The Depth of Molecular Recognition: Voltage-Sensitive Blockage of Synthetic Multifunctional Pores with Refined Architecture

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Supporting Information

General. As in (S1), Supporting Information. ANTS and DPX were purchased from Molecular Probes, valinomycin, melittin, ADP, ATP, phytate and safranin O from Sigma or Fluka-Aldrich.

Abbreviations. ANTS: 8-Aminonaphthalene-1,3,6-trisulfonate; Arg, R: L-Arginine; Asn, N: L-Asparagine; BLM: Bilayer lipid membrane; DMF: N,N-Dimethylformamide; DPX: p-Xylenebis(pyridinium)bromide; EDC: 1-[3-(Dimethylamino)propyl]-3-ethylcarbodiimide, EYPC-LUVs: Egg yolk phosphatidylcholine large unilamellar vesicles; Fmoc: 9-Fluorenylmethoxycarbonyl; Gla: -OCH₂CO- (H-Gla-OH: glycolic acid); HATU: N-[(dimethylamino)-1*H*-1,2,3-triazolo[4,5-*b*]pyridin-1-ylmethylene]-*N*-methylmethanaminium N - [(1H - benzotriazol - 1 hexafluorophosphate *N*-oxide; HBTU: yl)(dimethylamino)methylene]-N-methylmethanaminium hexafluorophosphate N-oxide; His, L-Histidine; HOBt: 1-Hydroxybenzotriazole; Leu, L: L-Leucine; MES: H: 2-Morpholinoethanesulfonic acid monohydrate; Pmc: 2,2,5,7,8,-Pentamethylchromane-6sulfonyl; TEA: Triethylamine; TES: N-[Tris(hydroxymethyl)methyl]-2-aminoethanesulfonic acid; TFA: Trifluoroacetic acid; Trt: Trityl.

Fmoc-Asn(Trt)-Leu-NH₂, general procedure A. EDC•HCl (288 mg, 1.5 mmol), HOBt (200 mg, 1.3 mmol), and H-Leu-NH₂ (166 mg, 1.0 mmol) and TEA (840 ml, 6.0 mmol) were added to a solution of Fmoc-Asn(Trt)-OH (657 mg, 1.1 mmol) in CH₂Cl₂ (8 ml) at 0 °C. After stirring for 4 hours in the dark at rt, the reaction mixture was diluted with CH₂Cl₂, washed successively with saturated aqueous NaHCO₃, brine, 1 M aqueous KHSO₄, and brine, dried over anhydrous Na₂SO₄, and concentrated in vacuo. Purification of the crude product by column chromatography (CH₂Cl₂/MeOH 50:1) yielded pure Fmoc-Asn(Trt)-Leu-NH₂ (479 mg, 68%) as colorless solid. TLC (CH₂Cl₂/MeOH 20:1): R_f 0.33; $[\alpha]^{20}_{D}$ = -22.7 (c = 0.77, CH₂Cl₂/MeOH 1:1); mp: 203-204 °C; IR: υ 3286 (m), 2958 (w), 1689 (s), 1652 (s), 1530 (s), 1448 (s), 1299(m), 1235 (m), 1038 (m), 739 (s), 696 (s); ¹H NMR (500 MHz, CD₃OD): δ 7.70 (d, ³*J* (H,H) = 7.3 Hz, 2H), 7.64 (d, ³*J* (H,H) = 7.3 Hz, 2H), 7.39 (t, ³*J* (H,H) = 7.3 Hz, 2H), 7.29-7.17 (m, 17H), 4.46 (dd, ³*J* (H,H) = 7.7 Hz, ³*J* (H,H) = 6.4 Hz, 1H), 4.42 (d, ³*J* (H,H) = 6.8 Hz, 2H), 4.33 (dd, ³*J* (H,H) = 7.0 Hz, ³*J* (H,H) = 5.0 Hz, 1H), 4.23 (t, ³*J* (H,H) = 6.8 Hz, 1H), 2.89 (dd, ²*J* (H,H) = 15.4 Hz, ³*J* (H,H) = 6.4 Hz, 1H), 2.75 (dd, ²*J* (H,H) = 15.4 Hz, ³*J* (H,H) = 6.4 Hz, 1H), 2.75 (dd, ²*J* (H,H) = 15.4 Hz, ³*J* (H,H) = 6.2 Hz, 3H), 0.85 (d, ³*J* (H,H) = 7.7 Hz, 3H), 0.88 (d, ³*J* (H,H) = 6.2 Hz, 3H), 0.85 (d, ³*J* (H,H) = 7.7 Hz, 3H), 0.85 (d, ³*J* (H,H) = 7.7 Hz, 3H), 0.85 (d, ³*J* (H,H) = 7.7 Hz, 3H), 0.88 (d, ³*J* (H,H) = 6.2 Hz, 3H), 0.85 (d, ³*J* (H,H) = 7.7 Hz, 3H), 0.85 (d, ³*J* (H,H)

(H,H) = 6.0 Hz, 3H); ¹³C NMR (125 MHz, CD₃OD): δ 173.7 (s), 171.7 (s), 156.1 (s), 145.9 (s), 145.3 (s), 142.6 (s), 130.1 (d), 128.8 (d), 128.7 (d), 128.2 (d), 127.8 (d), 126.2 (d), 121.0 (d), 71.8 (s), 67.9 (t), 53.0 (d), 52.7 (d), 48.0 (d), 41.4 (t), 39.1 (t), 25.6 (d), 23.3 (q), 21.4 (q); ESI-MS (MeOH/CH₂Cl₂ 1:1): m/z 731 (100, [M + Na]⁺).

