

Supporting Information

Synthesis of Huge Macrocycles using Two Calix[4]arenes as Templates

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General.

Solvents and other chemicals were purchased from commercial sources and were used without further purification. Column chromatography was performed on silica gel (Merck, 0.040-0.063 mm). ^1H NMR spectra were recorded on Bruker DRX400 Avance Instrument (400 MHz). Chemical shifts were calibrated to the residual signal of a deuterated solvent. ESI-mass spectra were obtained on a Q-TOF Ultima3 (from Micromass) instrument.

Multimacrocycle **6** ($\mathbf{m} = \mathbf{10}$, $\mathbf{Y} = \mathbf{C}_5\mathbf{H}_{11}$)

A suspension of **4** ($n = 6$, $\mathbf{Y} = \mathbf{C}_5\mathbf{H}_{11}$, 200 mg, 98.7 μmol) and **5** (230 mg, 148 μmol) in a mixture of benzene and dichloromethane (1:1 v/v, 400 mL) was stirred during 1.5 hours (55 °C, oil bath). After cooling, the mixture was purged with nitrogen during 30 minutes and a solution of the Grubbs' catalyst (bis(tricyclohexylphosphine)benzylidine ruthenium(IV) dichloride, 32.5 mg, 39.5 μmol) in dichloromethane (10 mL) was added in one portion. The mixture was stirred at ambient temperature during 6 days. Triethylamine (1 mL) was added and after 1 hour the mixture was evaporated. The hydrogenation was carried out with platinum (IV) oxide (85%, 40.3 mg) in THF (20 mL) under hydrogen atmosphere (12 h). Then the mixture was filtered and evaporated. The crude product was purified by the column chromatography on silica (THF / hexane = 1:2). After crystallisation from chloroform / methanol (5 : 50 mL), 160.9 mg (85%) of the desired product were obtained. m. p. > 290°C (partly decomposed); ^1H NMR (dmso-d_6) δ 7.38 (s, 4H, NH), 7.17 (s, 4H, NH), 6.89 (s, 8H, Calix-ArH), 6.51 (d, $^4J = 1.8$ Hz, 8H, ArH), 6.04 (br t, $^4J = 1.8$ Hz, 4H, ArH), 4.35 and 3.14 (d, $^2J=12.3$ Hz, 4H + 4H, ArCH₂Ar), 3.86 (t, $^3J = 6.5$ Hz, 16H, -OCH₂-), 3.82 (t, $^3J = 7.6$ Hz, 8H, -OCH₂-), 1.95 (t, $^3J = 7.3$ Hz, 8H, -OCH₂CH₂-), 1.64 (t, $^3J = 7.3$ Hz, 16H, -OCH₂CH₂-), 1.45-1.20 (two m, 64H, -

$\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2$), 0.95 (t, $^3J = 6.8$, 12H, - CH_3); ESI-MS m/z: calcd for $\text{C}_{116}\text{H}_{160}\text{N}_8\text{O}_{16}\text{Na}$ ($\text{M} + \text{Na}$) 1945.6, found 1944.9.

General procedure for the synthesis of macrocycles 7:

The compound **6** ($m = 8, 10, 14, 20$) (0.025 mmol) was refluxed in acetic acid (20 mL) for 24 hours. After cooling, acetic acid was removed in vacuo. The residue was separated and purified by column chromatography on silica gel ($\text{CH}_2\text{Cl}_2/\text{MeOH} = 20:1$) giving **7** as the first eluted compound and **8**.

7 ($m = 8$):

Yield: 50 %. m.p. $> 300^\circ\text{C}$, phase transition 136-140°C; ^1H NMR (dmso-d_6): $\delta = 9.79$ (s, 4H, -NH), 6.77 (br d, 8H, -ArH), 6.14 (br t, 4H, -ArH), 3.87 (t, $^3J = 6.5$ Hz, 16H, - OCH_2 -), 1.99 (s, 12H, -COCH₃), 1.66 (m, 16H, - OCH_2CH_2 -), 1.38-1.24 (m, 32H, - $\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2$); ESI-MS m/z: calcd for $\text{C}_{64}\text{H}_{92}\text{N}_4\text{O}_{12}\text{Na}$ ($\text{M} + \text{Na}$) 1132.5, found 1132.7.

7 ($m = 10$):

Yield: 66 %. m.p. $> 300^\circ\text{C}$, phase transition 130-135°C; ^1H NMR (dmso-d_6): $\delta = 9.78$ (s, 4H, -NH), 6.77 (d, $^4J = 2.0$ Hz, 8H, ArH), 6.14 (br t, 4H, -ArH), 3.87 (t, $^3J = 6.5$ Hz, 16H, - OCH_2 -), 1.99 (s, 12H, -COCH₃), 1.67-1.63 (m, 16H, - OCH_2CH_2 -), 1.36-1.23 (m, 48H, - $\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2$); ESI-MS m/z: calcd for $\text{C}_{72}\text{H}_{108}\text{N}_4\text{O}_{12}\text{Na}$ ($\text{M} + \text{Na}$) 1244.7, found 1243.6.

7 ($m = 14$):

Yield: 72 %. m.p. $> 300^\circ\text{C}$, phase transition 100-105°C; ^1H NMR (dmso-d_6): $\delta = 9.78$ (s, 4H, -NH), 6.77 (br d, 8H, -ArH), 6.12 (br t, 4H, -ArH), 3.86 (t, $^3J = 6.3$ Hz 16H, - OCH_2 -), 1.99 (s, 12H, -COCH₃), 1.65 (m, 16H, - OCH_2CH_2 -), 1.35-1.22 (m, 80H, - $\text{OCH}_2\text{CH}_2\text{CH}_2\text{CH}_2$); ESI-MS m/z: calcd for $\text{C}_{88}\text{H}_{140}\text{N}_4\text{O}_{12}\text{Na}$ ($\text{M} + \text{Na}$) 1469.1, found 1467.8.

