

Supporting Information

A Doubly Folded Spacer in a Self-Assembled Hybrid Material

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Experimental

1: tin tetrachloride (0.9 ml, 7.65 mmol) was added to a solution of 4,4'-bis(5-(tricyclohexylstannyl)pentyloxymethyl)biphenyl (3 g, 2.82 mmol) in dry toluene (100 ml) under a nitrogen atmosphere in a Schlenk tube. After stirring for 72h at room temperature, the liquid phase was removed and white precipitated obtained was taken up in dry acetonitrile. The solution was extracted with pentane (5x50ml) and evaporated. The obtained solid was dissolved in dry THF (100 ml). The solution was subsequently evaporated to give pure **1**. Crystals suitable for X-ray analysis were grown from a toluene solution.

Yield: 2.09 g, 96%; m.p. 159°C; ^1H NMR (300 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$): δ = 1.76 (m, 4H), 2.19 (m, 4H, $^1J_{\text{Sn-H}}$ = 314 Hz), 2.58 (m, 4H, $^2J_{\text{Sn-H}}$ = 92 Hz), 3.78 (t, 4H), 4.78 (s, 4H), 7.44 (d, 4H), 7.63 (d, 4H); ^{13}C NMR (62.9 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$): 24.7 ($^2J_{\text{Sn-C}}$ = 75 Hz), 29.9 ($^3J_{\text{Sn-C}}$ = 34 Hz), 38.9 ($^1J_{\text{Sn-C}}$ = 801 Hz), 72, 72.6, 127.1, 128.7, 134.7, 140.3; ^{119}Sn NMR (74.6 MHz, $\text{C}_2\text{D}_2\text{Cl}_4$, 25°C): -142.2; ^{117}Sn CPMAS NMR (89.13 MHz): δ = -163; elemental analysis calc (%) for $\text{C}_{22}\text{H}_{28}\text{Cl}_6\text{O}_2\text{Sn}_2$: C 34.12, H 3.64, Cl 27.46, O 4.13, Sn 30.65; found: C 33.72, H 3.63, Cl 27.14, O 4.57, Sn 30.97.

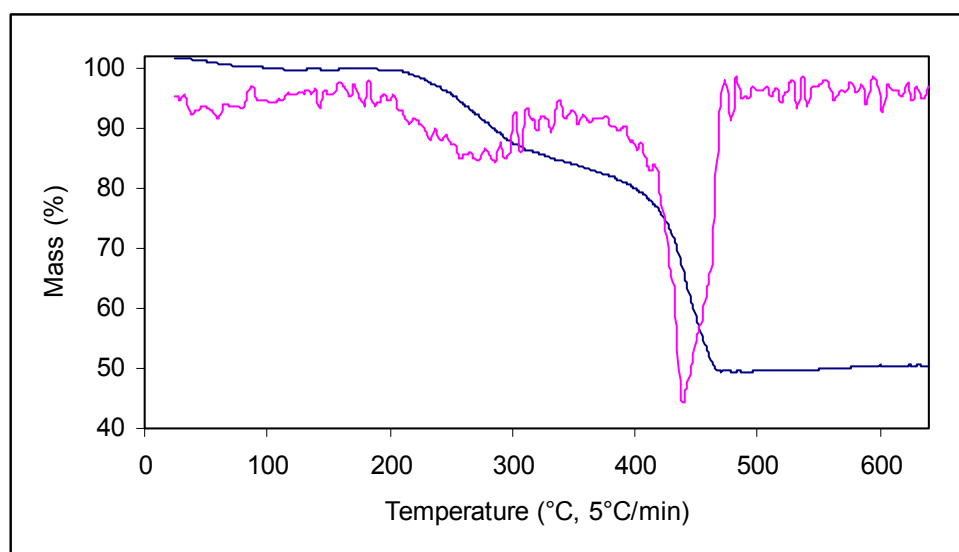
2: a 2.5M solution of BuLi (8 ml, 20 mmol) was added to a solution of propyne (5ml, 87.5 mmol) in dry toluene (50 ml) in a 500-ml 3-necked flask under nitrogen at -78°C. After stirring for 15 min, a solution of **1** (2 g, 2.58 mmol) in 200 ml of dry toluene was added at -78°C. After stirring for 15 h at room temperature, the suspension was filtered over dry MgSO_4 , the solvent was evaporated under vacuum to give **2** as an oil.

Yield: 1.29 g, 63%; ^1H NMR (300 MHz, CDCl_3): δ = 1.26 (t, 4H, $^2J_{\text{Sn-H}} = 77$ Hz), 1.76 (m, 8H), 1.90 (s, 18H, $^4J_{\text{Sn-H}} = 15$ Hz), 3.53 (m, 4H) 4.54 (s, 4H), 7.41 (d, 4H), 7.57 (d, 4H); ^{13}C NMR (62.9 MHz, CDCl_3): 5.3 ($^3J_{\text{Sn-C}} = 15$ Hz), 14.9 ($^1J_{\text{Sn-C}} = 635$ Hz), 22.4 ($^3J_{\text{Sn-C}} = 29$ Hz), 33.0 ($^2J_{\text{Sn-H}} = 87$ Hz), 69.7, 72.6, 76.8 ($^1J_{\text{Sn-C}} = 806$ Hz), 107.8 ($^2J_{\text{Sn-C}} = 168$ Hz), 127.1, 128.2, 137.8, 140.3; ^{119}Sn NMR (74.6 MHz, CDCl_3 , 25°C): δ = -252.1; elemental analysis calc (%) for $\text{C}_{40}\text{H}_{46}\text{O}_2\text{Sn}_2$: C 60.34, H 5.82, O 4.02, Sn 29.81; found: C 60.14, H 6.01, O 4.30, Sn 28.76.

3: a solution of HCl and water (0.137 ml of 1 N HCl in 1.35 ml H_2O , 76.4 mmol) in THF (3.39 g) was added to a solution of **2** (0.76 g, 1 mmol) in dry THF (6 g) in a Schlenk tube under nitrogen. The mixture gelified after 6 days at room temperature and was aged for 28 days. The solid was then filtered, washed with THF (3x50 ml) and dried under vacuum at 120°C for 3 h.

Yield: 0.40 g, 65%; elemental analysis calc (%) for $\text{C}_{22}\text{H}_{28}\text{O}_5\text{Sn}_2$: C 43.33, H 4.63, O 13.12, Sn 38.92; found: C 44.31, H 4.95, O 13.51, Sn 37.22.

Thermogravimetric analysis of **3**



The mass of residual SnO_2 at 640°C corresponds to a molecular mass of 608 to be compared to a theoretical mass of 609.85 ($\text{C}_{22}\text{H}_{28}\text{O}_5\text{Sn}_2$) for **6**.

Solid state CPMAS spectrum of 1 (blue line: spectrum; red line simulated spectrum).

