

Supplementary Material (ESI) for Chemical Communications
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Supplementary Information for:

**Encapsulated Transition Metal Catalysts Comprising
Peripheral Zn(II)salen Building Blocks: Template-Controlled
Reactivity and Selectivity in Hydroformylation Catalysis**

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Crystal Structure Determination

X-ray intensities were measured on a Nonius KappaCCD diffractometer with a rotating anode (Mo-K α , $\lambda = 0.71073 \text{ \AA}$) at a temperature of 150 K. The structures were solved with direct methods with the program SHELXS-97^a and SIR-97^b and refined with the program SHELXL-97^c against F² of all reflections. The drawings, structure calculations, and checking for higher symmetry were performed with the program PLATON^d

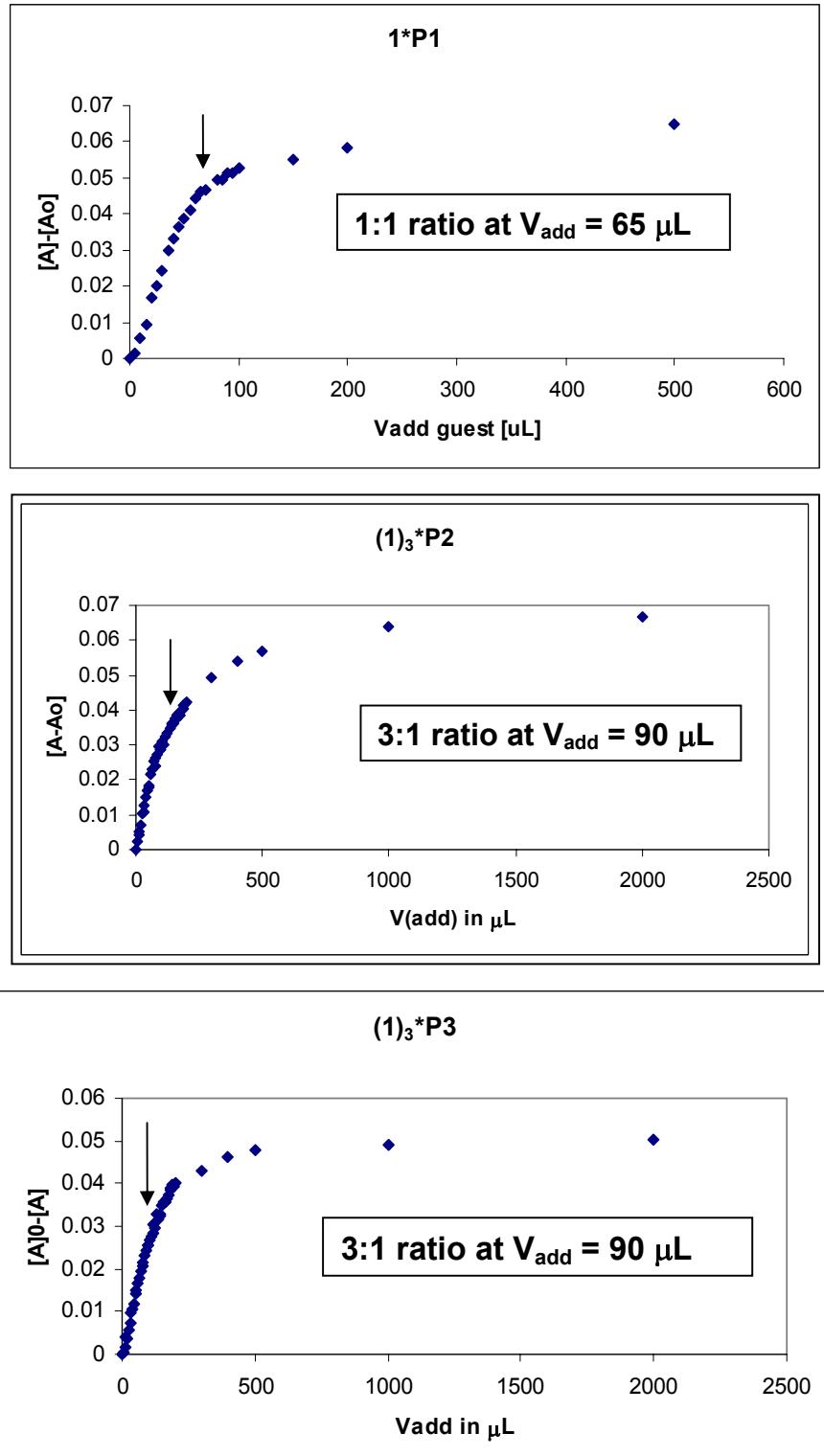
^aG. M. Sheldrick, SHELXS-97. Program for crystal structure solution, University of Göttingen, Germany, 1997