7 ($m = 20$):

Yield: 67 %. m.p. $>300^\circ\text{C}$, phase transition 95-100°C; ^1H NMR (dmso-d₆, 80 °C): $\delta = 9.42$ (s, 4H, -NH), 6.78 (d, $^4J = 2.0$ Hz, 8H, -ArH), 6.14 (t, $^4J = 2.0$ Hz, 4H, -ArH), 3.91 (t, $^3J = 6.4$ Hz, 16H, -OCH₂-), 2.00 (s, 12H, -COCH₃), 1.70 (m, 16H, -OCH₂CH₂-), 1.40-1.26 (m, 128H, -OCH₂CH₂CH₂CH₂-); ESI-MS m/z: calcd for C₁₁₂H₁₈₈N₄O₁₂Na (M + Na) 1805.8, found 1805.4.

Multimacrocycle 10 (n = 1, Y = C₅H₁₁)

10 was synthesized from **9** (n = 1, 204mg, 0.1mmol) and tetratosyl urea calix[4]arene **5** (230 mg, 0.15mmol) in a way similar to **6** (see above). After purification of the crude product on the silica column (eluant: CH₂Cl₂/THF = 4:1) a white solid (120mg, 62%) was obtained.

m.p. $>300^\circ\text{C}$ (partly decomposed); ^1H NMR (THF-d₈) δ 7.57 (d, 8H, NH), 6.91(s, 8H, Calix-ArH), 6.62 (s, 8H, ArH), 6.03 (s, 4H, ArH), 4.47 and 3.10 (d, 4H + 4H, ArCH₂Ar), 3.97-3.89 (m, 16H + 8H, OCH₂-), 3.63 and 3.45 (m, 32H, -OCH₂CH₂O-), 2.01 (m, 8H, -OCH₂CH₂-), 1.58 (m, 16H, -OCH₂CH₂CH₂CH₂O-), 1.45-1.23(m, 16H, -OCH₂CH₂CH₂CH₂CH₃), 0.99 (t, 12H, -CH₃); ESI-MS m/z: calcd for C₁₀₈H₁₄₄N₈O₂₄Na (M + Na) 1961.4, found 1960.9.

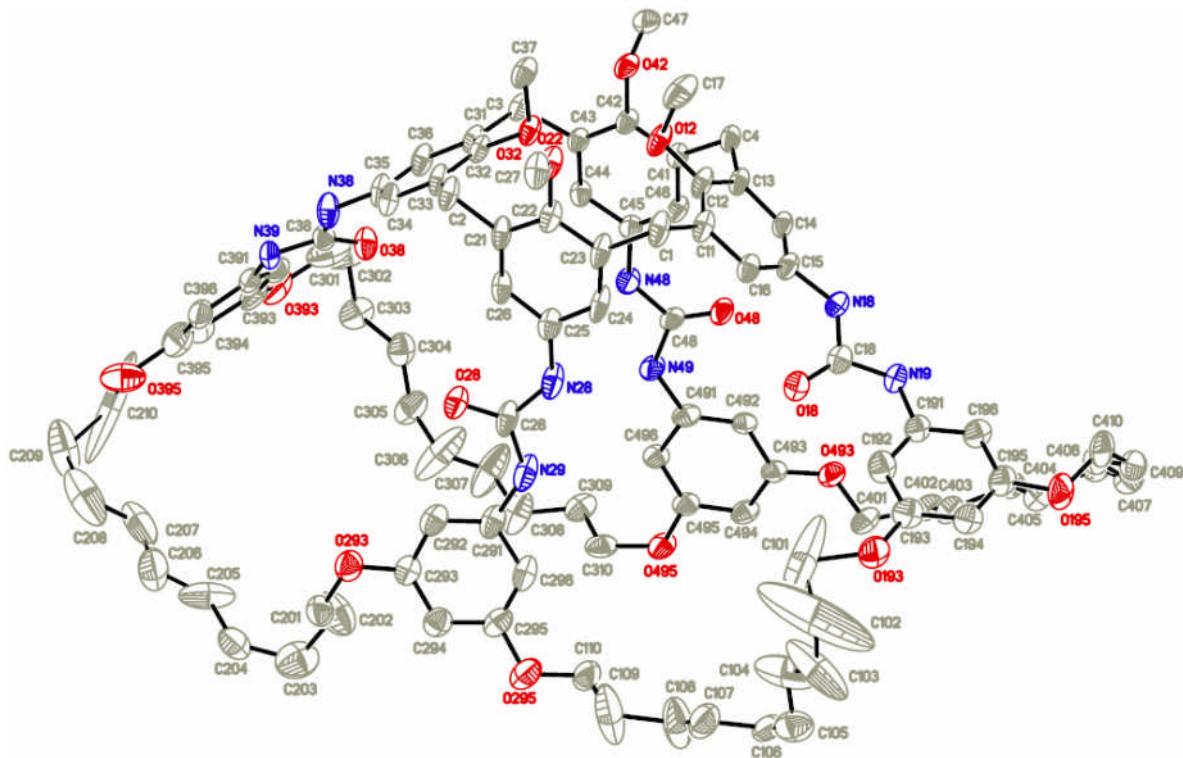
11 (n = 1): Macrocyclic **11** was prepared in a similar way to **7**.

