

Supplementary Information

Structure and properties of an Fe(III) complex containing a novel amide functionalized polyimidazole ligand.

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Physical Measurements:

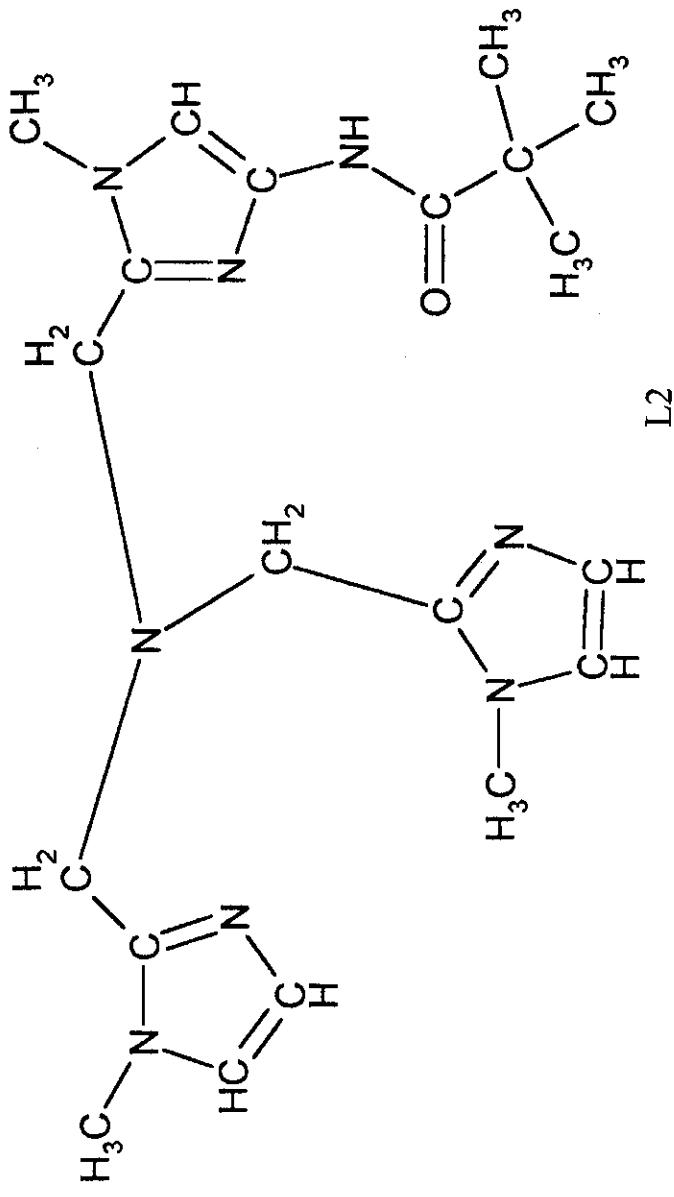
Compounds L1, L2, 1 and 2 have been characterized by elemental analysis (Midwest Analytical, Inc.) X-ray diffraction, ^1H , ^{13}C NMR, UV-visible and FT-IR spectroscopies, electrospray mass spectrometry, and cyclic voltammetry.

Single crystal X-ray diffraction studies were performed using a Brucker Smart Apex diffractometer equipped with a cryoflex low-temperature apparatus, or on an Enraf-Nonius CAD4 diffractometer. Crystallographic data of compounds 1 and 2 were recorded at 293 and 100K, respectively.

^1H , ^{13}C NMR spectra of L2 in CDCl_3 were recorded using a Varian Unity Inova 500 MHz NMR spectrometer at room temperature. UV-vis spectra of compounds 1 and 2 ($1 \times 10^{-4}\text{M}$ in CH_3CN) were recorded using a Cary 50 Bio spectrophotometer. FT-IR spectrum of L2 was recorded using a Mattson Galaxy Series FTIR 5000 spectrophotometer. A chloroform solution of L2 was evaporated on Ag/AgCl plates prior to recording the spectrum. Positive ion mode electrospray mass spectra of compounds L2 and 1 were recorded using Micromass Mass Spectrometer Electrospray Triple Quadrupole Quattro LCZ.

Cyclic voltammograms of compounds 1 and 2 ($1 \times 10^{-3}\text{M}$ in CH_3CN) containing 0.1 M TBAP were recorded using a standard three-electrode cell which included a glassy carbon working electrode, Pt-coil auxillary electrode and a Ag/AgCl reference electrode. The electrochemical measurements were recorded using a PARC Model 175 programmer; Model 173 potentiostat/gavanostat and Houston 2000 X-Y recorder.

