

## Supporting Information

### Carbon based Y-type molecules for application in nonlinear optics

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#### Supporting Information list:

**Figure S1.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with methyl group or isobutyl group at one azulene end in Y-B-C<sub>3h</sub>-3.

**Figure S2.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with F or NO<sub>2</sub> group at one azulene end in Y-B-C<sub>3h</sub>-3.

**Figure S3.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with naphthalene ends without (D<sub>3h</sub>) and with CN group (C<sub>s</sub>) at one naphthalene.

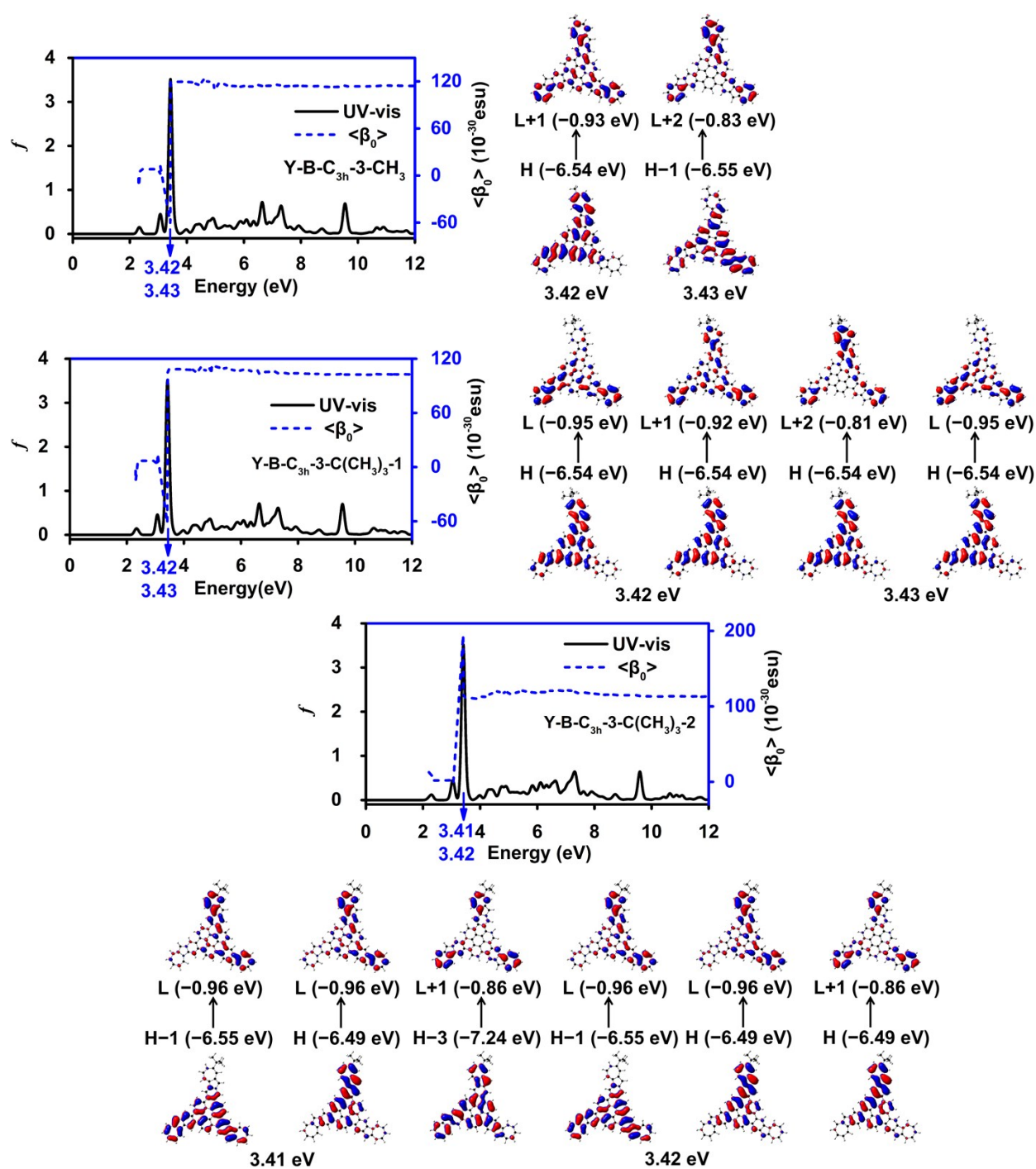
**Figure S4.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-Cs molecules.