Yield: 65 %. m.p. $>300^\circ\text{C}$, phase transition 65-70°C; ^1H NMR (dmso-d₆): $\delta = 9.80$ (s, 4H, -NH), 6.79 (br d, 8H, -ArH), 6.20 (br t, 4H, -ArH), 3.99 (br t, 16H, -OCH₂CH₂OCH₂-), 3.63 (br t, 16H, -OCH₂CH₂O-), 3.43(br t, 16H, -OCH₂CH₂O-), 1.99 (s, 12H, -COCH₃), 1.53 (m, 16H, -OCH₂CH₂CH₂-); ESI-MS m/z: calcd for C₆₄H₉₂N₄O₂₀Na (M + Na) 1260.4, found 1259.6.

Table S1. Crystal data and structure refinement for **6** (Y = CH₃, m = 10).

| | |
|-----------------------------------|--|
| Identification code | boeh33 |
| Empirical formula | C ₁₀₈ H ₁₆₃ N ₈ O _{25.50} |
| Formula weight | 1981.46 |
| Temperature | 173(2) K |
| Wavelength | 0.71073 Å |
| Crystal system | Triclinic |
| Space group | P -1 |
| Unit cell dimensions | a = 18.8106(15) Å α = 98.409(6)°. b = 23.1537(18) Å β = 104.443(6)°. c = 28.968(2) Å γ = 102.706(6)°. |
| Volume | 11645.5(15) Å ³ |
| Z | 4 |
| Density (calculated) | 1.130 Mg/m ³ |
| Absorption coefficient | 0.080 mm ⁻¹ |
| F(000) | 4284 |
| Crystal size | 0.42 x 0.40 x 0.36 mm ³ |
| Theta range for data collection | 3.56 to 25.03°. |
| Index ranges | -22<=h<=22, -27<=k<=27, -34<=l<=34 |
| Reflections collected | 93167 |
| Independent reflections | 38675 [R(int) = 0.1984] |
| Completeness to theta = 25.00° | 94.0 % |
| Absorption correction | None |
| Max. and min. transmission | 0.9718 and 0.9672 |
| Refinement method | Full-matrix least-squares on F ² |
| Data / restraints / parameters | 38675 / 42 / 2548 |
| Goodness-of-fit on F ² | 1.252 |
| Final R indices [I>2sigma(I)] | R1 = 0.1560, wR2 = 0.3756 |
| R indices (all data) | R1 = 0.2387, wR2 = 0.4450 |
| Largest diff. peak and hole | 1.379 and -0.590 e.Å ⁻³ |

a



b

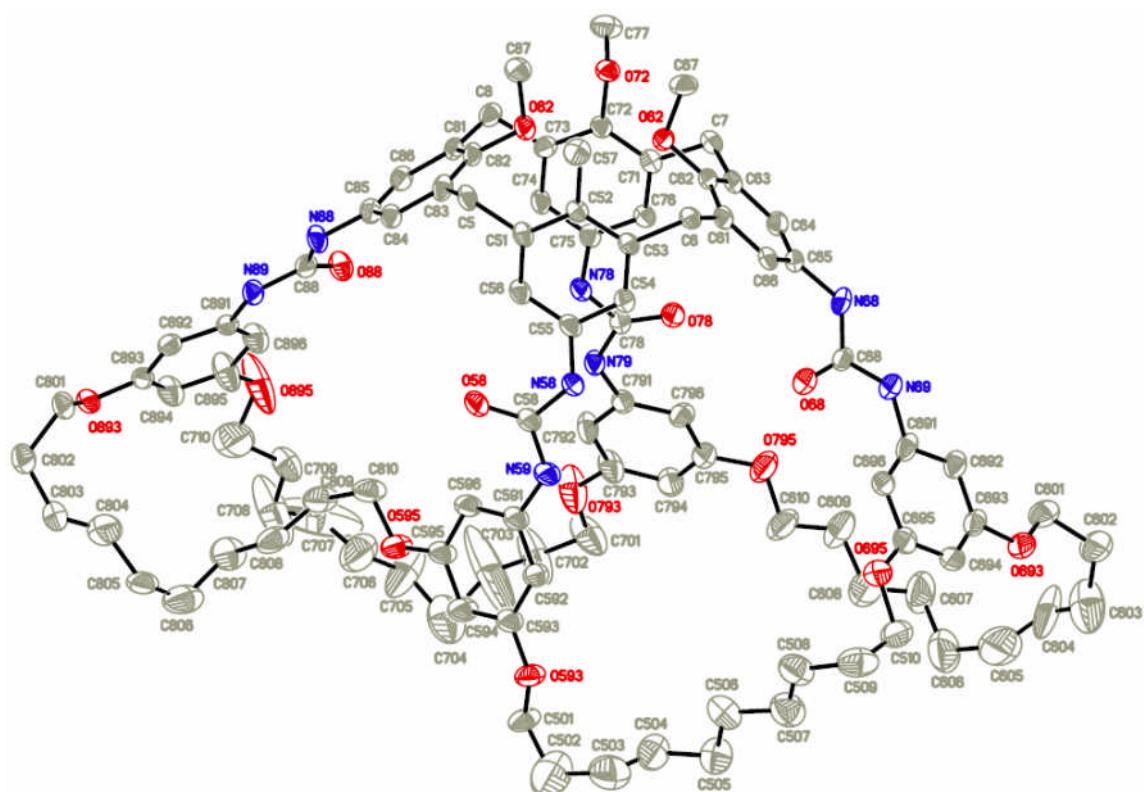


Figure S1. Single crystal X-ray structure (ORTEP-diagrams) of **6** ($m=10$, $Y=CH_3$). The two independent molecules (a and b), are shown.