Structure and Formula



Molecular Weight
Molecular Formula
Elemental Analysis

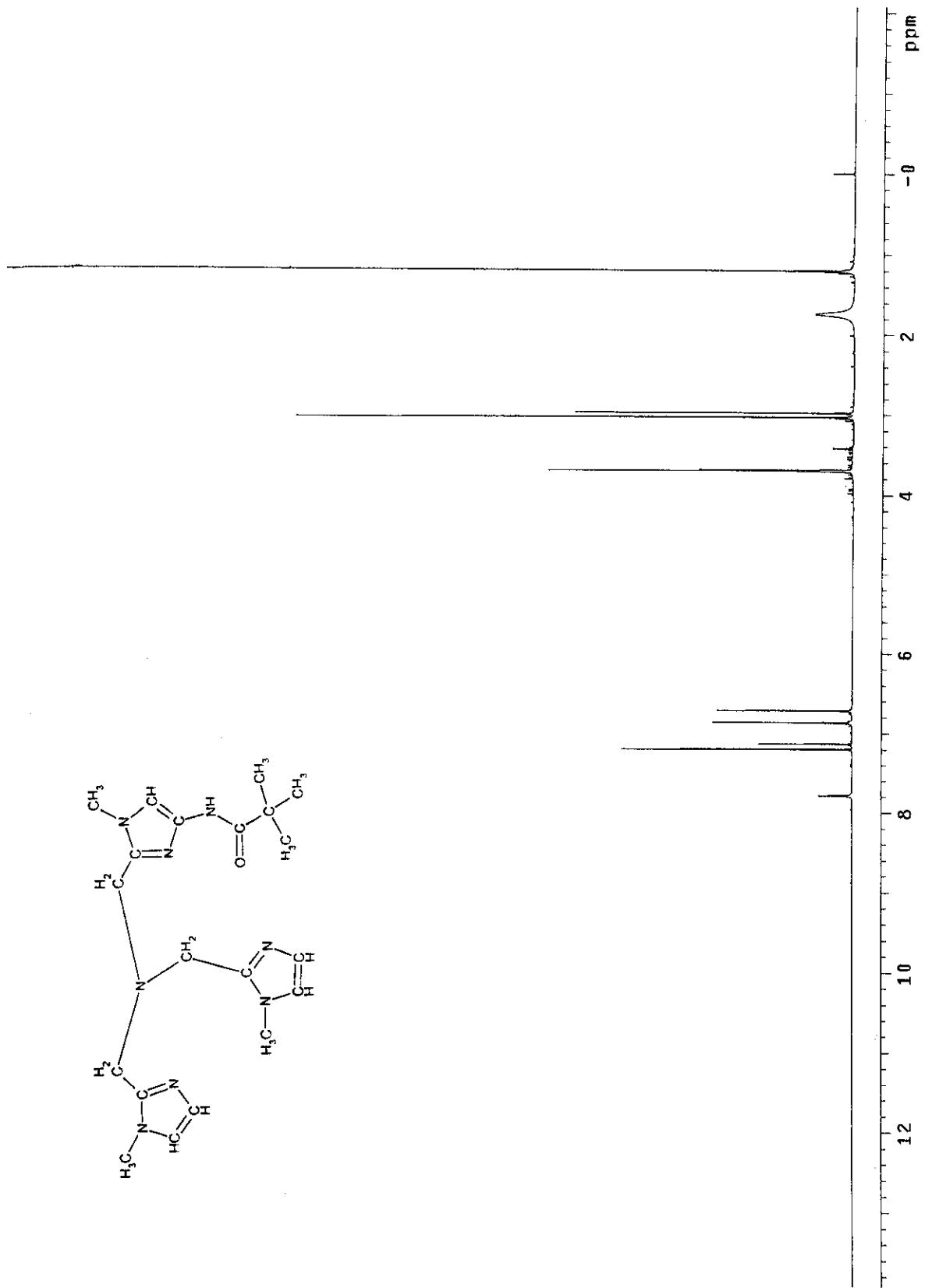
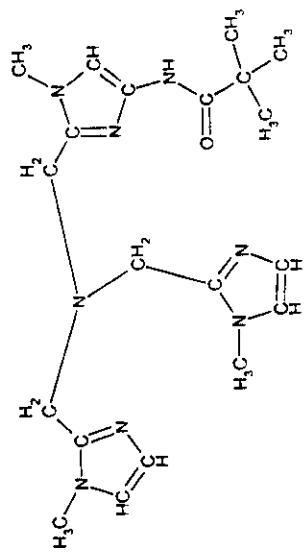
398.52

$\text{C}_{20}\text{H}_{30}\text{N}_8\text{O}$
 $\text{C}_{20}\text{H}_{30}\text{N}_8\text{O} \cdot \text{H}_2\text{O}$

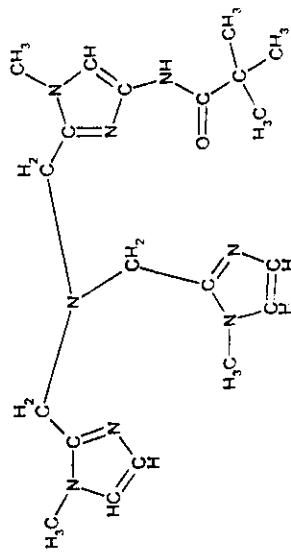
Cald. : C 57.69, H 7.69, N 26.92
Found: C 58.00, H 7.94, N 27.04

