

Unprecedented, Fully Recyclable, Solid-Supported Reagent for the Kinetic Resolution of Racemic Amines through Enantioselective *N*-Acetylation

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Alain Valleix

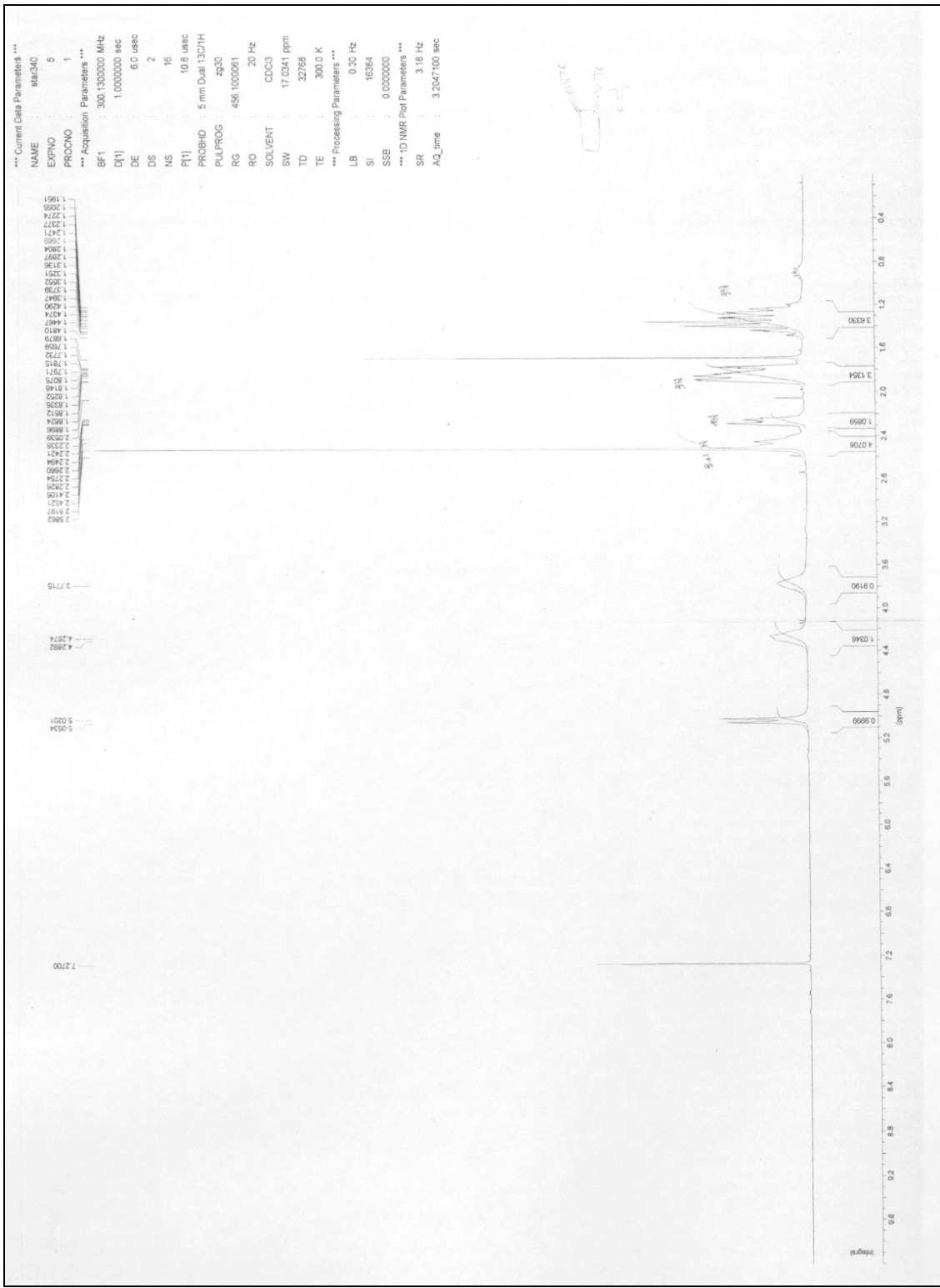
*CEA-Saclay, Service des Molécules Marquées, Bât. 547
Département de Biologie Cellulaire et Moléculaire, F-91191, Gif-sur-Yvette.*

Supporting Information

General Considerations. Glassware was oven-dried before use and cooled to room temperature under N₂. All solvents were commercially available and used without further purification unless noted otherwise. Yields and ee's in the tables refer to isolated yields (average of at least two runs) of compounds which are ≥95% pure as determined by ¹H NMR. All products were characterized by ¹H NMR, ¹³C NMR, and infrared (IR) spectroscopy. Analytical thin-layer chromatography (TLC) were performed using 2.5 cm x 5 cm plates coated with a 0.25 mm thickness of silica gel 60F-254 Merck, and visualization was accomplished with ultraviolet light or with ethanolic bromocresol green. Flash chromatography were performed using silica gel 60 Merck (40-60 mm). IR spectra were taken on a Perkin Elmer 1600 FT-IR spectrophotometer with samples prepared as KBr pellets (2-8 mg of product was pulverized with 100 mg of KBr and pressed into a pellet in standard fashion). ¹H and ¹³C NMR spectra were recorded on a Bruker WP-200 SY or on a Bruker DPX 300 spectrometer at ambient temperature. All ¹H NMR spectra (200 or 300 MHz) are reported as follows: chemical shift in parts per million, ppm, downfield from tetramethylsilane as an internal standard (□ scale), multiplicity (br=broad, s=singlet, d=doublet, t=triplet, q=quartet, and m=multiplet, integration, and coupling constant (Hz). All ¹³C NMR spectra (50 or 75 MHz) are determined with complete proton decoupling and reported in ppm relative to the central line of the triplet for CDCl₃ at 77 ppm. Analytical chiral HPLC were performed on Daicel Chiralcel OD column (250 mm x 4.6 mm). Mass spectra were