**Figure S5.** Two-dimensional second order NLO spectra of azulene scanned from -7.00 eV to 7.00 eV with step size of 0.05 eV.

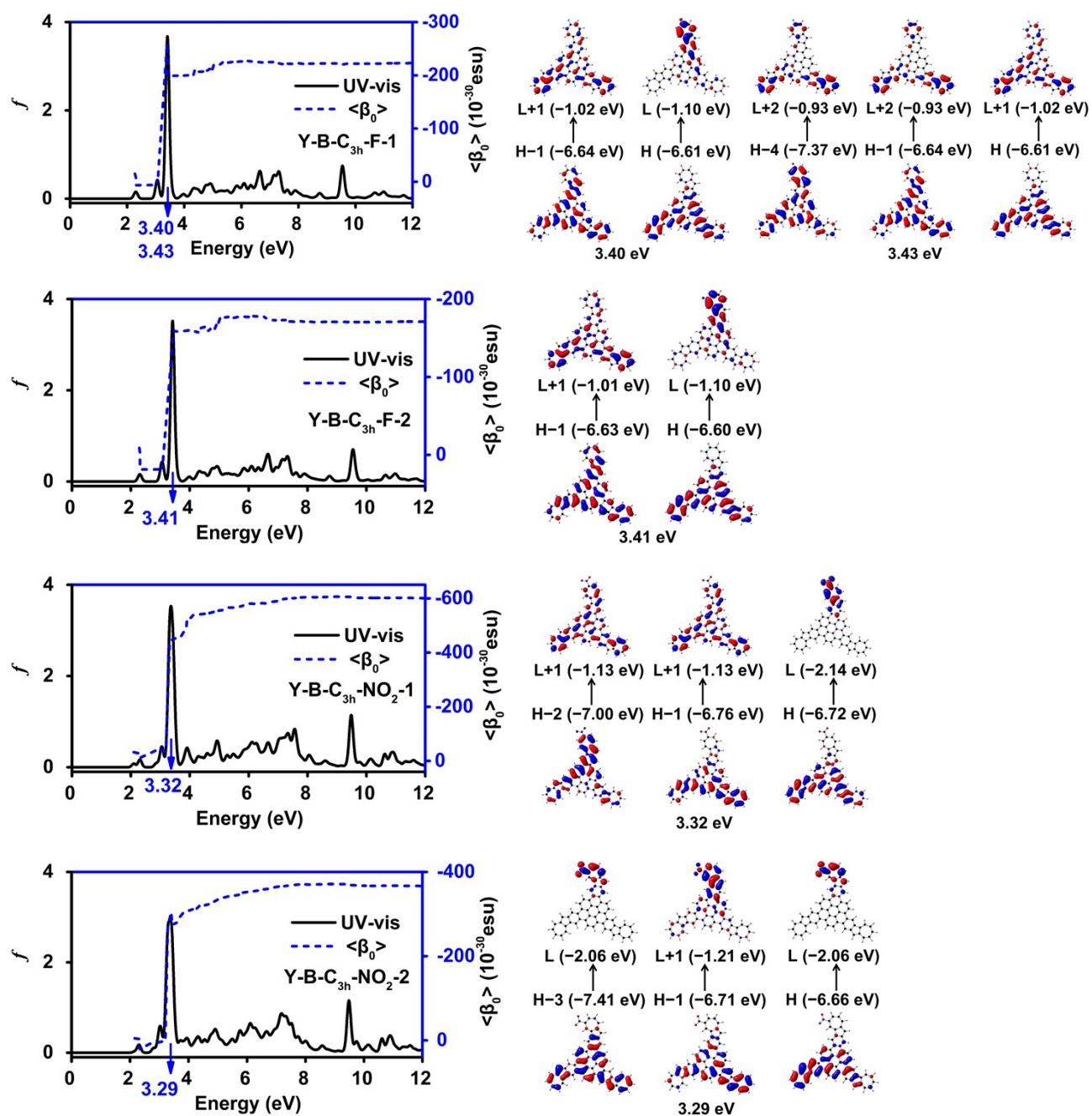
**Figure S6.** Two-dimensional second order NLO spectra of azulene with step size of 0.005 eV[(a)  $\omega_1$  scanned from 4.48 eV to 5.48 eV and  $\omega_2$  scanned from -5.48 eV to -4.48 eV; (b)  $\omega_1$  scanned from -0.5 eV to 0.5 eV and  $\omega_2$  scanned from -5.48 eV to -4.48 eV].