H-Asn(Trt)-Leu-NH₂, general procedure *B*. A solution of Fmoc-Asn(Trt)-Leu-NH₂ (1.76 g, 2.48 mmol) in 8 ml DMF containing 5% piperidine was stirred for 25 min at rt. Concentration in vacuo and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 20:1, then 7:1) yielded pure H-Asn(Trt)-Leu-NH₂ (1.19 g, 98%) as colorless powder. TLC (CH₂Cl₂/MeOH 10:1): R_f 0.30; $[\alpha]^{20}_{D}$ = -30.3 (c=1.01, CH₂Cl₂); mp = 102-104 °C; IR: v 3404 (m), 3350 (m), 2961 (m), 1683 (s), 1594 (w), 1492 (s), 1204 (w), 1018 (m), 810 (m); ¹H NMR (500 MHz, CD₃OD): δ 7.22-7.19 (m, 15H), 4.33 (dd, ³*J* (H,H) = 10.0 Hz, ³*J* (H,H) = 4.9 Hz, 1H), 3.63 (dd, ³*J* (H,H) = 7.3 Hz, ³*J* (H,H) = 5.8 Hz, 1H), 2.78 (dd, ²*J* (H,H) = 15.7 Hz, ³*J* (H,H) = 5.8 Hz, 1H), 2.63 (dd, ²*J* (H,H) = 15.7 Hz, ³*J* (H,H) = 7.3 Hz, 1H), 1.61-1.45 (m, 3H), 0.94 (d, ³*J* (H,H) = 6.6 Hz, 3H), 0.90 (d, ³*J* (H,H) = 6.3 Hz, 3H); ¹³C NMR (125 MHz, CD₃OD): δ 177.6 (s), 176.6 (s), 172.7 (s), 146.0 (s), 130.0 (d), 128.7 (d), 127.9 (d), 71.6 (s), 53.1 (d), 52.8 (s), 42.1 (t), 42.0 (t), 25.9 (d), 23.5 (q), 21.8 (q); ESI-MS (CH₂Cl₂): m/z 487 (100, [M + H]⁺).

Z-Leu-Asn(Trt)-Leu-NH₂. Coupling of H-Asn(Trt)-Leu-NH₂ (582 mg, 1.20 mmol) with Z-Leu-OH (433 mg, 1.63 mmol) following the *general procedure A* and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 30:1) yielded pure Z-Leu-Asn(Trt)-Leu-NH₂ (800 mg, 89%) as colorless powder. TLC (CH₂Cl₂/MeOH 15:1): R_f 0.31; $[\alpha]^{20}_{\ D} = -42.9$ (c = 0.99, CH₂Cl₂/MeOH 1:1); mp > 230 °C; IR: υ 3289 (m), 2957 (m), 1658 (s), 1492 (s), 1448 (w), 1234 (m), 1038 (m), 740 (m), 697 (m); ¹H NMR (500 MHz, CD₃OD): δ 7.27-7.16 (m, 20H), 5.05 (d, ²*J* (H,H) = 12.4 Hz, 1H), 4.98 (d, ²*J* (H,H) = 12.4 Hz, 1H), 4.57 (dd, ³*J* (H,H) = 7.6 Hz, ³*J* (H,H) = 5.8 Hz, 1H), 4.25 (dd, ³*J* (H,H) = 10.6 Hz, ³*J* (H,H) = 4.0 Hz, 1H), 4.07 (dd, ³*J* (H,H) = 7.6 Hz, ³*J* (H,H) = 15.8 Hz, ³*J* (H,H) = 5.8 Hz, 1H), 1.59-1.48 (m, 6H), 0.89-0.80 (m, 12H); ¹³C NMR (125 MHz, CD₃OD): δ 176.9 (s), 174.7 (s), 172.2 (s), 171.1 (s), 157.9 (s), 145.0 (s), 137.0 (s), 129.4 (d), 129.0 (d), 128.6 (d), 128.3 (d), 128.2 (d), 127.4 (d), 71.2 (s), 67.5 (t), 54.6 (d), 52.6 (d), 51.0 (d), 41.5 (t), 40.6 (t), 37.9 (t), 25.3 (d), 23.5 (q), 23.2 (q), 21.8 (q), 21.3 (q); ESI-MS (MeOH/CH₂Cl₂ 1/1): m/z 756 (100, [M + Na]⁺).

H-Leu-Asn(Trt)-Leu-NH₂, general procedure *C*. Pd(OH)₂/C (25 mg) was added to a solution of Z-Leu-Asn(Trt)-Leu-NH₂ (800 mg, 1.09 mmol) in CH₂Cl₂/MeOH 1:1 (16 ml). The suspension was degassed and set under a H₂ atmosphere. After stirring for 4 h, the reaction mixture was filtered through celite and aluminum oxide (basic), and the filtrate was concentrated *in vacuo*. Column chromatography (CH₂Cl₂/MeOH 30:1, then 7:1) yielded pure H-Leu-Asn(Trt)-Leu-NH₂ (442.0 g, 68%) as colorless powder. TLC (CH₂Cl₂/MeOH 10:1): R_f 0.26; [α]²⁰_D = -30.0 (c=0.7, CH₂Cl₂/MeOH 1:1); mp > 230 °C; IR: υ 3268 (m), 2957 (m), 1655 (s), 1636 (s), 1528 (s), 1444 (m), 1303 (m), 901 (w), 751 (w), 696 (s); ¹H NMR (500 MHz, CD₃OD): δ 7.28-7.19 (m, 15H), 4.67 (dd, ³J (H,H) = 7.8 Hz, ³J (H,H) = 6.3 Hz, 1H), 4.30 (dd, ³J (H,H) = 10.9 Hz, ³J (H,H) = 4.0 Hz, 1H), 3.33 (dd, 1H), 2.97 (dd, ²J (H,H) = 15.8 Hz, ³J (H,H) = 6.3 Hz, 1H), 1.71-1.47 (m, 4H), 1.37-1.30 (m, 2H), 0.94-0.86 (m, 12H); ¹³C NMR (125 MHz, CD₃OD): δ 177.9 (s), 145.9 (s), 130.1 (d), 128.7 (d), 127.8 (d), 71.7 (s), 54.3 (d), 52.9

(d), 51.2 (d), 45.4 (t), 41.5 (t), 38.9 (t), 25.8 (d), 25.7 (d), 23.7 (q), 23.4 (q), 22.5 (q), 21.5 (q); ESI-MS (CH₂Cl₂): m/z 1222 (100, $[2M + Na]^+$), 622 (40, $[M + Na]^+$), 600 (25, $[M + H]^+$).