recorded on a Finnigan-Mat 4600 spectrometer. Melting points were obtained using a Reichert-Jung melting point apparatus.

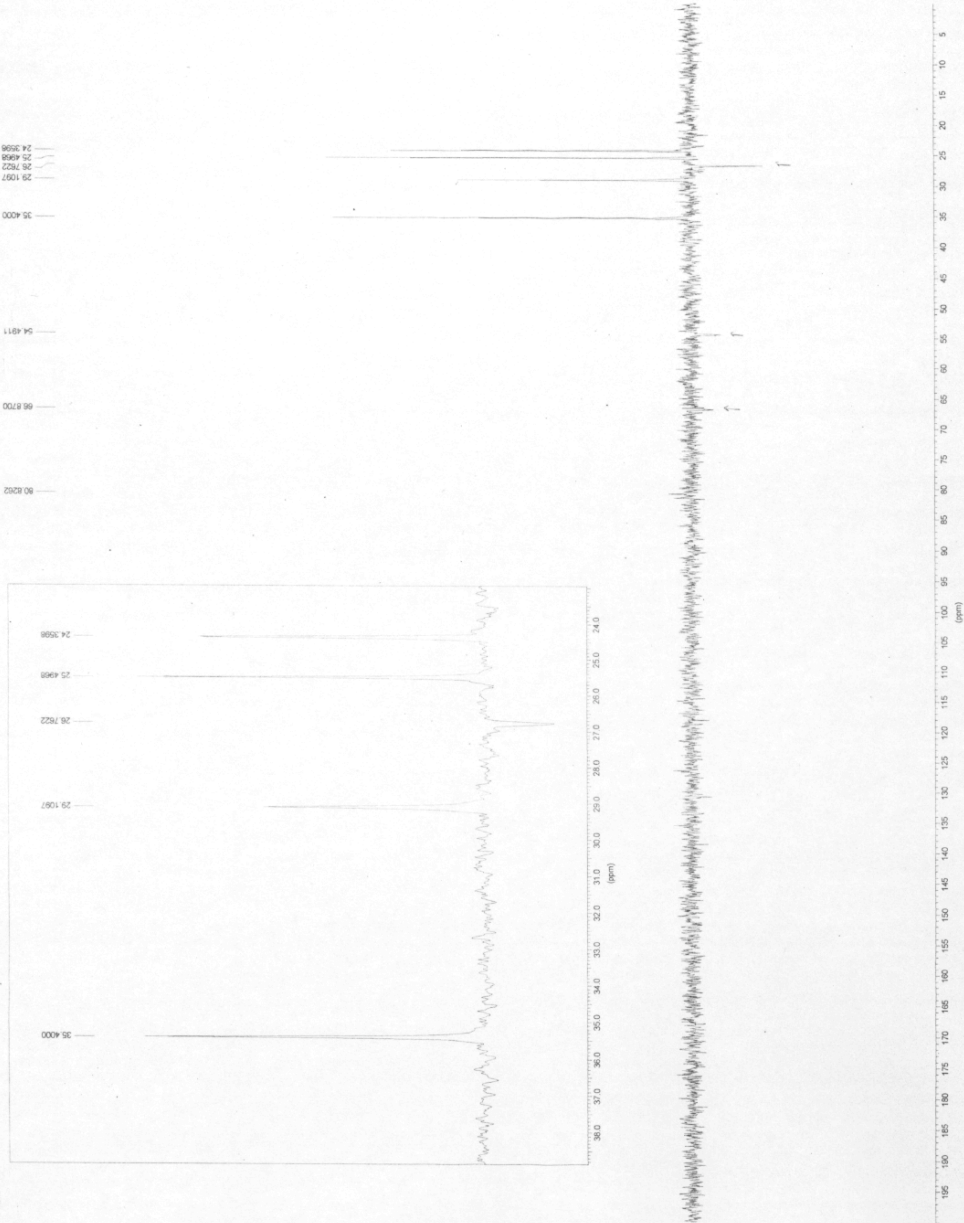
(1*S*,2*S*)-*N*-Acetyl-1,2-*bis*-trifluoromethanesulfonamidocyclohexane (1). Freshly distilled acetyl chloride (1.88 mL, 2.64 mmol) was added dropwise by syringe to a stirred solution of 1,2-*bis*-(trifluoromethanesulfonamido)-cyclohexane (Aldrich; 1.00 g, 2.64 mmol) and triethylamine (5.50 mL, 3.96 mmol) in Et₂O (20 mL) at -20 °C. The mixture was stirred 3h at 0 °C, time after which the solvent was removed under reduced pressure. The resulting residue was subjected to flash chromatography on silica gel (EtOAc/*n*-Hexane: 1/9) to afford the desired product as a white solid in 85% yield. mp = 111 °C; $[\alpha]_D^{20} = +17.46^\circ$ (c = 1.5; CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 1.10 to 1.60 (m, 3H), 1.70 to 2.00 (m, 3H), 2.05 to 2.35 (m, 1H), 2.35 to 2.60 (m, 1H), 2.52 (s, 3H), 3.60 to 4.05 (m, 1H), 4.10 to 4.60 (m, 1H), 5.04 (d, *J*=10 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 24.4, 25.5, 26.8, 29.1, 35.4, 54.5, 66.9, 119.4, 119.5, 170.0; FTIR (CsI) ν 3296, 3221, 2947, 2868, 1737, 1716, 1456, 1385, 1233, 1200, 1131, 1071, 1016, 971, 942, 918, 896, 737, 611, 532. MS (IC/NH₃, m/z) = 438 [M+NH₄]⁺.

