Is phenyl a good migrating group in the rearrangement of organoborates generated from sulfur ylides?

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Supporting information

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- IV. Cartesian coordinates and energies (a.u.) of the considered structures

I. Experimental procedures and analysis data

General information

All air and water sensitive reactions were carried out in oven dried glassware under an argon atmosphere. Solvents were dried by standard methods.¹ NMR spectra were recorded on Jeol 270 MHz or Jeol 400 MHz spectrometers using tetramethylsilane as the internal standard (0.00 ppm). CDCl₃ was used as the internal standard for ¹³C NMR spectra (0.0 ppm). CI mass spectra were obtained using a VG Platform mass spectrometer. IR spectra were obtained on a Perkin-Elmer Spectrum One FT-IR spectrometer. Flash chromatography was performed using silica gel (Merck Kieselgel 60). Melting points were determined with a Kofler hot stage apparatus and were not corrected.

Synthesis of Dihexylphenylborane

After extensive study, we found that dihexylphenylborane could be conveniently synthesised through the reaction of PhMgCl with methyl dihexylborinic ester rather than chlorodihexylborane which had been used previously.² The selection of reaction solvent is important. Reactions carried out in pentane as indicated in literature led to formation of a mixture of boranes,² while the reactions in diethyl ether gave the desired compound in high purity and good yield. We also found that the concentration of the methyl borinic ester in certain cases (e.g. synthesis of dihexylphenylborane) was important as well. The reaction using 1 M solution of methyl dihexylborinic ester in diethyl ether as indicated in the literature² gave a mixture of boranes, while the reaction using 0.5 M solution of methyl dihexylborinic ester in diethyl ether in diethyl ether furnished the desired compound in high yield and purity.

Synthesis of chlorodihexylborane

We synthesised chlorodihexylborane from 1-hexene and boron trichloride following Matteson's procedure.³

To a vigorously stirred boron trichloride solution (1 M in heptane, 35 mL) was added a mixture of 1hexene (2.94 g, 35 mmol) and diethylsilane (3.08 g, 35 mmol) at -78 °C. After addition, the mixture

¹ A. B. Pangborn, M. A. Giardello, R. M. Grubbs, R. K. Rosen, F. T. Timmers, *Organometallics*, 1996, **15**, 1518.

² W. K. Gary, H. C. Brown, J. Organomet. Chem. 1974, 73, 1

³ R. Soundararajan, D. S. Matteson, J. Org. Chem., 1990, 55, 2274.

was stirred for 30 mins at -78 °C and for another 30 mins at room temperature. After the solvent was removed under vacuum a crude product was obtained as a colourless oil, which, according to the ¹¹B NMR and ¹H NMR, was in high purity and was used without further purification in the next transformation. ¹H NMR (300 MHz, CDCl₃) 0.86-0.90 (6H, m), 1.26-1.38 (16H, m), 1.48-1.50 (4H, m). ¹¹B NMR (96 MHz, CDCl₃) 77.

Synthesis of methyl dihexylborinic ester

The crude chlorodihexylborane (2.9 g, 14.0 mmol) obtained above was dissolved in hexane (20 mL). To the solution was added slowly pre-degassed methanol (0.7 mL, 16.8 mmol) at -78 °C. After addition, the reaction mixture was allowed to warm to room temperature under reduced pressure. As the temperature increased hydrochloride, hexane and the excess methanol were removed, and the residue was the crude product, which, examined by ¹H NMR and ¹¹B NMR, was in high purity and was used without further purification. ¹H NMR (300 MHz, CDCl₃) 0.81-0.90 (10H, m), 1.25-1.37 (16H, m), 3.68 (3H, s). ¹¹B NMR (96 MHz, CDCl₃) 54.

Synthesis of Dihexylphenylborane

A pre-dried 100 mL Schlenk flask was charged with methyl dihexylborinic ester (1.0 M in diethyl ether, 20 mL, 20 mmol). The solution was diluted to 0.5 M by addition of anhydrous diethyl ether (20 mL). Phenyl magnesium chloride (2 M in THF, 20 mmol) was added dropwise to the solution at -78 °C. After addition, the reactants were allowed to worm to room temperature and stirred for 3 hours. The solvent was removed under reduced pressure a white solid, which presumably is the ate complex of the borane, was obtained. Hexane (60 mL) was introduced to crack the ate complex. The mixture was vigorously stirred for 5 hours and was then the slurry mixture was allowed to settle. The supernatent was transferred into a Schlenk flask by cannular. The solid was extracted with hexane (60 mL) again. The hexane solutions were combined. After removing the solvent the crude product (4.18g, 81%) was obtained as a colourless oil, which was used in next transformation without further purification. ¹H NMR (300 MHz, CDCl₃) 0.88 (6H, t, *J*=7.0 Hz), 1.26-1.58 (16H, m), 1.59 (4H, t, *J*=7.3 Hz), 7.35-7.45 (3H, m), 7.80-7.84 (2H, m). ¹³C NMR (75 MHz, CDCl₃) 14.3, 22.8, 25.3, 32.0, 33.1, 127.9, 131.7, 134.1. ¹¹B NMR (96 MHz, CDCl₃) 78.

Synthesis of *B*-phenyl-9-borabicyclo(3,3,1)nonane⁴

B-Methoxy-9-BBN (1 M solution in hexane, Aldrich, 20 mL, 20 mmol) was introduced into a Schlenk flask. The hexane was removed under reduced pressure and the borinic ester was redissolved in anhydrous diethyl ether (40 mL). To the solution of the borinic ester in diethyl ether was added phenyl magnesium chloride (2 M in THF, 10 mL, 20 mmol) at -78 °C. The reaction mixture was then stirred at room temperature for 3 hours. The same workup procedure as the one which was applied in the synthesis of dihexylphenylborane was followed to give a colourless oil (2.81g, 75%). ¹H NMR (300 MHz, CDCl₃) 1.29-1.40 (m, 4H), 1.78-1.93 (m, 4H), 2.00-2.08 (m, 4H), 2.31-2.37 (m, 2H), 7.48-7.52 (m, 2H), 7.56-7.63 (m, 1H), 8.01 (d, J = 8.4 Hz, 2H). ¹³C NMR (75 MHz, CDCl₃) 23.7, 34.4, 128.3, 133.0, 134.9. ¹¹B NMR (96 MHz, CDCl₃) 80.

Synthesis of diethylphenylborane

Methyl diethylborinic ester (2.6 mL, 20.0 mmol) was dissolved in diethyl ether (40 mL). To the solution was added phenyl magnesiumchloride (2 M in THF, 10 mL, 20 mmol) at -78 °C. The reaction mixture was then allowed to room temperature and stirred for 3 hours. The same workup procedure as the one which was applied in the synthesis of dihexylphenylborane was followed which result in a colourless oil in (1.9 g, 65% yield). ¹H NMR (300 MHz, CDCl₃) 1.06 (6H, t, J = 7.5 Hz), 1.61 (4H, q, J = 7.5 Hz), 7.35-7.45 (3H, m), 7.80-7.84 (2H, m). ¹³C NMR (75 MHz, CDCl₃) 7.5, 126.7, 133.0, 134.6. ¹¹B NMR (96 MHz, CDCl₃) 77.

