Supporting Information

For

Microwave-Enhanced Multiple Suzuki Couplings toward Highly Luminescent Starburst Monodisperse Macromolecules

Wen-Yong Lai, Qing-Quan Chen, Qi-Yuan He, Qu-Li Fan, and Wei Huang* Institute of Advanced Materials (IAM), Fudan University, Shanghai 200433, P. R. China E-mail: wei-huang@fudan.edu.cn / chehw@nus.edu.sg.

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2006



Scheme 1. Reagents and conditions: (a) (i) *n*-BuLi, -78 °C; (ii) 2-isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane, -78 °C to rt. (b) Conventional heating, Pd(PPh₃)₄, 2 M K₂CO₃/H₂O, THF, reflux, 2 d. (c) Conventional heating, KOH, THF, reflux, 12 h. (d) Microwave heating, pressurized-vessel, Pd(PPh₃)₄, aqueous K₂CO₃-THF, 150 °C, 15–30 min.

Experimental Section.

Hexabromo-triazatruxene (1),¹ 2-bromo-9,9-dihexylfluorene (3),² 2,7-dibromo-9,9-dihexylfluorene (4),³ 2-(2-bromo-9,9-dihexyl-9H-fluoren-7-yl)-9,9-dihexyl-9H-fluorene (6),⁴ and 2-(2-bromo-9,9-dihexyl-9H-fluoren-7-yl)-9,9-dihexyl-7-(9,9-di-hexyl-9H-fluoren-2-yl)-9H-fluorene $(8)^4$ were obtained similarly to the procedures described in the literatures.

Gerneral procedure for synthesis of oligofluorene boronates 5, 7, and 9. Into a solution of the corresponding 2-bromofluorene derivatives (3, 6 or 8) in anhydrous THF (12–15 mL per gram) was added dropwise *n*-BuLi (1.6 M solution in *n*-hexane, 1.1–1.3 equivalents) at –78 °C, where the reaction mixture was stirred 1 h before adding 2-isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3.0 equiv.). The mixture was allowed to warm to room temperature and stirred for 12 h. It was poured onto crushed ice and extracted several times with ethyl ether. The extracts were washed with brine three times, and finally dried over anhydrous Na₂SO₄. After filtration and evaporation of the solvents, the resulting product was purified by flash chromatography on a silica gel column with elution hexane/ethyl acetate (95:5) to afford oligofluorene boronates (5, 7, or 9).

5: Pale-yellow oil, yield 91%. ¹H NMR (400 MHz; CDCl₃, δ, ppm): 7.81-7.80 (d, 1H), 7.74–7.69 (m, 3H), 7.34–7.31 (m, 2H), 1.99–1.95 (m, 4H), 1.39 (s, 12H), 1.11–1.01 (m, 12H), 0.77–0.73 (t, 6H), 0.60–0.57 (m, 4H). ¹³C NMR (100 MHz, DMSO, δ, ppm): 151.5, 150.1, 144.3, 141.1, 133.9, 129.1, 127.7, 126.9, 123.1, 120.3, 119.6, 119.2, 83.9, 55.3, 40.5, 31.7, 29.9, 25.2, 25.0, 23.9, 22.8, 14.2. MALDI-TOF MS (*m/z*): Calcd for C₃₁H₄₅BO₂, Exact Mass: 460.35, Mol. Wt.: 460.5; Found: 460.3 (M⁺).

7: White solid, yield 86%. ¹H NMR (400 MHz; CDCl₃, δ, ppm): 7.85–7.73 (m, 6H), 7.66–7.61 (m, 4 H), 7.38–7.30 (m, 3H), 1.97–1.93 (m, 8H), 1.33 (s, 12H), 1.06–0.99 (m, 24H), 0.70–0.61 (m, 20H). ¹³C NMR (100 MHz, CDCl₃, δ, ppm): 152.3, 151.7, 151.2, 150.4, 144.0, 141.2, 141.0, 140.7, 140.6, 140.3, 134.0, 129.2, 127.3, 127.1, 126.4, 123.2, 121.8, 120.7, 120.2, 120.0, 119.4, 84.1, 55.6, 55.5, 40.7, 40.6, 31.8, 30.0, 25.3, 24.1, 24.0, 22.9, 14.4. MALDI-TOF MS (*m/z*): Calcd for C₅₆H₇₇BO₂, Exact Mass: 792.6, Mol. Wt.: 793.02; Found: 792.9 (M⁺).