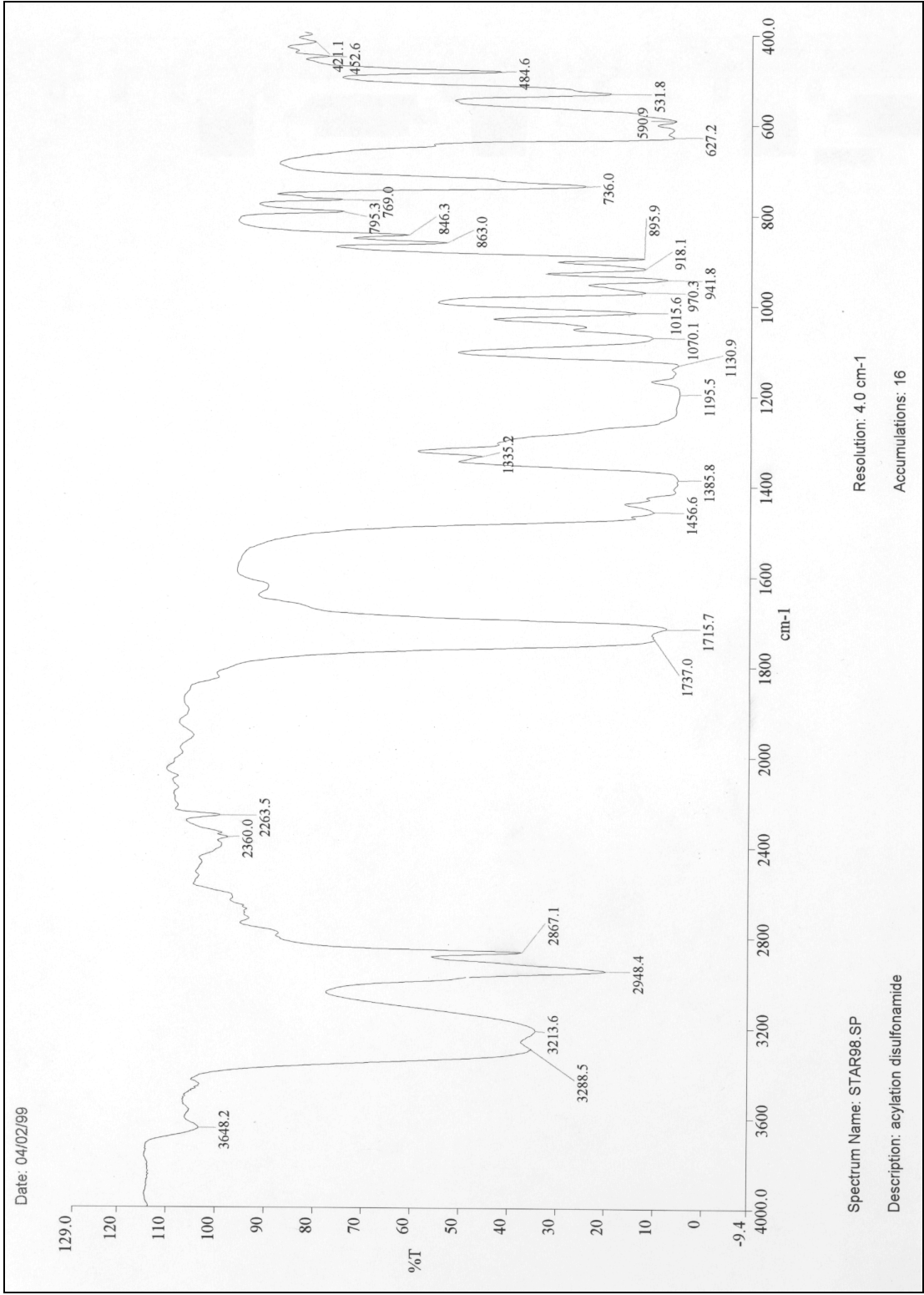




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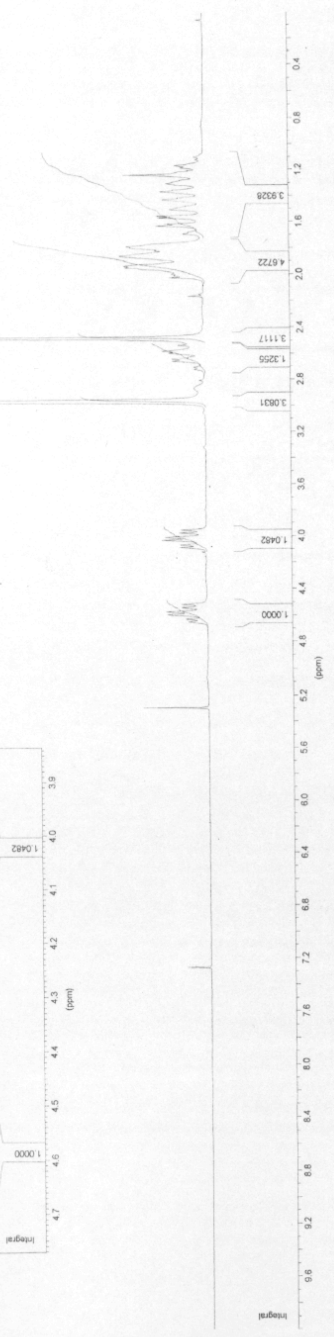
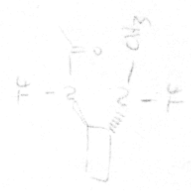
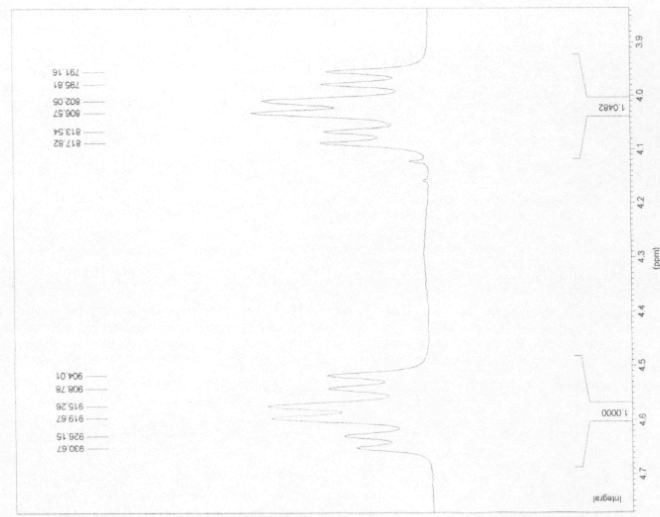