General Procedure for Borane-Ylide Reactions

To the mixture of sulfonium salt (0.50 mmol) in THF (4 mL) was added LHMDS (1 M in THF, 0.55 mL) at -78 °C. The mixture was stirred for 2 hours before being cooled to the temperature around -100-110 °C. The borane (1M in THF, 0.55 mL) was added within 10 mins. Half an hour later after addition, the cooling bath was removed, and the reaction temperature was allowed to rise to 0 °C. Aqueous NaOH solution (2 M, 2 mL) and H₂O₂ solution (30%, 1 mL) were added subsequently. The mixture was stirred at room temperature for 8 hours. Then, the reaction mixture was diluted with NaOH (2 M, 4 mL) and water (10 mL) and extracted with diethyl ether twice (2 × 30 mL). The ethereal solutions were combined and dried over anhydrous MgSO₄. After filtration and removal of the solvent under reduced pressure, the residue was further purified by chromatography or directly analysed by GC.

⁴ E. Kalbarczk, S. Pasynkiewicz, J. Organomet. Chem., 1985, 292, 119.

The reaction of 4-methylbenzyl sulfonium tetrafluoroborate and diethylphenylborane⁵

1-(4-Methylphenyl)-1-propanol and 1-(4-methylphenyl)-1-phenylmethanol were obtained. 1-(4-Methylphenyl)-1-propanol was obtained as a colourless oil in 65% yield: ¹H NMR (400 MHz, CDCl₃) 0.89 (3H, t, J = 7.6 Hz), 1.67-1.84 (2H, dq, J = 7.6, 7.2 Hz), 2.33 (3H, s), 4.52 (1H, t, J=7.2 Hz). 7.13-7.15 (2H, d, J = 7.8 Hz), 7.20-7.23 (2H, d, J = 7.8 Hz). ¹³C NMR (101 MHz, CDCl₃) 10.1, 21.0, 31.7, 75.7, 125.9, 129.0, 137.0, 141.5. 1-(4-Methylphenyl)-1-phenylmethanol was obtained as a white crystalline solid in 28% yield: m.p. 62-3 °C (ethyl acetate-petrol). ¹H NMR (400 MHz, CDCl₃) 2.30 (s, 3H), 5.72 (s, 1H), 7.08-7.34 (m, 9H).

The reaction of benzyl sulfonium tetrafluoroborate and dihexylphenylborane

Both 1-phenylheptan-1-ol and diphenylmethane were obtained. 1-Phenylheptan-1-ol was obtained as colourless oil in 63% isolated yield, and diphenylmethane was obtained as colourless oil in 29% yield. Data obtained for 1-phenylheptan-1-ol: ¹H NMR⁶ (400 MHz, CDCl₃) 0.86 (3H, t, *J*=6.8 Hz), 1.21-1.43 (8H, m), 1.64-1.73 (1H, m), 1.74-1.82 (1H, m), 4.63 (1H, dt, *J*=3.0, 6.2 Hz), 7.24-7.34 (5H, m). ¹³C NMR (101 MHz, CDCl₃) 14.2, 22.7, 25.9, 29.3, 31.9, 39.2, 74.8, 126.0, 127.5, 128.3, 145.0. Data obtained for diphenylmethane: ¹H NMR⁷ (400 MHz, CDCl₃) 3.91 (2H, s), 7.10-7.22 (10H, m). ¹³C NMR (101 MHz, CDCl₃) 42.0, 126.2, 128.5, 129.0, 141.2.

The reaction of 4-methylbenzyl sulfonium tetrafluoroborate and *B*-phenyl-9-borabicyclo(3,3,1)nonane

1-(4-Methylphenyl)-1-phenylmethanol⁸ and *cis*-5-(Hydroxy-4-tolylmethyl)cyclooctanol were obtained in a 5/2 ratio (crude). *cis*-5-(Hydroxy-*p*-tolylmethyl)cyclooctanol was obtained as a white crystalline solid in 24% yield. M.p 108-110 °C; v_{max}/cm^{-1} (neat) 3273, 2919, 2852. ¹H NMR (400 MHz, CDCl₃) 1.15-1.24 (1H, m), 1.29-1.48 (5H, m), 1.54-1.66 (2H, m), 1.72-1.92 (7H, m), 2.34 (3H, s), 3.91-3.97 (1H, m), 4.36 (1H, d, *J*=7.3 Hz), 7.14 (2H, d, *J*=7.8 Hz), 7.20 (2H, d, *J*=7.8 Hz). ¹³C NMR (101 MHz, CDCl₃) 21.2, 22.9, 23.3, 28.8, 30.5, 36.2, 44.3, 51.1, 72.2, 79.4, 126.6, 129.1, 137.2, 140.8. *m/z* (ESI) 271 (M + Na⁺); HRMS found 271.1673, C₁₆H₁₈NaO₂ required 271.1667. 1-(4-Methylphenyl)-1-

⁵ Y.-J. Chen, R.-X. Lin, C. Chen, *Tetrahedron Asymmetry*, 2004, **15**, 3561.

⁶G. Kabalka, W. George, Z. Wu, Y. Ju, *Tetrhedron*, 2001, **57**, 1663.

⁷ T. Nishino, Y. Nishiyama, N. Sonoda, Bull. Chem. Soc. 2003, 76, 635.

⁸ C. Krug.J. F. Hartwig, J. Am. Chem. Soc., 2002, 124, 1674.

phenylmethanol was obtained as a white crystalline solid in 65% yield. M.p. 62-3 °C (ethyl acetatepetrol). ¹H NMR (400 MHz, CDCl₃) 2.30 (s, 3H), 5.72 (s, 1H), 7.08-7.34 (m, 9H).

II. Computational details

All species have been fully geometry optimized using the Jaguar 4.0 pseudospectral program package, except transition states for which the Gaussian 03 program package was used. Geometry optimization was carried out using the well established B3LYP hybrid density functional and the standard split valence polarized 6-31G* basis set (optimization of the model reaction, $BMe_3 + CH_2SMe_2$, at MP2/6-31G* level using the Gaussian 03 program package did not lead to significant structure change).

Frequency calculations have been carried out on every transition state structures using the Gaussian 03 program package in order to check the correct nature of the point on the potential energy surface.

For the model reaction (BMe₃ + CH₂SMe₂), single point calculations using MP2/6-311+G** (Gaussian 03) and B3LYP/6-311+G** (Jaguar 4.0) methods, and G3(MP2)//B3LYP calculations (Gaussian 03) were carried out. A comparison of the different methods is given in the figure below.



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No significant difference in structure was observed between B3LYP/6-31G* and MP2/6-31G* optimized structures. One can notice however a systematic energetic underestimation of the migration barrier obtained at B3LYP level compared to the value obtained using G3(MP2)//B3LYP method. Accordingly, energies reported in this study were obtained after single point calculations at the MP2/6-311+G** level of theory on every optimized B3LYP/6-31G* structures.

Several local minima and saddle points corresponding to each intermediate or transition state were frequently found. This is due to the possibility of multiple conformations of phenyl substituent and around the C1-B and C1-S bonds. We have systematically varied the relevant diehedral angles, and optimized several different local minima and saddle points (up to ten different calculations for each species). The date presented refer to the lowest energy form unless mentioned otherwise.

III. NBO analysis

NBO analysis has been performed on every transition structure at the B3LYP/6-31G* level of theory using Jaguar 5.0 pseudospectral program package.