Fmoc-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂. Coupling of H-Leu-Asn(Trt)-Leu-NH₂ (516 mg, 0.86 mmol) with Fmoc-Asn(Trt)-OH (693 mg, 1.16 mmol) following the general procedure A and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 40:1) yielded pure Fmoc-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (801 mg, 79%) as colorless powder. TLC (CH₂Cl₂/MeOH 15:1): $R_f 0.44$; $[\alpha]^{20}_{D} = -21.2$ (c = 0.66, CH₂Cl₂/MeOH 20:1); mp > 230 °C; IR: v 3276 (m), 1657 (s), 1633 (s), 1490 (s), 1446 (m), 1225 (m), 1036 (m), 740 (m), 698 (m); ¹H NMR (500 MHz, CD₃OD/CDCl₃ 1:4): δ 7.81 (d, ³J (H,H) = 7.5 Hz, 2H), 7.61 (d, ${}^{3}J(H,H) = 7.5$ Hz, 2H), 7.44 (t, ${}^{3}J(H,H) = 7.5$ Hz, 2H), 7.34 (t, ${}^{3}J(H,H) = 7.5$ Hz, 2H), 7.30-7.10 (m, 30H), 4.55 (dd, ${}^{3}J$ (H,H) = 7.6 Hz, ${}^{3}J$ (H,H) = 4.6 Hz, 1H), 4.49-4.34 (m, 4H), 4.23 $(t, {}^{3}J(H,H) = 6.8 \text{ Hz}, 1\text{H}), 4.10 \text{ (dd}, {}^{3}J(H,H) = 9.1 \text{ Hz}, {}^{3}J(H,H) = 5.3 \text{ Hz}, 1\text{H}), 2.80-2.58 \text{ (m}, 10.10 \text{ Hz})$ 4H), 1.84-1.78 (m, 1H), 1.74-1.62 (m, 2H), 1.59-1.48 (m, 2H), 1.36-1.28 (m, 1H), 0.96 (d, ³J $(H,H) = 6.3 Hz, 3H), 0.92 (d, {}^{3}J (H,H) = 6.3 Hz, 3H), 0.86 (d, {}^{3}J (H,H) = 6.1 Hz, 3H), 0.84 (d,$ ${}^{3}J$ (H,H) = 6.1 Hz, 3H); ${}^{13}C$ NMR (125 MHz, CDCl₃): δ 175.0 (s), 172.9 (s), 172.1 (s), 171.0 (s), 170.9 (s), 170.3 (s), 156.2 (s), 144.3 (s), 144.2 (s), 143.8 (s), 143.7 (s), 141.5 (s), 141.4 (s), 128.8 (d), 128.7 (d), 128.1 (d), 128.1 (d), 127.9 (d), 127.3 (d), 127.2 (d), 125.1 (d), 125.0 (d), 120.2 (d), 71.1 (s), 71.0 (s), 67.4 (t), 53.7 (d), 52.7 (d), 51.1 (d), 51.0 (d), 47.0 (d), 40.1 (t), 40.0 (t), 38.9 (t), 36.8 (t), 24.9 (d), 24.8 (d), 23.3 (q), 23.0 (q), 21.6 (q), 21.5 (q); ESI-MS $(MeOH/CH_2Cl_2 1:1): m/z 1200 (100, [M + Na]^+).$

H-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂. Deprotection of Fmoc-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (1.31 g, 1.11 mmol) following the general procedure B and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 25:1) yielded pure H-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (984 mg, 93%) as colorless powder. TLC (CH₂Cl₂/MeOH 10:1): R_f 0.33; $[\alpha]_{D}^{20} = -19.3 \text{ (c}=1.01, \text{CH}_2\text{Cl}_2/\text{MeOH 5:1}); \text{mp} > 230 \text{ °C}; \text{IR: } \upsilon 3278 \text{ (m)}, 2957 \text{ (w)}, 1653$ (s), 1629 (s), 1491 (s), 1445 (m), 1185 (m), 1036 (w), 751 (w), 697 (s); ¹H NMR (500 MHz, DMSO-d₆/D₂O 50:1): δ 7.28-7.15 (m, 30H), 4.51 (dd, ³J (H,H) = 7.9 Hz, ³J (H,H) = 6.3 Hz, 1H), 4.32 (dd, ${}^{3}J$ (H,H) = 10.9 Hz, ${}^{3}J$ (H,H) = 4.0 Hz, 1H), 4.16 (dd, ${}^{3}J$ (H,H) = 14.5 Hz, ${}^{3}J$ $(H,H) = 8.8 Hz, 1H), 3.53 (dd, {}^{3}J(H,H) = 9.8 Hz, {}^{3}J(H,H) = 6.3 Hz, 1H), 2.74 (dd, {}^{2}J(H,H) =$ 14.8 Hz, ${}^{3}J$ (H,H) = 6.3 Hz, 1H), 2.55 (dd, ${}^{2}J$ (H,H) = 14.8 Hz, ${}^{3}J$ (H,H) = 7.9 Hz, 1H), 2.48 $(dd, {}^{2}J (H,H) = 15.0 Hz, {}^{3}J (H,H) = 6.3 Hz, 1H), 2.42 (dd, {}^{2}J (H,H) = 15.0 Hz, {}^{3}J (H,H) = 9.8$ Hz, 1H), 1.63-1.40 (m, 6H), 0.87-0.80 (m, 12H); ¹³C NMR (125 MHz, CD₃OD): δ 174.1 (s), 173.9 (s), 172.1 (s), 170.5 (s), 170.2 (s), 169.2 (s), 145.0 (s), 144.7 (s), 128.6 (d), 128.5 (d), 127.5 (d), 127.4 (d), 126.3 (d), 126.3 (d), 69.4 (s), 69.3 (s), 52.0 (d), 50.9 (d), 50.9 (d), 50.0 (d), 41.4 (t), 41.1 (t), 40.7 (t), 37.3 (t), 24.1 (d), 24.0 (d), 23.2 (q), 23.1 (q), 21.7 (q), 21.4 (q); ESI-MS (MeOH/CH₂Cl₂ 1:1): m/z 1935 (36, [2M + Na]⁺), 979 (100, [M + Na]⁺).

