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Supporting data:

## An outstanding catalyst for asymmetric transfer hydrogenation in aqueous solution and formic acid/triethylamine

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## Synthesis of catalyst 11, enantiomeric excess determination for products, $1 \mathrm{H}-\mathrm{NMR}$ spectrum of catalyst 11 and details of X-ray crystal structure.

Synthesis of 2-(2,3,4,5-tetramethylcyclopentadienyl)-benzyl (1R,2R)-toluenesulfonyl-cyclohexyldiamine 13; 2-(2,3,4,5-Tetramethylcyclopentadienyl)-benzylaldehyde $\mathbf{1 2}(3.08 \mathrm{~g}, 0.0136 \mathrm{~mol})$ was dissolved in dry methanol $(95 \mathrm{~mL})$. To the solution was added R,R TsCYDN $3(4.36 \mathrm{~g}, 0.0163 \mathrm{~mol})$, followed by the addition of 6 g of molecular sieves and 6 drops of glacial acetic acid. After formation of the imine was confirmed by TLC, sodium cyanoborohydride ( $1.11 \mathrm{~g}, 0.0177 \mathrm{~mol}$ ) was added and the reaction left to stir overnight at room temperature. The molecular sieves were filtered through filter paper and the solution was concentrated under reduced pressure to remove the methanol. The residue was redissolved in ethyl acetate ( 80 $\mathrm{mL})$. The organic layer was washed with saturated $\mathrm{NaHCO}_{3}(80 \mathrm{~mL})$ and saturated brine $(80 \mathrm{~mL})$ and then dried $\left(\mathrm{MgSO}_{4}\right)$. The solvent was removed to give a crude solid which was purified by silica gel column chromatography ( $0 \rightarrow 30 \% \mathrm{v} / \mathrm{v}$ ethyl acetate/hexane) to afford the product 13 as a yellow solid ( $2.50 \mathrm{~g}, 5.23 \mathrm{mmol}, 38 \%$ ). mp $48-50{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{22}-27\left(c 0.3, \mathrm{CHCl}_{3}\right)$; $v_{\max }($ neat $) / \mathrm{cm}^{-1}: 3261,2926,2855,1599,1446,1325,1159,1092,899,813,758,662 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.76-7.65$ $(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.44-7.07(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.06-6.95(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 5.42(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H \mathrm{Ts}), 3.97-3.33(2 \mathrm{H}, \mathrm{m}, \mathrm{ArCH} 2), 3.12-2.44$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{NHCH}_{2}+\mathrm{CpH}\right), 2.38\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{Ar}\right), 2.25-0.70\left(22 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{CpCH}_{3}+\mathrm{CHNTs}+\mathrm{CH} \mathrm{NHCH}_{2}+4 \times \mathrm{CH}_{2}\right.$ of cyclohexyl ring); $\delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): 143.1,140.3,138.6,138.0,137.3,130.1,129.8,129.6,129.5,128.7,128.4,127.3,127.3,126.7$ (10 Ar- C and 4 CpC ), $60.2(\mathrm{CH}), 58.0(\mathrm{CH}), 57.7(\mathrm{CH}), 47.8\left(\mathrm{CH}_{2}\right), 32.7\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $31.3\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $24.6\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $24.3\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $21.5\left(\mathrm{CH}_{3}\right), 14.2\left(\mathrm{CpCH}_{3}\right), 11.8\left(\mathrm{CpCH}_{3}\right), 11.3\left(\mathrm{CpCH}_{3}\right)$, $11.1\left(\mathrm{CpCH}_{3}\right) ; m / z($ LSIMS $) 479\left(\mathrm{MH}^{+}, 100 \%\right), 211(16 \%)$ HRMS (LSIMS) calc for $\mathrm{MH}^{+}=479.2732$ found 479.2743 (2.3 ppm error).