	Ph Me ⁻ Me	H H H	Me Me Ph	H H	Me Me	H. H.		Ph H +	Me Me Ph	Ph H S	Me Me Me	Ph H S
barrier ^a	13.19 ko	cal/mol	19.27 ko	cal/mol	15.40 ko	cal/mol	14.72 ko	cal/mol	16.36 ko	cal/mol	12.47 k	cal/mol
						NPA Ch	arge (a.u.)					
	ate complex	TS	ate complex	TS	ate complex	TS	ate complex	TS	ate complex	TS	ate complex	TS
R	-0.3324	-0.1588	-0.2725	-0.1929	-0.2940	-0.1305	-0.3320	-0.2424	-0.2640	-0.3187	-0.2878	-0.2090
В	0.4377	0.4061	0.4319	0.3840	0.4482	0.4104	0.4547	0.4136	0.4470	0.4326	0.4610	0.4160
C^1	-0.8122	-0.5314	-0.8188	-0.5841	-0.8251	-0.5310	-0.5531	-0.2901	-0.5629	-0.2956	-0.5694	-0.2882
S	0.8959	0.4714	0.9055	0.6100	0.9024	0.4662	0.8615	0.5407	0.8766	0.4443	0.8827	0.5376
	Stabilization (kcal/mol)											
$\pi_{\rm Ph} \rightarrow \sigma^*_{\rm CS}$	1.23	18.00	-	-	-	-	0.82	7.10	-	-	-	-
$\sigma_{B-R} \rightarrow \sigma^*_{CS}$	5.34	32.88	3.75	17.98	4.37	40.28	4.69	16.34	3.97	53.57 ^b	3.66	15.90
	•		-		π_{n}	$\rightarrow \sigma^*$	5 38	20.92	4 88	10.39	5.02	21.45

^a obtained at MP2/6-311+G**//B3LYP/6-31G* level of theory ^b this point appears to be out of line compared to other values. This could be due to the need in that precise case to force the NBO analysis to consider a structure with a σ_{C2-S} bond for the TS.

	Ph Me ^{nut}	H H H H	Me Me ^{nnuC1}	H H H H	
barrier ^a	3.58 kc	al/mol	11.27 k	cal/mol	
	NPA Charge (a.u.)				
	ion	TS	ion	TS	
R	0.1060	0.2198	0.0948	0.2416	
C^1	-0.1149	-0.1081	-0.1206	-0.0152	
C^2	-0.0802	-0.0741	-0.0765	-0.1023	
0	-0.7068	-0.8122	-0.6898	-0.9282	
Stabilization (kcal/mol)					
$\pi_{\rm Ph} \rightarrow \sigma^*_{\rm CO}$	3.55	22.88	-	-	
$\sigma_{C1-R} \rightarrow \sigma^*_{CO}$	7.85	19.05	6.77	102.66	

Wagner-Meerwein rearrangement

^a obtained at MP2/6-311+G**//B3LYP/6-31G* level of theory

IV. Cartesian coordinates (in Å) and energies (a.u.) of the considered structures

A. Model reaction: BMe₃ + CH₂SMe₂ BMe₃ (B3LYP 6-31G*) E(B3LYP 6-31G*) = -144.60852 $E(B3LYP 6-311+G^{**}) = -144.65648$ $E(MP2 6-311+G^{**}) = -144.13564$ E(G3(MP2)) = -144.29865B 0.00468 .01980 .00000 C 1.58647 -.02570 .00000 H 2.07487 .95468 .00000 Н 1.95379 -.58774 .87204 Н 1.95379 -.58774 -.87204 C -.79653 1.38017 .00000 H -.18479 2.28865 .00000 H-1.46906 1.41791 -.87099 H-1.46906 1.41791 .87099 C -.79093 -1.35060 .00000 H-1.88220 -1.24964 .00000 Н -.50834 -1.95818 -.87272 H -.50834 -1.95818 .87272 BMe₃ (MP2 6-31G*) $E(MP2 \ 6-31G^*) = -144.01029$ $E(MP2 \ 6-311+G^{**}) = -144.13564$ B 0 00000 0 02841 0 00000

в 0.000	00 0.0	JZ841	0.00000
C 1.559	945 0.2	28099	0.00000
H 1.848	318 1.	33476	0.00000
H 2.018	363 -0 .	20106	0.87248
H 2.018	363 -0.	20106	-0.87248
C -1.051	182 1.	20271	0.00000
H -0.623	319 2.	20757	0.00000
H -1.712	205 1.	10502	-0.87171
H -1.712	205 1.	10502	0.87171
C -0.509	985 -1.	46873	0.00000
H -1.597	777 -1.	57648	0.00000
H -0.113	350 -2.	00285	-0.87280
H-0.113	350 -2.	00285	0.87280

CH₂SMe₂ (B3LYP 6-31G*)

E(B3LYP 6-31G*) = -517.228444 E(B3LYP 6-311+G**) = -517.29857 E(MP2 6-311+G**) = -516.33055 E(G3(MP2)) = -516.56482

H 2.31962 1.57603 -.02079 C 2.66753 2.60119 .02968 H 2.00559 3.41440 .28820 S 4.22434 3.02760 -.28623 C 4.80358 2.36410 -1.92800 H 5.86718 2.58221 -2.06899

Η	4.62969	1.28482	-1.97669
Η	4.20900	2.86905	-2.69164
С	5.43086	2.02333	.71637
Η	6.45757	2.26145	.42000
Η	5.26533	2.29517	1.76055
Н	5.23528	.95583	.57602

CH₂SMe₂ (MP2 6-31G*)

 $E(MP2 \ 6-31G^*) = -144.01029$ $E(MP2 \ 6-311+G^{**}) = -144.13564$

H -1.74224 -0.00864 1.38926 C -1.64397 -0.00706 0.31189 H -2.50168 -0.01865 -0.34024 S -0.18838 -0.00146 -0.43458 C 0.89407 -1.34575 0.18478 H 1.89463 -1.26557 -0.24947 H 0.95418 -1.29971 1.27580 H 0.42692 -2.28454 -0.11524 C 0.87754 1.35533 0.18408 H 1.87951 1.28558 -0.24866 H 0.40027 2.28837 -0.11785 H 0.93663 1.31144 1.27523

Ate complex (B3LYP 6-31G*)

E(B3LYP 6-31G*) = -661.86941 E(B3LYP 6-311+G**) = -661.98423 E(MP2 6-311+G**) = -660.52132 E(G3(MP2)) = -660.90290

B-1.84656 -1.86169 0.60192 C -0.15476 -1.92806 0.43155 H 0.39977 -1.54914 1.30048 C -2.15882 -0.26785 0.33873 H-3.23547 -0.07205 0.45405 H-1.64772 0.40679 1.04494 H-1.88804 0.06881 -0.67519 S 0.51204 -3.58504 0.06807 H 0.15060 -1.36550 -0.46023 C -2.56999 -2.79444 -0.54070 H -3.66007 -2.65133 -0.50912 H -2.25894 -2.54836 -1.56929 H-2.41766 -3.88372 -0.42445 C -2.30602 -2.26722 2.12915 H-3.34856 -1.95290 2.28700 H-2.29094 -3.34117 2.37735 H-1.71804 -1.75694 2.91079 C 0.03666 -4.60687 1.49372 H-1.03834 -4.76510 1.42362 H 0.57088 -5.55730 1.42114 H 0.27293 -4.08949 2.42591 C 2.31703 -3.49410 0.34347 H 2.75115 -4.49101 0.22808 Н 2.72755 -2.82676 -0.4175

Н 2.52650 -3.09585 1.3397

Ate complex (MP2 6-31G*)

