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Supporting Information for:

Photoactuators based on the dynamic molecular crystals of naphthalene acrylic acids driven by sterospecific [2+2] cycloaddition reactions

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Measurements and characterizations

¹H NMR and ¹³C NMR spectra were recorded with a Bruker-AvanceIII 400 MHz and 101 MHz spectrometer using DMSO-d₆and CDCl₃ as solvents and tetramethylsilane (TMS) as the internal standard. The samples for irradiation time-dependent ¹H NMR measurements were obtained via the irradiation of the crystals of 1FNaAA, 1CINaAA, 1BrNaAA, 1INaAA and 6BrNaAA by 365 nm (3 W) light for different times, followed by dissolving in DMSO-d₆. FT-IR spectra were obtained with a Nicolet-360 FT-IR spectrometer by the incorporation of samples into KBr disks. High-resolution mass analyses were measured on Bruker micro TOF-QII apparatus. UV-vis absorption spectra were measured by a Shimadzu UV-1601PC spectrophotometer. Fluorescence emission spectra were taken on a Shimadzu RF-5301 Luminescence Spectrometer. The samples of 1FNaAA, 1ClNaAA, 1BrNaAA, 1INaAA and 6BrNaAA for electronic spectra measurements were prepared by grinding the crystals into microcrystals, which were smeared on the quartz slice and irradiated by 365 nm (3 W) light for different times. To reveal the geometrical structures, frontier molecular orbitals and energy band gaps of the target compounds, density functional theory (DFT) calculations were performed by using the Gaussian 09W program with the B3LYP/3-21G(d) basis set for 1FNaAA, 1CINaAA, 1BrNaAA, 1INaAA and 6BrNaAA.

The single crystals of **1FNaAA**, **D-1FNaAA**, **1CINaAA**, **D-1CINaAA**, **1BrNaAA**, **1INaAA** and **6BrNaAA**were selected for X-ray diffraction analysis on a Rigaku RA XIS-RA PID diffractometer using graphite-monochromated MoK α radiation (λ = 0.71073 Å), and the crystals were kept at 99.97 K, 100.0 K, 100.02 K, 100.0 K, 293.76 K, 150.0 K and 293.76 K during data collection, respectively. The single crystals of 1FNaAA, 1ClNaAA and 1BrNaAA were obtained in CH₂Cl₂, methanol and acetone, respectively, under petroleum ether steam by vapor diffusion method. The single crystals of 1INaAA and 6BrNaAA were obtained in THF under petroleum ether steam by vapor diffusion method. The single crystals of β -type **D-1FNaAA** and D-1CINaAA were obtained in acetone and THF, respectively, under petroleum ether steam by vapor diffusion method. In addition, the β -type **D-1FNaAA** and D-1CINaAA were isolated from the microcrystals of 1FNaAA and 1CINaAA, which were firstly irradiated by a hand-held 365 nm lamp (3 W) for 30 min, via column chromatography (silica gel) using ethylacetate/petroleum ether (v/v = 1/4) containing 3% acetic acid as the eluent. It should be noted that the absorption at 365 nm for the naphthalene acrylic acids in THF were quite low, but the microcrystals gave broad peaks with tails covering 365 nm. Indeed, the photochemical reactions could take place in microcrystals under 365 nm light. Therefore, the hand-held 365 nm lamp (3 W) was used as light source for [2+2] cycloaddition reactions and photomechanical studies.

The photomechanical effects were observed under an optical microscope at 298 K under ambient conditions. The crystals were put on the glass slices and then irradiated by a hand-held 365 nm lamp (3 W) for different times. The crystals of **1FNaAA** and **1ClNaAA**were obtained by slow evaporation from the solutions in CH₂Cl₂. The crystals of **1BrNaAA** and **6BrNaAA**were obtained by slow evaporation from the

solutions in CH₂Cl₂/petroleum ether (v/v = 1/4).

THF was dried over sodium and benzophenone. DMF was dried over phosphorus pentoxide. CH₂Cl₂ was dried over calcium hydride. The other chemicals and reagents were used as received without further purification.

Syntheses

1-Fluoro-2-naphthaldehyde (1)

The solution of *n*-butyllithium (8.5 mL, 13.6 mmol, 1.6 M in hexane) in cyclohexane was added to a solution of 1-fluoronaphthalene (2.0 g, 13.6 mmol) in dry THF (40 mL) at -78 °C. The mixture was stirred at -78 °C for 2 h, and then dry DMF (2.2 mL, 27.2 mmol) was added. After stirring for 5 min at -78 °C, the mixture was quenched with water and the crude product was extracted with ethyl acetate (20 mL × 3). The combined organic phase was dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 20/1) to afford **1** (1.8 g) as a yellow solid. Yield 75%. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 10.49 (s, 1H), 8.31-8.19 (m, 1H), 8.10 (d, J = 8.1 Hz, 1H), 7.94-7.86 (m, 1H), 7.86-7.72 (m, 3H) (Figure S23).

1-Chloro-3,4-dihydronaphthalene-2-carbaldehyde (2)

A flame-dried flask containing DMF (5.3 mL, 68.4 mmol) was cooled to 0 °C and POCl₃ (5 mL, 54.7 mmol) was added dropwise under stirring. The mixture was warmed to room temperature and stirred for 15 min. Then, it was cooled to 0 °C and 3,4-dihydronaphthalen-1-(2H)-one (4.6 mL, 34.2 mmol) was added dropwise. After that, the mixture was warmed to room temperature and stirred for another 1 h. The

mixture was poured into ice water (100 mL) and the saturated aqueous NaHCO₃ was added to adjust pH to ca. 6.0. The mixture was extracted with ethyl acetate (30 mL × 3) and the combined organic phase was dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 20/1) to afford **2** (5.9 g) as a slightly yellow oil. Yield 91%. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 10.28 (s, 1H), 7.82 (dd, J = 7.7, 1.1 Hz, 1H), 7.46 (td, J = 7.4, 1.4 Hz, 1H), 7.44-7.37 (m, 1H), 7.34 (dd, J = 7.3, 0.8 Hz, 1H), 2.88-2.78 (m, 2H), 2.58-2.52 (m, 2H) (Figure S24).

1-Chloro-2-naphthaldehyde (3)

The mixture of compound 2 (3.0 15.6 DDO g, mmol) and (2,3-dichloro-5,6-dicyano-1,4-benzoquinone, 7.1 g, 31.2 mmol) in chlorobenzene (100 mL) was refluxed for 72 h. It was then poured into saturated aqueous solution of NaHCO₃ and extracted with ethyl acetate (20 mL \times 3). The combined organic phase was dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v= 20/1) to afford 3 (1.9 g) as a pale vellow solid. Yield 63%. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 10.68 (s, 1H), 8.49 (dd, J = 5.8, 3.4 Hz, 1H), 8.16 (dd, J = 5.8, 3.6 Hz, 1H), 8.11 (d, J = 8.5 Hz, 1H), 7.92 (d, J = 8.6 Hz, 1H), 7.87 (dd, J = 6.3, 3.3 Hz, 2H) (Figure S25).

