

Supplementary materials for:

Lanthanide(III) complexes of a pyridine *N*-oxide analogue of DOTA: exclusive *M* isomer formation induced by six-membered chelate ring

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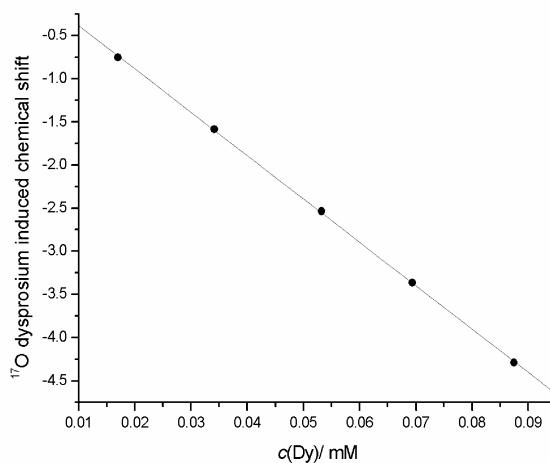


Figure S1. Induced ^{17}O chemical shift of water in presence of $[\text{Dy}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ (pH = 7, 25 °C, 400 MHz). The slope obtained from least-square was found to be -50 .

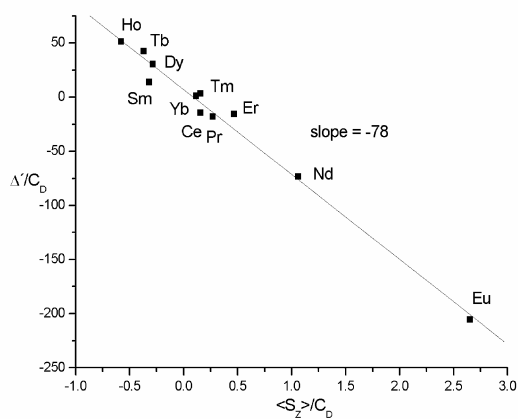


Figure S2A. Water ^{17}O chemical shifts in presence of $[\text{Ln}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ (pH = 7; 25 °C, 400 MHz).

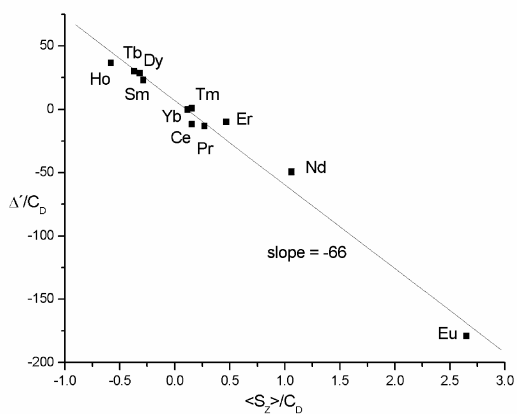


Figure S2B. Water ^{17}O chemical shifts in presence of $[\text{Ln}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ (pH = 7; 70 °C, 400 MHz).

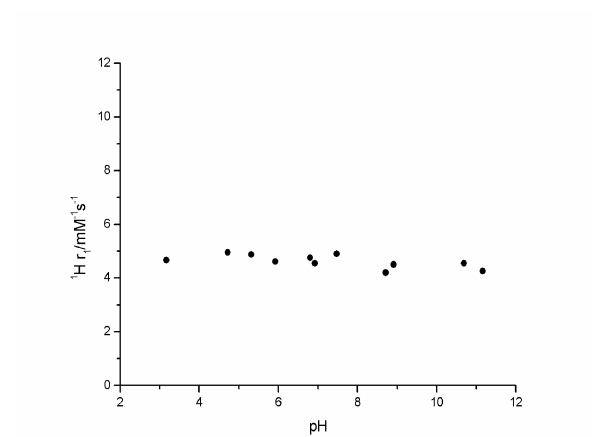


Figure S3. pH dependence of water proton relaxivity in presence of $[\text{Gd}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ complex.