E(MP2 6-31G*) = -660.24588 E(MP2 6-311+G**) = -660.522746

B -1.81746 -1.87094 0.61182
C -0.14070 -1.93211 0.45293
Н 0.40079 -1.57278 1.33801
C -2.15028 -0.29655 0.33741
H -3.22670 -0.10964 0.44943
H-1.64275 0.38243 1.03871
H -1.87890 0.02925 -0.67766
S 0.48305 -3.56249 0.06411
Н 0.16153 -1.33976 -0.41942
C -2.52902 -2.80870 -0.52124
H -3.61600 -2.65816 -0.49089
H -2.21156 -2.56392 -1.54578
H -2.38050 -3.89506 -0.39929
C -2.28870 -2.25845 2.12873
H -3.34286 -1.97713 2.25296
H -2 23622 -3 32002 2 40717
H -1 73348 -1 70131 2 89878
C 0 01100 -4 57833 1 46748
H -1 05884 -4 75819 1 38961
H 0 56033 -5 51993 1 40485
H 0.23094 -4 05561 2 39970
C 2 26839 -3 49234 0 32870
H 2 69813 -4 48569 0 18199
H 2 68183 -2 80397 -0 41015
H 2 48315 -3 12689 1 33522
11 2.10010 0.12000 1.000244

TS (B3LYP 6-31G*)

E(B3LYP 6-31G*) = -661.85734 E(B3LYP 6-311+G**) = -661.97203 E(MP2 6-311+G**) = -660.49676 E(G3(MP2)) = -660.881397

B -1.64350 0.03628 0.04089 C -0.38781 -0.82712 -0.30346 Н 0.00491 -0.85187 -1.32478 C -2.55242 -1.15168 -0.85380 Н -3.55482 -0.72697 -0.71266 Н -2.36344 -1.19969 -1.93169 Н -2.56778 -2.16914 -0.44862 S 1.66717 -0.06909 0.56259 Н -0.25881 -1.79156 0.19350 C -2.10269 0.01964 1.59809 Н -3.14109 0.36047 1.72158 Н -2.03922 -0.97459 2.06607 Н -1.48630 0.69532 2.21183 C -1.78584 1.47444 -0.70636 Н -2.83763 1.79184 -0.75491 Н -1.26077 2.28056 -0.17025

H -1.41087 1.48156 -1.74197 C 1.79947 1.53063 -0.29936 H 1.04352 2.18768 0.13254 H 2.79415 1.95635 -0.14288 H 1.59806 1.41294 -1.36741 C 2.93788 -1.02160 -0.33764 H 3.92471 -0.57273 -0.19257 H 2.94213 -2.03254 0.07699 H 2.70440 -1.06946 -1.40547

TS (MP2 6-31G*)

E(MP2 6-31G*) = -660.22459 E(MP2 6-311+G**) = -660.498554

B-1.33335-2.01444 0.895673 C -0.16888 -2.19622 -0.106194 H 0.71213 -1.55402 -0.034950 C -1.54134 -0.51431 0.049544 H -2.45041 -0.20767 0.579392 H-0.77580 0.24542 0.225807 H -1.78113 -0.55048 -1.016411 S 1.19746 -4.07664 0.127702 H -0.39791 -2.43607 -1.146193 C -2.65842 -2.91089 0.667888 H -3.54500 -2.39863 1.063851 H -2.85927 -3.11400 -0.393025 H -2.61864 -3.88120 1.177662 C -0.97746 -1.67568 2.435055 H -1.80994 -1.14802 2.918324 H-0.80188-2.57458 3.042326 H -0.09207 -1.03378 2.544688 C 0.04665 - 5.14023 1.016069 H -0.67459 -5.52494 0.293660 Н 0.58359 - 5.97468 1.472389 H-0.48307-4.56067 1.771832 C 2.17724 -3.46936 1.515227 H 2.65631 -4.30225 2.034043 H 2.95073 - 2.81223 1.113137 H 1.54218 -2.90706 2.201863

BMe₂Et (B3LYP 6-31G*)

E(B3LYP 6-31G*) = -183.920035 E(B3LYP 6-311+G**) = -183.978090 E(MP2 6-311+G**) = -183.328722 E(G3(MP2)) = -183.532019

В	-1.34093	-1.82764	.83056
С	.16235 -	2.26855	.60420
Η	.71085 -	2.10830	1.54746
С	-2.54090	-2.73969	.35586
Η	-2.41876	-3.78214	.68074
Η	-3.53345	-2.39220	.66397
Η	-2.53545	-2.78413	74546
С	-1.63820	43739	1.52065
Η	-2.38408	.13991	.95600
Η	-2.10720	62556	2.49996
Η	76155	.19818	1.69176

C .44689 -3.67398 .05675 H 1.51926 -3.84962 -.09132 H .08062 -4.44996 .73996 H -.04832 -3.83628 -.90809 H .59810 -1.50844 -.06952

BMe2Et (MP2 6-31G*)

 $E(MP2 \ 6-31G^*) = -183.17320$ $E(MP2 \ 6-311+G^{**}) = -183.32878$

B 0.56177 -0.00185 -0.00292 C -0.78617 -0.82500 -0.04322 H -0.73759 -1.58496 0.75155 C 0.55316 1.57579 -0.00750 H-0.12162 1.97496 0.75850 Н 1.53829 2.03215 0.11977 H 0.14219 1.92432 -0.96528 C 1.93605 -0.77692 0.02007 Н 2.61390 -0.39328 -0.75195 Н 2.43860 -0.57226 0.97544 H 1.85310 -1.86139 -0.09231 C -2.10815 -0.05992 0.03429 H -2.97040 -0.73070 -0.03264 H-2.19247 0.49013 0.97633 H -2.19303 0.66861 -0.77737 H -0.74917 -1.42195 -0.96921

SMe₂ (B3LYP 6-31G*)

E(B3LYP 6-31G*) = -478.01395 E(B3LYP 6-311+G**) = -478.06684 E(MP2 6-311+G**) = -477.22982 E(G3(MP2)) = -477.40917

 S
 .00000
 0.00000
 -.59219

 C
 .00000
 1.39545
 .58519

 H
 .00000
 2.31459
 -.00670

 H
 -.89456
 1.38350
 1.21609

 H
 .89456
 1.38350
 1.21609

 C
 .00000
 -1.39545
 .58519

 H
 .00000
 -2.31459
 -.00670

 H
 .89456
 -1.38350
 1.21609

 H
 .90000
 -2.31459
 -.00670

 H
 .89456
 -1.38350
 1.21609

 H
 -.89456
 -1.38350
 1.21609

SMe2 (MP2 6-31G*)

E(MP2 6-31G*) = -477.12111 E(MP2 6-311+G**) = -477.23035

S 0.00000 0.00000 0.66470 C 0.00000 1.36879 -0.51421 H 0.00000 2.29842 0.05800 H 0.89199 1.34818 -1.14516 H -0.89199 1.34818 -1.14516 C 0.00000 -1.36879 -0.51421 H 0.00000 -2.29842 0.05800 H -0.89199 -1.34818 -1.14516 H 0.89199 -1.34818 -1.14516

