

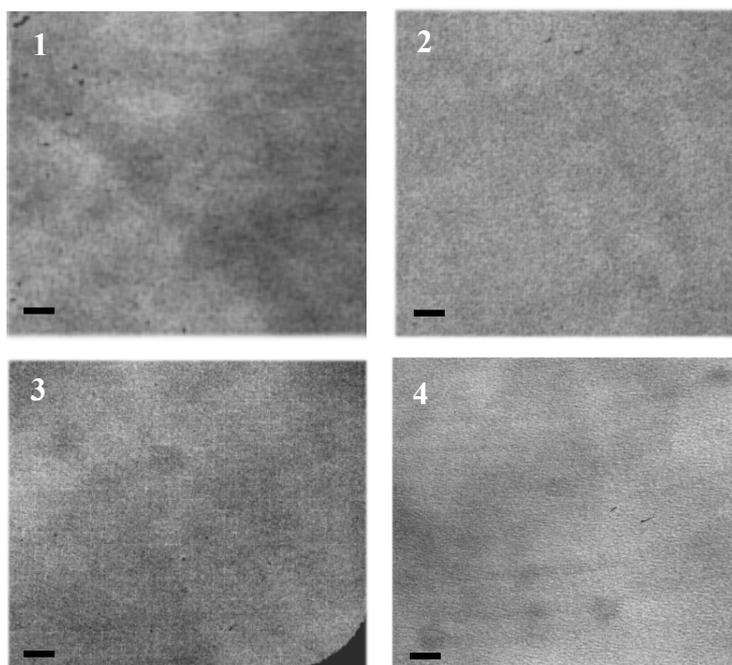
# Novel synthetic approach to PtCo alloy nanoparticles by reduction of nanometer-sized metal coordination polymers

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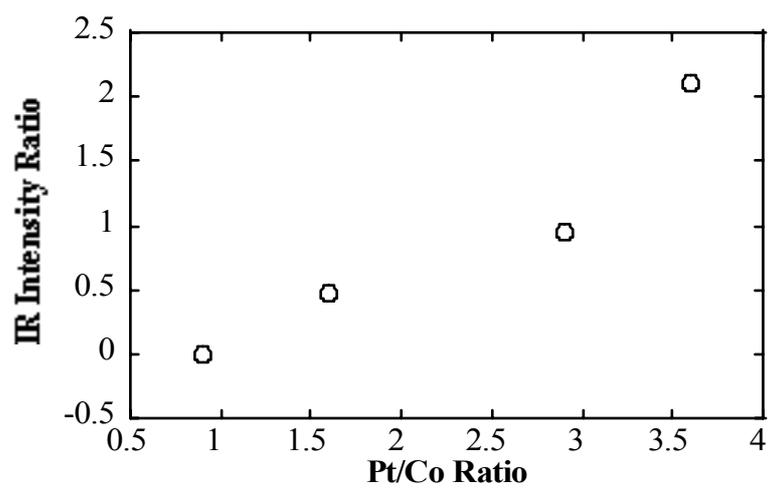
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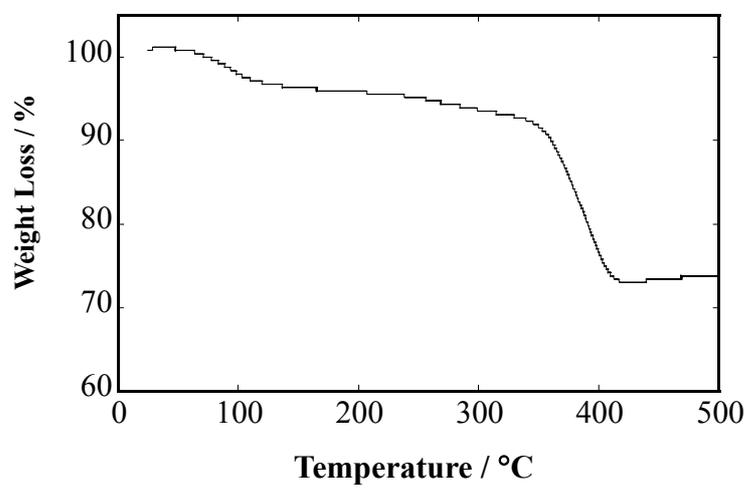
## Electronic Supplementary Information (ESI)