(water detected in proton spectrum of compound in CDCl_3)
ESI-MS 399.2 ($\text{M}+\text{H}^+$)

¹H NMR in CDCl₃



^{13}C NMR in CDCl_3



NMR spectroscopic data

¹H NMR in CDCl₃

L2

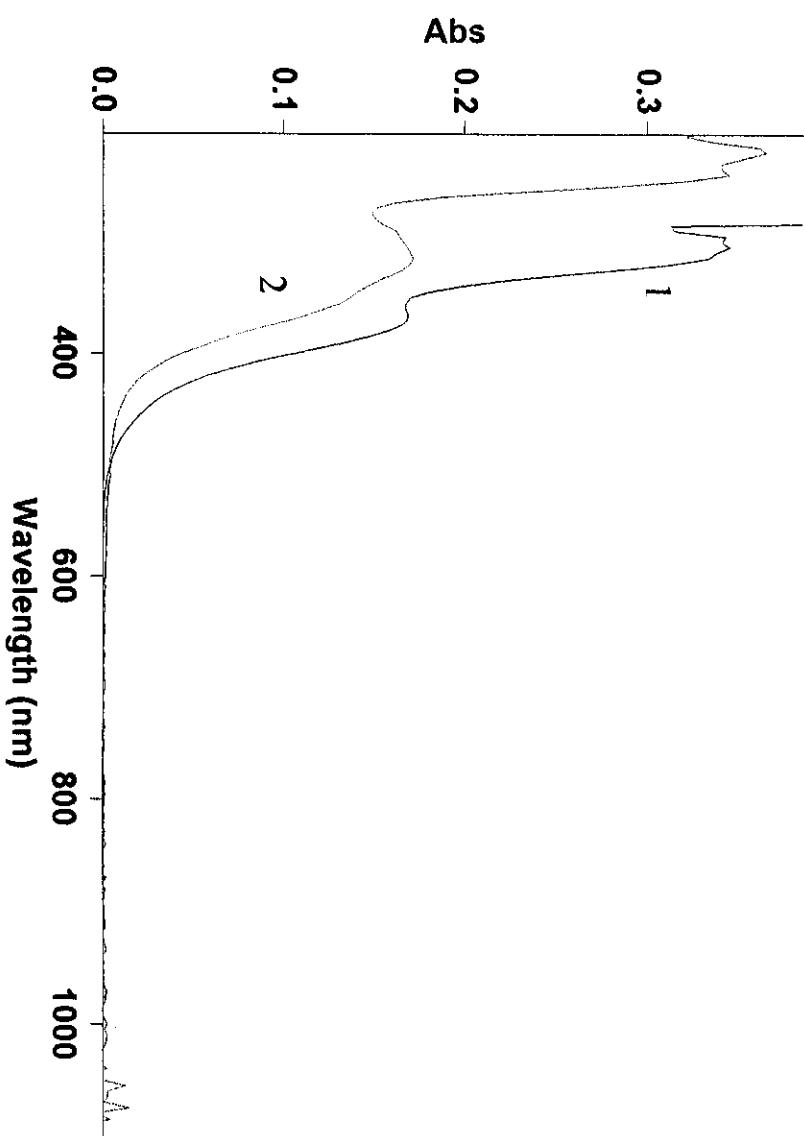
ppm	Assignment	Integral	Multiplicity
7.8	^{Am} Im(-NH-)	1	s
7.21	^{Am} Im(C5-H)	1	s
6.95	Im(C5-H)	1	s
6.8	Im(C4-H)	1	s
3.78	Im(-CH ₂ -)	4	s
3.77	^{Am} Im(-CH ₂ -)	2	s
3.11	Im(N-CH ₃ -)	6	s
3.05	^{Am} Im(N-CH ₃ -)	3	s
1.29	^{Am} Im(^t Bu)	9	s

