Synthesis and Characterization of the First Trimetallic Nitride Template Endohedral Metallofulleropyrrolidines

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Experimental Details

♦ Synthesis and Isolation

1. Sc₃N@C₈₀ pyrrolidine derivative **2a** (natural abundance carbon):

To a solution of 5.0 mg (4.5 µmol) of Sc₃N@C₈₀ in 50 mL of *o*-dichlorobenzene, 1.2 mg (14 µmol) of N-methylglycine and 1.6 mg (53 µmol) of paraformaldehyde were added. The mixture was heated at 110°C in oil bath for 10 h. The solvent was removed *in vacuo* and the crude solid was purified by silica gel column chromatography (eluent: *o*-dichlorobenzene). The collected solid was dissolved in CS₂ and injected into an HPLC for final purification. HPLC condition: PYE [2-1-(pyrenyl)ethyl silica] column (10 × 250 mm), mobile phase (CS₂), flow rate: 0.5mL/min, λ = 390nm. This led to the isolation of 2 mg (40%) of pure **2a** as a black solid.

2. Sc₃N@C₈₀ pyrrolidine derivative **2a** (99% ¹³C labeled)

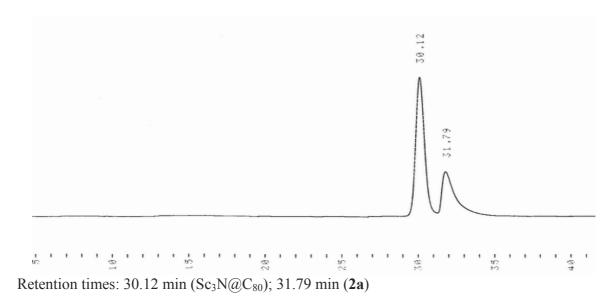
To a solution of 2.0 mg (1.8 μ mol) of Sc₃N@C₈₀ in 30 mL of *o*-dichlorobenzene, 0.7 mg (9 μ mol) of N-methylglycine and 3 μ l (0.03 mmol) of formaldehyde (99% ¹³C, ~20% W/W in H₂O) (Cambridge Isotope Laboratory, Inc.) were added. The product was purified as indicated above. This led to the isolation of 0.6 mg (30%) of pure **2a** as a black solid.

3. $Er_3N@C_{80}$ pyrrolidine derivative **2b** (99% ¹³C labeled)

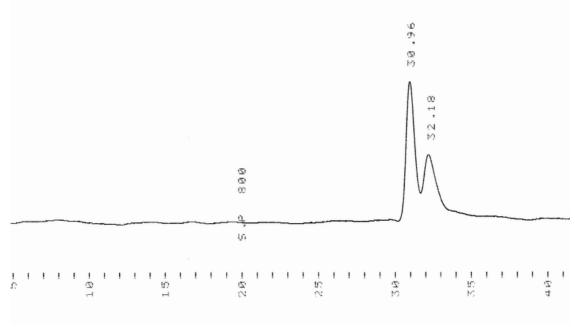
To a solution of 2.2 mg (1.5 μ mol) of Er₃N@C₈₀ in 30 mL of *o*-dichlorobenzene, 0.6 mg (7 μ mol) of N-methylglycine and 2~3 μ l (0.02 mmol) of formaldehyde (99% ¹³C, ~20% W/W in H₂O) (Cambridge Isotope Laboratory, Inc.) were added. The product was purified as indicated above. This led to the isolation of 0.7 mg (30%) of pure **2b** as a black solid.

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• HPLC trace of monoadduct 2a and unreacted $Sc_3N@C_{80}(1a)$.

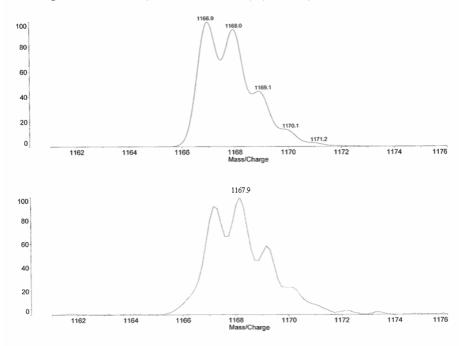


♦ HPLC trace of monoadduct **2b** and unreacted Er₃N@C₈₀(**1b**).



Retention times: 30.96 min (Er₃N@C₈₀); 32.18 min (2b)

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- ◆Comparison of computer simulated mass spectra with experimental mass spectra (Assuming A₃N₂C₈₃H₈ is equivalent to A₃N₂C₈₃H₇ with one carbon which is 100% ¹³C labeled)
- 1. Computer simulated mass spectrum of the molecule with formula of Sc₃N₂C₈₃H₈ (top) Experimental mass spectrum of **2a** (99% ¹³C labeled) (bottom)



2. Computer simulated mass spectrum of the molecule with formula of Er₃N₂C₈₃H₈ (top) Experimental mass spectrum of **2b** (99% ¹³C labeled) (bottom)