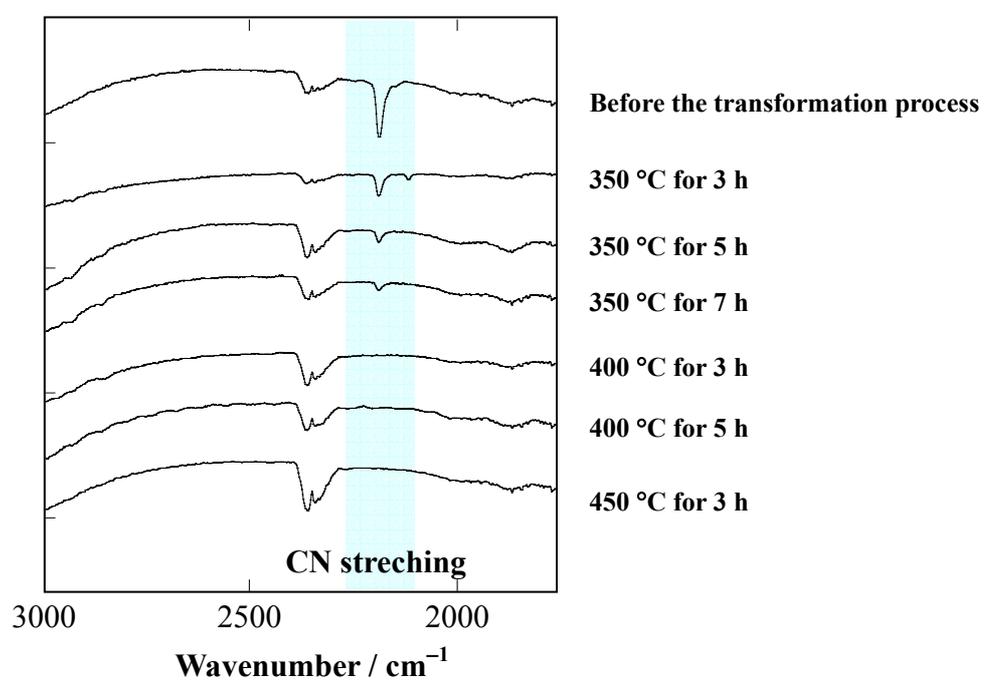
**Fig. S1** TEM images of the compounds 1-4. The numbers in the figure refer to those of the compound. Scale bar = 20 nm.



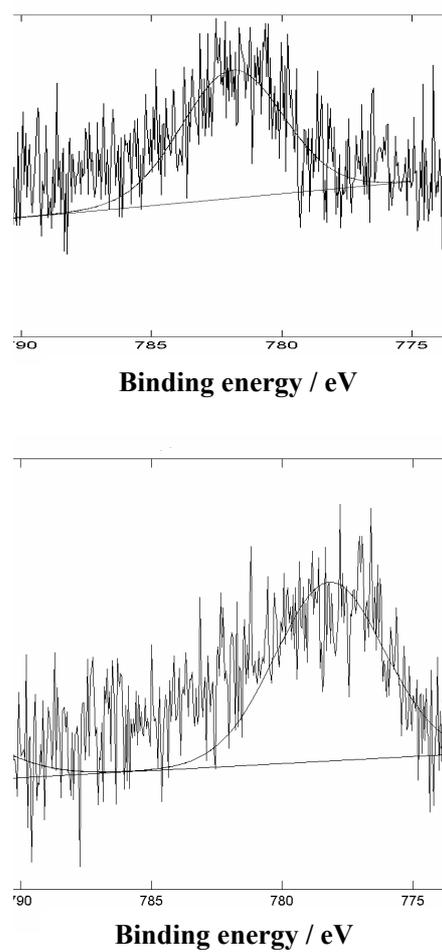
**Fig. S2** The relation between Pt/Co ratio and the IR intensity ratio of Pt<sup>II</sup>-CN-Pt<sup>IV</sup> to Pt<sup>II</sup>-CN-Co.



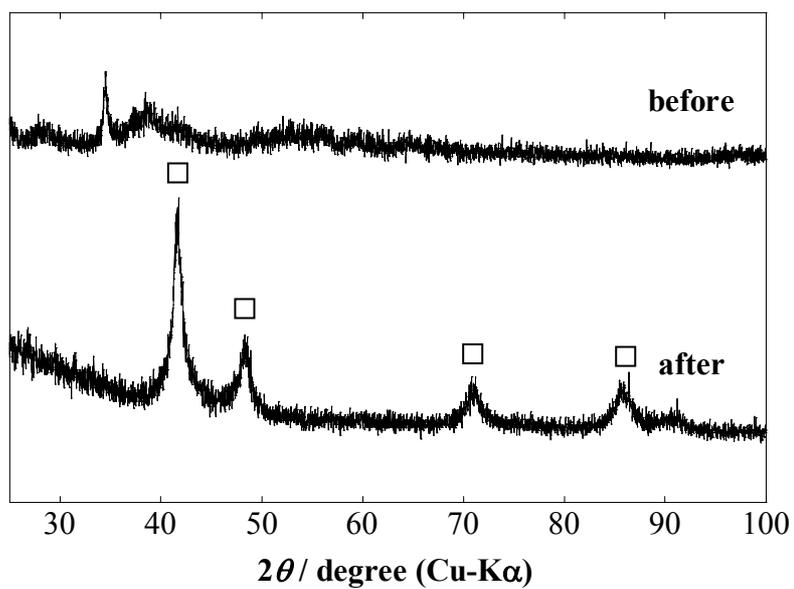
**Fig. S3** TGA curve of the compound **1** in the H<sub>2</sub> atmosphere (N<sub>2</sub>/H<sub>2</sub> = 10, total flow rate is 110 mL/min.).



**Fig. S4** IR spectra of the compound **1** before and after the transformation reaction in the  $\text{H}_2$  atmosphere. The reaction temperature and the reaction time are noted in the figure.



**Fig. 5** XPS curve in the Co<sub>2</sub>p energy range of compound **1** after the transformation reaction in H<sub>2</sub> atmosphere at 350 °C for 3h (top) and at 400 °C for 3h (bottom). The peak at 781.9 eV attributed to the Co sites in Pt<sup>II</sup>-CN-Pt<sup>IV</sup>/Co negatively shifts to the peak at 778.2 eV due to metal Co.



**Fig. S6** XRD patterns of the compound **1** before and after the transformation reaction in the  $\text{H}_2$  atmosphere at  $400\text{ }^\circ\text{C}$  for 3h. The peaks marked by squares are assigned to PtCo.