¹³C NMR in CDCl₃

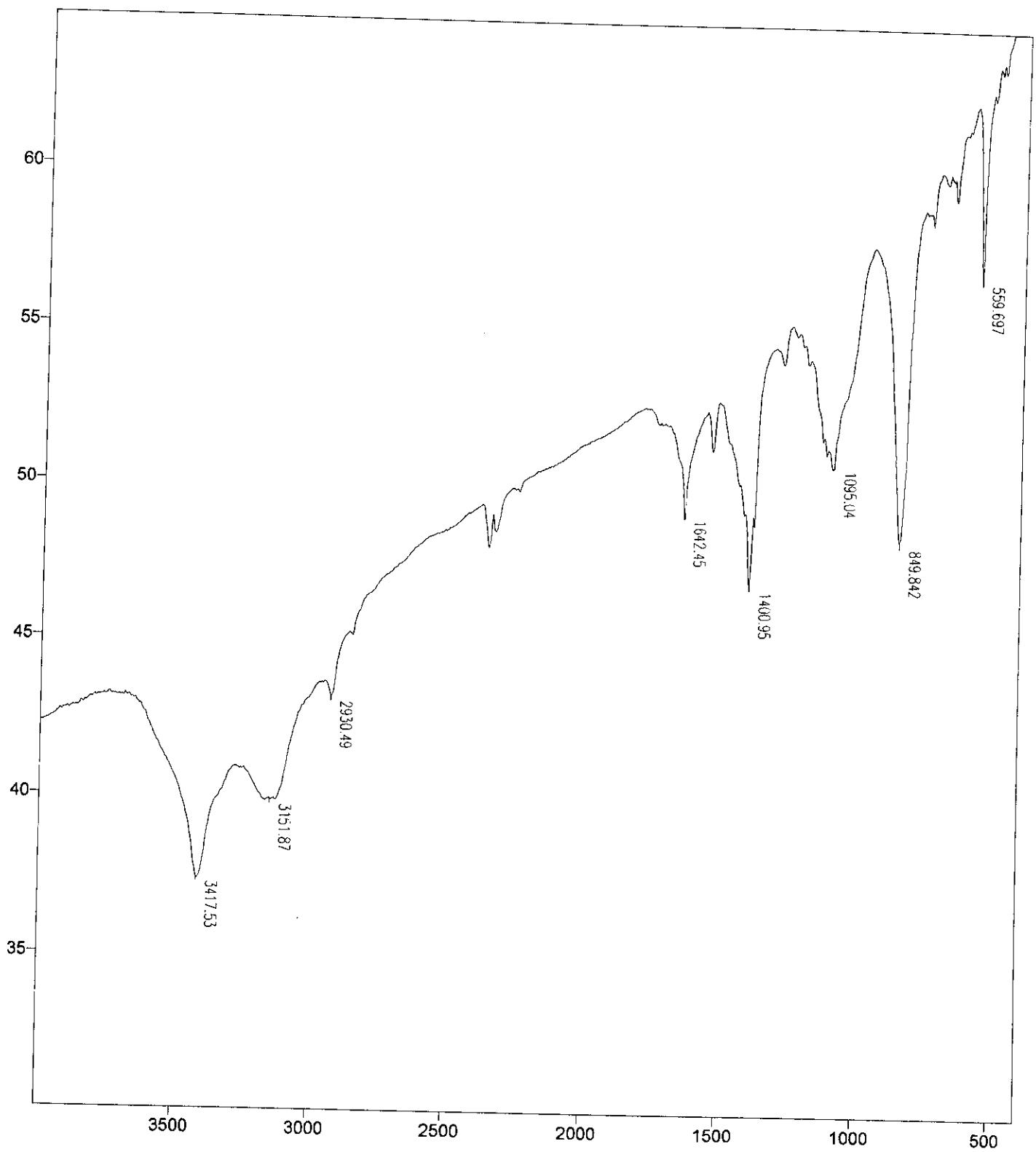
ppm	Assignment
175.8	^{Am} Im(-C=O)
145.4	Im(C2)
141.2	^{Am} Im(C2)
136.3	^{Am} Im(C4)
127.7	Im(C5)
121.7	Im(C4)
109.2	^{Am} Im(C5)
48.9	Im(-CH ₂ -), ^{Am} Im(-CH ₂ -)
39.2	^{Am} Im(^t Bu-C)
32.1	Im(N-CH ₃ -), ^{Am} Im(N-CH ₃ -)
27.8	^{Am} Im(^t Bu)