Z-Leu-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂. Coupling of H-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (319 mg, 0.33 mmol) with Z-Leu-OH (127 mg, 0.48 mmol) following the *general procedure* A and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 25:1) and PTLC (CH₂Cl₂/MeOH 20:1) yielded pure Z-Leu-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (314 mg, 79%) as colorless powder. TLC (CH₂Cl₂/MeOH 15:1): R_f 0.26; HPLC (YMC-Pack SIL, 250 **x** 4.6 mm, CH₂Cl₂/MeOH 98:2, 1 ml/min, t_R = 4.1 min); [α]²⁰_D = -29.1 (c=0.1.29, CH₂Cl₂/MeOH 5:1); mp > 230 °C; IR: υ 3280 (m), 2957 (m), 1658 (s), 1636 (s), 1491 (s), 1447 (m), 1214 (m), 1037 (m), 695 (m); ¹H NMR (500 MHz, CD₃OD): δ 7.25-7.09 (m, 35H), 5.01 (d, ²J (H,H) = 12.2 Hz, 1H), 4.94 (d, ²J (H,H) = 12.2 Hz, 1H), 4.44 (dd, ³J (H,H) = 8.1

Hz, ${}^{3}J$ (H,H) = 5.9 Hz, 1H), 4.42 (dd, ${}^{3}J$ (H,H) = 7.2 Hz, ${}^{3}J$ (H,H) = 5.4 Hz, 1H), 4.29 (dd, ${}^{3}J$ (H,H) = 10.5 Hz, ${}^{3}J$ (H,H) = 3.9 Hz, 1H), 4.11 (dd, ${}^{3}J$ (H,H) = 9.7 Hz, ${}^{3}J$ (H,H) = 5.2 Hz, 1H), 4.02 (dd, ${}^{3}J$ (H,H) = 7.6 Hz, ${}^{3}J$ (H,H) = 7.3 Hz, 1H), 2.86 (dd, ${}^{2}J$ (H,H) = 15.2 Hz, ${}^{3}J$ (H,H) = 7.2 Hz, 1H), 2.69 (dd, ${}^{2}J$ (H,H) = 15.2 Hz, ${}^{3}J$ (H,H) = 5.4 Hz, 1H), 2.60 (dd, ${}^{2}J$ (H,H) = 15.7 Hz, ${}^{3}J$ (H,H) = 8.1 Hz, 1H), 1.68-1.43 (m, 9H), 0.93 (d, ${}^{3}J$ (H,H) = 5.8 Hz, 6H), 0.89 (d, ${}^{3}J$ (H,H) = 4.6 Hz, 6H), 0.82 (d, ${}^{3}J$ (H,H) = 6.6 Hz, 6H); ${}^{13}C$ NMR (125 MHz, CD₃OD): δ 176.9 (s), 174.8 (s), 174.2 (s), 172.9 (s), 172.5 (s), 170.9 (s), 170.4 (s), 157.9 (s), 145.1 (s), 145.0 (s), 136.9 (s), 129.4 (d), 129.3 (d), 129.0 (d), 128.6 (d), 128.4 (d), 128.3 (d), 128.2 (d), 127.6 (d), 127.5 (d), 71.3 (s), 71.1 (s), 67.5 (t), 54.7 (d), 53.9 (d), 52.6 (d), 51.7 (d), 51.4 (d), 41.3 (t), 40.5 (t), 40.3 (t), 38.0 (t), 38.0 (t), 25.4 (d), 25.3 (d), 25.2 (d), 23.5 (q), 23.3 (q), 23.2 (q), 21.8 (q), 21.6 (q), 21.5 (q); ESI-MS (MeOH/CH₂Cl₂ 1:1): m/z 2430 (75, [2M + Na]⁺), 1226 (100, [M + Na]⁺).

H-Leu-Asn(Trt)-Leu-Asn(Trt)-Leu-NH, 5. Deprotection of Z-Leu-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂ (82.8 mg, 0.069 mmol) following the general procedure C and purification of the crude product by column chromatography (CH₂Cl₂/MeOH 12:1 then 7:1) and PTLC (CH₂Cl₂/MeOH 10:1) yielded pure 5 (984 mg, 93%) as colorless powder. TLC (CH₂Cl₂/MeOH 10:1): R_f 0.25; HPLC (YMC-Pack SIL, 250 x 4.6 mm, CH₂Cl₂(0.1% TEA)/MeOH 97:3, 1 ml/min, $t_R = 3.4$ min); $[\alpha]_D^{20} = -17.4$ (c=0.66, CH₂Cl₂/MeOH 5:1); mp > 230 °C; IR: v 3286 (m), 2956 (m), 1634 (s), 1490 (s), 1446 (m), 1186 (m), 1036 (w), 697 (s); ¹H NMR (500 MHz, DMSO-d₆/D₂O 50:1): δ7.33-7.27 (m, 12H), 7.25-7.17 (m, 18H), 4.61 $(dd, {}^{3}J(H,H) = 8.9 Hz, {}^{3}J(H,H) = 4.8 Hz, 1H), 4.52 (dd, {}^{3}J(H,H) = 7.3 Hz, {}^{3}J(H,H) = 7.3 Hz,$ 1H), 4.30 (dd, ${}^{3}J$ (H,H) = 9.1 Hz, ${}^{3}J$ (H,H) = 4.9 Hz, 1H), 4.19 (dd, ${}^{3}J$ (H,H) = 9.1 Hz, ${}^{3}J$ $(H,H) = 6.0 Hz, 1H), 3.25 (dd, {}^{3}J (H,H) = 9.4 Hz, {}^{3}J (H,H) = 5.0 Hz, 1H), 2.82-2.72 (m, 2H),$ 2.65 (dd, ${}^{2}J$ (H,H) = 14.9 Hz, ${}^{3}J$ (H,H) = 4.8 Hz, 1H), 2.48 (dd, ${}^{2}J$ (H,H) = 15.2 Hz, ${}^{3}J$ (H,H) = 7.3 Hz, 1H), 1.82-1.72 (m, 1H), 1.66-1.54 (m, 2H), 1.52-1.40 (m, 5H), 1.30-1.22 (m, 1H), 0.94-0.82 (m, 18H); ¹³C NMR (125 MHz, CD₃OD): δ 175.6 (s), 175.3 (s), 172.5 (s), 171.0 (s), 170.9 (s), 169.4 (s), 169.1 (s), 143.4 (s), 127.9 (d), 126.9 (d), 126.0 (d), 69.8 (s), 69.7 (s), 52.0 (d), 51.6 (d), 51.1 (d), 50.0 (d), 49.5 (d), 39.2 (t), 39.2 (t), 39.1 (t), 39.0 (t), 37.0 (t), 36.5 (t), 23.8 (d), 23.7 (d), 23.6 (d), 22.0 (q), 21.8 (q), 21.6 (q), 20.5 (q), 20.0 (q), 20.0 (q); ESI-MS (MeOH/CH₂Cl₂ 1:1): m/z 1091 (40, [M + Na]⁺), 1069 (100, [M + H]⁺).