**Figure S7.** Two-dimensional second order NLO spectra of Y-B-NC-C<sub>2</sub>-3 scanned from -7.00 eV to 7.00 eV with step size of 0.05 eV.

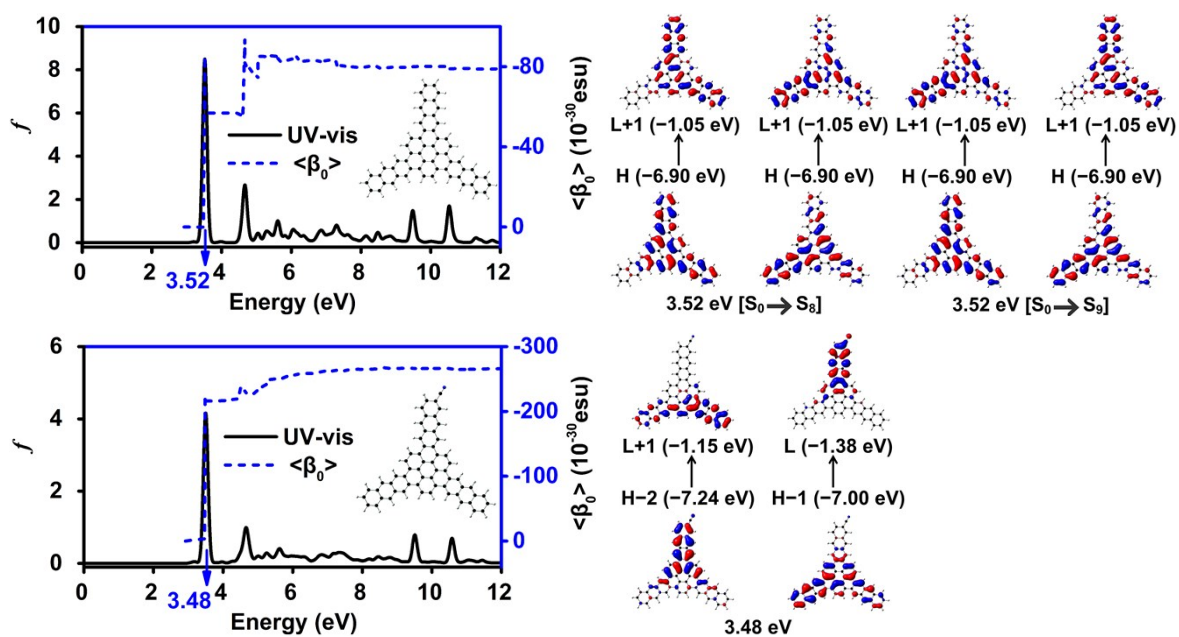
**Table S1.** Major electronic spectra absorption peaks with transition nature in all molecules.