^{Am}Im corresponds to the amide functionalized imidazole ring

UV-visible Spectra of Compounds 1 and 2

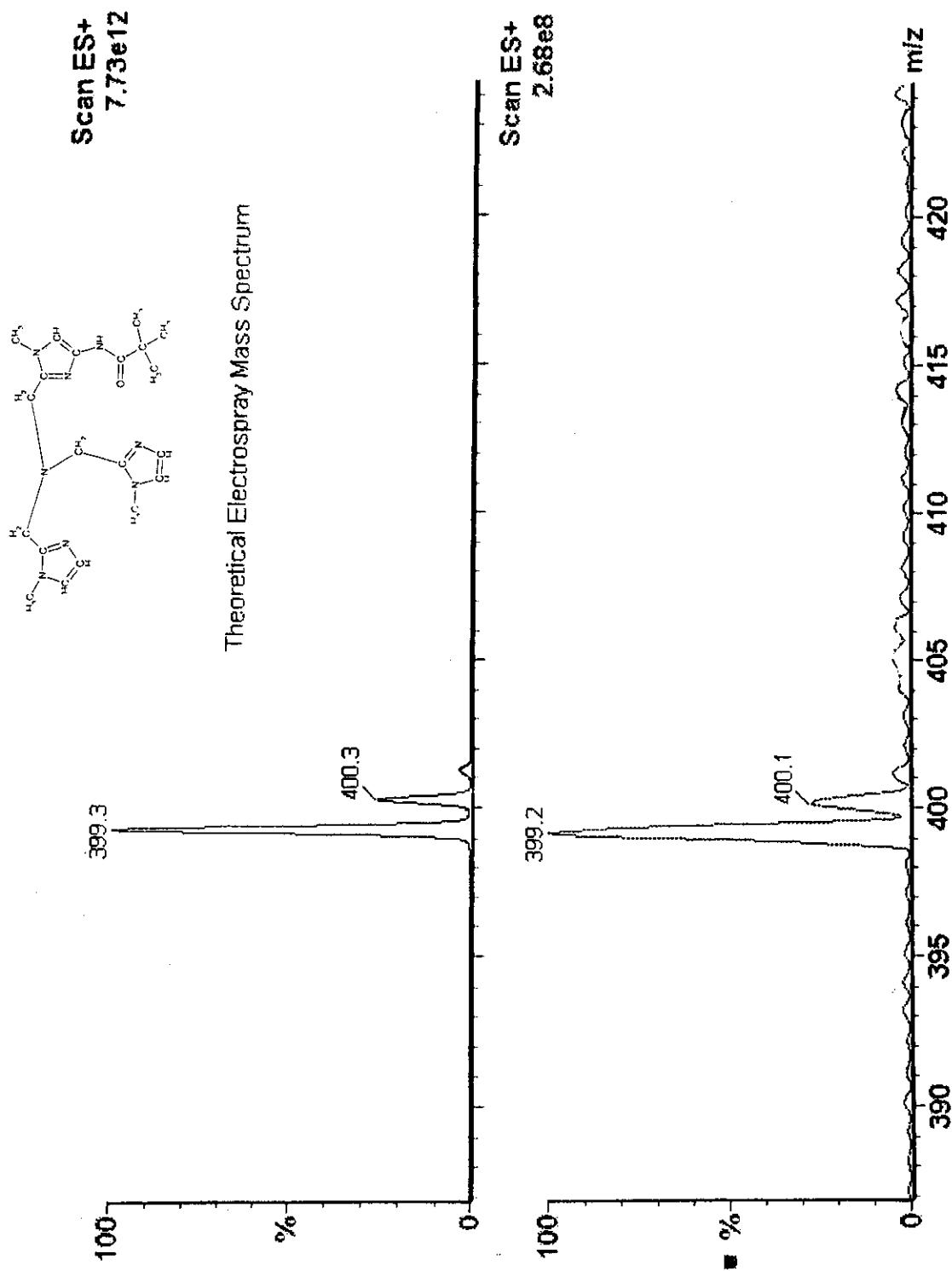