2-(1-Bromonaphthalen-2-yl)-1,3-dioxolane (4)

1-Bromo-2-naphthaldehyde(2.0 g, 8.5 mmol) was dissolved in toluene (60 mL), and 4-methylbenzenesulfonic acid monohydrate (17.4 mg, 0.091 mmol) as well as ethylene glycol (2.4 mL, 43.5 mmol) were added. Then, the mixture was stirred at 125 °C for 18 h under nitrogen atmosphere. After that, the excess toluene was removed under vacuum. The crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 20/1) to afford **4** (2.2 g) as a yellow liquid. Yield 93%. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 8.33 (d, J = 8.5 Hz, 1H), 8.06 (d, J = 8.6 Hz, 2H), 7.81-7.74 (m, 1H), 7.74-7.67 (m, 2H), 6.30 (s, 1H), 4.27-4.05 (m, 4H) (Figure S26).

1-Iodo-2-naphthaldehyde (5)

Compound 4 (2.2 g, 7.9 mmol) was dissolved in dry THF (20 mL) and the solution was cooled to -78 °C under nitrogen atmosphere. Then, n-butyllithium (6.5 mL, 15.8 mmol, 2.5 M in hexane) was added dropwise and stirred for 20 min. Iodine (3 g, 11.9 mmol) in THF (15 mL) was added at -78 °C. After that, the solution was allowed to warm up to room temperature and stirred for 14 h. The reaction was quenched by saturated aqueous solution of NH₄Cl and NaHSO₃. The mixture was extracted with ethyl acetate (20 mL \times 3). The combined organic phase was dried over anhydrous Na₂SO₄. After removal of the solvent. the crude product of 2-(1-iodonaphthalen-2-yl)-1,3-dioxolane was not further purified. It was dissolved in MeOH/toluene (21 mL, v/v = 4/17), to which HCl (15 mL, 20%) was added. The mixture was stirred at room temperature for 18 h under nitrogen. The saturated aqueous NaHCO₃ was added to adjust pH to ca. 7.0. The mixture was extracted by ethyl acetate (20 mL \times 3), and the combined organic phase was dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified by column

chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 20/1) to afford **5** (1.2 g) as a yellow solid. Yield 55%. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 10.37 (d, J = 0.6 Hz, 1H), 8.47 (dd, J = 6.3, 3.4 Hz, 1H), 8.15-8.02 (m, 2H), 7.91-7.78 (m, 3H) (Figure S27).

Methyl 6-bromo-2-naphthoate (6)

6-Bromo-2-naphthoic acid (5.0 g, 19.9 mmol) was suspended in methanol (100 mL), and concentrated sulphuric acid (2 mL) was added. After the mixture was refluxed for 12 h, it was cooled and the precipitate was collected by filtration. The crude product was purified by column chromatography (silica gel, dichloromethane/petroleum ether, v/v = 1/1) to afford 6 (4.9 g) as a white solid. Yield 93%. ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.56 (s, 1H), 8.07 (dd, J = 8.6, 1.5 Hz, 1H), 8.03 (s, 1H), 7.79 (t, J = 9.4 Hz, 2H), 7.61 (dd, J = 8.7, 1.8 Hz, 1H), 3.98 (s, 3H) (Figure S28). ¹³C NMR (101 MHz, CDCl₃, ppm) δ 166.88, 136.40, 130.91, 130.85, 130.18, 129.92, 127.85, 127.24, 126.36, 122.63, 52.35 (Figure S29).

(6-Bromonaphthalen-2-yl)methanol (7)

To a solution of diisobutylaluminium hydride (DIBALH) (1.0 M in heptane, 99.6 mL, 99.6 mmol) at -78 °C under nitrogen atmosphere a solution of compound **6** (8.8 g, 33.2 mmol) in anhydrous THF (40 mL) was added dropwise. The mixture was allowed to warm up to room temperature and stirred for 14 h. The reaction was quenched by methanol (20 mL), followed by adding saturated solution of sodium potassium tartrate, and the mixture was stirred at room temperature for another 2 h. After extracted by ethyl acetate (20 mL \times 3), the combined organic phase was washed

with saturated aqueous solution of NH₄Cl and brine, dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 1/1) to afford **7** (7.7 g) as a white solid. Yield 98%. ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.03 (s, 1H), 7.81 (s, 1H), 7.78 (d, J = 8.5 Hz, 1H), 7.73 (d, J = 8.7 Hz, 1H), 7.59 (dd, J = 8.7, 1.9 Hz, 1H), 7.53 (dd, J = 8.5, 1.3 Hz, 1H), 4.88 (s, 2H) (Figure S30). ¹³C NMR (101 MHz, CDCl₃, ppm) δ 138.85, 133.99, 131.80, 129.79, 129.60, 129.54, 127.43, 126.18, 125.30, 119.84, 65.27 (Figure S31).

6-Bromo-2-naphthaldehyde (8)

A solution of compound 7 (2.1 g, 8.7 mmol) in dry CH_2Cl_2 was added to a suspension of pyridinium chromate (2.1 g, 9.7 mmol) in dry CH_2Cl_2 (30 mL). The mixture was refluxed for 5 h, and then CH_2Cl_2 was removed under vacuum. The crude product was purified by column chromatography (silica gel, petroleum ether/ethyl acetate, v/v = 3/1) to afford **8** (1.9 g) as a white solid. Yield 90%. ¹H NMR (400 MHz, $CDCl_3$, ppm) δ 10.21 (s, 1H), 8.37 (s, 1H), 8.14 (s, 1H), 8.04 (d, J = 8.5 Hz, 1H), 7.93 (dt, J = 13.7, 6.8 Hz, 2H), 7.73 (dd, J = 5.3, 3.3 Hz, 1H) (Figure S32). ¹³C NMR (101 MHz, $CDCl_3$, ppm) δ 191.82, 137.31, 134.37, 134.06, 131.10, 130.99, 130.68, 130.29, 128.19, 124.07, 123.62 (Figure S33).

(E)-3-(1-Fluoronaphthalen-2-yl)acrylic acid (1FNaAA)

Compound 1 (1.0 g, 5.7 mmol), malonic acid (1.7 g, 16.1 mmol) and piperidine (1.0 mL) were dissolved in pyridine (60 mL), and the mixture was stirred at 90 °C for 8 h. Then, the mixture was poured into water (200 mL) and dilute HCl was added to

adjust the pH to around 1.0. The white solid **1FNaAA** (0.9 g) was collected by filtration and recrystallized in ethanol. Yield 75%. m.p. 161.0-162.8 °C. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 12.63 (s, 1H), 8.13 (dd, J = 5.9, 3.7 Hz, 1H), 8.04 (dd, J = 4.8, 2.4 Hz, 1H), 7.94 (dd, J = 17.6, 12.3 Hz, 2H), 7.83 (d, J = 8.7 Hz, 1H), 7.73-7.65 (m, 2H), 6.76 (d, J = 16.1 Hz, 1H) (Figure S34). ¹³C NMR (101 MHz, DMSO-d₆, ppm) δ 167.32, 157.54, 154.97, 135.07, 135.02, 134.88, 134.83, 128.34, 127.79, 127.76, 127.47, 124.41, 124.37, 123.76, 123.73, 122.69, 122.52, 121.79, 121.75, 120.68, 120.62, 116.53, 116.43 (Figure S35). FT-IR (KBr, cm⁻¹): 3451, 2989, 1700, 1620, 1420, 1381, 1277, 1075, 982, 869, 750, 559. HRMS (ESI): m/z calcd. for C₁₃H₉FO₂ [M–H]⁻ 215.0541; found 215.0502 (Figure S36).