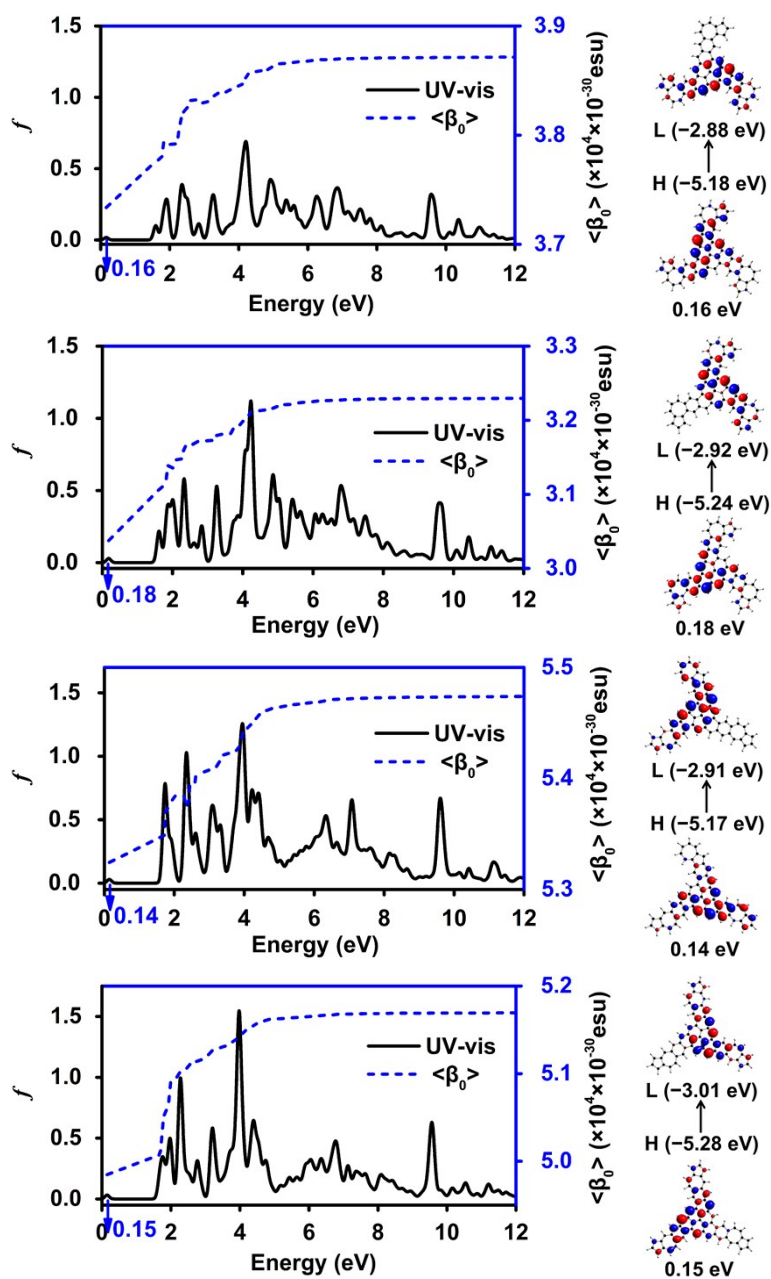
**Figure S1.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with methyl group or isobutyl group at one azulene end in Y-B-C<sub>3h</sub>-3.



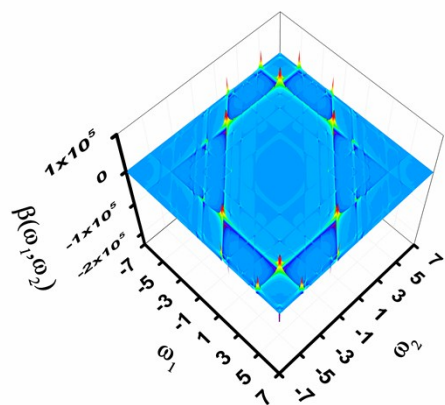
**Figure S2.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with F or NO<sub>2</sub> group at one azulene end in Y-B-C<sub>3h</sub>-3.