Synthesis of 2-(2,3,4,5-tetramethylcyclopentadienyl)-benzyl (1R,2R)-toluenesulfonyl-cyclohexyldiamine rhodium (III) chloride 11; Rhodium (III) chloride hydrate ( $0.90 \mathrm{~g}, 4.31 \mathrm{mmol}$ ) was added to a stirred solution of 2-(2,3,4,5-tetramethylcyclopentadienyl)-benzyl (1R,2R)- toluenesulfonyl-cyclohexyldiamine $\mathbf{1 3}$ ( $2.06 \mathrm{~g}, 4.31 \mathrm{mmol}$ ) in methanol ( 90 mL ). The reaction mixture was heated under reflux and stirred for 24 hours. Triethylamine ( $1.20 \mathrm{~mL}, 8.62 \mathrm{mmol}$ ) was added to the reaction mixture and the reactants were stirred at reflux temperature for a further 24 hours. The reaction mixture was cooled to room temperature and the solvent removed under reduced pressure. The crude residue was triturated with water ( 50 mL ) for 10 minutes, collected by filtration (filter paper), washed with water ( 50 mL ) and then allowed to dry on the filter paper. The redbrown solid was purified by silica gel column chromatography $(50 \rightarrow 100 \% \mathrm{v} / \mathrm{v}$ ethyl acetate/hexane and then $0 \rightarrow 10 \% \mathrm{v} / \mathrm{v}$ methanol/ethyl acetate) to afford the product 11 as a dark red solid ( $1.14 \mathrm{~g}, 1.86 \mathrm{mmol}, 43 \%$ yield); decomposition temperature $244-246^{0} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{23}-201.3\left(c 0.3, \mathrm{CHCl}_{3}\right) ; v_{\max }($ neat $) / \mathrm{cm}^{-1}: 3450,3230,2927,1448,1264,1252,1139,1129,1098,1022,928$, $899,828,769,714,667 ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.95(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{ArH}), 7.55-7.46(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.40(1 \mathrm{H}, \mathrm{d}, J 7.0$, $\mathrm{Ar} H), 7.16(2 \mathrm{H}, \mathrm{d}, J 8.2, \mathrm{Ar} H), 4.36(1 \mathrm{H}, \mathrm{dd}, J 14.2$ and $2.9, \mathrm{ArCHH}), 4.24(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 14.2, \mathrm{ArCH} H), 4.01(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 11.3$, NH ), 2.38-2.29 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CHNTs}+\mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{a} \mathrm{H}_{\mathrm{b}} \mathrm{CHNTs}$ ), $2.34\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3} \mathrm{Ar}\right), 2.10-1.98\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CHNHCH}_{2}+\right.$ $\left.\mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{a} \mathrm{H}_{\mathrm{b}} \mathrm{CHNHCH}_{2}\right), 1.96\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CpCH}_{3}\right), 1.88\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CpCH}_{3}\right), 1.57\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CpCH}_{3}\right), 1.56-1.50(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{a} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CHNTs}$ ), 1.49 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{CpCH}_{3}$ ), $1.44-1.36$ ( $1 \mathrm{H}, \mathrm{m}, \quad \mathrm{CH}_{a} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CHNHCH}_{2}$ ), 1.03-0.73 (4H, m, $\mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{a} \boldsymbol{H}_{b} \mathrm{CHNHCH}_{2}+\mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CH}_{\mathrm{a}} \boldsymbol{H}_{b} \mathrm{CHNTs}+\mathrm{CH}_{\mathrm{a}} \boldsymbol{H}_{b} \mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CHNHCH}_{2}+\mathrm{CH}_{\mathrm{a}} \boldsymbol{H}_{b} \mathrm{CH}_{\mathrm{a}} \mathrm{H}_{\mathrm{b}} \mathrm{CHNTs}$ ); $\delta_{\mathrm{C}}$ ( $100.6 \mathrm{MHz} ;$ $\mathrm{CDCl}_{3}$ ): 141.4, 140.1, 135.4, 131.1, 130.0, 129.9, 129.7, 128.6, 128.1, 126.9 ( $\mathrm{Ar}-C$ ), $104.2\left(J^{\mathrm{RhC}} 6.5, \mathrm{CpC}\right), 101.0\left(J^{\mathrm{RhC}} 6.9\right.$, $\mathrm{CpC}), 97.4\left(J^{\mathrm{RhC}} 9.6, \mathrm{Cp} C\right), 86.6\left(J^{\mathrm{RhC}} 9.2, \mathrm{CpC}\right), 81.0\left(J^{\mathrm{RhC}} 8.4, \mathrm{CpC}\right), 67.7(\mathrm{CH}), 63.5(\mathrm{CH}), 50.8\left(\mathrm{CH}_{2}\right), 35.5\left(\mathrm{CH}_{2}\right.$ of
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cyclohexyl ring), $30.3\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $24.6\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring), $24.3\left(\mathrm{CH}_{2}\right.$ of cyclohexyl ring $)$, $21.4\left(\mathrm{CH}_{3} \mathrm{Ar}\right)$, $11.0\left(\mathrm{CpCH}_{3}\right), 9.9\left(\mathrm{CpCH}_{3}\right), 9.7\left(\mathrm{CpCH}_{3}\right), 8.0\left(\mathrm{CpCH}_{3}\right) ; m / z(\mathrm{FAB}) 579(\mathrm{M}-\mathrm{Cl}, 100 \%), 423(13 \%), 154(17 \%)$. HRMS (FAB) calc for $\mathrm{M}-\mathrm{Cl}=579.1553$ found 579.1551 ( 0.2 ppm error).