(E)-3-(1-Chloronaphthalen-2-yl)acrylic acid (1ClNaAA)

By following the synthetic procedure for **1FNaAA**, **1CINaAA** was prepared from compound **3** (1.0 g, 5.2 mmol) and malonic acid (1.5 g, 14.7 mmol) in pyridine (60 mL) in the presence of piperidine (1.0 mL). The crude product was purified by crystallization in ethanol to afford white solid **1CINaAA** (1.2 g). Yield 98%. m.p. 157.5-160.0 °C. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 12.70 (s, 1H), 8.30 (d, J = 8.3 Hz, 1H), 8.17 (d, J = 15.9 Hz, 1H), 8.07-8.00 (m, 2H), 7.97 (d, J = 8.7 Hz, 1H), 7.79-7.61 (m, 2H), 6.76 (d, J = 15.9 Hz, 1H) (Figure S38). ¹³C NMR (101 MHz, DMSO-d₆, ppm) δ 167.25, 139.09, 134.52, 131.60, 130.21, 129.39, 128.43, 128.30, 128.07, 127.71, 124.58, 123.97, 123.04 (Figure S39). FT-IR (KBr, cm⁻¹): 3415, 2985, 1695, 1618, 1419, 1338, 1281, 1233, 979, 809, 749, 695. HRMS (ESI): m/z calcd. for C₁₃H₉ClO₂ [M–H]⁻ 231.0218; found 231.0206 (Figure S40).

(E)-3-(1-Bromonaphthalen-2-yl)acrylic acid (1BrNaAA)

By following the synthetic procedure for **1FNaAA**, **1BrNaAA** was prepared from 1-bromo-2-naphthaldehyde (0.56 g, 2.4 mmol) and malonic acid (0.7 g, 6.7 mmol) in pyridine (50 mL) in the presence of piperidine (0.5 mL). The crude product was purified by crystallization in ethanol to afford white solid **1BrNaAA** (0.57 g). Yield 86%. m.p. 176.0-178.0 °C. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 12.71 (s, 1H), 8.31 (d, J = 8.3 Hz, 1H), 8.20 (d, J = 15.9 Hz, 1H), 8.08-7.93 (m, 3H), 7.81-7.61 (m, 2H), 6.74 (d, J = 15.8 Hz, 1H) (Figure S42). ¹³C NMR (101 MHz, DMSO-d₆, ppm) δ 167.19, 142.27, 134.59, 131.67, 128.53, 128.47, 128.40, 128.03, 127.41, 125.52, 124.39, 123.20 (Figure S43). FT-IR (KBr, cm⁻¹): 3392, 2984, 1693, 1620, 1418, 1331, 1276, 1228, 978, 810, 749, 692. HRMS (ESI): m/z calcd. for C₁₃H₉BrO₂ [M–H]⁻ 274.9713; found 274.9717 (Figure S44).

(E)-3-(1-Iodonaphthalen-2-yl)acrylic acid (1INaAA)

By following the synthetic procedure for **1FNaAA**, **1INaAA** was prepared from compound **5** (1.2 g, 4.3 mmol) and malonic acid (1.3 g, 11.9 mmol) in pyridine (80 mL) in the presence of piperidine (1.0 mL). The crude product was purified by crystallization in ethanol to afford white solid **1INaAA** (0.8 g). Yield 57%. mp: 178.2-180.5 °C. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 12.70 (s, 1H), 8.25 (d, J = 8.3 Hz, 1H), 8.15 (d, J = 15.7 Hz, 1H), 7.98 (t, J = 7.5 Hz, 2H), 7.91 (d, J = 8.6 Hz, 1H), 7.68 (ddd, J = 21.2, 13.9, 6.9 Hz, 2H), 6.63 (d, J = 15.7 Hz, 1H) (Figure S45). ¹³C NMR (101 MHz, DMSO-d₆, ppm) δ 168.06, 151.24, 136.03, 134.49, 134.05, 132.85, 129.19, 128.71, 128.57, 127.89, 124.44, 123.20, 108.68, 99.49 (Figure S46). FT-IR

(KBr, cm⁻¹): 3415, 2985, 1695, 1618, 1419, 1338, 1281, 1233, 979, 809, 749, 695. HRMS (ESI): m/z calcd. for C₁₃H₉IO₂ [M–H]⁻ 322.9574; found 322.9553 (Figure S47).

(E)-3-(6-Bromonaphthalen-2-yl)acrylic acid (6BrNaAA)

By following the synthetic procedure for **1FNaAA**, **6BrNaAA** was prepared from compound **8** (0.5 g, 2.2 mmol) and malonic acid (0.6 g, 6.2 mmol) in pyridine (40 mL) in the presence of piperidine (0.5 mL). The crude product was purified by crystallization in ethanol to afford white solid **6BrNaAA** (0.5 g). Yield 88%. m.p. 174.0-176.8 °C. ¹H NMR (400 MHz, DMSO-d₆, ppm) δ 12.54 (s, 1H), 8.35-8.11 (m, 2H), 7.94 (dd, J = 10.1, 4.8 Hz, 3H), 7.82-7.60 (m, 2H), 6.71 (d, J = 16.0 Hz, 1H) (Figure S48). ¹³C NMR (101 MHz, DMSO-d₆, ppm) δ 167.52, 143.37, 134.73, 132.48, 131.41, 130.57, 129.71, 129.63, 129.38, 127.71, 125.24, 120.44, 120.24 (Figure S49). FT-IR (KBr, cm⁻¹): 3365, 2919, 2849, 2596, 1702, 1620, 1424, 1316, 1286, 1219,1154, 1123, 1058, 973, 857, 810, 687, 626, 545, 471, 417. HRMS (ESI): m/z calcd. for C₁₃H₉BrO₂ [M–H]⁻ 274.9713; found 274.9711 (Figure S50).

Compound	Absorption ^a (nm) ($\epsilon / \times 10^4 \text{ M}^{-1} \text{cm}^{-1}$)	Emission ^b (nm)
1FNaAA	266 (5.9), 297 (3.7), 308 (3.4)	382, 368(shoulder)
1ClNaAA	263 (2.2), 301 (1.3), 312 (1.1)	387
1BrNaAA	263 (1.5), 302 (1.8), 312 (0.7)	403
1INaAA	267 (6.2), 307 (2.7), 319 (2.7)	392
6BrNaAA	271 (4.8), 303 (2.9), 315 (3.0)	376, 359(shoulder)

Table S1 Photophysical data of 1FNaAA, 1ClNaAA, 1BrNaAA, 1INaAA and 6BrNaAA in THF.

 a Measured in THF (1.0 \times 10 $^{-5}$ M); b Excited at 320 nm.