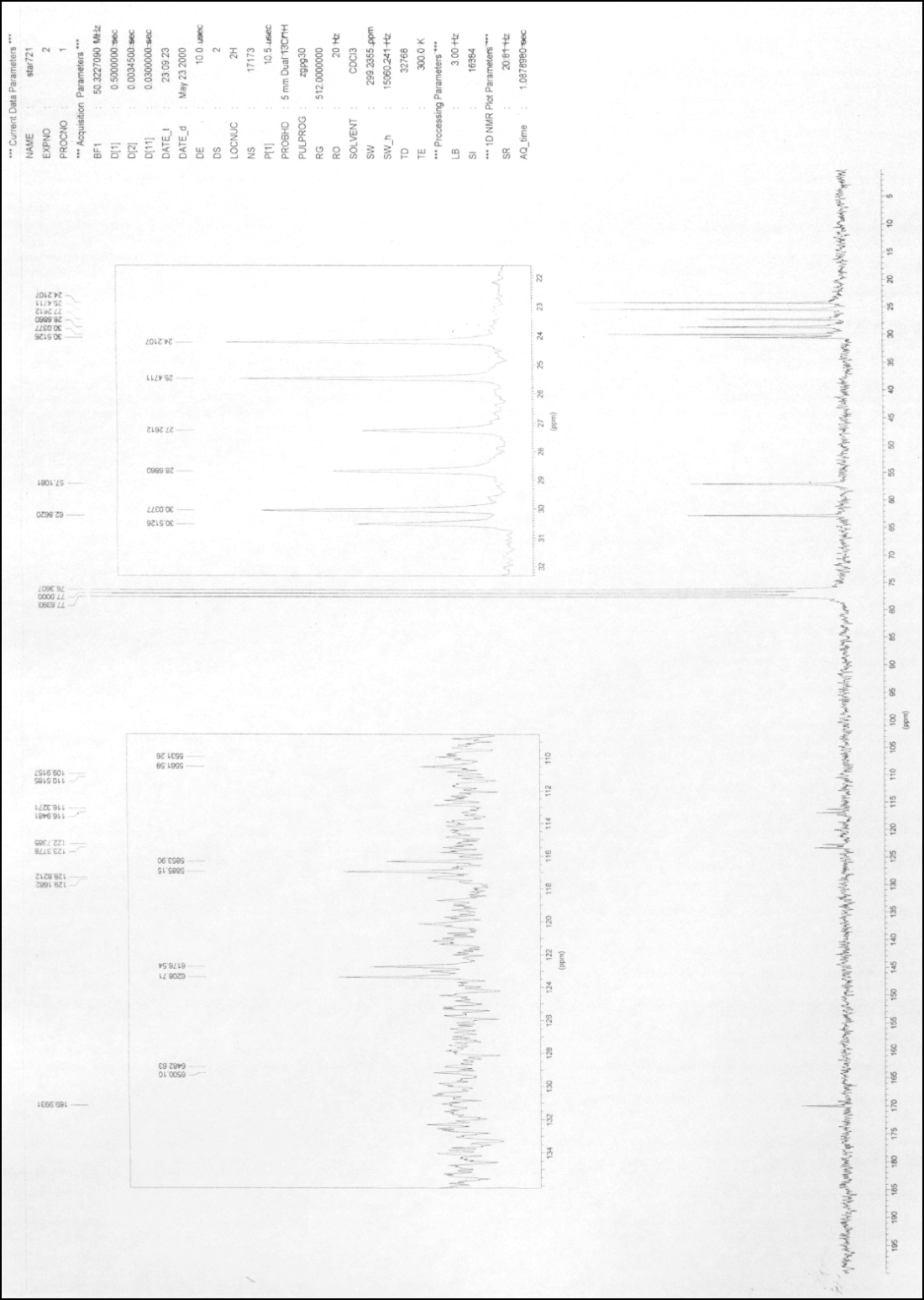


(1*S*,2*S*)-*N*-Acetyl-*N'*-methyl-1,2-*bis*-trifluoromethanesulfonamidocyclohexane (2). A solution of diazomethane (25 mmol) in Et₂O (20 mL) prepared from 1-methyl-3-nitro-1-nitrosoguanidine was added cautiously dropwise to a stirred solution of (1*S*,2*S*)-*N*-acetyl-1,2-*bis*-(trifluoromethanesulfonamido)-cyclohexane **(1*S*,2*S*)-1** (1.00 g, 2.38 mmol) in Et₂O (20 mL) at 0 °C. The resulting mixture was maintained under an inert atmosphere (N₂) and stirred at room temperature overnight. TLC analysis (EtOAc/*n*-Hexane: 2/8) showed complete conversion of the precursor. Evaporation of the solvent under reduced pressure, followed by flash chromatography on silica gel (EtOAc/*n*-Hexane: 5/95), afforded the desired *N*-methyl derivative as a colourless oil in quantitative yield. mp 38-40 °C; $[\alpha]_D^{20} = +6.22^\circ$ (c = 4.1; CHCl₃); ¹H NMR (200 MHz, CDCl₃) δ 1.00 to 1.65 (m, 3H), 1.68 to 2.05 (m, 4H), 2.48 (s, 3H), 2.40 to 2.75 (m, 1H), 2.96 (s, 3H), 4.02 (td, *J*=11.37 Hz and *J*=4.28 Hz, 1H), 5.58 (td, 1H, *J*=10.95 Hz et *J*=4.52 Hz, 1H); ¹³C RMN (50 MHz, CDCl₃) δ = 24.2, 25.5, 27.3, 28.7, 30.0, 30.5, 57.1, 62.8, 119.5, 120.2, 169.9 ppm. FTIR (CsI) ν = 3455, 2950, 2869, 1740, 1398, 1190, 1131, 1098, 1065, 981, 951, 933, 895, 774, 609, 533, 491; MS (IC/NH₃, *m/z*) = 452 [M+NH₄]⁺.