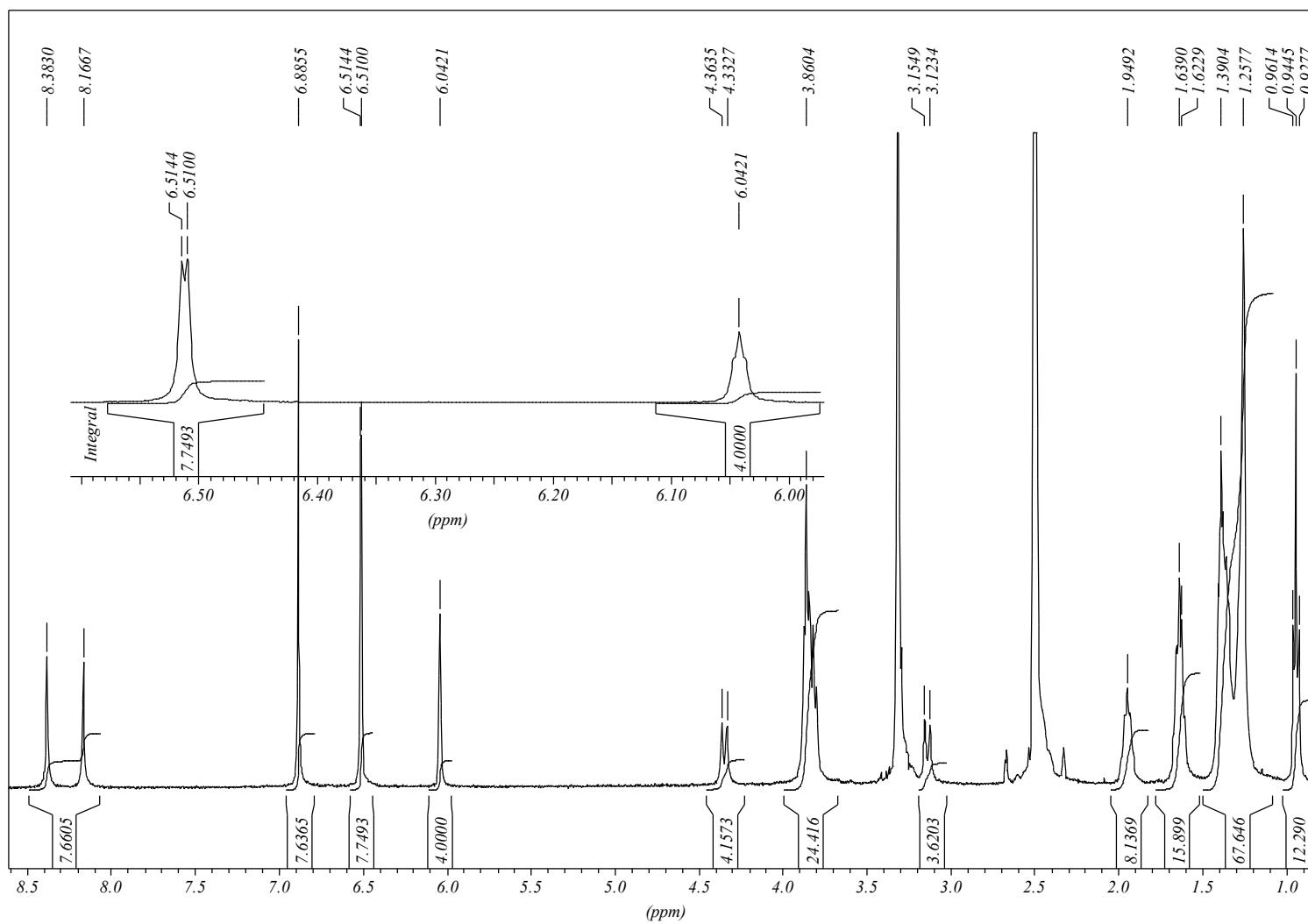


Figure S2. ^1H NMR spectrum of **6** ($m = 10$, $\text{Y} = \text{C}_5\text{H}_{11}$, 400 MHz, THF-d_8 , 25°C).