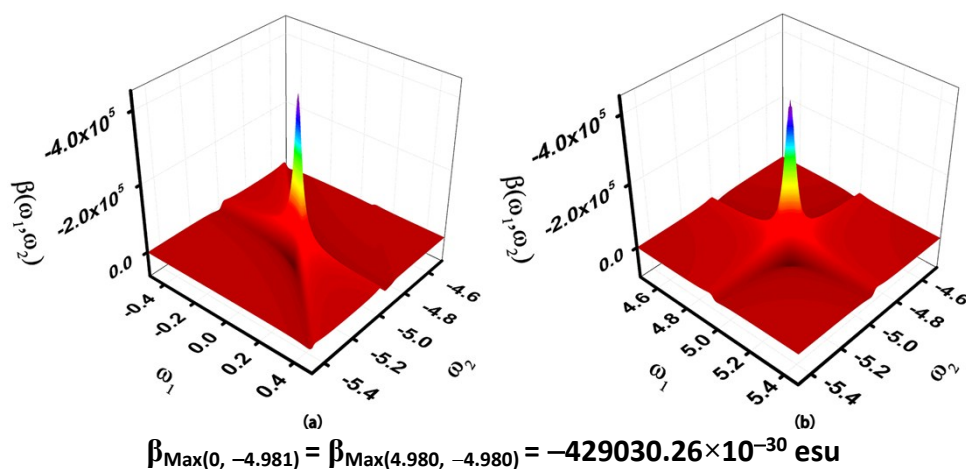
**Figure S3.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-B type molecules with naphthalene ends without ( $D_{3h}$ ) and with CN group ( $C_s$ ) at one naphthalene.



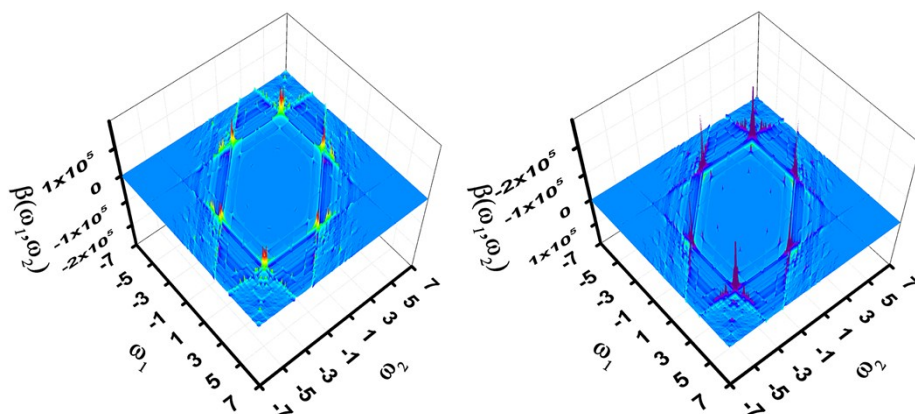
**Figure S4.** Evolution of the static first hyperpolarizability ( $\langle\beta_0\rangle$ ) with electron excitation in Y-Cs molecules.



**Figure S5.** Two-dimensional second order NLO spectra of azulene scanned from  $-7.00$  eV to  $7.00$  eV with step size of  $0.05$  eV.



**Figure S6.** Two-dimensional second order NLO spectra of azulene with step size of  $0.005$  eV [(a)  $\omega_1$  scanned from  $-0.5$  eV to  $0.5$  eV and  $\omega_2$  scanned from  $-5.48$  eV to  $-4.48$  eV; (b)  $\omega_1$  scanned from  $4.48$  eV to  $5.48$  eV and  $\omega_2$  scanned from  $-5.48$  eV to  $-4.48$  eV].



**Figure S7.** Two-dimensional second order NLO spectra of Y-B-NC-C<sub>2</sub>-3 scanned from  $-7.00$  eV to  $7.00$  eV with step size of  $0.05$  eV.