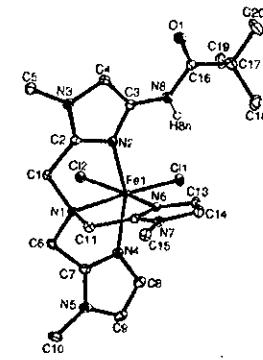
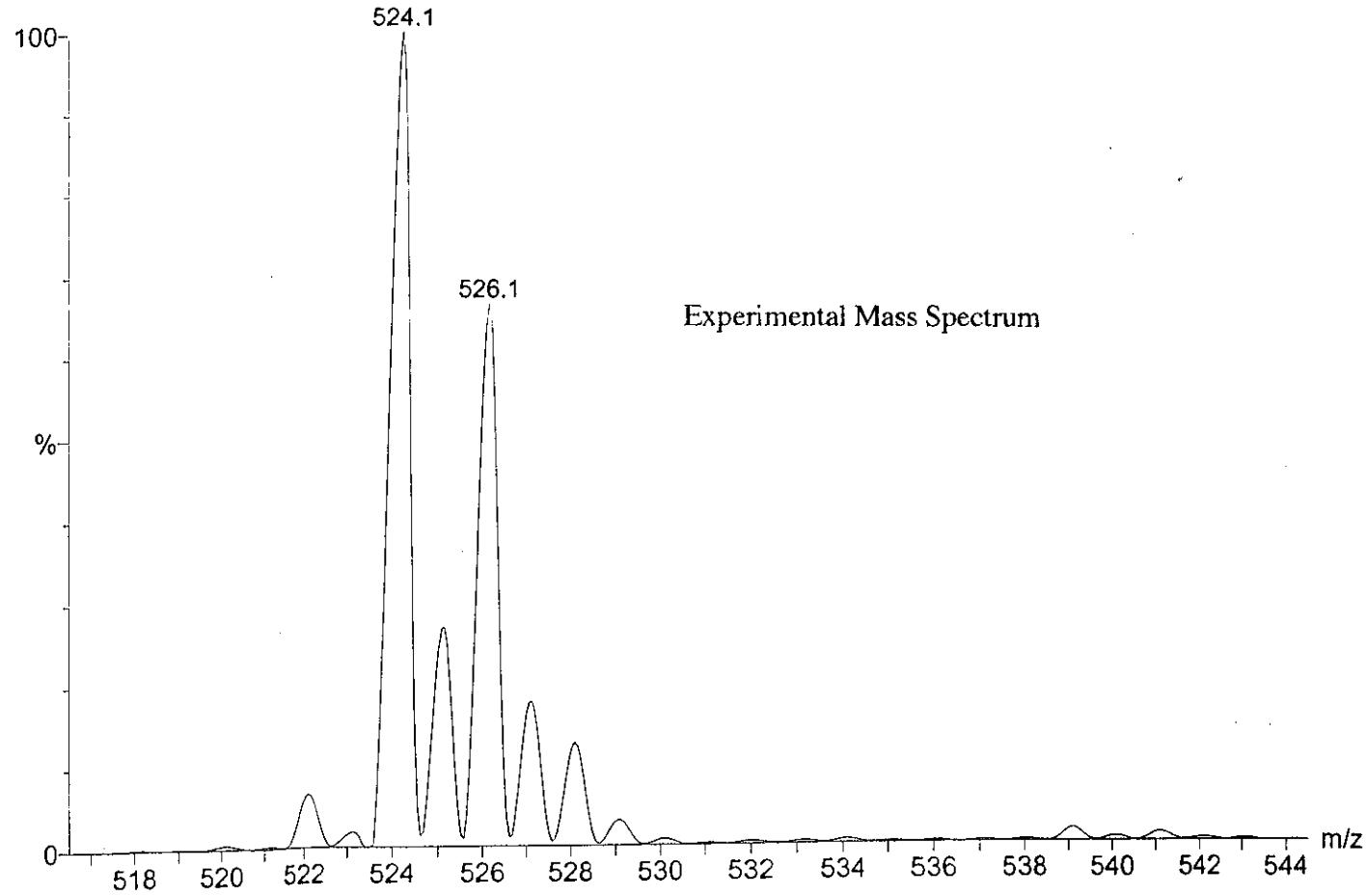
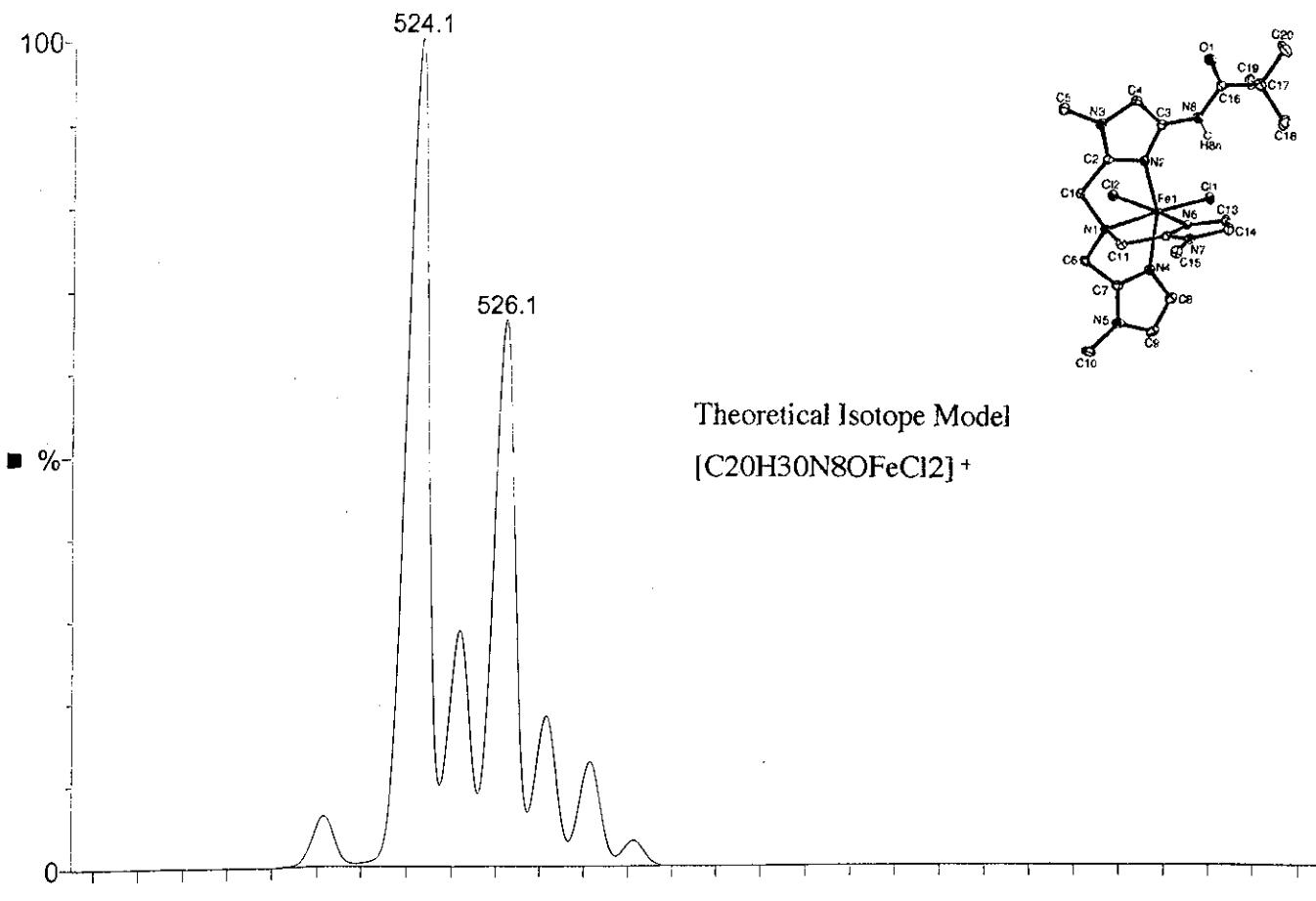
FT-IR Spectrum of L2