H-Leu-Arg(Pmc)-Leu-His(Trt)-Leu-NH₂ 6. This compound was prepared in 8 steps following previously reported procedures.^{S2}

1³,2³,3²,4³,5²,6³,7²,8³-Octakis(*Gla*-Leu-Arg-Leu-His-Leu-NH₂)-*p*-octiphenyl. The monomer of pore 2 was prepared in 19 steps following previously reported procedures.^{S2}

1³,2³,7²,8³-Tetrakis(*Gla*-OBn)-3²,4³,5²,6³-tetrakis(*Gla*-OtBu)-*p*-octiphenyl 3. This compound was prepared in 10 steps following previously reported procedures.⁸³

1³,**2**³,**7**²,**8**³-**Tetrakis**(*Gla*-OH)-**3**²,**4**³,**5**²,**6**³-**tetrakis**(*Gla*-OtBu)-*p*-octiphenyl 7. $Pd(OH)_2/C$ (1 mg) was added to a solution of **3** (13 mg, 7.3 µmol) in a mixture of MeOH (5 ml) and THF (5 ml), and the suspension was degassed before being set under a H₂-atmosphere. After stirring for 2 h, the reaction mixture was filtered through celite and basic alumina. Concentration of the filtrate *in vacuo* yielded pure **7** (10.5 mg, 100%) as a colorless powder. ¹H NMR (300 MHz, CDCl₃/CD₃OD 1:1): δ 7.51-7.43 (m, 6H), 7.39-7.15 (m, 18H), 6.94 (d, ³*J* (H,H) = 7.6 Hz, 2H), 4.67 (s, 4H), 4.65 (s, 4H), 4.62 (s, 4H), 4.61 (s, 4H), 1.48 (s, 36H).



Scheme S1. (a-j) see ref. S3; (k) H_2 , $Pd(OH)_2/C$, MeOH/THF, 2 h, rt, quant; (l) 5, HATU, TEA, DMF, 3 h, rt, 57%; (m) TFA, 3 h, rt, quant; (n) 6, HATU, TEA, DMF, 3 h, rt, 65%; (o) TFA, 3 h, rt, quant; (p) self-assembly in lipid bilayers.

1³,**2**³,**7**²,**8**³-**Tetrakis**(*Gla*-Leu-Asn(Trt)-Leu-Asn(Trt)-Leu-NH₂)-**3**²,**4**³,**5**²,**6**³-tetrakis(*Gla*-O*t***Bu**)-*p*-octiphenyl **8**. A solution of **5** (27 mg, 25.2 μmol) in DMF (610 μl) was added to **7** (6 mg, 4.2 μmol), HATU (13 mg, 33.6 μmol) and TEA (14 μl, 100 μmol). After stirring for 3 h at rt, the solvent was evaporated and the crude product dried *in vacuo*. Purification of the crude product by PTLC (first with CHCl₃/MeOH 4:1, R_f = 0.9, then with CHCl₃/ MeOH 20:1, R_f = 0.1) yielded HPLC-pure (YMC-Pack SIL, 250 x 4.6 mm, CHCl₃/ MeOH 93:7, 1 ml/min, t_R = 4.77 min) **8** (13.5 mg, 57%) as a colorless solid. ¹H NMR (300 MHz, CDCl₃/CD₃OD 1:1): δ 8.33-8.29 (m, ≤ 28 NH (slow exchange)), 7.93-7.73 (m, 8H), 7.49-7.45 (m, 4H), 7.42-7.39 (m, 10H), 7.26-7.09 (m, 122H), 6.89 (d, ³*J* (H,H) = 8.3 Hz, 2H), 4.58-4.36 (m, 26H), 4.35-4.27 (m, 4H), 4.24-4.15 (m, 6H), 2.88-2.60 (m, 12H), 2.32-2.21 (m, 4H), 1.75-1.48 (m,

30H), 1.46 (s, 18H), 1.44 (s, 18H), 1.40-1.28 (m, 6H), 0.99-0.80 (m, 60H), 0.63 (d, ${}^{3}J(H,H) = 5.52$ Hz, 6H), 0.59 (d, ${}^{3}J(H,H) = 6.49$ Hz, 6H); ESI-MS (CH₂Cl₂/MeOH 9:1): m/z 1902 (100, [M+3Na⁺]³⁺), 1432 (57, [M+4Na⁺]⁴⁺).