**Table S1.** Major electronic absorption peaks with transition nature in all molecules ( $f$  is the oscillator strength in arbitrary unit,  $E$  is the transition energy in eV, contribution to  $\langle\beta_0\rangle$  in  $10^{-30}$  esu).

Compounds	$f$	$E$	Transition	Transition nature	Contribution to $\langle\beta_0\rangle$
PNA(C <sub>6</sub> H <sub>6</sub> N <sub>2</sub> O <sub>2</sub> )	0.4960	4.47	S <sub>0</sub> →S <sub>4</sub>	H→L (95.40%)	-31.84
	0.0234	2.21	S <sub>0</sub> →S <sub>1</sub>	H→L (91.51%)	1.94
	0.4145	3.56	S <sub>0</sub> →S <sub>2</sub>	H-1→L (35.00%) H→L+1 (64.60%)	2.97
Azulene(C <sub>10</sub> H <sub>8</sub> )	0.2331	4.40	S <sub>0</sub> →S <sub>3</sub>	H-1→L (80.28%)	2.64
	2.0474	4.98	S <sub>0</sub> →S <sub>4</sub>	H-1→L (63.28%) H→L+1 (34.05%)	52.86
	0.2752	6.85	S <sub>0</sub> →S <sub>11</sub>	H-2→L+1 (34.49%) H-1→L+2 (61.42%)	-0.85
Y-B-NC-C <sub>3</sub> -1(C <sub>36</sub> H <sub>24</sub> )	1.2405	4.99	S <sub>0</sub> →S <sub>25</sub>	H-2→L (16.56%)	9.54
				H-2→L (16.56%) H-1→L+2 (28.04%)	
Y-B-NC-C <sub>1</sub> -2(C <sub>36</sub> H <sub>24</sub> )	0.9562	4.80	S <sub>0</sub> →S <sub>23</sub>	H→L+5 (51.36%)	-22.87
Y-B-NC-C <sub>2</sub> -3(C <sub>36</sub> H <sub>24</sub> )	3.0325	4.18	S <sub>0</sub> →S <sub>9</sub>	H-4→L+1 (13.15%)	74.72
				H-3→L (18.94%) H-2→L+3 (36.02%)	
Y-B-NC-C <sub>2</sub> -4(C <sub>36</sub> H <sub>24</sub> )	3.1090	4.18	S <sub>0</sub> →S <sub>11</sub>	H-4→L+2 (16.90%)	44.57
				H→L (13.05%) H-3→L+4 (10.39%) H→L+5 (14.58%)	
Y-B-NC-C <sub>2</sub> -5(C <sub>36</sub> H <sub>24</sub> )	2.5249	4.16	S <sub>0</sub> →S <sub>11</sub>	H-3→L+3 (11.75%) H-2→L+3 (25.34%)	39.01
	2.5333	4.16	S <sub>0</sub> →S <sub>12</sub>	H-2→L+3 (11.28%) H-1→L+4 (16.84%)	-9.89
	0.2723	4.87	S <sub>0</sub> →S <sub>21</sub>	H-3→L (68.61%)	5.68
	0.2642	4.87	S <sub>0</sub> →S <sub>22</sub>	H-3→L+1 (67.91%) H-3→L+5 (10.04%)	-17.38
Y-B-C <sub>5</sub> -1(C <sub>60</sub> H <sub>30</sub> )	4.6117	3.42	S <sub>0</sub> →S <sub>8</sub>	H-3→L (15.93%)	-58.78
				H-2→L (14.13%) H-1→L+1 (18.12%)	
Y-B-C <sub>3h</sub> -2(C <sub>60</sub> H <sub>30</sub> )	3.9794	3.44	S <sub>0</sub> →S <sub>8</sub>	no major transition	-50.41
				H-2→L+1 (14.28%)	-95.41
				H→L (16.58%) H→L (16.58%)	
Y-B-C <sub>3h</sub> -3(C <sub>60</sub> H <sub>30</sub> )	3.5007	3.42	S <sub>0</sub> →S <sub>8</sub>	H-2→L+1 (14.28%)	28.17
				H→L (16.58%) H→L (16.58%)	
				H→L+1 (11.22%) H→L (14.93%) H→L (14.93%)	
Y-B-C <sub>3h</sub> -2-CN (C <sub>61</sub> H <sub>29</sub> N)	4.4728	3.40	S <sub>0</sub> →S <sub>8</sub>	H→L+1 (11.22%)	-125.79
				H→L (14.93%) H→L (14.93%)	
Y-B-C <sub>3h</sub> -2-CN (C <sub>61</sub> H <sub>29</sub> N)	4.4728	3.40	S <sub>0</sub> →S <sub>8</sub>	H-2→L+1 (23.68%)	-98.97