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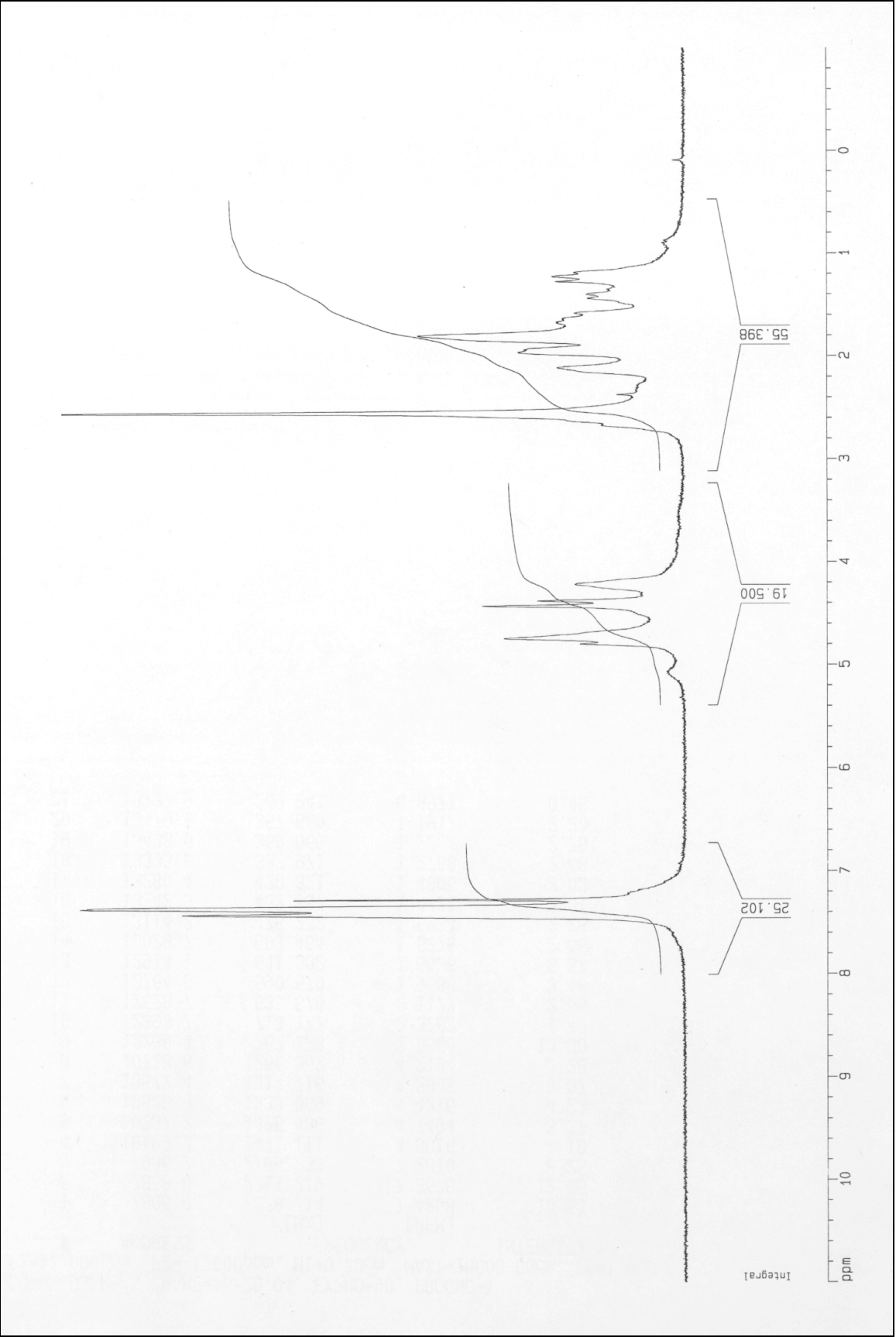
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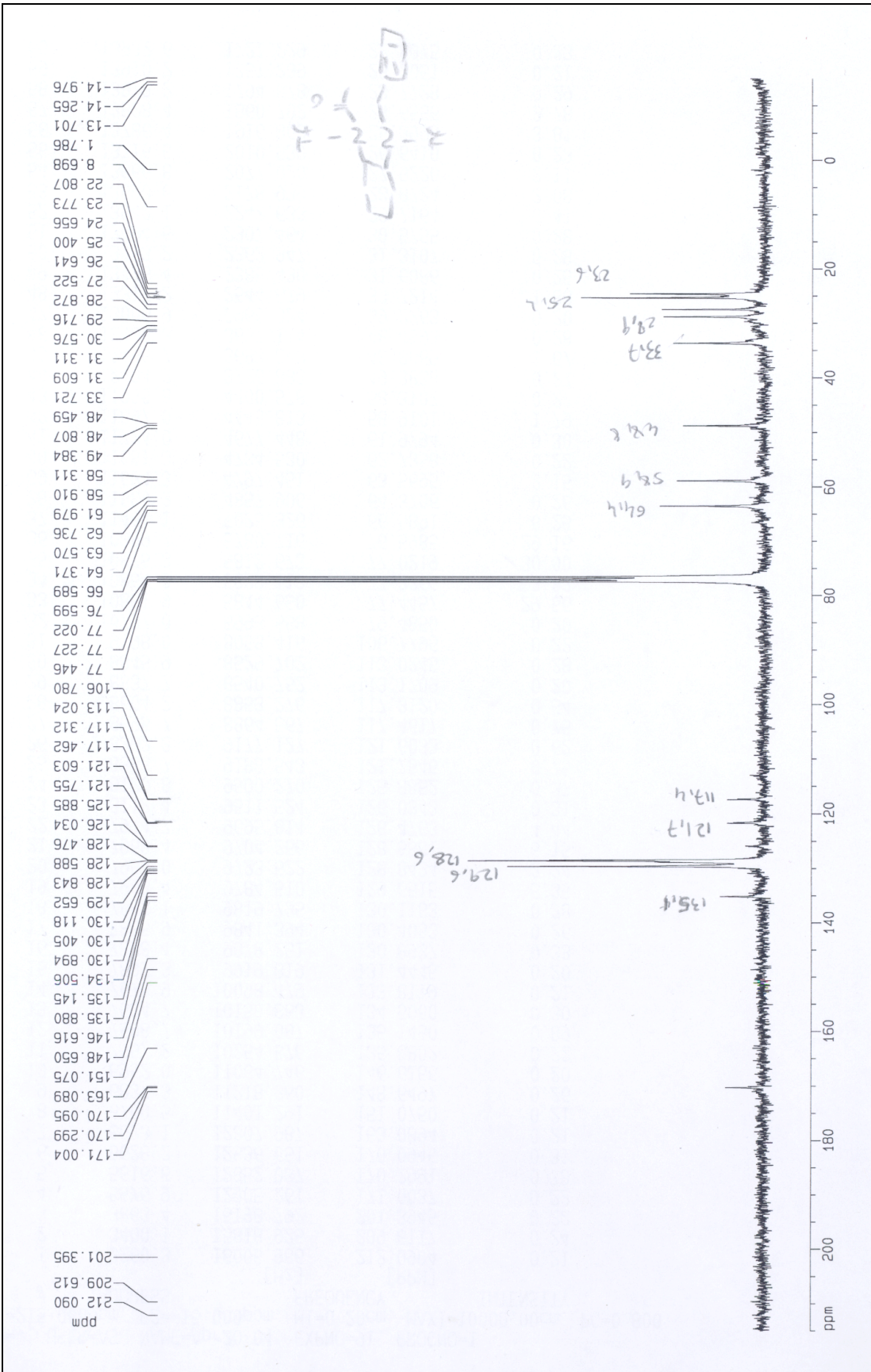