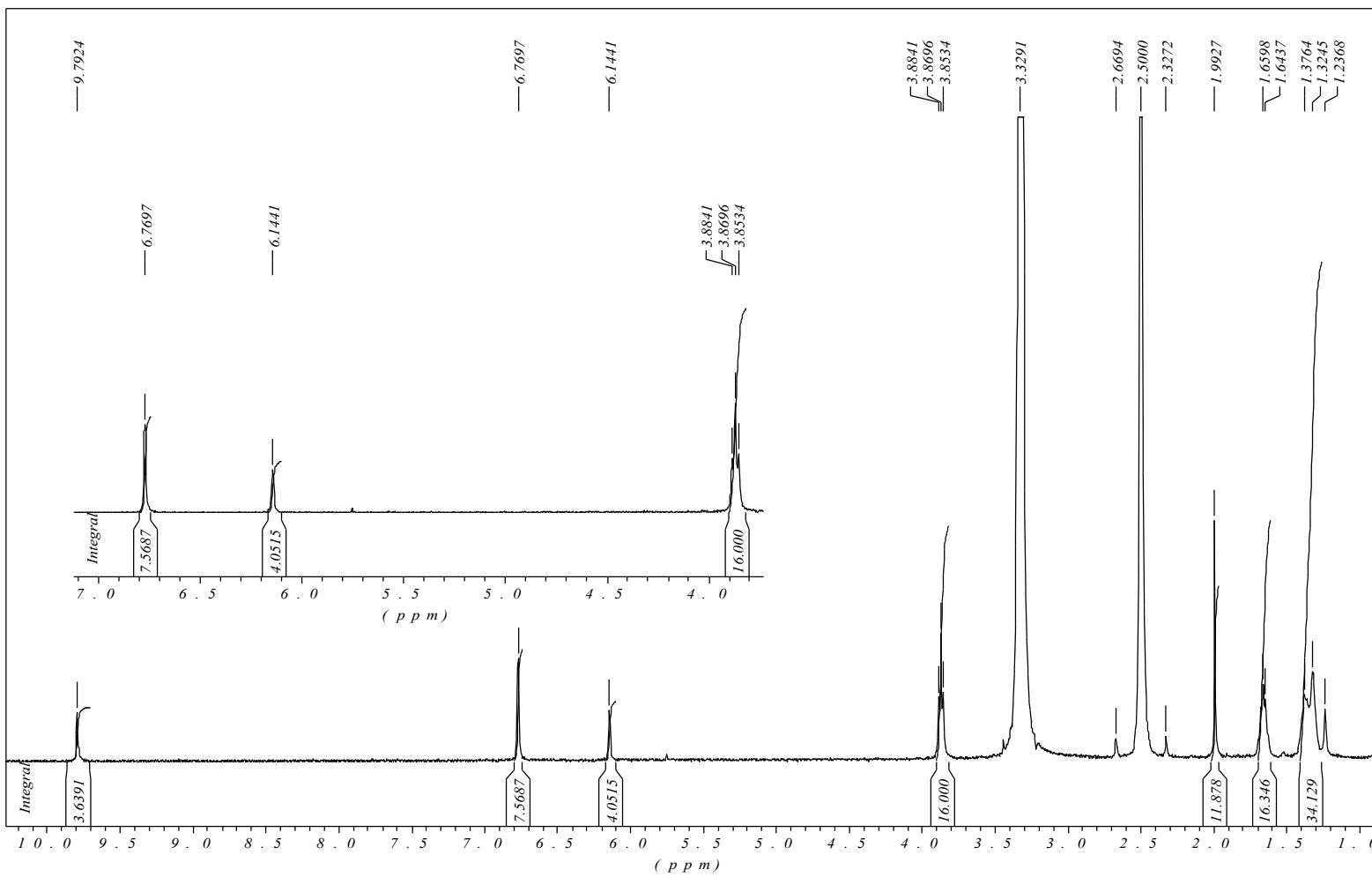


Figure S3. ¹H NMR spectrum of **7** ($m = 8$, 400 MHz, dmso-d₆, 25 °C).