## Reduction of ketones using catalyst 11 in Formic acid/triethylamine

Reduction of ketones using catalyst 11 in Formic acid:Triethylamine (5:2); A solution of $\mathbf{1 1}(0.016 \mathrm{mmol})$ in formic acid : triethylamine $5: 2$ azeotrope ( 1.5 mL ) was stirred in a flame dried schlenk tube at $28^{0} \mathrm{C}$ for 15 minutes. The ketone substrate ( 3.2 mmol ) was added and the reaction mixture was stirred at $25^{\circ} \mathrm{C}$ for a period of time over which time the conversion was monitored by GC. After full conversion had been achieved the reaction was filtered (silica), washed ( $20 \% \mathrm{EtOAc} / 80 \%$ hexane) and concentrated under vacuum to give the reduction product. The residue was purified by flash column chromatography where necessary.

Reduction of ketones using catalyst 11 in Water; To a solution of $11(0.016 \mathrm{mmol})$ in water ( 5.5 mL ) was added HCOONa $(16.2 \mathrm{mmol})$ and stirred in a flame dried schlenk tube at $40^{\circ} \mathrm{C}$ for 15 minutes. The ketone substrate ( 3.2 mmol ) was added and the reaction mixture was stirred at $40^{\circ} \mathrm{C}$ for a period of time over which time the conversion was monitored by GC. After full conversion had been achieved the reaction was filtered through silica and $\mathrm{MgSO}_{4}$, washed ( $20 \% \mathrm{EtOAc} / 80 \%$ hexane) and concentrated under vacuum to give the reduction product. The residue was purified by flash column chromatography where necessary.

1-Phenylethanol 14: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 $50 \mathrm{~m}, \mathrm{~T}=115{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $9.7 \mathrm{~min}, \mathrm{R}$ isomer 14.2 min ., S isomer 15.1 min .); $96 \%$ ee $(\mathrm{R})\left(\mathrm{lit} .{ }^{1}[\alpha]_{\mathrm{D}}{ }^{26}+45.4\right.$ (c 0.50 in $\mathrm{CHCl}_{3}$ ) $98 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(3'-Chlorophenyl)ethanol 15: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=150{ }^{\circ} \mathrm{C}, \mathrm{P}=10 \mathrm{psi}$, ketone 11.1 min , R isomer 16.3 min ., S isomer 16.9 min .); $95 \%$ ee ( R ) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{24}+38.2$ (c 0.9 in $\mathrm{CHCl}_{3}$ ) $96 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(3'-Trifluoromethylphenyl)ethanol 16: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=130{ }^{\circ} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone $8.5 \mathrm{~min}, \mathrm{R}$ isomer $15.0 \mathrm{~min} ., \mathrm{S}$ isomer 15.7 min .); $96 \%$ ee (R) (lit. ${ }^{1}$ $[\alpha]_{\mathrm{D}}{ }^{26}+27.1\left(\mathrm{c} 1.60\right.$ in $\left.\mathrm{CH}_{3} \mathrm{OH}\right) 96 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(3'-Methoxyphenyl)ethanol 17: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin-$\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=140{ }^{0} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $12.9 \mathrm{~min}, \mathrm{R}$ isomer 17.9 min ., S isomer 18.5 min .); Assigned as R configuration (see results for reductions in $\mathrm{H}_{2} \mathrm{O}$ for $[\alpha]_{\mathrm{D}}$ value and ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data).