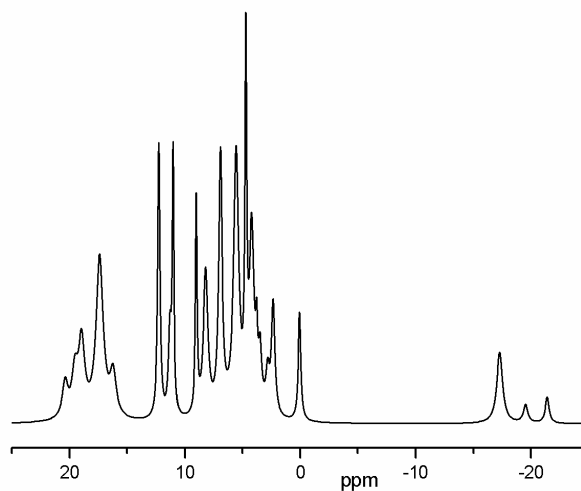


Figure S4A. ^1H NMR spectrum of $[\text{Nd}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ ($\text{CD}_3\text{OD}/\text{D}_2\text{O}$ 4/1; 25 °C, 400 MHz).

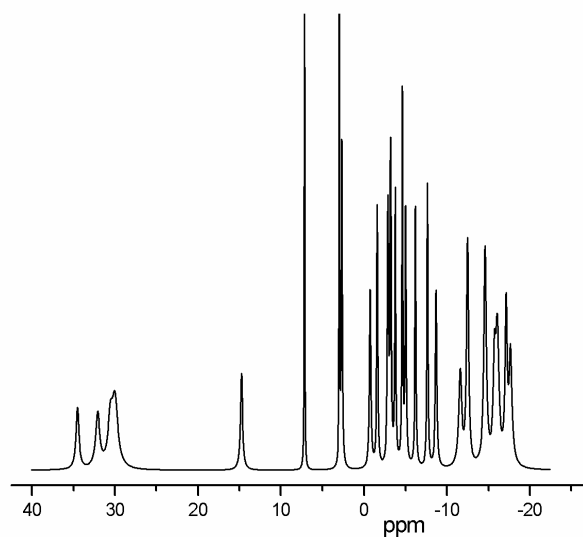


Figure S4B. ^1H NMR spectrum of $[\text{Eu}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ ($\text{CD}_3\text{OD}/\text{D}_2\text{O}$ 4/1; 25 °C, 400 MHz).

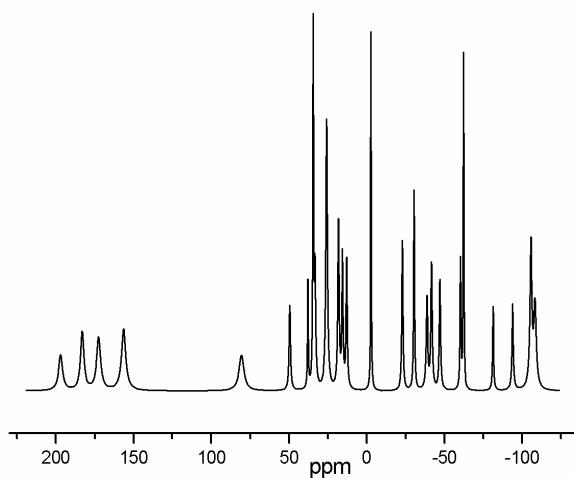


Figure S4C. ^1H NMR spectrum of $[\text{Yb}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ ($\text{CD}_3\text{OD}/\text{D}_2\text{O}$ 4/1; -25 °C, 400 MHz).