1³,2³,7²,8³-Tetrakis(Gla-Leu-Asn-Leu-Asn-Leu-NH₂)-3²,4³,5²,6³-tetrakis(Gla-OH)-p-

octiphenyl 9. A solution of 8 (12.5 mg, 2.2 μ mol) in TFA (2 ml) was stirred for 1.5 h at rt. After evaporation, impurities were removed by solid-liquid extraction with hexane (5 x 2 ml) and dried. Then, TFA (2 ml) was added and the solution was stirred for 1.5 h at rt. Following the addition of CH₂Cl₂ (1 ml) and evaporation of the solvent, impurities were removed by solid-liquid extraction with hexane (1 ml, 4x), toluene (1 ml, 2x) and CH₂Cl₂ (1 ml, 3x) to give HPLC-pure 9 (7.7 mg, 100%). ¹H NMR (300 MHz, CDCl₃/CD₃OD 1:1): δ 7.52-7.23 (m, 22H), 7.19-7.13 (m, 4H), 6.88 (d, ³J (H,H) = 7.81 Hz, 2H), 5.06-4.42 (m, 36H), 3.12-2.89 (m, 16H), 1.78-1.23 (m, 36H), 0.92-0.62 (m, 72H).

1³,2³,7²,8³-Tetrakis(Gla-Leu-Asn-Leu-Asn-Leu-NH₂)-3²,4³,5²,6³-tetrakis(Gla-Leu-

Arg(Pmc)-Leu-His(Trt)-Leu-NH₂)-*p*-octiphenyl 10. A solution of 6 (15.4 mg, 13.3 μmol) in DMF (500 μl) was added to 9 (7.7 mg, 2.2 μmol), HATU (6.7 mg, 17.7 μmol) and TEA (7.4 μl, 53.2 μmol). After stirring for 3 h at rt, the solvent was evaporated and the product dried *in vacuo*. Impurities were removed by solid-liquid extraction with EtOAc (1 ml, 3x). Purification of the crude by column chromatography (LH-20 sephadex column, DMF) and RP-HPLC (YMC-Pack ODS-A, 250 x 410 mm, MeOH, 2 ml/min, t_R = 4.18 min) gave HPLC-pure 10 (11.5 mg, 65%) as a colorless solid. ¹H NMR (300 MHz, CDCl₃/CD₃OD/DMF 20:20:1): δ 7.58-6.97 (m, 94H), 5.06-4.17 (m, 56H), 3.24-2.65 (m, 32H), 2.64-2.44 (m, 32H), 2.07-1.97 (m, 12H), 1.82-1.24 (m, 96 H), 1.28-1.22 (m, 24H), 1.01-0.57 (m, 144H).

1³,2³,7²,8³-Tetrakis(*Gla*-Leu-Asn-Leu-Asn-Leu-NH₂)-3²,4³,5²,6³-tetrakis(*Gla*-Leu-Arg-

Leu-His-Leu-NH₂)-p-octiphenyl 4. A solution of 10 (2.0 mg, 0.25 µmol) in TFA (2 ml) was stirred for 1.5 h at rt. After addition of CH₂Cl₂ (1 ml), evaporation of the solvents and solidliquid extraction of impurities with hexane (2 ml, 5x), TFA (2 ml) was added and the solution was stirred for 1.5 h at rt, diluted with CH₂Cl₂ (1 ml) and the solvents were evaporated. Impurities were removed by solid-liquid extraction with hexane (1 ml, 4x), toluene (1 ml, 2x) and CH₂Cl₂ (1 ml, 3x). The crude product was purified by RP-HPLC (YMC-Pack ODS-A, 250 x 10 mm, MeOH/ H_2O / TFA 92:8:1, 2 ml/min, $t_R = 5.20$) to give HPLC-pure (Fig. S1) 4 (1.5 mg, 100%) as a colorless solid. ¹H NMR (300 MHz, CD₃OD): δ 7.64-7.20 (m, 36H), 7.09-7.04 (m, 2H), 4.80-4.20 (m, 56H), 3.26-2.55 (m, 32H), 1.82-1.27 (m, 88H), 1.04-0.64 (m, 144H); ESI-MS (MeOH/CH₃CN/AcOH 74:24:2): m/z 2038 (11, [M+3H+TFA]³⁺), 2033 (8, [M+3H+H₃PO₄]³⁺), 3032 (11, [M+3H+TFA-H₂O/NH₃]³⁺), 1586 (21, [M+4H+3TFA]⁴⁺), 1581 (17, $[M+4H+2TFA+H_3PO_4]^{4+}$), 1580 (17, $[M+4H+3TFA-H_2O/NH_3]^{4+}$), 1577 (12, $[M+4H+TFA+2H_3PO_4]^{4+}$, 1577 (12, $[M+4H+2TFA+H_3PO_4-H_2O/NH_3]^{4+}$), 1557 (38, $[M+4H+2TFA]^{4+}$, 1553 (33, $[M+4H+TFA+H_3PO_4]^{4+}$), 1552 (33, $[M+4H+2TFA-H_3PO_4]^{4+}$) H₂O/NH₃]⁴⁺), 1548 (12, [M+4H+TFA+H₃PO₄-H₂O/NH₃]⁴⁺), 1529 (30, [M+4H+TFA]⁴⁺), 1525 $(16, [M+4H+H_3PO_4]^{4+}), 1524 (39, [M+4H+TFA-H_2O/NH_3]^{4+}), 1520 (13, [M+4H+H_3PO_4 H_2O/NH_3^{4+}$, 1500 (75, $[M+4H]^{4+}$), 1496 (34, $[M+4H-H_2O/NH_3]^{4+}$), 1491 (15, $[M+4H-H_2O/NH_3]^{4+}$) 2H₂O/NH₃]⁴⁺), 1268 (14, [M+5H+3TFA]⁵⁺), 1265 (17, [M+5H+2TFA+H₃PO₄]⁵⁺), 1265 (17, $[M+5H+3TFA-H_2O/NH_3]^{5+}),$ 1262 (11, $[M+5H+TFA+2H_3PO_4]^{5+}),$ 1261 (20, $(26, [M+5H+2TFA]^{5+}),$ $[M+5H+2TFA+H_3PO_4-H_2O/NH_3]^{5+}),$ 1246 1242 (21, $[M+5H+TFA+H_3PO_4]^{5+}),$ 1242 $(21, [M+5H+2TFA-H_2O/NH_3]^{5+}),$ 1239 (13,