Compound	$\lambda_{abs}^{[a]}(nm)$	$\lambda_{abs}^{[b]}(nm)$	$f^{[c]}$	Transition (%) ^[d]
1FNaAA	353	336.83	0.1823	H→L (88.0)
				H→L+1 (2.5)
	308	305.50	0.0005	H−2→L (92.5)
				H−2→L+1 (2.9)
	297	294.32	0.2830	H−1→L (60.1)
				$H \rightarrow L(10.3)$
1ClNaAA	361	336.54	0.1193	H→L (88.4)
				H→L+1 (2.1)
	312	304.28	0.0863	H−2→L (60.2)
				H−2→L+1 (2.9)
				H−1→L (26.4)
	301	295.54	0.2508	H−1→L (42.5)
				$H \rightarrow L(7.8)$
1BrNaAA	347	340.76	0.1286	$H \rightarrow L(91.4)$
	312	301.41	0.1298	H−2→L (37.9)
				H−2→L+1 (2.3)
				H−1→L (43.4)
				$H \rightarrow L(2.6)$
	302	294.66	0.1274	H−2→L (46.7)
				H→L+1 (14.9)
1INaAA	362	344.05	0.1464	H→L (90.0)
				H→L+1 (2.2)
	319	300.87	0.1589	H−4→L (30.7)
				H−1→L (47.8)
				$H \rightarrow L(3.1)$
	307	292.86	0.0427	H−4→L (28.4)
				H−3→L (11.6)
				H−2→L (7.8)
				H→L+1 (28.2)
6BrNaAA	351	339.90	0.3627	$H \rightarrow L (92.5)$
				$H-1 \rightarrow L(4.5)$
	315	310.01	0.2387	H−1→L (69.0)
				$H \rightarrow L+1 (22.9)$
	303	305.52	0.0005	H−2→L (92.1)
				H−2→L+1 (3.7)

 Table S2 Main electronic transitions calculated with TD-DFT.

^aExperimental absorption in THF; ^bCompound transition wavelength *in vacuo*; ^cCompound oscillator strength *in vacuo*; ^dH represents HOMO, L represents LUMO.

	1FNaAA	D-1FNaAA	1CINaAA	D-1CINaAA
Formula	C ₁₃ H ₉ FO ₂	$C_{29}H_{24}F_2O_5$	C ₁₃ H ₉ ClO ₂	$C_{30}H_{26}Cl_2O_5$
Formula weight	216.20	490.48	232.65	537.41
Space group	P21/c	P21/n	P1	P21/c
Crystal system	Monoclinic	Monoclinic	Triclinic	Monoclinic
<i>a</i> (Å)	6.7599(8)	7.4331(7)	3.7709(7)	13.8413(5)
<i>b</i> (Å)	3.7695(4)	21.337(2)	7.9900(14)	6.6231(2)
<i>c</i> (Å)	38.685(4)	15.4508(17)	17.205(3)	27.4197(9)
α (deg)	90	90	91.451(7)	90
β (deg)	93.899(4)	98.686(5)	90.871(7)	91.7890(10)
γ (deg)	90	90	91.865(7)	90
$V(\text{\AA}^3)$	983.46(18)	2422.4(4)	517.87(16)	2512.40(14)
Ζ	4	4	2	4
$D_{\rm calc}({\rm g/cm}^3)$	1.460	1.345	1.492	1.421
μ (mm ⁻¹)	0.110	0.101	0.347	0.299
Final R indices	$R_1 = 0.0558$	$R_1 = 0.0562$	$R_1 = 0.1117$	$R_1 = 0.0292$
[I>2sigma(I)]	$wR_2 = 0.1367$	$wR_2 = 0.0934$	$wR_2 = 0.2969$	$wR_2 = 0.0724$
R indices(all data)	$R_1 = 0.0674$	$R_1 = 0.1494$	$R_1 = 0.1782$	$R_1 = 0.0335$
	$wR_2 = 0.1436$	$wR_2 = 0.1275$	$wR_2 = 0.3717$	$wR_2 = 0.0766$
GoF	1.131	1.019	1.135	1.067
CCDC	1948533	1955133	1948658	1955134

 Table S3 Single crystal data of 1FNaAA, D-1FNaAA, 1ClNaAA and D-1ClNaAA.

	1BrNaAA	1INaAA	6BrNaAA
Formula	$C_{13}H_9BrO_2$	$C_{13}H_9IO_2$	C ₁₃ H ₉ BrO ₂
Formula weight	277.11	324.10	277.11
Space group	P1	PĪ	PĪ
Crystal system	Triclinic	Triclinic	Triclinic
<i>a</i> (Å)	3.8948(4)	4.8142(2)	3.9738(2)
<i>b</i> (Å)	8.1149(8)	8.2211(4)	6.2790(4)
<i>c</i> (Å)	17.1274(16)	14.3363(7)	21.6685(13)
α (deg)	90.194(3)	102.141(2)	97.582(2)
β (deg)	91.725(4)	91.410(2)	93.911(2)
γ (deg)	92.396(4)	93.280(2)	92.391(2)
$V(\text{\AA}^3)$	540.60(9)	553.41(4)	533.98(5)
Ζ	2	2	2
$D_{\rm calc}({\rm g/cm}^3)$	1.702	1.945	1.723
$\mu (\mathrm{mm}^{-1})$	3.781	2.873	3.828
Final R indices	$R_1 = 0.0264$	$R_1 = 0.0302$	$R_1 = 0.0383$
[I>2sigma(I)]	$wR_2 = 0.0609$	$wR_2 = 0.0601$	$wR_2 = 0.0996$
R indices(all data)	$R_1 = 0.0325$	$R_1 = 0.0408$	$R_1 = 0.0405$
	$wR_2 = 0.0633$	$wR_2 = 0.0633$	$wR_2 = 0.1009$
GoF	1.050	1.086	1.092
CCDC	1948592	1948594	1948602

Table S4 Single crystal data of 1BrNaAA,1INaAA and 6BrNaAA.

	<i>d</i> (Å)	θ_1 (deg)	θ_2 (deg)	θ_3 (deg)
1FNaAA	3.769	0	103.73	69.39, 70.25
1CINaAA	3.771	0	112.94	68.35, 87.98
1BrNaAA	3.895	0	112.17	67.69, 85.68
1INaAA	4.814	0	121.07	58.00, 52.82
6BrNaAA	3.974	0	96.35	64.04, 62.78
Ideal	<4.2	0	90	90

Table S5 Geometric parameters of 1FNaAA, 1ClNaAA, 1BrNaAA,1INaAA and6BrNaAA in single crystals (parameter definitions are given in Chart S1).

 Table S6 Calculated attachment energies (Eatt) for different crystalline planes in the single crystal of 1ClNaAA.

hkl	N (1/ 1 · · ·	$d_{hkl}/ {\rm \AA}$	Eatt (total)	Total facet
	Multiplicity		$/ \text{kcal} \cdot \text{mol}^{-1}$	area/ %
(001)	2	17.197	-6.662	57.541
(010)	2	7.983	-13.072	29.925
(100)	2	3.768	-31.389	8.069
(10-1)	2	3.693	-31.218	4.167
(11-1)	2	3.318	-33.015	0.217
(111)	2	3.287	-33.408	0.081

hkl	Multiplicity	$d_{hkl}/ {\rm \AA}$	Eatt (total)	Total facet
	Multiplicity		$/ \text{kcal} \cdot \text{mol}^{-1}$	area/ %
(001)	2	21.419	-4.669	67.063
(010)	2	6.216	-16.088	29.925
(01-1)	2	6.198	-16.134	9.311
(100)	2	3.960	-25.257	7.188
(10-1)	2	3.946	-25.341	4.836
(110)	2	3.420	-29.241	0.936
(1-11)	2	3.382	-29.565	0.045

Table S7 Calculated attachment energies (Eatt) for different crystalline planes in the single crystal of **6BrNaAA**.