1-(2'-Trifluoromethylphenyl)ethanol 20: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=120^{\circ} \mathrm{C}, \mathrm{P}=10 \mathrm{psi}$, ketone 11.1 min , R isomer 17.7 min ., S isomer 18.7 min .); Assigned as R configuration (see results for reductions in $\mathrm{H}_{2} \mathrm{O}$ for $[\alpha]_{\mathrm{D}}$ value); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(2'-Methoxyphenyl)ethanol 19: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin-$\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=140^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $12.5 \mathrm{~min}, \mathrm{~S}$ isomer 14.4 min ., R isomer 14.8 min .); $94 \%$ ee ( R ) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{27}$ +37.7 (c 0.28 in toluene) $90 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-Phenylpropan-1-ol 22: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M$1950 \mathrm{~m}, \mathrm{~T}=115^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone 14.1 min , R isomer 21.2 min ., S isomer 22.3 min .); $[\alpha]_{\mathrm{D}}{ }^{26}+46.9$ (c 1.4 in $\left.\mathrm{CHCl}_{3}\right) 96 \%$ ee (R) (lit. ${ }^{2}[\alpha]_{\mathrm{D}}{ }^{20}+47.0\left(\mathrm{c}^{2} .4\right.$ in $\left.\mathrm{CHCl}_{3}\right) 95 \%$ ee $\left.(\mathrm{R})\right)$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.36-7.24(5 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 4.57(1 \mathrm{H}, \mathrm{t}, J 6.5$, $\left.\mathrm{PhCH}(\mathrm{OH}) \mathrm{CH}_{2}\right), 2.00(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.86-1.68\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 0.90\left(3 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 144.6$, 128.4, 127.5, $126.0(\mathrm{Ar}-\mathrm{C}), 76.0(\mathrm{CH}), 31.9\left(\mathrm{CH}_{2}\right), 10.2\left(\mathrm{CH}_{3}\right)$.

1-(1'-Naphthyl)ethanol 21: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=170^{\circ} \mathrm{C}, \mathrm{P}=10 \mathrm{psi}$, ketone 23.0 min , S isomer $31.9 \mathrm{~min} ., \mathrm{R}$ isomer 32.7 min .) ; $[\alpha]_{\mathrm{D}}{ }^{22}+71.4\left(\mathrm{c} 0.35 \mathrm{in} \mathrm{Et}_{2} \mathrm{O}\right)$ $84 \%$ ee (R) $\left(\right.$ lit. $^{3}[\alpha]_{\mathrm{D}}{ }^{28}+77.2$ (c 0.67 in $\left.\mathrm{Et}_{2} \mathrm{O}\right) 99 \%$ ee (R)); $8.09(1 \mathrm{H}, \mathrm{d}, J 8.0, \mathrm{Ar} H), 7.87-7.83(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.76(1 \mathrm{H}, \mathrm{d}, J$ 8.3, $\mathrm{Ar} H), 7.65(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{Ar} H), 7.53-7.43(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 5.64\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 2.05(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.65(3 \mathrm{H}, \mathrm{d}$, $\left.J 6.5, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 141.5,133.8,130.3,128.9,127.9,126.0,125.6,125.5,123.2,122.1$ (Ar-C), 67.1 $(\mathrm{CH}), 24.4\left(\mathrm{CH}_{3}\right)$.

2-Thienylethanol 26: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 $50 \mathrm{~m}, \mathrm{~T}=125^{\circ} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone $13.2 \mathrm{~min}, \mathrm{R}$ isomer 15.6 min ., S isomer 16.4 min .); $[\alpha]_{\mathrm{D}}{ }^{22}+25.6\left(\mathrm{c} 0.50 \mathrm{in} \mathrm{CHCl}_{3}\right) 97 \%$ ee (R) $\left(\right.$ lit. $^{3}[\alpha]_{\mathrm{D}}{ }^{20}+15.2\left(\mathrm{c} 0.50\right.$ in $\left.\mathrm{CHCl}_{3}\right) 52 \%$ ee (R) ); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.24(1 \mathrm{H}, \mathrm{dd}, J 4.8$ and 1.5, $\mathrm{Ar} H), 6.99-6.95$ $(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 5.13\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 2.03(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}) 1.60\left(3 \mathrm{H}, \mathrm{d}, J 6.5, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}:$ 149.9, 126.7, 124.5, $123.2(\mathrm{Ar}-\mathrm{C}), 66.3(\mathrm{CH}), 25.3\left(\mathrm{CH}_{3}\right)$.