^bA. Altomare, M. C. Burla, M. Camalli, G. L. Cascarano, C. Giacovazzo, A. Guagliardi, A. G. G. Moliterni, G. Polidori, R. Spagna, *J. Appl. Crystallogr.*, 1999, **32**, 115

^cG. M. Sheldrick, SHELXL-97. Program for crystal structure refinement. University of Göttingen, Germany, 1997

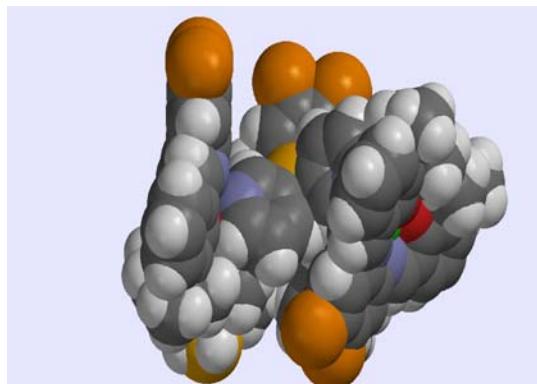
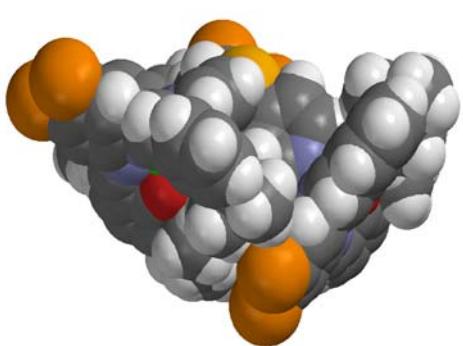
^dA. L. Spek, *J. Appl. Crystallogr.*, 2003, **36**, 7

UV-vis Titration curves for assemblies **1•P1**, **(1)₃•P2** and **(1)₃•P3**.