B. BMe₃ + PhCHSMe₂

PhCHSMe₂

E(B3LYP 6-31G*) = -748.29197 E(MP2 6-311+G**) = -746.75294

С	2.60933	.08803	28422
С	1.98349	1.31912	.02277
С	.60275	1.26389	.33855
С	09801	.06402	.34618
С	.54181	-1.14144	.03907
С	1.90099	-1.11216	27495
Η	3.66615	.05998	53344
С	2.63943	2.60783	.03705
Η	.08489	2.18962	.58057
Η	-1.15708	.06882	.59471
Η	00671	-2.07902	.04497
Η	2.42229	-2.03580	51745
Η	2.06802	3.49413	.28502
S	4.22639	3.00478	28970
С	4.81055	2.37892	-1.94479
Η	5.86038	2.66057	-2.06877
Η	4.69244	1.29768	-2.02998
Η	4.19533	2.88133	-2.69331
С	5.44671	2.03509	.73139
Η	6.45690	2.34184	.44454
Η	5.25291	2.30414	1.77125
Η	5.32081	.95995	.59493

Ate complex

E(B3LYP 6-31G*) = -892.91730
$E(MP2 \ 6-311+G^{**}) = =-890.94140$
B -1.82879 -1.83520 .56598
C11210 -1.87682 .39738
Н .34283 -1.61426 1.36188
C -2.2333926141 .32409
Н -3.3112113744 .50813
Н -1.72344 .43678 1.00754
Н -2.04461 .1010369636
S .52070 -3.60098 .06622
C 1.66477 .62263 -2.66445
C 1.92400 .85720 -1.31369
C 1.36794 .0270333905
C .52243 -1.0378268916
C .26558 -1.25851 -2.05391
C .8358644096 -3.02871
H 2.10103 1.26347 -3.42586
H 2.56282 1.68451 -1.01518
H 1.57384 .21835 .71192
Н40009 -2.06310 -2.35043
Н .6214762878 -4.07758
C -2.58981 -2.8061852120
Н -3.67559 -2.7416835735
Н -2.42941 -2.52308 -1.57285
Н -2.35212 -3.8840845271

С	-2.21932 -2.23966	2.11768
Η	-3.25769 -1.93172	2.31022
Η	-2.18706 -3.30884	2.37962
Η	-1.60903 -1.71330	2.87156
С	01656 -4.62398	1.46998
Η	-1.08296 -4.80361	1.34631
Η	.53842 -5.56450	1.42429
Η	.16513 -4.10405	2.41238
С	2.31287 -3.52408	.41233
Η	2.74970 -4.51343	.25479
Η	2.74187 -2.80932	29234
Η	2.48374 -3.18558	1.43770

TS

E(B3LYP 6-31G*) = -892.91327 E(MP2 6-311+G**) = -890.92152

В	-1.48692	-1.53141	0.00109
С	-0.49890	-0.34625	-0.46035
Η	-0.75880	0.05714	-1.44670
С	-0.95967	-2.68207	-1.11490
Η	-1.60850	-3.55533	-0.95713
Η	-1.06873	-2.39038	-2.16913
Η	0.07602	-3.01174	-0.96383
S	-1.23724	1.60587	0.59121
С	-1.32157	-2.10098	1.52515
Η	-2.08648	-2.86729	1.71852
Η	-0.35567	-2.58391	1.73368
Η	-1.46973	-1.32782	2.29743
С	-3.05025	-1.15135	-0.32897
Η	-3.68386	-2.04708	-0.26113
Η	-3.48186	-0.43259	0.38880
Η	-3.21037	-0.73971	-1.33977
С	-2.36492	2.29366	-0.66619
Η	-3.22084	1.62023	-0.72545
Η	-2.69482	3.29062	-0.36167
Η	-1.87117	2.34282	-1.64101
С	0.14527	2.78525	0.44489
Η	-0.19670	3.78898	0.71182
Η	0.91907	2.46609	1.14569
Η	0.55897	2.77987	-0.56736
С	3.76425	-0.15273	0.06711
С	3.19054	0.14532	-1.16990
С	1.80473	0.10560	-1.31936
С	0.96476	-0.26031	-0.24943
С	1.56304	-0.56177	0.99038
С	2.94575	-0.49962	1.14656
Η	4.84332	-0.11598	0.19105
Η	3.82008	0.41288	-2.01436
Η	1.36020	0.33673	-2.28498
Η	0.93333	-0.84152	1.82653
Η	3.38955	-0.73247	2.11087

PhCHB(Me)₂Me

E(B3LYP 6-31G*) = -414.96820 E(MP2 6-311+G**) = -413.74332

С	.18022	-2.23090 .48364
Η	.79339	-1.82042 1.29801
С	-2.55543	-2.62545 .11952
Н	-2.32546	-3.6207727456
Н	-3.37020	-2.72321 .85039
Н	-2.98168	-2.0340570588
С	-1.64937	53507 1.60285
Н	-2.61182	05918 1.37949
Н	-1.70619	86671 2.65405
Η	86057	.22565 1.56609
С	.46644	-3.74016 .38200
Н	1.52768	-3.93102 .18148
Η	.20464	-4.24687 1.31883
Η	11150	-4.2191341463
С	1.21780	.02781 -3.09082
С	1.63616	.46173 -1.83154
С	1.30776	2723869181
С	.55843	-1.4567177871
С	.13914	-1.87544 -2.05280
С	.46762	-1.14462 -3.19514
Η	1.47471	.59601 -3.98081
Η	2.22392	1.37120 -1.73511
Η	1.64547	.07012 .28391
Η	44824	-2.78380 -2.15571
Н	.13564	-1.49379 -4.16970

B -1 34346 -1 81679 73233

C. $BPhMe_2 + CH_2SMe_2$

BPhMe₂

 $E(B3LYP \ 6-31G^*) = -336.34926$ $E(MP2 \ 6-311+G^{**}) = -335.35198$

B -1.34339 -1.86315 .79336 .15775 -2.32405 .56013 C .71452 -2.36635 1.50736 Η C -2.48788 -2.83445 .29957 Н -2.36294 -3.81523 .78346 Н -3.52317 -2.51459 .45331 Н -2.35331 -3.03363 -.77456 Н .24472 -3.30536 .08065 Н .69989 -1.60071 -.06647 C -2.15196 1.99814 2.74668 C -3.21316 1.17631 2.36061 C -2.95441 -.04580 1.74110 C -1.63940 -.48912 1.48841 C -.59118 .36618 1.89273 C -.83607 1.59059 2.51092 Н -2.34870 2.95221 3.22936 H -4.23801 1.48963 2.54243 Н -3.79306 -.67068 1.44804 Н .43805 .06434 1.71784 Н -.00695 2.22699 2.80960

Ate complex (Me anti)

E(B3LYP 6-31G*) = -853.61505 E(MP2 6-311+G**) = -851.75073

В-	1.82945	-1.79131	.52717
С	14264	-1.83885	.39936
Η	.29744	-1.06733	1.04385
С-	2.18477	21420	.21753
Н -	3.27273	06086	.25169
Н -	1.74633	.47171	.95961
Н -	1.85559	.13319	77393
S	.60800 -	3.36708	1.06271
Η	.24768	-1.71891	61890
С-	2.26958	-2.19555	2.05446
Н -	3.34572	-2.03379	2.21091
Η・	2.08118	-3.24512	2.34207
Н -	1.75717	-1.57491	2.80613
С	2.35342	-3.33531	.52435
Н	2.83868	-2.50234	1.03784
Н	2.83471	-4.27372	.81253
Н	2.40952	-3.19102	55778
С	01377	-4.70860	.00248
Н	.53951	-5.61439	.26291
Н.	1.07726	-4.82695	.21110
Н	.11420	-4.44930 -	-1.05031
С-	3.62634	-4.40672	-2.66194
С-	4.01511	-4.55290	-1.32979
С-	3.45518	-3.73768	33911
С-	2.49215	-2.74984	62517
С-	2.13864	-2.62144	-1.98534
С-	2.68380	-3.42693	-2.98728
Н.	4.05668	-5.03733	-3.43638
Н.	4.75840	-5.30098	-1.06005
Н -	3.78372	-3.87126	.68960
Н -	1.42211	-1.85243	-2.27579
H ·	2.38094	-3.28750	-4.02363