				H-1→L+1 (12.74%) H→L (12.59%)	
	3.9006	3.45	S <sub>0</sub> →S <sub>9</sub>	H-4→L+2 (16.54%) H-1→L+2 (14.83%) H→L+1 (14.83%)	45.96
Y-B-C <sub>3h</sub> -2-NC (C <sub>61</sub> H <sub>29</sub> N)	4.5311	3.41	S <sub>0</sub> →S <sub>8</sub>	H-2→L+1 (18.23%) H-1→L+1 (15.03%) H→L (12.59%)	-33.77
	3.9141	3.45	S <sub>0</sub> →S <sub>9</sub>	H-4→L+2 (18.84%) H-1→L+2 (13.20%) H→L+1 (15.25%)	76.08
Y-B-C <sub>3h</sub> -3-CN1(C <sub>61</sub> H <sub>29</sub> N)	0.1208	2.19	S <sub>0</sub> →S <sub>1</sub>	H-4→L (19.92%) H-2→L (54.21%) H-1→L (13.01%)	-31.38
	3.8662	3.35	S <sub>0</sub> →S <sub>8</sub>	H-2→L (14.29%) H-1→L (13.09%) H→L (18.34%)	-341.73
	3.2017	3.43	S <sub>0</sub> →S <sub>9</sub>	H-4→L+2 (11.80%) H-1→L+2 (22.38%) H→L+1 (14.46%)	24.96
Y-B-C <sub>3h</sub> -3-CN2(C <sub>61</sub> H <sub>29</sub> N)	0.2494	2.86	S <sub>0</sub> →S <sub>4</sub>	H-2→L+1 (22.79%) H-2→L+1 (15.74%)	-32.97
	0.1939	3.38	S <sub>0</sub> →S <sub>8</sub>	H-3→L (10.29%) H→L (28.07%)	-451.66
Y-B-C <sub>3h</sub> -3-CH <sub>3</sub> (C <sub>61</sub> H <sub>32</sub> )	3.5159	3.42	S <sub>0</sub> →S <sub>8</sub>	H→L+1 (14.18%)	-72.54
	3.4927	3.43	S <sub>0</sub> →S <sub>9</sub>	H-1→L+2 (14.99%)	179.86
Y-B-C <sub>3h</sub> -3-C(CH <sub>3</sub> ) <sub>3</sub> -1 (C <sub>64</sub> H <sub>38</sub> )	3.5591	3.42	S <sub>0</sub> →S <sub>8</sub>	H→L (12.49%) H→L+1 (19.20%)	-76.83
	3.5108	3.43	S <sub>0</sub> →S <sub>9</sub>	H→L+2 (19.04%) H→L (12.49%)	170.41
Y-B-C <sub>3h</sub> -3-C(CH <sub>3</sub> ) <sub>3</sub> -2 (C <sub>64</sub> H <sub>38</sub> )	3.4659	3.41	S <sub>0</sub> →S <sub>8</sub>	H-1→L (16.82%) H→L (15.90%)	195.35
	3.5228	3.42	S <sub>0</sub> →S <sub>9</sub>	H-3→L+1 (10.38%) H-1→L (15.50%) H→L (10.36%) H→L+1 (11.35%)	-82.45
Y-B-C <sub>3h</sub> -3-F-1(C <sub>60</sub> H <sub>29</sub> F)	3.6772	3.40	S <sub>0</sub> →S <sub>8</sub>	H-1→L+1 (15.04%) H→L (21.10%)	-262.56
	3.4096	3.43	S <sub>0</sub> →S <sub>9</sub>	H-4→L+2 (12.04%) H-1→L+2 (19.62%) H→L+1 (15.16%)	62.88
Y-B-C <sub>3h</sub> -3-F-2(C <sub>60</sub> H <sub>29</sub> F)	3.5192	3.41	S <sub>0</sub> →S <sub>8</sub>	H-1→L+1 (13.17%) H→L (23.04%)	-159.21
Y-B-C <sub>3h</sub> -3-NO <sub>2</sub> -1 (C <sub>60</sub> H <sub>29</sub> NO <sub>2</sub> )	3.5323	3.32	S <sub>0</sub> →S <sub>9</sub>	H-2→L+1 (21.42%) H-1→L+1 (11.94%) H→L (17.23%)	-407.36
Y-B-C <sub>3h</sub> -3-NO <sub>2</sub> -2 (C <sub>60</sub> H <sub>29</sub> NO <sub>2</sub> )	2.9487	3.29	S <sub>0</sub> →S <sub>9</sub>	H-3→L (10.73%) H-1→L+1 (12.80%)	-268.18