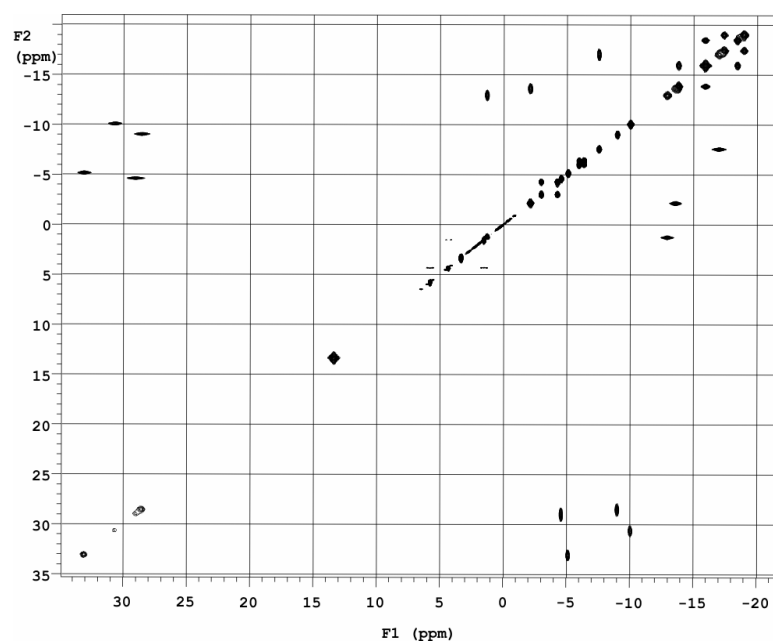


Figure S5. ^1H EXSY spectrum of $[\text{Eu}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ ($\text{CD}_3\text{OD}/\text{D}_2\text{O}$ 4/1; 25 °C, 400 MHz). The FID was accumulated 4 h with mixing time $t_{\text{mix}} = 20$ ms.

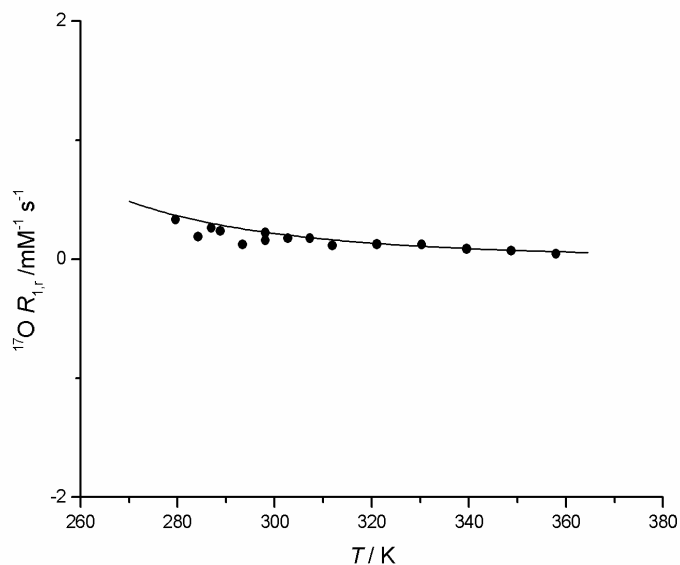


Figure S6A. Variable-temperature water ^{17}O R_{1r} relaxation rates measured in presence of $[\text{Gd}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ (pH = 7, 400 MHz). The curve represents the best result of simultaneous fitting according to BMS equation.

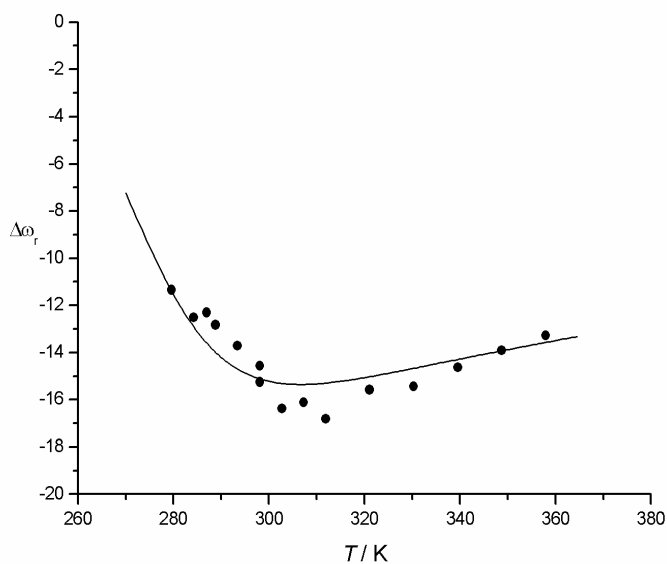


Figure S6B. Variable-temperature water ^{17}O reduced resonance frequencies measured in presence of $[\text{Gd}(\text{do3a-pyNox})(\text{H}_2\text{O})]$ (pH = 7, 400 MHz). The curve represents the best result of simultaneous fitting according to BMS equation.