Chart S1 Geometric parameters for [2+2] topochemical dimerization. The angle θ_1 corresponds to the rotational angle of one double bond with respect to the other, while θ_2 corresponds to the obtuse angle of the parallelogram formed by the double bond carbons, whereas θ_3 measures the angle between the alkene substituents and cyclobutane planes, *d* is the distance between the two reactive olefin centers and the reaction double bonds of the adjacent molecules satisfy the parallel arrangement. The ideal values for θ_1 , θ_2 , and θ_3 for the best overlap of π -orbitals are 0°, 90° and 90° and d = 3.5-4.2 Å, respectively.¹



Figure S1 The frontier molecular orbital plots and energy levels for the HOMOs and LUMOs of 1FNaAA, 1ClNaAA, 1BrNaAA, 1INaAA and 6BrNaAA.



Figure S2 Optical microscopy photos of the bending of the slice-like crystal (700 μ m × 110 μ m × 15 μ m) attached by a needle-like crystal (360 μ m × 33 μ m × 15 μ m) of **1CINaAA**upon irradiation with 365 nm light (3W) for different times (the irradiation direction: from left to right).



Figure S3 UV-vis absorption (a) and fluorescence emission (b; λ_{ex} = 320 nm) spectra of **1CINaAA** in the microcrystals before and after irradiation by 365 nm light (3 W) for different times.



Figure S4 ¹H NMR spectra of the samples before (a) and after irradiation of the microcrystals **1CINaAA** by 365 nm light (3 W) for 30 s (b), 1 min (c), 3 min (d), 5 min (e) and 10 min (f), followed by dissolving in DMSO-d₆;(g) ¹H NMR spectrum of **D-1CINaAA** in DMSO-d₆.



Figure S5 (a) The H-bonds in the single crystal of 1ClNaAA; the single crystal structure of viewed along the b-axis (b) and a-axis (c); (d) the packing of H-bonded dimer in single crystal; (e) the dihedral angle between the plane of the double bond and naphthalene and distance between two adjacent parallel molecules; (f) the molecular configuration of β -type D-1ClNaAA in the single crystal.



Figure S6 (a) Hirshfeld surface mapped with d_{norm} of **1CINaAA**; (b) the two-dimensional fingerprint plots in crystal stacking for **1CINaAA**; (c) the percentage of individual atomic contacts contribution to Hirshfeld surface. The surfaces are shown as transparent to allow visualization of the orientation and conformation of the functional groups in the molecules.



Figure S7 Optical microscopy photos of the bending, rotating and coiling of the slice-like crystal of **1BrNaAA** (700 μ m × 110 μ m × 15 μ m) upon irradiation with 365 nm light (3W) for different times (the irradiation direction: from left to right).



Figure S8 UV-vis absorption spectra of **1BrNaAA** in the microcrystals before and after irradiation by 365 nm light (3 W) for different times.



Figure S9 ¹H NMR spectra of the samples before (a) and after irradiation of the microcrystals **1BrNaAA** by 365 nm light (3 W) for 1 min (b), 5 min (c) and 10 min (d), followed by dissolving in DMSO- d_6 .



Figure S10 (a) The H-bonds in the single crystal of **1BrNaAA**; the single crystal structure of viewed along the a-axis (b) and b-axis (c); (d) the packing of H-bonded dimer in single crystal; (e) the dihedral angle between the plane of the double bond and naphthalene and distance between two adjacent parallel molecules.



Figure S11 (a) Hirshfeld surface mapped with d_{norm} of **1BrNaAA**; (b) the two-dimensional fingerprint plots in crystal stacking for **1BrNaAA**; (c) the percentage of individual atomic contacts contribution to Hirshfeld surface. The surfaces are shown as transparent to allow visualization of the orientation and conformation of the functional groups in the molecules.



Figure S12 Optical microscopy photos of the bending and twisting of the slice-like crystals of **6BrNaAA** (2400 μ m × 80 μ m × 40 μ m) upon irradiation with 365 nm light (3W) for different times(the arrows indicate the irradiation direction).



Figure S13 UV-vis absorption (a) and fluorescence emission (b; $\lambda_{ex} = 320$ nm) spectra of **6BrNaAA** in the microcrystals before and after irradiation by 365 nm light (3 W) for different times.



Figure S14 ¹H NMR spectra of the samples before (a) and after irradiation of the microcrystals **6BrNaAA** by 365 nm light (3 W) for 1 min (b), 5 min (c) and 10 min (d), followed by dissolving in DMSO-d₆.



Figure S15 (a) The H-bonds in the single crystal of **6BrNaAA**; the single crystal structure of viewed along the a-axis (b) and b-axis (c); (d) the packing of H-bonded dimer in single crystal; (e) the dihedral angle between the plane of the double bond and naphthalene and distance between two adjacent parallel molecules.



Figure S16 (a) Hirshfeld surface mapped with d_{norm} of **6BrNaAA**; (b) the two-dimensional fingerprint plots in crystal stacking for **6BrNaAA**; (c) the percentage of individual atomic contacts contribution to Hirshfeld surface. The surfaces are shown as transparent to allow visualization of the orientation and conformation of the functional groups in the molecules.



Figure S17 Calculated growth morphology of **1ClNaAA**in single crystal (a, b) and microscopic images showing the widest face (c) and the bending of the widest face of **1ClNaAA**crystal(d).



Figure S18 Calculated growth morphology of **6BrNaAA**in single crystal (a, b) and microscopic images showing the widest face (c) and the bending of the widest face of **6BrNaAA**crystal(d).



Figure S19 UV-vis absorption (a) and fluorescence emission (b; $\lambda_{ex} = 320$ nm) spectra of **11NaAA** in the microcrystals before and after irradiation by 365 nm light (3 W) for different times.



Figure S20 ¹H NMR spectra of the samples before (a) and after irradiation of the microcrystals **1INaAA** by 365 nm light (3 W) for 5 min (b), 10 min (c) and 1 h (d), followed by dissolving in DMSO-d₆.



Figure S21 (a) The H-bonds in the single crystal of **1INaAA**; the single crystal structure of viewed along the a-axis (b) and b-axis (c); (d) the packing of H-bonded dimer in single crystal; (e) the dihedral angle between the plane of the double bond and naphthalene and distance between two adjacent parallel molecules.



Figure S22 (a) Hirshfeld surface mapped with d_{norm} of 1INaAA; (b) the two-dimensional fingerprint plots in crystal stacking for 1INaAA; (c) the percentage of individual atomic contacts contribution to Hirshfeld surface. The surfaces are shown as transparent to allow visualization of the orientation and conformation of the functional groups in the molecules.



Figure S23 ¹H NMR (400 MHz) spectrum of compound 1 in DMSO-d₆.



Figure S24 ¹H NMR (400 MHz) spectrum of compound 2 in DMSO-d₆.



Figure S25 ¹H NMR (400 MHz) spectrum of compound 3 in DMSO-d₆.



Figure S26¹H NMR (400 MHz) spectrum of compound 4 in DMSO-d₆.



Figure S27 ¹H NMR (400 MHz) spectrum of compound 5 in DMSO-d₆.



Figure S28 ¹H NMR (400 MHz) spectrum of compound 6 in CDCl₃.


Figure S29¹³C NMR (101 MHz) spectrum of compound 6 in CDCl₃.



Figure S30 ¹H NMR (400 MHz) spectrum of compound 7 in CDCl₃.



Figure S31 ¹³C NMR (101 MHz) spectrum of compound 7 in CDCl₃.



Figure S32 ¹H NMR (400 MHz) spectrum of compound 8 in CDCl₃.



Figure S33 ¹³C NMR (101 MHz) spectrum of compound 8 in CDCl₃.



Figure S34 ¹H NMR (400 MHz) spectrum of 1FNaAA in DMSO-d₆.