Ate complex (Ph anti)

E(B3LYP 6-31G*) = -853.60925 E(MP2 6-311+G**) = -851.74264

B -2.04851 -2.04793 .66002 C -.35842 -1.99486 .53289 Н .16919 -1.61602 1.41844 C -2.55678 -3.01455 1.88821 Н -3.61348 -2.81363 2.11390 H -2.52031 -4.08810 1.63508 H -2.01604 -2.88871 2.84093 S .42555 -3.61103 .14473 H -.07820 -1.37950 -.33132 C -2.67604 -2.55343 -.76884 Н -3.77313 -2.48788 -.75661 Н -2.34374 -1.97114 -1.64262 Н -2.43899 -3.60726 -.99301 С .48093 -4.44721 1.75922 Н -.54959 -4.63948 2.05588 Н 1.02992 -5.38434 1.64202 Η .96559 -3.79975 2.49444 С 2.20937 -3.27919 -.08220 Н 2.74361 -4.22952 -.16687 Н 2.31847 -2.71549 -1.01134

Н 2.	59332	-2.69329 .75701
C -2.	83512	2.30833 1.45328
C -2.	89246	1.37948 2.49324
C -2.	65380	.02578 2.24068
C -2.	35561	46025 .95328
C -2.	30309	.5061007129
C -2.	53802	1.86232 .16369
Н-3.	02252	3.36286 1.64311
Н-3.	12707	1.70903 3.50399
Н -2.	71240	67382 3.07184
Н -2.	08459	.18872 -1.09014
Н -2.	49494	2.5720966070

TS Me migration

E(B3LYP 6-31G*) = -853.60452 E(MP2 6-311+G**) = -851.71999

В	-1.31652	-1.88821	0.82768
С	-0.12143	-2.11912	-0.13980
Η	0.76452	-1.48575	-0.06392
С	-1.58383	-0.44116	-0.13454
Η	-2.47161	-0.08097	0.39920
Η	-0.81319	0.33238	-0.06347
Η	-1.86317	-0.58199	-1.18237
S	1.25345	-4.08269	0.22948
Η	-0.28435	-2.43802	-1.16973
С	-0.95192	-1.42879	2.34266
Η	-1.77188	-0.86674	2.81143
Η	-0.75041	-2.27796	3.01357
Η	-0.06758	-0.77616	2.38233
С	0.00104	-5.22164	0.90621
Η	-0.59769	-5.57869	0.06590
Η	0.49705	-6.06735	1.39006
Η	-0.65617	-4.69785	1.60216
С	2.10484	-3.57411	1.76084
Η	2.54707	-4.44672	2.24924
Η	2.90228	-2.88363	1.47528
Η	1.40898	-3.06963	2.43440
С	-4.82951	-4.61889	0.36926
С	-4.46852	-4.12915	1.62417
С	-3.37036	-3.27373	1.75543
С	-2.59043	-2.87456	0.65378
С	-2.98908	-3.38341	-0.59908
С	-4.08230	-4.23787	-0.74800
Η	-5.68285	-5.28402	0.26049
Η	-5.04284	-4.41120	2.50431
Η	-3.11566	-2.90614	2.74684
Η	-2.43609	-3.09852	-1.49373
Η	-4.35459	-4.60466	-1.73568

TS Ph migration

E(B3LYP 6-31G*) = -853.60030 E(MP2 6-311+G**) = -851.72160 B -1.07578 -1.60204 0.63931 C 0.10183 -1.53613 -0.42486

Н 0.92980 -0.83708 -0.28270

S 1.65195 -3.31232 -0.51512
C 0.67436 -4.71601 0.11099
Н -0.02965 -4.99614 -0.67516
Н 1.33657 -5.55899 0.32590
Н 0.11713 -4.42006 1.00057
C 2.64330 -2.91504 0.96333
Н 3.26266 -3.77297 1.23809
Н 3.29242 -2.07579 0.70205
Н 1.99023 -2.62989 1.79065
C -2.28388 2.50458 -0.84342
C -1.35448 2.37621 0.19122
C -0.94185 1.10984 0.60591
C -1.46124 -0.06718 0.03017
C -2.37177 0.09533 -1.03315
C -2.79029 1.35690 -1.45713
Н -2.60680 3.48967 -1.17135
Н -0.95061 3.26365 0.67351
Н -0.21824 1.03315 1.41345
Н -2.78024 -0.78506 -1.52206
Н -3.51062 1.44556 -2.26736
Н -0.12340 -1.63940 -1.48851
C -2.26176 -2.68135 0.33524
Н -3.21948 -2.34439 0.75581
Н -2.44201 -2.88035 -0.73217
Н -2.05910 -3.65751 0.79963
C -0.63384 -1.55711 2.20855
Н -1.46240 -1.22106 2.84732
Н -0.36106 -2.55860 2.57840
Н 0.21998 -0.90073 2.43519

EtBMePh

E(B3LYP 6-31G*) = -375.65890 E(MP2 6-311+G**) = -374.54363

В	-1.33732	-1.86972	.80676
С	.17352 ·	-2.32954	.60600
Η	.69865	-2.24375	1.57090
С	-2.49382	-2.81640	.29279
Η	-2.40836	-3.80360	.77098
Η	-3.52036	-2.46508	.43320
Η	-2.35462	-3.01545	78051
С	.42976 ·	-3.72282	.01161
Η	1.50168	-3.92969	09226
Η	.00449	-4.51289	.64211
Η	02112	-3.82715	98225
Η	.66600	-1.57111	02681
С	-2.15623	2.00622	2.73846
С	-3.21288	1.16922	2.37434
С	-2.94895	05472	1.76002
С	-1.63321	48826	1.49397
С	58967	.38202	1.87885
С	83972	1.60998	2.48769
Η	-2.35656	2.96242	3.21541
Η	-4.23887	1.47182	2.56777
Η	-3.78598	68946	1.48434
Η	.44125	.09189	1.69518
Н	01319	2.25770	2.76865

PhCH₂BMe₂

E(B3LYP 6-31G*) = -375.65831
$E(MP2 \ 6-311+G^{**}) = -374.54692$

В	-1.30770	-1.90185	.86971
С	.20586	-2.21680	.48308
Н	.81805	-1.97590	1.36737
Η	.49985	-1.45341	25653
С	1.07805	-6.20331	97289
С	.72848	-5.18786	-1.86485
С	.46223	-3.90108	-1.39485
С	.53762	-3.59770	02745
С	.89221	-4.62836	.85532
С	1.15934	-5.91693	.39100
Η	1.28837	-7.20544	-1.33688
Η	.66718	-5.39552	-2.93033
Η	.19820	-3.11558	-2.10082
Η	.96514	-4.41536	1.92038
Η	1.43627	-6.69699	1.09591
С	-2.43357	-2.98967	.69472
Η	-3.43792	-2.66834	.99124
Η	-2.47009	-3.32572	35203
Η	-2.17456	-3.89546	1.26221
С	-1.66012	46855	1.43137
Η	-2.44727	00650	.81704
Η	-2.10572	56553	2.43312
Η	82111	.23337	1.49893