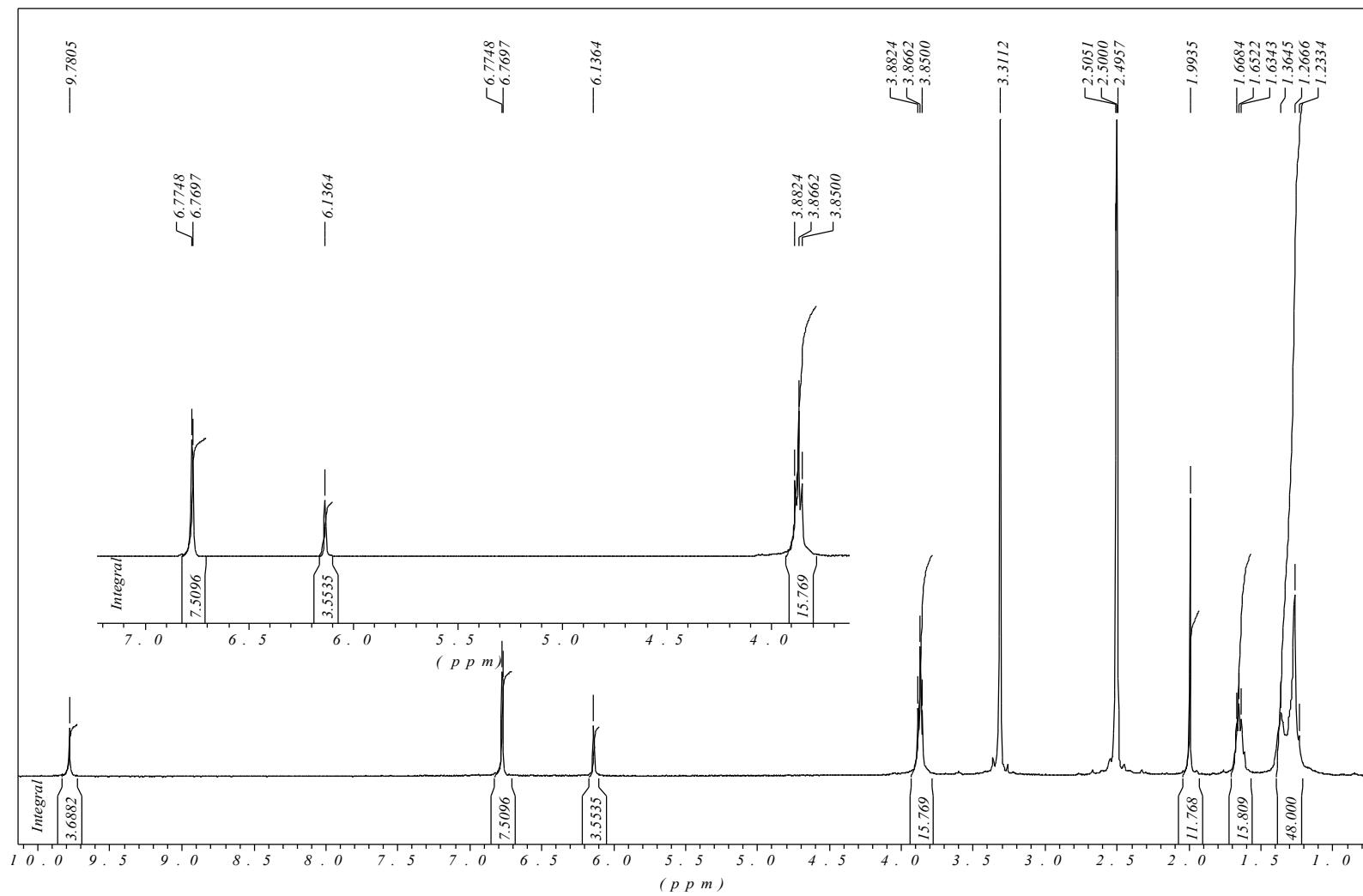


Figure S4. ^1H NMR spectrum of **7** ($m = 10$, 400 MHz, dmso- d_6 , 25 °C).

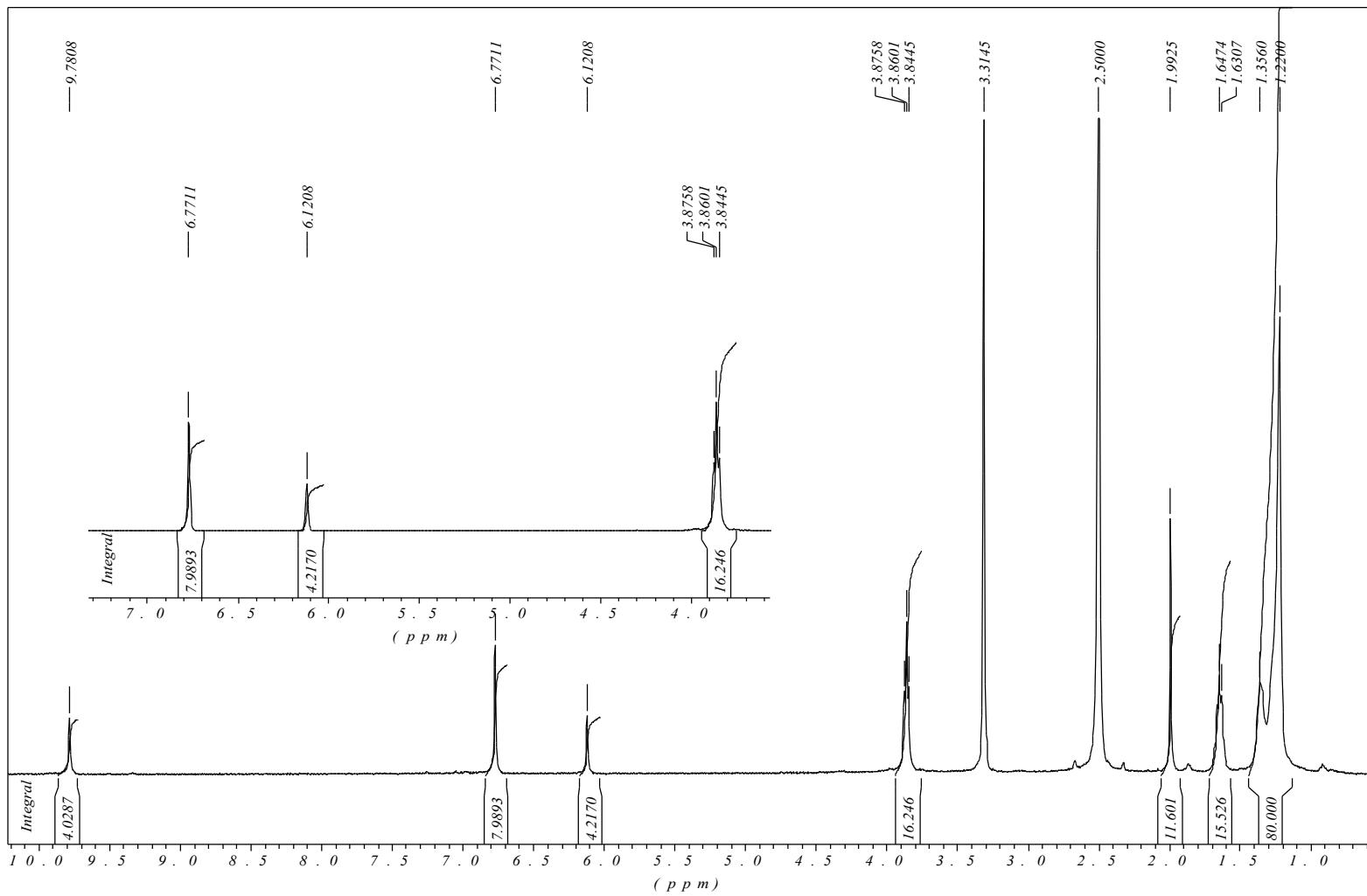


Figure S5. ^1H NMR spectrum of 7 ($m = 14$, 400 MHz, dmso-d_6 , 25 °C).