Figure S35 ¹³C NMR (101 MHz) spectrum of 1FNaAA in DMSO-d₆.



Figure S36 The HRMS of 1FNaAA.



Figure S37 ¹H NMR (400 MHz) spectrum of D-1FNaAA in DMSO-d₆.



Figure S38 ¹H NMR (400 MHz) spectrum of 1ClNaAA in DMSO-d₆.



Figure S39 ¹³C NMR (101 MHz) spectrum of 1ClNaAA in DMSO-d₆.



Figure S40 The HRMS of 1ClNaAA.



Figure S41 ¹H NMR (400 MHz) spectrum of D-1ClNaAA in DMSO-d₆.



Figure S42 ¹H NMR (400 MHz) spectrum of 1BrNaAA in DMSO-d₆.



Figure S43 ¹³C NMR (101 MHz) spectrum of 1BrNaAA in DMSO-d₆.



Figure S44 The HRMS of 1BrNaAA.



Figure S45¹H NMR (400 MHz) spectrum of 1INaAA in DMSO-d₆.



Figure S46¹³C NMR (101 MHz) spectrum of 1INaAA in DMSO-d₆.



Figure S47 The HRMS of 1INaAA.



Figure S48 ¹H NMR (400 MHz) spectrum of 6BrNaAA in DMSO-d₆.



Figure S49 ¹³C NMR (101 MHz) spectrum of 6BrNaAA in DMSO-d₆.



Figure S50 The HRMS of 6BrNaAA.

Reference

1. K. Biradha and R. Santra, Chem. Soc. Rev., 2013, 42, 950.

Structure factors have been supplied for datablock(s) 1FNaAA

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1FNaAA

Bond precision: C-C = 0.0021 AWavelength=0.71073 Cell: a=6.7599(8) b=3.7695(4) c = 38.685(4)alpha=90 beta=93.899(4) gamma=90 Temperature: 100 K Calculated Reported Volume 983.47(19) 983.46(18) P 21/c P 1 21/c 1 Space group Hall group -P 2ybc -P 2ybc Moiety formula C26 H18 F2 O4 C13 H9 F O2 Sum formula C26 H18 F2 O4 C13 H9 F O2 Mr 432.40 216.20 1.460 1.460 Dx,g cm-3 2 Ζ 4 Mu (mm-1) 0.110 0.110 F000 448.0 448.0 F000′ 448.27 h,k,lmax 9,5,53 9,5,53 2755 Nref 2766 Tmin,Tmax Tmin' Correction method= Not given Data completeness= 0.996 Theta(max) = 29.596R(reflections) = 0.0558(2339) wR2(reflections) = 0.1436(2755) S = 1.131Npar= 181

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level. Click on the hyperlinks for more details of the test

PLAT053_ALERT_1_C Minimum Crystal Dimension Missing (or Error)	Please	Check
PLAT054_ALERT_1_C Medium Crystal Dimension Missing (or Error)	Please	Check
PLAT055_ALERT_1_C Maximum Crystal Dimension Missing (or Error)	Please	Check
PLAT222_ALERT_3_C Non-Solv. Resd 1 H Uiso(max)/Uiso(min) Range	8.0	Ratio
PLAT303_ALERT_2_C Full Occupancy Atom H003 with # Connections	2.00	Check
PLAT906_ALERT_3_C Large K Value in the Analysis of Variance	6.133	Check
PLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L= 0.600	2	Report

Alert level G

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ	Please	Check
PLAT045_ALERT_1_G Calculated and Reported Z Differ by a Factor	0.50	Check
PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels	25	Note
PLAT910_ALERT_3_G Missing # of FCF Reflection(s) Below Theta(Min).	2	Note
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600	7	Note
PLAT955_ALERT_1_G Reported (CIF) and Actual (FCF) Lmax Differ by .	1	Units
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density.	13	Info