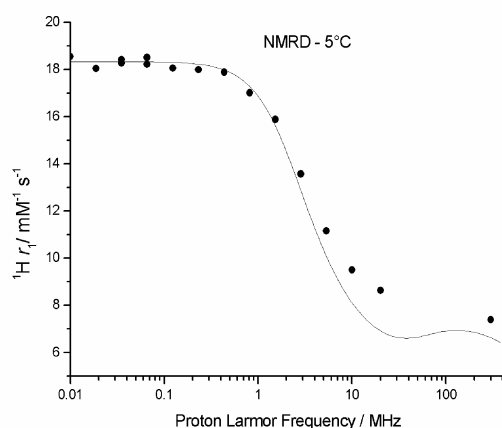


Figure S7A. ^1H NMRD profile at pH = 7 and 5 °C. The curve represents the best result of simultaneous fitting according to BSM equations.

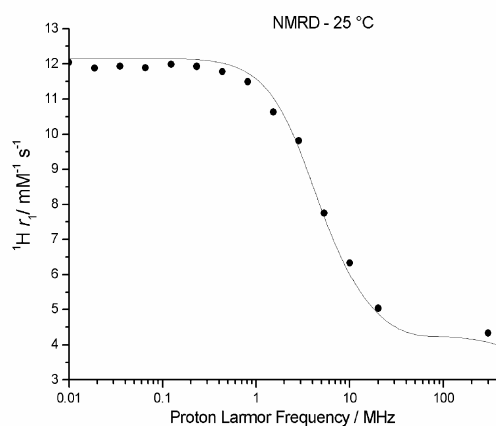


Figure S6B. ^1H NMRD profile at pH = 7 and 25 °C. The curve represents the best result of simultaneous fitting according to BMS equations.

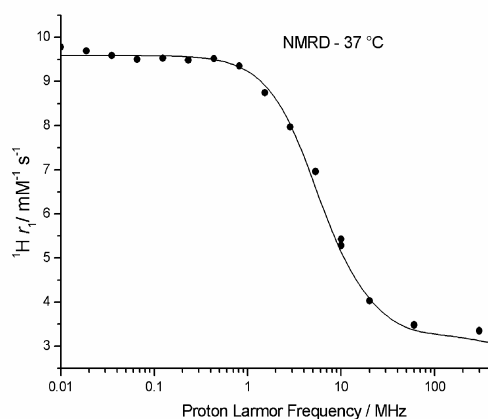


Figure S6C. ^1H NMRD profile at pH = 7 and 37 °C. The curve represents the best result of simultaneous fitting according to BSM equations.

Table S1. The complete set of results obtained from simultaneous least-square fitting of data.

Parameter	[Gd(DO3A-pyNox)(H ₂ O)]
$r_1^{310} [\text{s}^{-1} \cdot \text{mM}^{-1}]^a$	4.04
$\Delta^2 [10^{20} \text{s}^{-2}]$	0.69 ± 0.03
$\tau_M^{298} [\text{ns}]$	39 ± 1
$\Delta H_M^\# [\text{kJ} \cdot \text{mol}^{-1}]$	51 ± 2
$\tau_R^{298} [\text{ps}]$	74 ± 3
$\tau_v^{298} [\text{ps}]$	2.7 ± 0.1
$A/\hbar [10^6 \text{rad} \cdot \text{s}^{-1}]$	<u>-3.75</u>
$r_{\text{GdO}} [\text{\AA}]$	<u>2.5</u>
$r_{\text{GdH}} [\text{\AA}]$	<u>3.1</u>
$A [\text{\AA}]$	<u>3.5</u>
q	<u>1</u>

^a The value of millimolar relaxivity was obtained at 37 °C and 20 MHz, the value found for 25 °C and 10 MHz is $6.31 \text{ s}^{-1} \cdot \text{mM}^{-1}$.