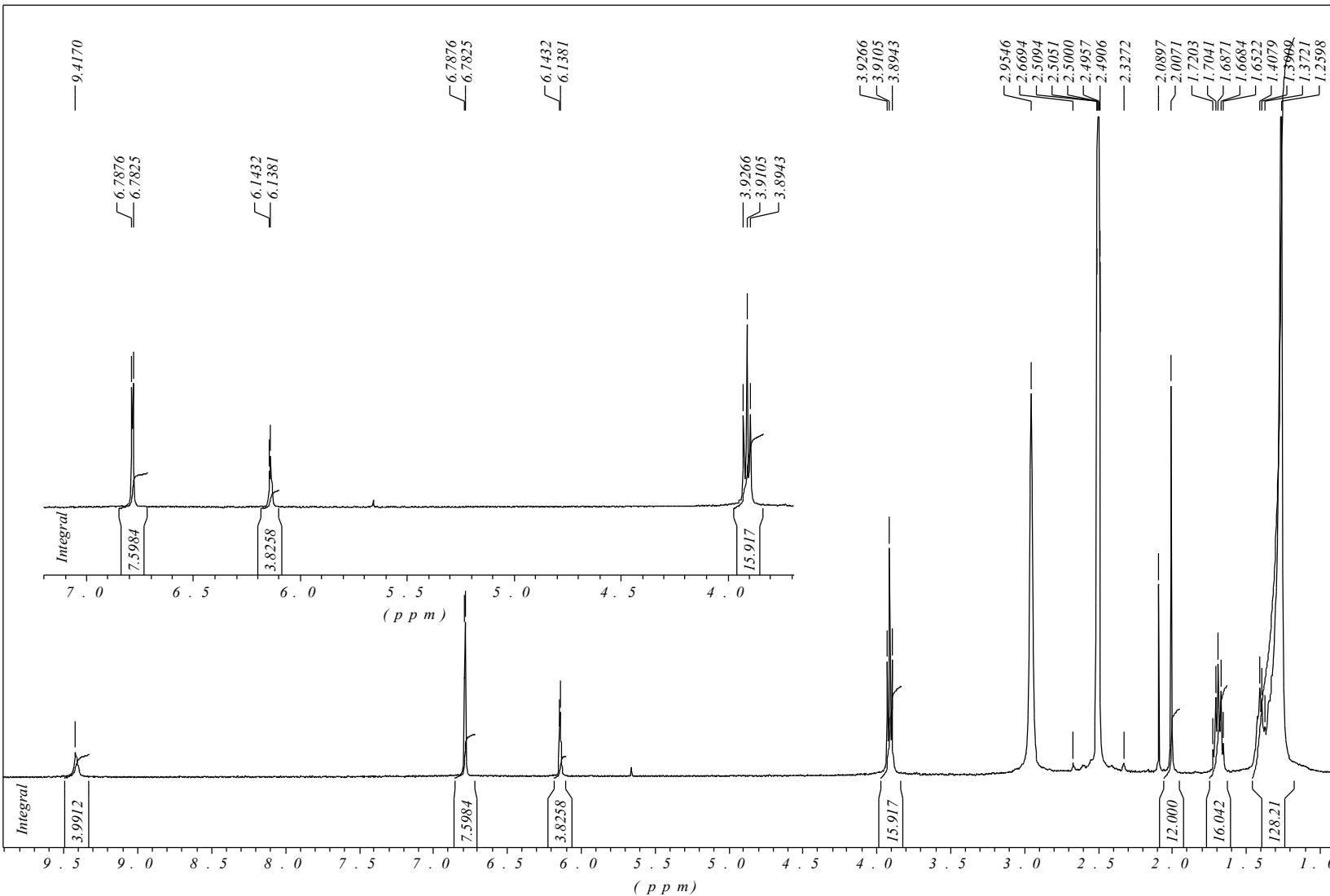


Figure S6. ^1H NMR spectrum of **7** ($m = 20$, 400 MHz, dmso-d_6 , 80 °C).