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0 ALERT level A = Most likely a serious problem - resolve or explain
0 ALERT level B = A potentially serious problem, consider carefully
7 ALERT level C = Check. Ensure it is not caused by an omission or oversight
7 ALERT level G = General information/check it is not something unexpected
6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
2 ALERT type 2 Indicator that the structure model may be wrong or deficient
4 ALERT type 3 Indicator that the structure quality may be low
2 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check
```

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Structure factors have been supplied for datablock(s) 1ClNaAA

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1ClNaAA

Bond precision: C-C = 0.0070 AWavelength=0.71073 Cell: a=3.7709(7) b=7.9900(14) c=17.205(3) alpha=91.451(7) beta=90.871(7) gamma=91.865(7) Temperature: 100 K Calculated Reported Volume 517.87(16) 517.87(16) P -1 Space group P -1 Hall group -P 1 -P 1 Moiety formula C13 H9 Cl O2 C13 H9 Cl O2 Sum formula C13 H9 Cl O2 C13 H9 Cl O2 Mr 232.65 232.65 1.492 1.492 Dx,g cm-3 2 2 Ζ Mu (mm-1) 0.347 0.347 F000 240.0 240.0 F000′ 240.40 h,k,lmax 5,11,25 5,11,24 3229 Nref 3432 0.951,0.963 Tmin,Tmax Tmin′ 0.949 Correction method= Not given Data completeness= 0.941 Theta(max) = 31.388R(reflections) = 0.1117(1859) wR2(reflections) = 0.3717(3229) S = 1.134Npar= 178

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level.

🎈 Alert level B

PLAT084_ALERT_3_B High wR2 Value (i.e. > 0.25) 0.37 Report

Alert level C

PLAT029_ALERT_3_C _diffrn_measured_fraction_theta_full value Low .	0.979	Why?
PLAT082_ALERT_2_C High R1 Value	0.11	Report
PLAT245_ALERT_2_C U(iso) H2 Smaller than U(eq) C2 by	0.013	Ang**2
PLAT245_ALERT_2_C U(iso) H11 Smaller than U(eq) C11 by	0.013	Ang**2
PLAT250_ALERT_2_C Large U3/U1 Ratio for Average U(i,j) Tensor	2.1	Note
PLAT340_ALERT_3_C Low Bond Precision on C-C Bonds	0.007	Ang.
PLAT351_ALERT_3_C Long C-H (X0.96,N1.08A) C10 - H10 .	1.11	Ang.
PLAT905_ALERT_3_C Negative K value in the Analysis of Variance	-0.282	Report
PLAT910_ALERT_3_C Missing # of FCF Reflection(s) Below Theta(Min).	5	Note
PLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L= 0.600	34	Report

Alert level G

PLAT007_ALERT_5_G Number of Unrefined Donor-H Atoms	1	Report
PLAT072_ALERT_2_G SHELXL First Parameter in WGHT Unusually Large	0.20	Report
PLAT154_ALERT_1_G The s.u.'s on the Cell Angles are Equal(Note)	0.007	Degree
PLAT169_ALERT_4_G The CIF-Embedded .res File Contains AFIX 1 Recds	1	Report
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600	161	Note
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density.	3	Info
PLAT992_ALERT_5_G Repd & Actual _reflns_number_gt Values Differ by	1	Check

0 ALERT level A = Most likely a serious problem - resolve or explain
1 ALERT level B = A potentially serious problem, consider carefully
10 ALERT level C = Check. Ensure it is not caused by an omission or oversight
7 ALERT level G = General information/check it is not something unexpected
1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
6 ALERT type 2 Indicator that the structure model may be wrong or deficient
7 ALERT type 3 Indicator that the structure quality may be low
2 ALERT type 4 Improvement, methodology, query or suggestion
2 ALERT type 5 Informative message, check

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1BrNaAA

Bond precision: C-C = 0.0033 A Wavelength=0.71073 Cell: a=3.8948(4) b=8.1149(8) c=17.1274(16)alpha=90.194(3) beta=91.725(4) qamma = 92.396(4)Temperature: 294 K Calculated Reported Volume 540.60(9) 540.60(9) Space group P -1 P -1 Hall group -P 1 -P 1 Moiety formula C13 H9 Br O2 C13 H9 Br O2 Sum formula C13 H9 Br O2 C13 H9 Br O2 Mr 277.10 277.11 1.702 1.702 Dx,g cm-3 2 2 Ζ Mu (mm-1) 3.781 3.781 F000 276.0 276.0 F000′ 275.56 h,k,lmax 4,10,21 4,10,21 2164 Nref 2174 0.573,0.635 Tmin,Tmax Tmin′ 0.562 Correction method= Not given Data completeness= 0.995 Theta(max) = 26.245R(reflections) = 0.0264(1929) wR2(reflections) = 0.0633(2164) S = 1.050Npar= 177

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level.

PLAT057_ALERT_3_C	Correction for Abs	sorption Required	l RT(exp)	 1.11	Do !
PLAT218_ALERT_3_C	Constrained U(ij)	Components(s) fo	or Hl	6	Check

Alert level G

PLAT007_ALERT_5_G Number of Unrefined Donor-H Atoms	1 Rep	ort
PLAT164_ALERT_4_G Nr. of Refined C-H H-Atoms in Heavy-Atom Struct.	8 Not	e
PLAT169_ALERT_4_G The CIF-Embedded .res File Contains AFIX 1 Recds	1 Rep	ort
PLAT910_ALERT_3_G Missing # of FCF Reflection(s) Below Theta(Min).	3 Not	e
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600	8 Not	e
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density.	5 Inf	0

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0 ALERT level A = Most likely a serious problem - resolve or explain
0 ALERT level B = A potentially serious problem, consider carefully
2 ALERT level C = Check. Ensure it is not caused by an omission or oversight
6 ALERT level G = General information/check it is not something unexpected
0 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
1 ALERT type 2 Indicator that the structure model may be wrong or deficient
3 ALERT type 3 Indicator that the structure quality may be low
3 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check
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It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1INaAA

Bond precision: C-C = 0.0042 A Wavelength=0.71073 Cell: a=4.8142(2) b=8.2211(4) c=14.3363(7)alpha=102.141(2) beta=91.410(2) qamma = 93.280(2)Temperature: 150 K Calculated Reported Volume 553.41(4) 553.41(4)P -1 Space group P -1 Hall group -P 1 -P 1 Moiety formula C13 H9 I O2 C13 H9 I O2 C13 H9 I O2 Sum formula C13 H9 I O2 Mr 324.10 324.10 1.945 1.945 Dx,g cm-3 2 2 Ζ Mu (mm-1) 2.873 2.873 F000 312.0 312.0 F000′ 311.18 h,k,lmax 6,11,19 6,11,19 2982 2953 Nref Tmin,Tmax Tmin' Correction method= Not given Data completeness= 0.990 Theta(max) = 29.104R(reflections) = 0.0302(2537) wR2(reflections) = 0.0633(2953) S = 1.086Npar= 149

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level.

PLAT053_ALERT_1_C Minimum Crystal Dimension Missing (or Error) ...Please CheckPLAT054_ALERT_1_C Medium Crystal Dimension Missing (or Error) ...Please CheckPLAT055_ALERT_1_C Maximum Crystal Dimension Missing (or Error) ...Please CheckPLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L=0.60010 Report

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0 ALERT level A = Most likely a serious problem - resolve or explain
0 ALERT level B = A potentially serious problem, consider carefully
4 ALERT level C = Check. Ensure it is not caused by an omission or oversight
5 ALERT level G = General information/check it is not something unexpected
4 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
2 ALERT type 2 Indicator that the structure model may be wrong or deficient
2 ALERT type 3 Indicator that the structure quality may be low
1 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check
```

It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

Publication of your CIF in IUCr journals

A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica, Journal of Applied Crystallography, Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E* or *IUCrData*, you should make sure that full publication checks are run on the final version of your CIF prior to submission.

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PLATON version of 07/08/2019; check.def file version of 30/07/2019

Datablock 1INaAA - ellipsoid plot



Structure factors have been supplied for datablock(s) 6BrNaAA

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 6BrNaAA

Bond precision: C-C = 0.0041 AWavelength=0.71073 Cell: a=3.9738(2) b=6.2790(4) c=21.6685(13)alpha=97.582(2) beta=93.911(2) qamma = 92.391(2)Temperature: 294 K Calculated Reported Volume 533.98(5) 533.98(5)P -1 Space group P -1 Hall group -P 1 -P 1 Moiety formula C13 H9 Br O2 C13 H9 Br O2 Sum formula C13 H9 Br O2 C13 H9 Br O2 Mr 277.10 277.11 1.723 1.723 Dx,g cm-3 2 2 Ζ Mu (mm-1) 3.828 3.828 F000 276.0 276.0 F000′ 275.56 h,k,lmax 4,7,26 4,7,26 2076 Nref 2083 Tmin,Tmax Tmin' Correction method= Not given Data completeness= 0.997 Theta(max) = 25.994R(reflections) = 0.0383(1950) wR2(reflections) = 0.1009(2076) S = 1.092Npar= 181

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level.

Alert level C		
PLAT053_ALERT_1_C Minimum Crystal Dimension Missing (or Error)	Please	Check
PLAT054_ALERT_1_C Medium Crystal Dimension Missing (or Error)	Please	Check
PLAT055_ALERT_1_C Maximum Crystal Dimension Missing (or Error)	Please	Check
PLAT094_ALERT_2_C Ratio of Maximum / Minimum Residual Density	3.23	Report
PLAT222_ALERT_3_C Non-Solv. Resd 1 H Uiso(max)/Uiso(min) Range	4.2	Ratio
PLAT790_ALERT_4_C Centre of Gravity not Within Unit Cell: Resd. #	1	Note
C13 H9 Br O2		
PLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L= 0.600	4	Report

PLAT154_ALERT_1_G The s.u.'s on the Cell Angles are Equal(Note)	0.002 Degree
PLAT164_ALERT_4_G Nr. of Refined C-H H-Atoms in Heavy-Atom Struct.	8 Note
PLAT434_ALERT_2_G Short Inter HLHL Contact Br1Br1	3.54 Ang.
1-x,-y,1-z =	2_656 Check
PLAT910_ALERT_3_G Missing # of FCF Reflection(s) Below Theta(Min).	2 Note
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600	2 Note
PLAT913_ALERT_3_G Missing # of Very Strong Reflections in FCF	3 Note
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density.	6 Info

0 ALERT level A = Most likely a serious problem - resolve or explain 0 ALERT level B = A potentially serious problem, consider carefully 7 ALERT level C = Check. Ensure it is not caused by an omission or oversight 7 ALERT level G = General information/check it is not something unexpected 4 ALERT type 1 CIF construction/syntax error, inconsistent or missing data 3 ALERT type 2 Indicator that the structure model may be wrong or deficient 4 ALERT type 3 Indicator that the structure quality may be low 3 ALERT type 4 Improvement, methodology, query or suggestion 0 ALERT type 5 Informative message, check