D. BPhMe₂ + PhCHSMe₂

Ate complex (Me anti)

E(B3LYP 6-31G*) = -1084.66200 E(MP2 6-311+G**) = -1082.17197

В	-1.85251	-1.85078	.41480
С	19212	-2.20468	.16687
Η	.31662	-2.24486	1.13756
С	-2.75931	-2.22390	89951
Η	-3.81203	-1.96532	71634
Η	-2.46758	-1.65187	-1.79111
Н	-2.76213	-3.28781	-1.19703
S	.05737	-3.92906	51951
С	-1.90079	23376	.71015
Η	-2.92856	.05662	.97049
Η	-1.26813	.08781	1.55200
Н	-1.60793	.37127	15950
С	58882	-5.08333	.73046
Η	-1.67590	-5.01419	.71987
Η	25869	-6.08525	.44321
Η	22450	-4.81193	1.72314
С	1.84611	-4.24250	32567
Η	2.06530	-5.25760	66595
Η	2.36956	-3.51672	95039
Η	2.13743	-4.11563	.72032
С	-3.18603	-3.99049	4.14814
С	-3.91014	-4.11986	2.96244
С	-3.48872	-3.45969	1.80304

С	-2.33695 -2.64897 1.76872	
С	-1.64385 -2.52634 2.99133	
С	-2.04722 -3.18074 4.15741	
Η	-3.50801 -4.50130 5.05263	
Η	-4.80875 -4.73395 2.93919	
Η	-4.08020 -3.57662 .89734	
Η	76680 -1.88017 3.04379	
Η	-1.48008 -3.05118 5.07772	
С	2.04048 .48884 -2.39004	
С	2.29237 .42850 -1.01861	
С	1.584664658121474	
С	.59305 -1.3002675589	
С	.34841 -1.22684 -2.13847	
С	1.0688534602 -2.94513	
Η	2.59442 1.17978 -3.01964	
Η	3.04421 1.0740657196	
Η	1.7876650579 .85319	
Η	41961 -1.85085 -2.58426	
Η	.8611930562 -4.01118	

Ate complex (Ph anti)

 $E(B3LYP 6-31G^*) = -1084.65629$ $E(MP2 6-311+G^{**}) = -1082.16617$

B -2.14036 -1.92920	.55709
C44385 -1.97137	.32202
Н .06673 -1.67049	1.24544
C -2.59854 -2.94308	1.77496
H -3.65323 -2.75099	2.01403
H -2.55624 -4.01241	1.50338
H -2.05214 -2.81988	2.72423
S .16193 -3.75534	.10216
C -2.95331 -2.33198	80424
H-4.03590 -2.30677	61267
H-2.77996 -1.67507	-1.66751
H -2.73540 -3.35969	-1.14121
C .36511 -4.35598	1.81007
Н62618 -4.42621	2.25386
Н .83576 -5.34112	1.76643
Н .98557 -3.65663	2.37653
$C \ 1.92844 \ \textbf{-3.65092}$	36242
H 2.34602 -4.66107	34890
H 1.97818 -3.23660	-1.36982
Н 2.46399 -2.99931	.33259
C -2.74553 2.32799	1.86967
C -2.07571 1.41748	2.68985
C -1.89471 .09931	2.26640
C -2.3632136976	1.02234
C -3.03092 .57501	.22197
C -3.22500 1.89697	.63190
H-2.89459 3.35586	2.19281
H-1.70111 1.73337	3.66212
Н -1.3823359067	2.93757
Н -3.41678 .26996	74761
H-3.75287 2.59259	01806
C 1.12890 .42287	-2.92440
C 1.33759 .79159	-1.59479
C .8465600404	55942

C .11413 -1.1705982890 C10245 -1.52259 -2.17290 C .4095073990 -3.20823 H 1.51813 1.03721 -3.73176 H 1.88725 1.69914 -1.35935 H 1.00589 .29571 .47318 H68951 -2.40494 -2.41134 H .23218 -1.03299 -4.23979	
TS Me migration	
E(B3LYP 6-31G*) = -1084.65550 E(MP2 6-311+G**) = -1082.14588	
B 0.11520 1.58710 0.20168 C 1.22592 1.20327 -0.87709 C -0.32052 3.06042 -0.52953 H -1.18835 3.40590 0.04774 H 0.46823 3.81588 -0.44124 H -0.62767 3.01721 -1.58168 S 1.86471 -1.15091 -0.33996 H 0.80070 0.84860 -1.82165 C 0.58950 1.88380 1.73631 H -0.28409 2.08568 2.37176 H 1.2528 1.04498 2.20767 H 1.23622 2.76715 1.82914 C 3.41981 -1.40647 -1.25756 H 4.18899 -0.80453 -0.76924 H 3.70731 -2.46106 -1.21915 H 3.32308 -1.08223 -2.29786 C 0.73752 -2.17785 -1.34243 H 1.08741 -3.21434 -1.34005 H 0.68095 -1.80560 -2.36917 C -3.17704 -1.45243 -0.13040 C -2.42604 -1.34165 1.04107 C -1.42026 -0.37452 1.14312 C -1.11539 0.51260 0.09182 C -1.89993 0.37547 -1.07167 C -2.91051 -0.58117 -1.18975 H -3.96426 -2.19815 -0.21377 H -2.62601 -2.00530 1.88020 H -0.85777 -0.30802 2.07130 H -1.72303 1.04257 -1.91427 H -3.49456 -0.64486 -2.10604 C 5.07657 2.94994 -1.59115 C 4.26533 2.56584 -2.66113 C 3.02624 1.97915 -2.41398 C 2.55776 1.78758 -1.09733 C 3.39107 2.18491 -0.03104 C 4.63839 2.75207 -0.27777 H 6.04728 3.40117 -1.77871 H 4.60018 2.71926 -3.68341 H 2.39495 1.67996 -3.24769 H 3.05210 2.03511 0.98648	
11 5.27019 5.04820 0.55520	

TS Ph migration

B	-1.03288	-1.86932	.99685
С	.39726	-2.24129	.36436
ч	1 1 2 2 7 0	2 12048	1 17183

С	.39726	-2.24129	.36436
Η	1.13379	-2.12048	1.17183
С	.73714	-6.37905	93444
С	.56247	-5.37452	-1.88548
С	.46107	-4.03637	-1.49502
С	.52983	-3.67292	14400
С	.71212	-4.69625	.80141