(1*S*,2*S*)-*N*-Acetyl-*N*'-benzyl-1,2-bis-trifluoromethanesulfonamidocyclohexane (3). A solution of **(1*S*,2*S*)-1** (650 mg, 1.55 mmol) in dry dichloromethane (10 mL) is added to a solution of benzyldiazonium (4.64 mmol) prepared according to the procedure developed by Reese,¹ in 5 mL of dichloromethane at room temperature. When the nitrogen evolution ceased, evaporation of the solvent under reduced pressure, followed by flash chromatography on silica gel (EtOAc/*n*-Hexane: 1/9), afforded the desired *N*-benzyl derivative as a colourless oil in 86 % yield. $[\alpha]_D^{20} = +10.43$ (c = 0.23; CHCl₃); ¹H NMR (300 MHz, CDCl₃) δ 1.00 to 2.30 (6m, 7H), 2.55 (s, 3H), 2.40 to 2.79 (m, 1H), 4.05 to 4.28 (m, 1H), 4.38 and 4.75 (2d, 2H, *J*=15 Hz), 4.55 to 4.75 (m, 1H), 7.18 to 7.60 (m, 5H); ¹³C RMN (75 MHz, CDCl₃) δ 23.6, 25.4, 27.5, 28.9, 33.7, 48.8, 58.9, 64.4, 119.6, 119.5, 128.6, 129.6, 135.1, 170.1. FTIR (CsI) ν 2945, 1738, 1398, 1204, 1128, 1027, 951, 898, 734, 607. MS (IC/NH₃, m/z) = 528 [M+NH₄]⁺.