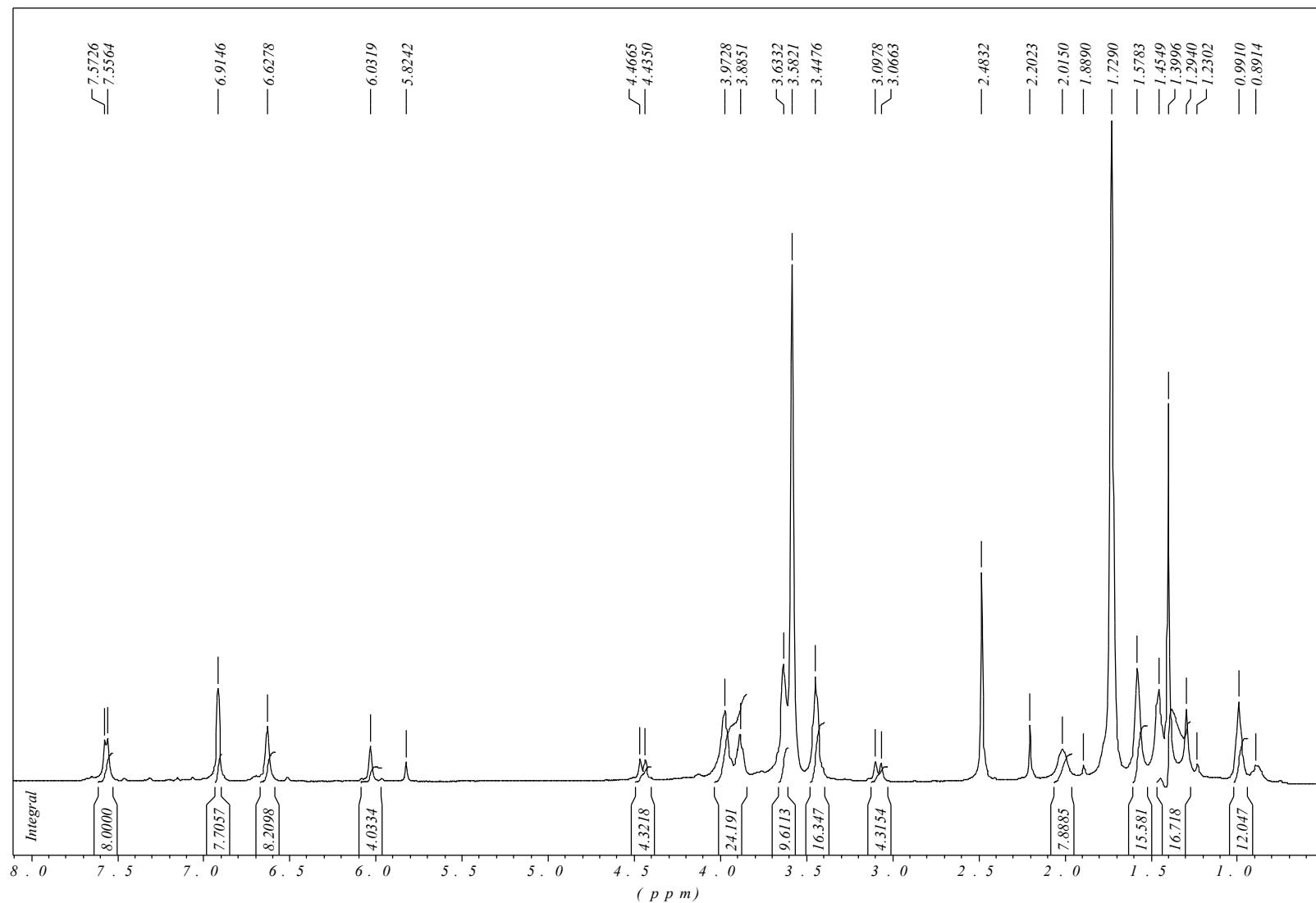


Figure S7. ^1H NMR spectrum of **10** ($n = 1$, 400 MHz, THF- d_8 , 25 °C).

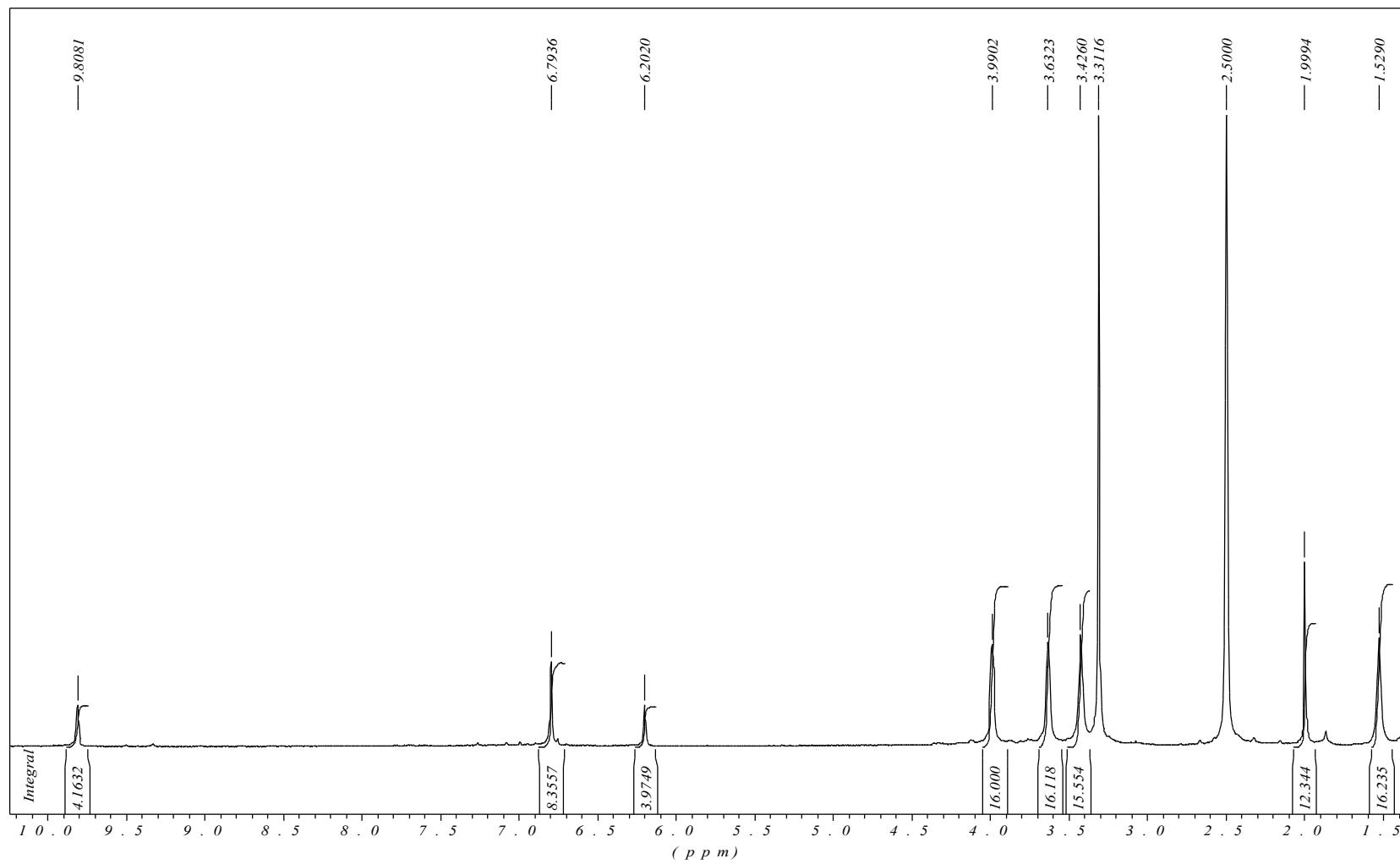


Figure S8. ^1H NMR spectrum of **11** ($n = 1$, 400 MHz, dmso- d_6 , 25 °C).