E(B3LYP 6-31G*) = -606.70272 E(MP2 6-311+G**) = -604.96171

$E(B3LYP \ 6-31G^*) = -1084.65264$
$E(MP2 6-311+G^{**}) = -1082.14261$

B -0.06638	0.53105	0.73292
C -0.53469	-0.14495	-0.62619
H -0.43496	-1.22458	-0.76356
S -2.89064	-0.16762	-0.74620
C -3 31434	-1 67789	0 17986
H -3 14849	-1 46574	1 23799
H -4 36663	-1.40574	0.01747
Н 2 66701	2 50002	0.01747
$C_{2,20202}$	-2.30093	2 42400
C -3.30293	1 02002	-2.42477
H -4.33839	-1.02983	-2.4/323
H -3.12390	0.083/0	-3.10920
H -2.00030	-1.592/1	-2.69956
C 4.165/9	-0.45186	-0.54961
C 3.41918	-1.37950	0.18033
C 2.09708	-1.09644	0.52364
C 1.48550	0.12664	0.18206
C 2.25330	1.02782	-0.58246
C 3.57584	0.75453	-0.93274
Н 5.19596	-0.66912	-0.82124
H 3.86761	-2.32401	0.48047
Н 1.53470	-1.83234	1.09266
H 1.81404	1.97372	-0.88797
Н 4.14659	1.48182	-1.50589
C -0.26130	2.14814	0.83767
Н 0.49525	2.59495	1.49771
H -0.19487	2.68404	-0.12141
H -1.23425	2.41826	1.27358
C -0.46545	-0.25506	2.10501
Н 0.19255	0.03351	2.93640
H -1.48637	-0.00028	2.43209
H -0.42564	-1.35315	2.04317
C 0.13415	1.89032	-4.36488
C 0 78669	0 67498	-4 10032
C 0 56868	0.01159	-2.88171
C -0 30186	0.56354	-1 92766
C -0 95441	1 77884	-2 19214
C -0 73639	2 44227	-3 41083
H 0 30542	2.11227	-5 32224
H 1 47061	0 24144	-4 84995
H 1 08132	-0.9/316	-2 67300
H_1 63876	2 21225	-2.07390
Н 1 24002	2.21233	3 61965
11-1.24703	5.57/05	-3.01003

Ph₂CHBMe₂

C .81414 -6.03157 .41631
Н .81821 -7.41881 -1.23997
Н .50985 -5.62782 -2.94156
Н .34268 -3.27058 -2.25524
Н .78267 -4.43725 1.85687
Н .95839 -6.80075 1.17108
C -2.33270 -2.67439 .61241
Н -3.26699 -2.12746 .78395
H -2.32701 -3.0782740687
H -2.36043 -3.56073 1.26875
C -1.1196569825 2.05211
H-1.85726 .05322 1.73730
H -1.50543 -1.09569 3.00406
Н1744018418 2.25859
C 1.23765 .95957 -2.46002
C 2.14857 .64651 -1.44884
C 1.875343876155384
C .69207 -1.1362864700
C2177380415 -1.66593
C .05333 .22982 -2.56449
H 1.44922 1.76461 -3.15863
Н 3.07550 1.20700 -1.35719
Н 2.5926262682 .22787
H -1.14649 -1.36065 -1.76275
Н66607 .46601 -3.34435

PhCH(Me)BMePh

E(B3LYP 6-31G*) = -606.70719 E(MP2 6-311+G**) = -604.96072

B -1.01034 -2.32144 1.75886 C .52632 -2.70773 1.52371 Н .62356 -3.48762 .75851 2.22102 .90316 -.21679 С C 2.44793 .61002 1.12810 C 1.91735 -.54880 1.69901 C 1.15187 -1.44418 .93670 .92950 -1.13208 -.41618 С С 1.45789 .02392 -.98822 Н 2.63598 1.80448 -.66000 Н 3.04451 1.28321 1.73886 Н 2.11570 -.76011 2.74539 Η .33773 -1.81133 -1.02611 Н 1.27512 .23689 -2.03853 C -1.38025 -1.34095 2.94831 Н -1.99634 -.49835 2.60764 Н -1.99085 -1.87182 3.69449 Н -.51470 -.92090 3.47007 C 1.22286 -3.22786 2.79553 Н 2.30067 -3.35667 2.63518 H 1.09183 -2.55649 3.65052 Н .80893 -4.20079 3.08427 C -4.24390 -3.88674 -.79527 C -4.51299 -2.91589 .17308 C -3.48008 -2.42710 .97010 C -2.14930 -2.87994 .83247 C -1.91539 -3.86224 -.15324 C -2.94067 -4.36169 -.95596

H -5.04617 -4.27172 -1.41973

TS Ph migration

E(B3LYP 6-31G*) = -465.02559 E(MP2 6-311+G**) = -463.76112

 $\begin{array}{cccccc} C & -1.03556 & -1.86734 & 0.38669 \\ C & 0.30603 & -2.01197 & -0.20728 \\ H & 1.12861 & -1.45703 & 0.23397 \\ O & 1.01511 & -3.71254 & 0.36289 \\ H & 0.39139 & -2.20687 & -1.27239 \\ H & 1.98977 & -3.74948 & 0.33196 \\ H & 0.69347 & -4.45025 & -0.18762 \\ C & -1.44238 & 2.08376 & -1.45571 \\ C & -0.86763 & 1.95865 & -0.18845 \\ C & -0.71715 & 0.70228 & 0.39050 \\ C & -1.16544 & -0.45620 & -0.28004 \\ C & -1.71532 & -0.31506 & -1.57265 \\ C & -1.86364 & 0.94406 & -2.14573 \\ \end{array}$

Η	-1.55993	3.06533	-1.90476
Η	-0.53809	2.84083	0.35145
Η	-0.27717	0.63249	1.37879
Η	-2.05898	-1.18258	-2.12471
Η	-2.31050	1.03587	-3.13076
С	-2.06134	-2.89027	-0.12810
Η	-3.07660	-2.54925	0.09293
Н	-1.99067	-3.06980	-1.20476
Η	-1.91577	-3.84748	0.38350
С	-1.04042	-1.83331	1.92277
Η	-1.99649	-1.44550	2.28534
Η	-0.91075	-2.84690	2.31224
Η	-0.23957	-1.21818	2.34074

Me₃CCH₂OH₂⁺

E(B3LYP 6-31G*) = -273.29431 E(MP2 6-311+G**) = -272.55230

С	41559	.32650	04239
С	.40393	52728	.90901
Η	1.12544	.01861	1.51737
Η	17133	-1.22465	1.51855
С	-1.33285	55507	91340
Η	77191	-1.18857	-1.61671
Η	-1.98065	-1.19603	30457
Η	-1.97962	.07523	-1.53035
С	.50328	1.20346	91644
Η	1.11016	.61502	-1.62102
Η	09959	1.87608	-1.53318
Η	1.17325	1.82351	31002
С	-1.27022	1.22065	.88974
Η	-1.89770	1.87510	.27712
Η	-1.93233	.62670	1.52881
Η	64711	1.85721	1.52701
0	1.35373	-1.51805	.14011
Η	.83679	-2.12995	42891
Η	1.97774	-1.02694	43791

TS Me migration

E(B3LYP 6-31G*) = -273.27352 E(MP2 6-311+G**) = -272.53432

С	-1.33074	-1.98416	0.92549
С	-0.28936	-2.06207	-0.01112
Η	0.65938	-1.56474	0.15873
С	-1.54446	-0.52785	0.00061
Η	-2.52534	-0.32225	0.44023
Η	-0.84865	0.26319	0.27304
Η	-1.68782	-0.60856	-1.07842
0	0.80510	-4.03279	0.57670
Η	-0.44042	-2.52356	-0.98071
С	-2.53160	-2.89667	0.74307
Η	-3.40833	-2.50290	1.26481
Η	-2.78609	-3.04578	-0.31071
Η	-2.28881	-3.87179	1.17807
С	-1.00333	-1.60092	2.35820

H-1.86796-1.149952.85279H-0.74184-2.514922.90182H-0.15780-0.909902.42427H0.80323-4.80959-0.00638H1.69304-4.019340.97004

F. Energy profiles



Torsional rotation around the B-C(1) profile, Me₂PhBCH₂SMe₂ (B3LYP 6-31G*)

Torsional rotation around the B-C(1) profile, Me₂PhBCH₂SMe₂ (B3LYP 6-31G*)