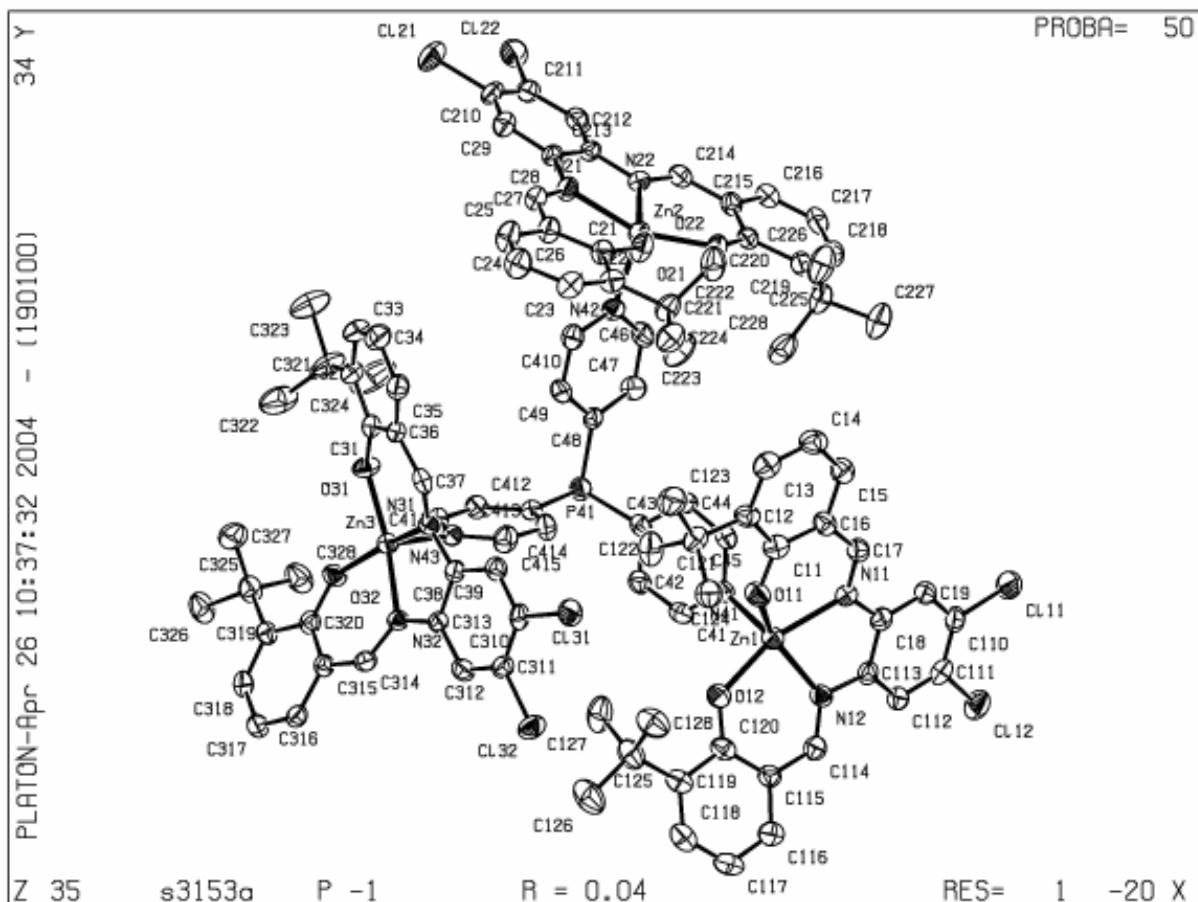
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Molecular modeling results (PM3 calculations) for assemblies **(1)₃•P2** (LEFT) and **(1)₃•P3** (RIGHT)



(Zn = red, Cl = orange, P = yellow, N = blue, C = dark grey, H = light grey)

Displacement ellipsoid plot for assembly **(1)₃•P2** with the adopted numbering scheme.



Selected bond distances (\AA) and angles ($^{\circ}$): $\text{Zn(1)-N(11)} = 2.097(3)$, $\text{Zn(1)-N(12)} = 2.076(3)$, $\text{Zn(1)-N(41)} = 2.138(3)$, $\text{Zn(1)-O(11)} = 1.946(3)$, $\text{Zn(1)-O(12)} = 1.964(2)$, $\text{Zn(2)-N(21)} = 2.085(3)$, $\text{Zn(2)-N(22)} = 2.060(3)$, $\text{Zn(1)-N(42)} = 2.132(3)$, $\text{Zn(2)-O(21)} = 1.953(3)$, $\text{Zn(2)-O(22)} = 1.949(2)$, $\text{Zn(3)-N(31)} = 2.118(3)$, $\text{Zn(3)-N(32)} = 2.069(3)$, $\text{Zn(3)-N(43)} = 2.124(3)$, $\text{Zn(3)-O(31)} = 1.948(2)$, $\text{Zn(3)-O(32)} = 1.975(2)$, $\text{C(43)-P(41)-C(48)} = 102.89(16)$, $\text{C(43)-P(41)-C(413)} = 98.44(16)$, $\text{C(48)-P(41)-C(413)} = 99.40(17)$.