 $\begin{bmatrix} M+5H+TFA+H_3PO_4-H_2O/NH_3\end{bmatrix}^{5+}, 1222 (40, [M+5H+TFA]^{5+}), 1220 (33, [M+5H+H_3PO_4]^{5+}), 1219 (30, [M+5H+TFA-H_2O/NH_3]^{5+}), 1216 (25, [M+5H+H_3PO_4-H_2O/NH_3]^{5+}), 1200 (100, [M+5H]^{5+}), 1197 (44, [M+5H-H_2O/NH_3]^{5+}), 1193 (16, [M+5H-2H_2O/NH_3]^{5+}), 1038 (13, [M+6H+2TFA]^{6+}), 1019 (13, [M+6H+TFA]^{6+}), 1017 (14, [M+6H+H_3PO_4]^{6+}), 1016 (21, [M+6H+TFA-H_2O/NH_3]^{6+}), 1014 (14, [M+6H+H_3PO_4-H_2O/NH_3]^{6+}), 1001 (88, [M+6H]^{6+}), 998 (36, [M+6H-H_2O/NH_3]^{6+}), 995 (13, [M+6H-2H_2O/NH_3]^{6+}) (Fig. S2).$



Figure S1. HPLC of target molecule 4 (detection at 254 nm).



Figure S2. ESI-MS of target molecule **4** (TFA salt) under denaturating acidic conditions (MeOH/CH₃CN/AcOH 74:24:2).

Fig. S2, comments: For counteranion scavenging, see ref (S4).

EYPC-LUVsDANTS/DPX. Solutions of EYPC (25 mg) in CHCl₃/MeOH 1/1 (2 ml) were dried using rotary evaporator and then under vacuum (>2 h) to form thin films. The resulting films were hydrated with 1 ml buffer (12.5 mM ANTS, 45.0 mM DPX, 5 mM TES, 20 mM KCl, pH 7.0) for more than 30 min and subjected to freeze-thaw cycles (5x) and extrusions (15x, Mini-Extruder with a polycarbonate membrane, pore size 100 nm). Extravesicular dyes were removed by gel filtration (Sephadex G-50; 5 mM TES, 100 mM KCl, pH 7.0; external cation exchange using 5 mM TES, 100 mM NaCl, pH 7.0 was done for polarized LUVs; see below). The LUV fractions were combined and diluted to 6 ml with the same buffers. Lipid concentrations were estimated from amount of entrapped dye. The estimated values were consistent with earlier results from phosphate analysis.⁸⁵ The final stock solutions had the following characteristics: ~1.3 mM EYPC; 12.5 mM ANTS, 45.0 mM DPX, 5 mM TES, 20 mM KCl, pH 7.0 inside, 5 mM TES, 100 mM KCl or NaCl, pH 7.0 outside.

Vesicle Polarization.⁸⁶ 50 µl of LUV-stock solution was diluted with a mixture of k µl of 100 mM KCl, 10 mM MES, pH 4.5-7.0, and 1950 - k µl of 100 mM NaCl, 10 mM MES, pH 4.5-7.0, containing safranin O (60 nM). Membrane potential V was calculated from the Nernst equation [S1]

$$V(mV) = 59 x \log \left(\left[K_{out}^{+} \right] / \left[K_{in}^{+} \right] \right)$$
 [S1],

$$= -153.5 + 59 \times \log k$$
 [S1']

assuming external $[K^+_{out}] \approx 0$ in the LUV-stock solution. For calibration, the emission of safranin O was monitored at λ_{em} 581 nm (λ_{ex} 522 nm) as a function of time during addition of valinomycin (20 µl of 60 µM, final 600 nM), incubation for ~100 sec (until emission of polarized vesicles was constant) and successive addition of external K⁺ buffer. Linear correlation was found between the emission intensities and the calculated Nernst potential from -80 to -190 mV.^{S6,S7} Extrapolation to $k = 0 \mu l$ gave V = -211 mV for a Nernst potential of $-\infty$ mV, indicative for residual external K⁺ in the LUV-stock solution or "saturation" (S6). The magnitude of vesicle polarization was determined in each experiment by comparison of the measured Safranin-O emission intensity after the addition of valinomycin with this calibration curve.

ANTS/DPX-assay. Unpolarized LUVs. EYPC-LUVs DANTS/DPX (50 µl) were added to gently stirred, thermostated buffer (1.95 ml; 100 mM KCl, 10 mM MES, pH 4.5-7.0, 25 °C). Changes in fluorescence emission of ANTS (F_{t} , $\lambda_{em} = 510$ nm, $\lambda_{ex} = 353$ nm; similar to top traces in Fig. 2A) was monitored (FluoroMax-3, Jobin Yvon) as a function of time during addition of blocker (at time 0, ADP, ATP, phytate, varied concentrations), barrel 1 (usually 250 nM; for concentration dependence 0-250 nM: see Fig. S3B) or 2 (75 nM) at 2 min and 40 µl of 1.2 % Triton X-100 at 9.5 min. Precise control of temperature and continuous stirring during fluorescence kinetics measurements were crucial for reproducible results. Polarized LUVs. EYPC-LUVs⊃ANTS/DPX (50 µl, external salts exchanged to Na⁺) were added to gently stirred, thermostated buffer (k µl of 100 mM KCl, 5 mM MES, 1950-k µl of 100 mM NaCl, 5 mM MES, pH 4.5-7.0, 25 °C). Changes in fluorescence emission of ANTS (λ_{em} = 510 nm, $\lambda_{ex} = 353$ nm; top traces in Fig. 2A) and safranin O (λ_{em} 581 nm, λ_{ex} 522 nm, bottom traces in Fig. 2A) were monitored simultaneously in two different channels as a function of time during addition of safranin O (60 nM final), blocker (ADP, ATP, phytate, varied concentrations), valinomycin (0.6 μ M; Fig. 2A: (c) = 1 min), barrel 1 (250 nM) or 2 (75 nM) at time (d) (Fig. 2A: (d) = 5 min), and 40 μ l melittin (1.3 mg / ml H₂O) at time (e) (Fig. 2A: (e) = 9.5 min).