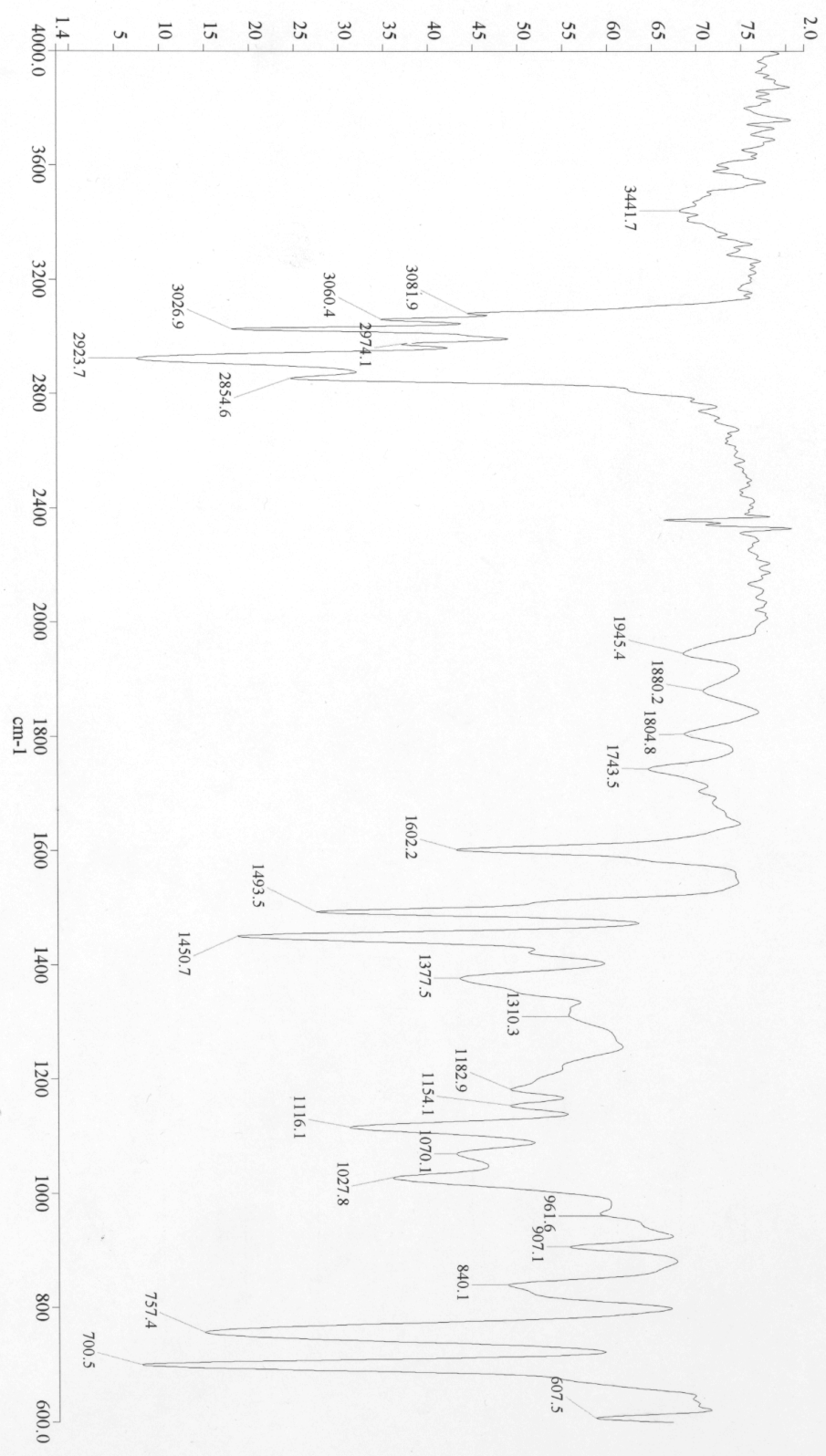




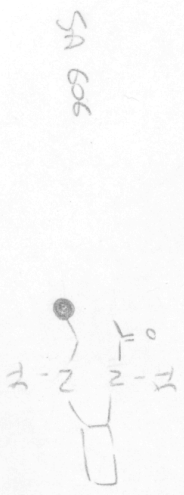
(1*S*,2*S*)-*N*-Acetyl-*N'*-polymer-supported-1,2-bis-trifluoromethanesulfonamidocyclohexane

(4). Polymer-supported reagent **4** was prepared in four steps starting Merrifield resin; each step being monitored using IR spectroscopy (recorded using a Perkin-Elmer 2000 FT-IR directly on the resin beads). The solid-phase synthesis was initiated by treating commercially available Merrifield resin (chloromethylated polystyrene, 1% cross-linked divinylbenzene, 1.58 mmol.g⁻¹) with 4-hydroxy-benzaldehyde and sodium hydroxyde in DMSO (90°C),² to afford the benzaldehyde resin **5** ($\nu_{\text{C=O}}=1695\text{ cm}^{-1}$) in excellent yield (>95% determined by elementary analysis; loading=1.39 mmol.g⁻¹). Resin **5** was then treated with 1,3,5-triisopropylbenzene-sulfonyl hydrazine at room temperature leading to the supported hydrazone **6** which was subsequently treated with potassium hydroxyde in a MeOH/THF mixture (90°C).³ The resulting supported benzyl diazonium salt **7** reacted at room temperature with (1*S*,2*S*)-*N*-acetyl-1,2-bis-trifluoromethanesulfonamidocyclohexane **(1*S*,2*S*)-1** in DMF leading to the desired supported chiral reagent **4** ($\nu_{\text{C=O}}=1743\text{ cm}^{-1}$). The loading of resin **4** was determined as 0.58 mmol.g⁻¹ by treating it with an excess of (±)-1-phenylethylamine and quantifying the recovered acetylated product. This value was also confirmed by fluorine elemental analysis. Polymer-supported reagent **4** was prepared in a 65% overall yield from readily available Merrifield resin.