9: White solid, yield 81%. ¹H NMR (400 MHz; CDCl₃, δ, ppm): 7.86–7.79 (m, 6H), 7.77–7.75 (d, 2H), 7.69–7.64 (m, 8H), 7.39–7.30 (m, 3H), 2.13–2.05 (m, 12H), 1.41 (s, 12H), 1.27–1.09 (m, 36H), 0.79–0.72 (m, 30H). ¹³C NMR (100 MHz, CDCl₃, δ, ppm): 152.4, 152.1, 151.8, 151.3, 150.5, 144.2,

141.1, 140.8, 140.7, 140.5, 140.4, 140.3, 134.2, 129.2, 127.3, 127.1, 126.5, 126.4, 123.3, 121.9, 121.8, 120.7, 120.3, 120.2, 120.0, 119.4, 84.1, 55.7, 55.5, 40.7, 40.6, 31.8, 30.0, 25.3, 22.9, 14.4. MALDI-TOF MS (*m/z*): Calcd for C₈₁H₁₀₉BO₂, Exact Mass: 1124.85, Mol. Wt.: 1125.54; Found: 1124.9 (M⁺).

2. A mixture of 2,3,7,8,12,13-hexabromo-5H-diindolo[3,2-a:3',2'-c]carbazole **1** (2.45 g, 3 mmol) and KOH (1.68 g, 30 mmol) was heated to reflux in THF (50 mL). Then, 1-bromohexane (2.23 g, 13.5 mmol) was added and the mixture was refluxing for 12 h before it was cooled to room temperature. The mixture was diluted with CH₂Cl₂ and washed with 10% aqueous HCl solution and with saturated aqueous NaCl solution, dried (Na₂SO₄) and evaporated. The resulting residue was purified by silica-gel column chromatography using hexane/CH₂Cl₂ (8:1) as the eluent to give **2** as a white solid (2.96 g, 92%). ¹H NMR (400 Hz, CDCl₃, δ , ppm): 8.02 (s, 3H), 7.55 (s, 3H), 4.08 (m, 6H), 1.74 (brs, 6H), 1.33–1.26 (m, 18H), 0.87–0.84 (m, 9H). ¹³C NMR (100 MHz, CDCl₃, δ , ppm): 140.3, 138.9, 125.2, 123.1, 118.7, 115.4, 114.8, 101.5, 47.2, 31.7, 30.4, 26.4, 22.9, 14.2. MALDI-TOF MS (*m*/*z*): Calcd for C₄₂H₄₅Br₆N₃, Exact Mass: 1064.87 Mol. Wt.: 1071.25; Found: 1069.0 (M⁺), 991.0 ([M–Br]⁺), 913.1 ([M–2Br]⁺), 833.2 ([M–3Br]⁺), 753.3 ([M–4Br]⁺), 675.3 ([M–5Br]⁺), 597.5 ([M–6Br]⁺).

Gerneral procedure for synthesis of six-arm triazatruxenes G1–G3. A solution of 2 (0.05 mmol), 5, 7, or 9 (0.39 mmol) and aqueous 2 M K₂CO₃ (375 μ L, 0.75 mmol) in THF (3 mL for 5, 4 mL for 7, 5 mL for 9) in a 10 mL pressurized vessel was carefully degassed before and after adding Pd(PPh₃)₄ (0.03 mmol). The vessel was then sealed and heated in the CEM Discover system. The initial microwave power was set at 100 W. After the set temperature of 150 °C is reached, the microwave power regulated itself to keep that temperature for corresponding reaction hold time as listed in Table 1 before cooling down to room temperature. The mixture was subsequently diluted with CH₂Cl₂ and then washed with 1 M aqueous HCl and with saturated brine, dried (Na₂SO₄) and evaporated. The resulting residue was purified with column chromatography using hexane–CH₂Cl₂ as eluant to give the final product. The temperature was monitored using a calibrated infrared temperature control mounted under the reaction vessel. The pressure was controlled by a load cell connected to the vessel via the septum.

G1: Yield 89%. ¹H NMR (400 MHz, CDCl₃, δ , ppm): 8.44 (s, 3H), 7.71 (s, 3H), 7.64–7.62 (m, 9H), 7.57–7.55 (d, 3H), 7.45–7.43 (d, 3H), 7.39 (s, 3H), 7.35–7.33 (d, 3H), 7.30–7.20 (m, 21H), 5.05 (m, 6H), 2.15 (m, 6H), 1.89–1.74 (m, 24H), 1.26 (m, 36H), 1.20–1.18 (m, 9H), 1.08–0.87 (m, 54H), 0.77–0.56 (m, 60H). ¹³C NMR (100 MHz, CDCl₃, δ , ppm): 151.1, 151.0, 150.8, 150.5, 142.1, 141.5, 141.2, 141.1, 140.9, 139.9, 139.5, 139.3, 136.7, 133.9, 129.3, 129.0, 127.1, 127.0, 126.9, 125.6, 125.2, 124.3, 123.0, 122.9, 119.9, 119.8, 119.5, 119.3, 112.4, 103.5, 55.1, 55.0, 47.6, 40.6, 32.2, 31.9, 30.0, 29.6, 26.9, 24.1, 22.9, 22.8, 14.4, 14.3, 14.2. MALDI-TOF MS (*m*/*z*): Calcd for C₁₉₂H₂₄₃N₃, Exact Mass: 2590.91, Mol. Wt.: 2593; Found: 2592.0 (M⁺), 2507.6 ([M–C₆H₁₃]⁺), 2423.8 ([M–2C₆H₁₃]⁺). Anal. Calcd for C₁₉₂H₂₄₃N₃: C, 88.93; H, 9.45; N, 1.62. Found: C, 88.86; H, 9.48; N, 1.66.

