Figure 4. The [100], [111] and [110] projections of a typical diamond structure with *Fd3m* space group are as follows:



Figure 5. ¹H NMR spectrum for the tri-head-group surfactant $C_{18}H_{37}N(CH_3)_2C_3H_6N(CH_3)_3\bullet 3Br$ ($C_{18-2-3-1}$). The spectrum was measured on Brucker DMX 500 spectrometer at 500MHz with D₂O as a solvent.



Table 1 Assignment and chemical shift data of ¹H NMR spectrum for the tri-head-group surfactant $C_{18}H_{37}N(CH_3)_2CH_2CH_2N(CH_3)_2C_3H_6N(CH_3)_3\bullet 3Br$ ($C_{18-2-3-1}$).

Assignment	Α	В	С	D	Е	F
Chemical	0.830	1.245	1.354	1.802	2.434	3.188
Shift (ppm)						
Assignment	G	Н	Ι	J	K	
Chemical	3.273	3.350	3.463	3.580	4.154	
Shift (ppm)						

The elemental analysis (%) calcd for the tri-head-group surfactant $C_{18}H_{37}N(CH_3)_2CH_2CH_2$ N(CH₃)₂C₃H₆N(CH₃)₃•3Br (C₁₈₋₂₋₃₋₁) is C 50.7, H 9.68, N 5.92; found: C 48.5, H 9.74, N 5.95. Figure 6. SEM image for as-synthesized mesoporous silica prepared by using $C_{18-2-3-1}$ as a structure-directing agent under base condition at 17°C.



Experimental Characterization:

X-ray powder diffraction (XRD) patterns were taken on a Bruker D4 Endeavor diffractometer by using Cu K α radiation with voltage 40 kV and current 40 mA. The nitrogen adsorption/desorption isotherms were measured at 77 K by using a Micromeritics ASAP 2010 system. The samples were degassed at 250 °C for 3 hour. The transmission electron micrographs (TEM) were taken on a 2010 JEOL electron microscope operating at 200 kV. The samples were mounted on a holy carbon copper grid by placing a droplet of a suspension of ground sample in Scan electron micrographs (SEM) were taken on a XL 30 Philips 25 kV.