Supplementary Material (ESI) for Chemical Communications

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Physical properties of initiators

Free-base initiator ¹H NMR (CDCl₃, 300.13 MHz) δ 8.88-8.81 (m, 8H, β-pyrrole*H*) δ 8.24 (d, 2H, Ar*H*O, ³J = 8.5 Hz) δ 8.09 (d, 6H, Ar*H*CH₃, ³J = 7.7) δ 7.56 (d, 6H, Ar*H*CH₃, ³J = 7.7) δ 7.54 (d, 2H, Ar*H*O, ³J = 8.22 Hz), δ 2.71 (s, 9H, ArC*H*₃)), δ 2.23 (s, 6H, C(C*H*₃)₂Br)), δ - 2.78 (s, 2H, N*H*) ppm. Maldi-TOF mass spectrometry (reflective mode) *m/z* 820, 821, 822, 823, 824, 825 (isotopic pattern M⁺). UV-vis (CHCl₃) λ 420, 517, 553, 594, 648 nm. IR (KBr pellet) $v_{(C=0)}$ 1756 cm⁻¹.

Cu(II) initiator	UV-vis (CHCl ₃) λ 417, 541, 574 nm.
Mn(III) initiator	UV-vis (CHCl ₃) λ 382, 486, 586, 626 nm.
Zn(II) initiator	UV-vis (CHCl ₃) λ 421, 549, 586 nm.

ATRP procedure

A schlenk tube was charged with the porphyrin initiator and CuBr and vacuum-nitrogen refilled three times. Subsequently, the schlenk tube was capped with a septum and styrene and anisole were added with a syringe. The resulting mixture was purged with nitrogen for 5 min. and placed in an ice-bath, after which PMDETA was added as the ligand. Next, the tube was placed in an oil bath with a temperature of 90 °C and the reaction mixture was sampled periodically to determine the ratio styrene/anisole with gas chromatography.

Characterization of Cu(II) porphyrin polystyrene



UV-VIS Cu(II) porphyrin polystyrene (toluene) λ 417, 540, 575 nm. ¹H NMR (CDCl₃, 300.13 MHz) δ 1.22-1.71 ((CH₂-CHPh)_n) δ 1.71-2.35 ((CH₂-CHPh)_n) δ 6.28-7.25 ((CH₂-CHPh)_n ppm, The paramagnetic Cu(II) porphyrin is not visible with NMR.

Sample treatment

After polymerization the mixture was dissolved in toluene and purified over a short size exclusion column (Bio-beads S-X1) with toluene as the eluents to remove traces of copper and unreacted monomer from the polymer batch. Preparation of the (Cu)porphyrin

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polystyrene samples for TEM and SEM studies was performed by injecting 100 μ l of a THF solution (0.3 mg/ml) of the compound into 1 ml of milliQ water while sonicating at 60 °C. A drop of the resulting cloudy solution was placed on a carbon coated copper grid and the water was drained off after 2 min.

For the (Mn) porphyrin polystyrene studies, to 100 μ l of a THF solution (3 mg/ml) of the compound was slowly added 200 μ l of water. A drop of the resulting solution was placed on a carbon coated copper grid and the drop was drained off after 2 min.

TEM images were obtained with a JEOL JEM-1010 microscope (60 kV) equipped with a CCD camera. SEM images were recorded with a JEOL JSM-6330F microscope.