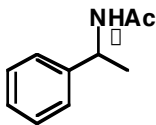
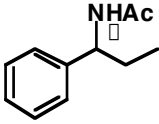
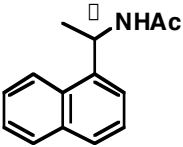
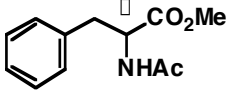
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General procedure for enantioselective acetylation of secondary alkylamines. Racemic amine (0.5 mmol) was added by syringe to an agitated solution of supported chiral acetylating agent (0.10 mmol) in the chosen solvent system at room temperature, and the mixture was agitated for 3 hr. The resin was then filtered, washed with dichloromethane, and the solvents were removed under reduced pressure. The resulting residue was then purified by flash chromatography on silica gel (EtOAc/*n*-Hexane: 1/1) and analyzed by HPLC using a chiral stationary phase.

Product	ee assay	Conditions	Retention Time of first isomer (min)	Retention Time of second isomer (min)
	Chiralcel OD	35°C 0.8 ml / min Pentane / EtOH 98/2	38.57 (<i>R</i>)	42.75 (<i>S</i>)
	Chiralcel OD	35°C 1 ml / min Hexane / EtOH 97/3	13.98	16.85
	Chiralcel OD	30°C 1.2 mL / min Hexane / EtOH 85/15	4.00	5.75
	Chiralcel OD	35°C 0.8 mL / min Hexane / EtOH 97/3	20.18	23.57

¹ Reese, C. B.; Dudman, C. C. *Synthesis* **1982**, 5, 419.

² Beebe, X.; Chiappari, C. L.; Olmstead, M. M.; Kurth, M. J.; Schore, N. E. *J. Org. Chem.*, **1995**, 60, 4204.

³ Bhalay, G.; Dunstan, A. R. *Tetrahedron Lett.*, **1998**, 39, 7803.