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Structure factors have been supplied for datablock(s) D-1FNaAA

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: D-1FNaAA

Bond precision: C-C = 0.0033 A Wavelength=0.71073 Cell: a=7.4331(7) b=21.337(2) c=15.4508(17)alpha=90 beta=98.686(5) gamma=90 Temperature: 100 K Calculated Reported Volume 2422.4(4)2422.4(4)P 21/n P 1 21/n 1 Space group Hall group -P 2yn -P 2yn Moiety formula C52 H36 F4 O8, 2(C3 H6 O) C26 H18 F2 O4, C3 H6 O Sum formula C58 H48 F4 O10 C29 H24 F2 O5 Mr 980.96 490.48 1.345 1.345 Dx,g cm-3 2 Ζ 4 Mu (mm-1) 0.101 0.101 F000 1024.0 1024.0 F000′ 1024.62 h,k,lmax 10,29,21 10,29,21 Nref 6753 6668 Tmin,Tmax Tmin' Correction method= Not given Data completeness= 0.987 Theta(max) = 29.488R(reflections) = 0.0562(3525) wR2(reflections) = 0.1275(6668) S = 1.019Npar= 421

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level. Click on the hyperlinks for more details of the test

RINTA01_ALERT_3_C Th	ne value of Rint i	ls greater than	0.12			
Rint give	en 0.175					
PLAT053_ALERT_1_C Mir	nimum Crystal Dime	ension Missing	(or Error)		Please	Check
PLAT054_ALERT_1_C Med	dium Crystal Dime	ension Missing	(or Error)		Please	Check
PLAT055_ALERT_1_C Max	kimum Crystal Dime	ension Missing	(or Error)		Please	Check
PLAT222_ALERT_3_C Nor	n-Solv. Resd 1 H	H Uiso(max)/U	iso(min) Ra	nge	10.0	Ratio
PLAT245_ALERT_2_C U(i	iso) HOOW Sma	aller than U(eq) COOW	by	0.011	Ang**2
PLAT303_ALERT_2_C Ful	ll Occupancy Atom	H003 with	# Connecti	ons	2.00	Check
PLAT355_ALERT_3_C Lor	ng O-H (X0.82,N0	0.98A) 0004	- H004		1.07	Ang.
PLAT391_ALERT_3_C Dev	viating Methyl C01	LO Н-С-Н В	ond Angle .		117	Degree
PLAT391_ALERT_3_C Dev	viating Methyl C01	LO Н-С-Н В	ond Angle .		101	Degree
PLAT905_ALERT_3_C Neg	gative K value in	the Analysis o	f Variance		-2.764	Report
PLAT910_ALERT_3_C Mis	ssing # of FCF Ref	Election(s) Bel	ow Theta(Mi	n).	5	Note

Alert level G

PLAT020_ALERT_3_G The Value of Rint is Greater That	an 0.12	0.175	Report
PLAT042_ALERT_1_G Calc. and Reported MoietyFormula	a Strings Differ	Please	Check
PLAT045_ALERT_1_G Calculated and Reported Z Differ	r by a Factor	0.50	Check
PLAT720_ALERT_4_G Number of Unusual/Non-Standard I	Labels	60	Note
PLAT793_ALERT_4_G Model has Chirality at COOC	(Centro SPGR)	R	Verify
PLAT793_ALERT_4_G Model has Chirality at COOG	(Centro SPGR)	S	Verify
PLAT793_ALERT_4_G Model has Chirality at COOI	(Centro SPGR)	S	Verify
PLAT793_ALERT_4_G Model has Chirality at COOJ	(Centro SPGR)	R	Verify
PLAT912_ALERT_4_G Missing # of FCF Reflections Abc	ove STh/L= 0.600	80	Note
PLAT978_ALERT_2_G Number C-C Bonds with Positive F	Residual Density.	6	Info

0 ALERT level A = Most likely a serious problem - resolve or explain 0 ALERT level B = A potentially serious problem, consider carefully 12 ALERT level C = Check. Ensure it is not caused by an omission or oversight 10 ALERT level G = General information/check it is not something unexpected 5 ALERT type 1 CIF construction/syntax error, inconsistent or missing data 3 ALERT type 2 Indicator that the structure model may be wrong or deficient 8 ALERT type 3 Indicator that the structure quality may be low 6 ALERT type 4 Improvement, methodology, query or suggestion 0 ALERT type 5 Informative message, check

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Structure factors have been supplied for datablock(s) D-1ClNaAA

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: D-1ClNaAA

Bond precision: C-C = 0.0017 AWavelength=0.71073 Cell: a=13.8413(5) b=6.6231(2) c=27.4197(9) beta=91.789(1) alpha=90 gamma=90 Temperature: 100 K Calculated Reported Volume 2512.40(14) 2512.40(14)P 21/c P 1 21/c 1 Space group Hall group -P 2ybc -P 2ybc Moiety formula C26 H18 Cl2 O4, C4 H8 O C26 H18 Cl2 O4, C4 H8 O Sum formula СЗО Н26 С12 О5 C30 H26 C12 O5 Mr 537.41 537.41 1.421 1.421 Dx,g cm-3 Ζ 4 4 Mu (mm-1) 0.299 0.299 F000 1120.0 1120.0 F000′ 1121.69 h,k,lmax 18,8,35 18,8,35 5819 Nref 5869 Tmin,Tmax Tmin' Correction method= Not given Data completeness= 0.991 Theta(max) = 27.666R(reflections) = 0.0292(5293) wR2(reflections) = 0.0766(5819) S = 1.067Npar= 438

The following ALERTS were generated. Each ALERT has the format test-name_ALERT_alert-type_alert-level.

CRYSC01_ALERT_1_C The word below has not been recognised as a standard identifier. dull CRYSC01_ALERT_1_C The word below has not been recognised as a standard identifier. whiteish PLAT053_ALERT_1_C Minimum Crystal Dimension Missing (or Error) ... Please Check PLAT054_ALERT_1_C Medium Crystal Dimension Missing (or Error) ... Please Check PLAT055_ALERT_1_C Maximum Crystal Dimension Missing (or Error) ... Please Check

Alert level G

PLAT398_ALERT_2_G Devia	ting C-O-C	Angle From 120	0 for 0004	10	5.0	Degree
PLAT720_ALERT_4_G Number	r of Unusual/N	on-Standard Lab	oels		63	Note
PLAT793_ALERT_4_G Model	has Chirality	at COOD	(Centro Si	PGR)	S	Verify
PLAT793_ALERT_4_G Model	has Chirality	at COOJ	(Centro Si	PGR)	R	Verify
PLAT793_ALERT_4_G Model	has Chirality	at COOK	(Centro Si	PGR)	R	Verify
PLAT793_ALERT_4_G Model	has Chirality	at COON	(Centro Si	PGR)	S	Verify
PLAT910_ALERT_3_G Missin	ng # of FCF Re	flection(s) Be	low Theta(M	in).	4	Note
PLAT912_ALERT_4_G Missin	ng # of FCF Re	flections Above	e STh/L= 0	.600	41	Note
PLAT953_ALERT_1_G Report	ted (CIF) and	Actual (FCF) Hr	max Differ]	by .	1	Units
PLAT978_ALERT_2_G Number	r C-C Bonds wi	th Positive Rea	sidual Dens	ity.	21	Info

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