				H→L (23.75%)	
Y-B-D <sub>3h</sub> (C <sub>60</sub> H <sub>30</sub> )	4.2492	3.52	S <sub>0</sub> →S <sub>8</sub>	H→L+1 (29.17%) H→L+1 (29.17%)	-85.34
	4.2492	3.52	S <sub>0</sub> →S <sub>9</sub>	H→L+1 (29.17%) H→L+1 (29.17%)	28.45
Y-B-D <sub>3h</sub> -CN(C <sub>61</sub> H <sub>29</sub> N)	4.1297	3.48	S <sub>0</sub> →S <sub>8</sub>	H-2→L+1 (29.17%) H-1→L (29.17%)	-220.72
Y-C-C <sub>3h</sub> -1(C <sub>58</sub> H <sub>30</sub> )	0.0478	0.16	S <sub>0</sub> →S <sub>1</sub>	H→L (91.30%)	37336.50
Y-C-C <sub>5</sub> -2(C <sub>58</sub> H <sub>30</sub> )	0.0511	0.18	S <sub>0</sub> →S <sub>1</sub>	H→L (91.91%)	30375.50
Y-C-C <sub>5</sub> -3(C <sub>58</sub> H <sub>30</sub> )	0.0397	0.14	S <sub>0</sub> →S <sub>1</sub>	H→L (89.26%)	53243.80
Y-C-C <sub>3h</sub> -4(C <sub>58</sub> H <sub>30</sub> )	0.0522	0.15	S <sub>0</sub> →S <sub>1</sub>	H→L (91.04%)	49846.40
Y-C-C <sub>5</sub> -5(C <sub>58</sub> H <sub>30</sub> )	0.1023	0.29	S <sub>0</sub> →S <sub>1</sub>	H→L (91.85%)	14705.50
Y-C-C <sub>3h</sub> -6(C <sub>58</sub> H <sub>30</sub> )	0.0628	0.17	S <sub>0</sub> →S <sub>1</sub>	H→L (90.86%)	42145.20
Y-Ref (C <sub>50</sub> H <sub>21</sub> N <sub>7</sub> O <sub>2</sub> S <sub>2</sub> )	3.0853	2.58	S <sub>0</sub> →S <sub>1</sub>	H-1→L+1 (35.28%) H→L (39.36%)	-115.47
	0.9635	2.70	S <sub>0</sub> →S <sub>2</sub>	H-1→L (40.16%) H→L+1 (35.41%)	-96.27