bars represent diffraction data of hexagonal β-NaYF₄ crystal (JCPDS No. 16-0334). The precipitated phase inside glass is hexagonal NaYF₄, and the particle sizes are in the range of 40~100 nm.

The glass composite for the Yb/Er: NaYF₄@glass sample is 55SiO₂-6Al₂O₃-12Na₂O-19NaF-8YF₃. The sample was prepared by melt-quenching and subsequent heat-treatment. About 15 g of high-purity raw materials were mixed and ground evenly, and then placed in an alumina crucible at 1500 °C for 30 minutes in the ambient atmosphere. The melt was poured into a pre-heated brass mold at 350 °C and then cooled to room temperature to obtain precursor glass. Finally, the Yb/Er: NaYF₄@glass was achieved by heating the precursor glass at 650 °C for 2 h to induce β-NaYF₄ crystallization.

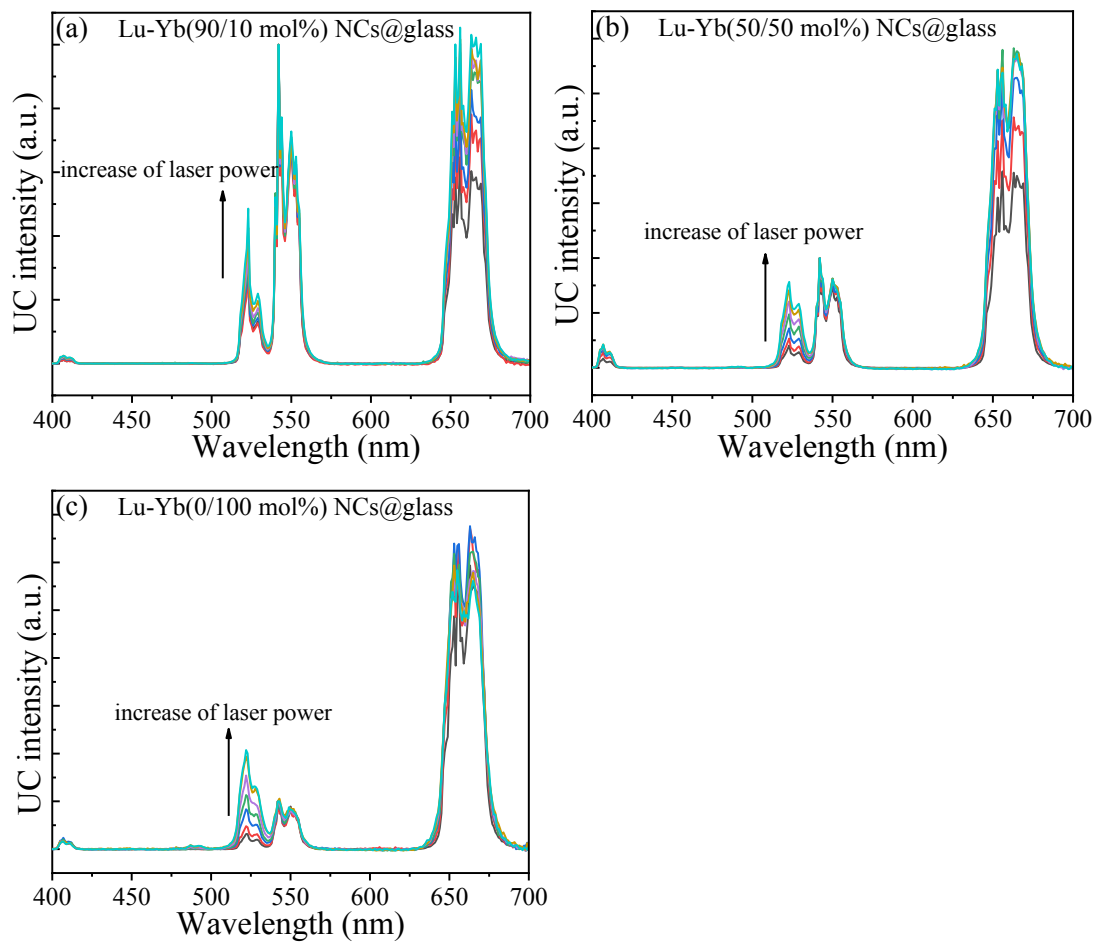


Figure S8 Laser power dependent UC emission spectra of Er: K(Lu/Yb)₂F₇ NCs@glass (all the spectra are normalized at Er³⁺ 545 nm emission assigned to ⁴S_{3/2}→⁴I_{15/2} transition): (a) 10 mol% Yb, (b) 50 mol% Yb and (c) 100 mol% Yb.