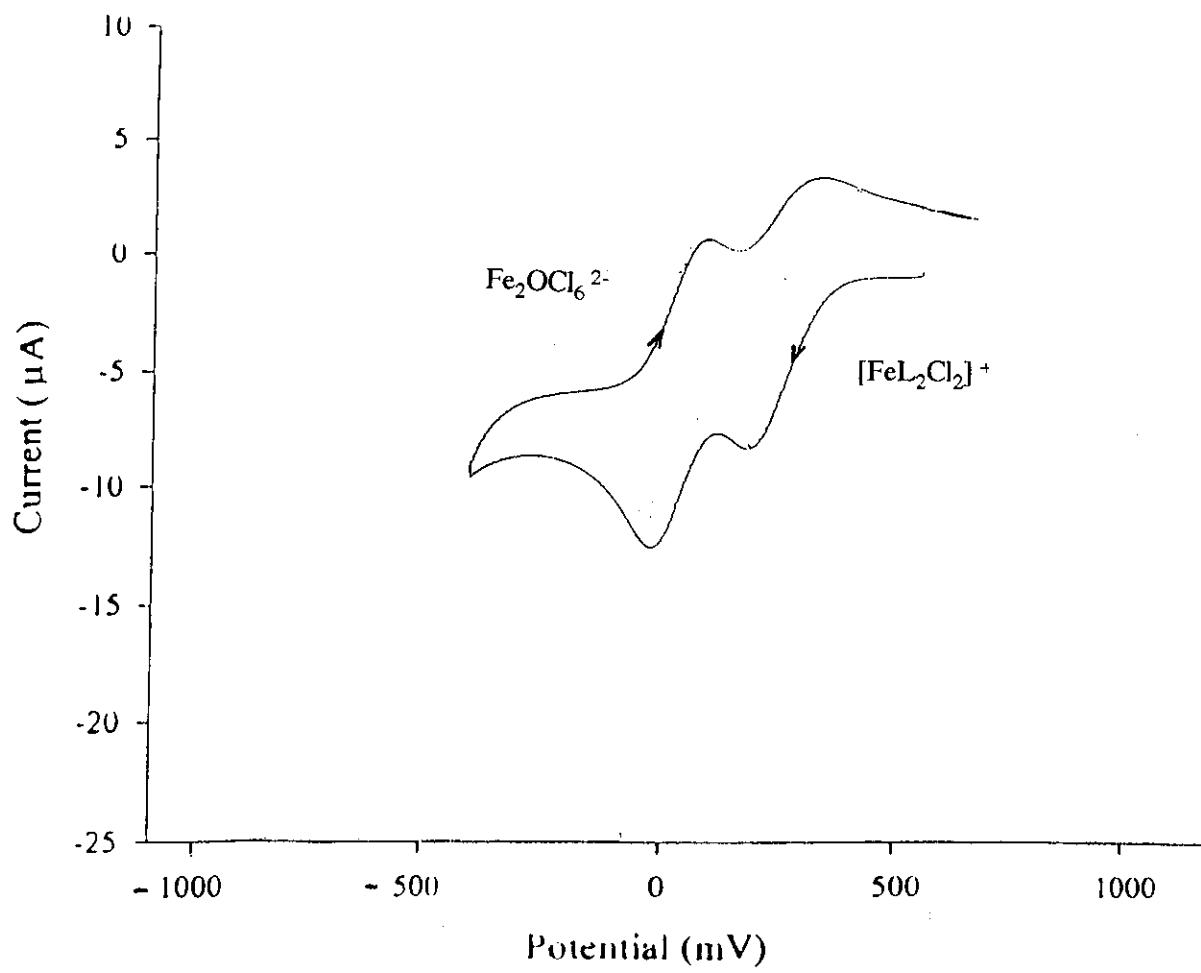
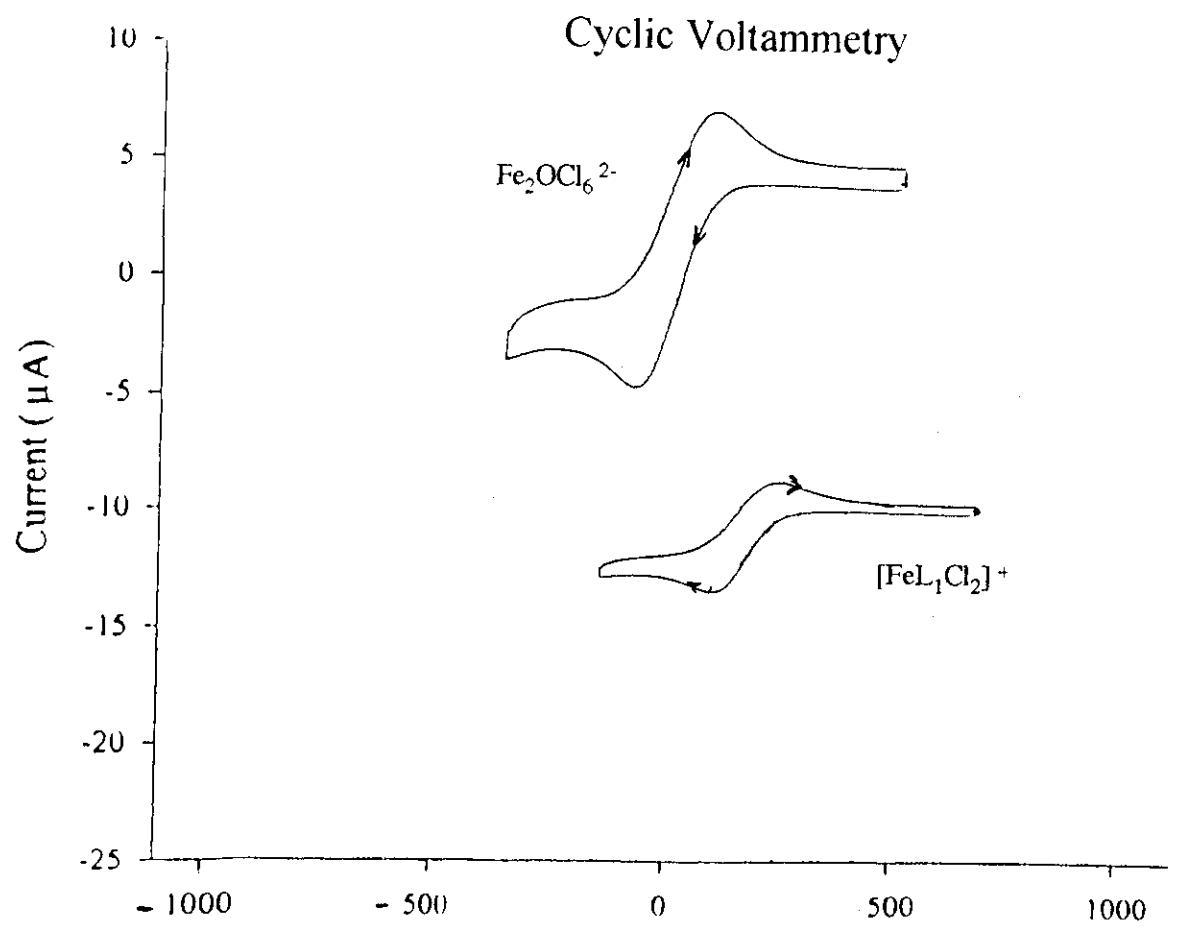