G2: Yield 86%. ¹H NMR (400 MHz, CDCl₃, δ, ppm): 8.51 (s, 3H), 7.77–7.70 (m, 24H), 7.63–7.50 (m, 30H), 7.46 (s, 3H), 7.39–7.29 (m, 24H), 5.11 (m, 6H), 2.21–1.81 (m, 54H), 1.38–1.07(m, 156H), 0.86–0.72 (m, 135H). ¹³C NMR (100 MHz, CDCl₃, δ, ppm): 151.6, 151.2, 142.2, 141.6, 141.0, 140.8, 140.5, 140.4, 136.7, 133.9, 132.7, 127.1, 127.0, 126.2, 123.1, 121.6, 120.0, 103.5, 55.3, 40.6, 32.7, 31.8, 31.7, 29.9, 24.0, 22.9, 22.8, 20.3, 14.4, 14.3, 14.2. MALDI-TOF MS (*m/z*): Calcd for C₃₄₂H₄₃₅N₃, Exact Mass: 4584.41, Mol. Wt.: 4588.13; Found: 4586.7 (M⁺), 4469.0 ([M–117.7]⁺), 4350.8 ([M–117.7-118.2]⁺). Anal. Calcd for C₃₄₂H₄₃₅N₃: C, 89.53; H, 9.56; N, 0.92. Found: C, 89.50; H, 9.63; N, 0.94.

G3: Yield 84%. ¹H NMR (400 MHz, CDCl₃, δ, ppm): 8.52 (s, 3H), 7.82–7.71 (m, 36H), 7.67–7.51

(m, 54H), 7.47 (s, 3H), 7.43–7.30 (m, 24H), 5.13 (m, 6H), 2.22–1.84 (m, 78H), 1.37–1.08(m, 228H), 0.90–0.75 (m, 195H). ¹³C NMR (100 MHz, CDCl₃, δ , ppm): 152.1, 151.8, 151.3, 141.1, 140.8, 140.6, 140.3, 126.4, 126.3, 123.3, 121.8, 120.3, 120.0, 55.5, 40.7, 32.0, 31.8, 30.2, 30.0, 27.0, 24.1, 23.1, 22.9, 14.5, 14.4. MALDI-TOF MS (*m*/*z*): Calcd for C₄₉₂H₆₂₇N₃, Exact Mass: 6577.92, Mol. Wt: 6583.26; Found: 6581.0 (M⁺), 6461.9 ([M-119.1]⁺), 6344.7 ([M-119.1-117.2]⁺). Anal. Calcd for C₄₉₂H₆₂₇N₃: C, 89.76; H, 9.60; N, 0.64. Found: C, 89.57; H, 9.58; N, 0.63.



Fig. S1 Temperature, pressure, and power profile after optimization for the preparation of G1 in aqueous K_2CO_3 -THF heated under sealed-vessel microwave irradiation conditions. The maximum pressure in the reaction vessel was ca. 75 psi. After the set temperature of 150 °C is reached, the power regulates itself down to ca.50 W.

References

- N. Robertson, S. Parsons, E. J. MacLean, R. A. Coxall and A. R. Mount, *J. Mater. Chem.*, 2000, 10, 2043.
- 2. J. F. Ding, M. Day, G. Robertson and J. Roovers, *Macromolecules*, 2002, 35, 3474.
- 3. M. Ranger, D. Rondeau and M. Leclerc, *Macromolecules*, 1997, 30, 7686.
- A. L. Kanibolotsky, R. Berridge, P. J. Skabara, I. F. Perepichka, D. D. C. Bradley and M. Koeberg, *J. Am. Chem. Soc.*, 2004, **126**, 13695; Y. Geng, A. Traykovska, D. Katsis, J. D. Ou, S. W. Culligan and S. H. Chen, *J. Am. Chem. Soc.*, 2002, **124**, 8337.