Data analysis. Fluorescence time courses were normalized to fractional emission intensity I^n using equation [S2]

$$I^{n} = (F_{t} - F_{0}) / (F_{\infty} - F_{0})$$
[S2],

where $F_0 = F_t$ at pore addition, $F_{\infty} = F_t$ at saturation after lysis. If appropriate, fractional pore activity *Y* was calculated using equation [S3]

$$Y = I^n / I^n_{MAX}$$
[S3],

where I_{MAX}^{n} is I^{n} at saturation obtained under the conditions giving the highest activity, and used instead of the fractional emission intensity I^{n} . Effective monomer concentrations EC_{50} and thermodynamic pore stability and/or^{S8} stoichiometry *n* were determined from a fit of the dependence of pore activity on monomer concentration c_{M} to the Hill equation [S4]

$$I^{n} = I^{n}_{MAX} + (I^{n}_{0} - I^{n}_{MAX}) / \{1 + c_{M} / EC_{50}\}^{n}\}$$
[S4],

where I_0^n is I^n without pore, I_{MAX}^n is I^n four minutes after addition of excess pore and n is the Hill coefficient (pH, V, T, = constant). Gating charges z_g were determined from the fit of the dependence of pore activity on vesicle polarization V (at constant pore concentration = 250 nM and constant pH = 5.5) to equation [S5]

$$Y \propto \exp\left(z_e \ e \ V / k \ T\right)$$
[S5]

where *e* is the elementary charge, *k* the Boltzmann constant, and *T* the absolute temperature (pH, T, $c_{\rm M}$ = constant).^{S6} Apparent dissociation constants $K_{\rm D}$ (more accurately: inhibitory concentrations IC_{50}) were determined from the dependence of fractional pore activities *Y* at 400 sec on the blocker concentration and fitted to the Hill equation [S6]

$$Y = Y_{\infty} + (Y_0 - Y_{\infty}) / \{1 + c_{\text{BLOCKER}} / K_{\text{D}})^n\}$$
[S6],

where Y_0 is Y without ligand, Y_{∞} is Y with excess ligand and n is the Hill coefficient (pH, V, T, c_M = constant). The Woodhull distance l_W from pore entrance to active site OR the effective charge z_{eff} of the blocker were determined by the fit of the dependence of the apparent dissociation constant K_D on the vesicle polarization V to the Woodhull equation [S7]

$$pK_{\rm D} = pK_{\rm D} (0 \text{ mV}) - (z_{\rm eff} l F / l_{\rm W} 2.303 R T) V$$
[S7],

where *l* is the length of the pore (34 Å assumed), *F* the Faraday constant, *R* the gas constant and *T* the absolute temperature.^{S9}



Figure S3. Dependence of the fractional activity of pore **1** on pH (A, 250 nM tetramer, V = 0 mV) on concentration (B, curve fit to eqn [S4], pH = 4.25, V = 0 mV), and on vesicle polarization (C, curve fit to eqn [S5], 250 nM tetramer, pH = 5.5).

<u>*Fig. S3, comments:*</u> (A) for pH profile of pore **2**, see (S10). (B) Based on extensive precedence, ^{S8,S10} pore concentration was approximated as monomer concentration $c_M/4$. Linear dependence indicated that pore **1** is formed by monomeric rods (unlikely) or by exergonic self-assembly (likely). ^{S8,S10} Absence of early saturation indicated that pore **1** has reduced thermodynamic stability compared to pore **2**. ^{S8,S10} (C) Curve fit to eqn [S5] resulted in the gating charge z_g 0.047. This value indicated voltage independence of the pore **1**. For comparison of results on voltage gating from polarized vesicles with IV profiles from planar bilayer conductance experiments, see (S6).

Planar Bilayer Conductance. BLMs were formed by painting a solution of EYPC in *n*-decane (42 mg/ml) on an aperture (d = 150 μ m, pretreated with the same solution) in a delrin cuvette separating two chambers containing 1 ml each of buffer (5mM MES 1.05 M KCl, pH 4.65) and agar bridge connection (2 M KCl) to Ag/AgCl electrodes (Warner Instrument Corp. Hamden, CT). Currents were recorded at different holding potentials (*trans* at ground) in a house-made Faraday cage, amplified (BC-525c, Warner Instrument Corp.), low-pass filtered with a 8-pole Bessel filter at 1 kHz (LPF-8, Warner Instrument Corp.), A-D converted (DigiData 1200, Axon Instruments, Union City, CA), and sampled at 10 kHz by computer (pClamp 8.0, Axon Instruments). Pore **1** (250 nM) was added to the *cis* chamber. All the conductance measurements were performed at room temperature (22 ± 1 °C). Conductance *g* was estimated from the histogram, multiplied by Sansom's correction factor (5.61)^{S11} and applied to Hille's equation^{S12}

$$1 / g = (l + \pi d / 4) \times (4\rho / \pi d^2)$$
[S8]

where *l* is the ion channel length (34 Å) and ρ the resistivity of the recording solution ($\rho = 7.8 \ \Omega \cdot cm$), to give the calculated channel diameter *d*, 0.7 nm (for pore **2**, see (S5)). To examine the effect of ADP to conductance of the pores, BLMs were prepared as described above in a buffer (10 mM MES, 0.1 M KCl, pH 4.5). Currents through multiple channels were measured at different holding potential in the presence of increasing amount of ADP in the *trans* chamber. The average currents over 100 ms (*Y*) were fitted to the Hill equation [S6] (Figure S4) to give apparent $K_{\rm D}s$. Thus obtained $K_{\rm D}s$ were applied to the Woodhull equation [S7] (Fig. 3B). Data points for pore **1** in Fig. 3B were the averages of two independent series of experiments \pm errors.



Figure S4. Representative multichannel dose response curves at 60 mV (open circles) and 100 mV (filled circles).



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