ESI-MS

Molecular Ion

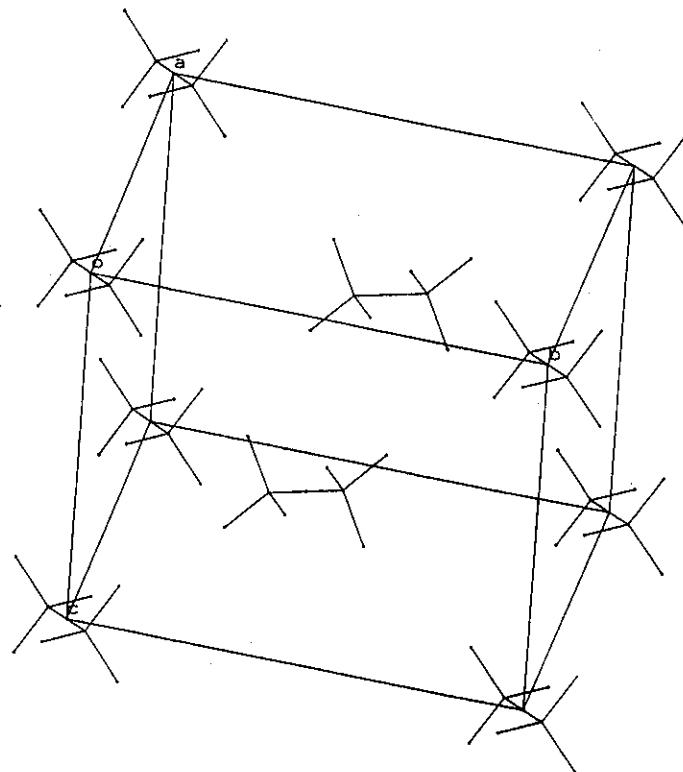






Packing diagrams

Packing diagram for the anionic species $\text{Fe}_2\text{OCl}_6^{2-}$



Packing diagram for the cationic species $\text{Fe}(\text{L}2)\text{Cl}_2^+$