3-Thienylethanol 27: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 $50 \mathrm{~m}, \mathrm{~T}=120^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $10.1 \mathrm{~min}, \mathrm{R}$ isomer 13.5 min ., S isomer 14.2 min .); $[\alpha]_{\mathrm{D}}{ }^{24}+23.3$ (c 0.78 in EtOH ) $93 \%$ ee (R) (lit. ${ }^{3}[\alpha]_{\mathrm{D}}{ }^{24}+33.8(\mathrm{c} 0.43 \mathrm{in} \mathrm{EtOH}) 91 \%$ ee (R)); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}$ : $7.27(1 \mathrm{H}, \mathrm{dd}, J 4.9$ and $2.9, \mathrm{Ar} H), 7.16-7.14$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 7.07(1 \mathrm{H}, \mathrm{dd}, J 5.0$ and 1.3, $\mathrm{Ar} H), 4.92\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 2.32(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.49(3 \mathrm{H}, \mathrm{d}, J 6.5$, $\left.\mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 147.4,126.1,125.7,120.2(\mathrm{Ar}-\mathrm{C}), 66.5(\mathrm{CH}), 24.5\left(\mathrm{CH}_{3}\right)$.

2-Furylethanol 25: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 50m, $\mathrm{T}=85{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $10.5 \mathrm{~min}, \mathrm{R}$ isomer 15.1 min ., S isomer 16.1 min .); $[\alpha]_{\mathrm{D}}{ }^{20}+18.9$ (c 0.56 in $\left.\mathrm{CHCl}_{3}\right) 98 \%$ ee ( R ) (lit. ${ }^{4}[\alpha]_{\mathrm{D}}{ }^{25}+20.8\left(\mathrm{c} 1.27\right.$ in $\left.\mathrm{CHCl}_{3}\right) 99 \%$ ee $(\mathrm{R})$ ); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.37(1 \mathrm{H}, \mathrm{dd}, J 1.9$ and $0.6, \mathrm{ArH}), 6.33(1 \mathrm{H}$, dd, $J 3.3$ and $2.0, \mathrm{ArH}), 6.23(1 \mathrm{H}, \mathrm{d}, J 3.3, \mathrm{Ar} H), 4.88\left(1 \mathrm{H}, \mathrm{q}, J 6.5, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 1.96(1 \mathrm{H}, \mathrm{br}$ s, OH$\left.) 1.54(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{CH})^{2}\right) ; \delta_{\mathrm{C}}$ (100.6MHz; $\left.\mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 157.6,141.9,110.1,105.1(\mathrm{Ar}-\mathrm{C}), 63.7(\mathrm{CH}), 21.3\left(\mathrm{CH}_{3}\right)$.

4-Pyridylethanol 28: Enantiomeric excess and conversion determined by HPLC analysis (Chiralcel OD-H column, ethanol:hexane $=3: 97(0.5 \mathrm{~mL} / \mathrm{min})$, ketone $28.4 \mathrm{~min}, \mathrm{~S}$ isomer 42.1 min ., R isomer 46.0 min.$)[\alpha]_{\mathrm{D}}{ }^{20}+22.1(\mathrm{c} 0.14 \mathrm{in} \mathrm{EtOH})$ $95 \%$ ee (R) (lit. ${ }^{3}[\alpha]_{\mathrm{D}}{ }^{21}+51.2$ (c 0.122 in EtOH$) 93 \%$ ee (R)); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 8.44(2 \mathrm{H}, \mathrm{dd}, J 4.5$ and $1.5, \mathrm{Ar} H)$, $7.30(2 \mathrm{H}, \mathrm{dd}, J 4.5$ and $1.5, \mathrm{Ar} H), 4.89\left(1 \mathrm{H}, \mathrm{q}, J 6.6, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 4.00(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.48\left(3 \mathrm{H}, \mathrm{d}, J 6.5, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}$ $\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 155.5,149.4,120.6(\mathrm{Ar}-\mathrm{C}), 68.5(\mathrm{CH}), 25.1\left(\mathrm{CH}_{3}\right)$.

1-(2'-Chlorophenyl)ethanol 18: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=150{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone 6.6 min , R isomer 9.8 min ., S isomer 10.5 min .); $85 \%$ ee ( R ) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{26}+48.8$ (c 1.0 in $\mathrm{CHCl}_{3}$ ) $77 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.
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1-Cyclohexylethanol 32: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}$ $1950 \mathrm{~m}, \mathrm{~T}=92{ }^{\circ} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone $25.0 \mathrm{~min}, \mathrm{R}$ isomer 40.5 min ., S isomer 41.0 min .); $[\alpha]_{\mathrm{D}}{ }^{21}+2.07$ (c $1.11 \mathrm{in} \mathrm{CHCl}_{3}$ ) $87 \%$ ee (S) (lit. ${ }^{3}[\alpha]_{\mathrm{D}}{ }^{29}+1.82\left(\mathrm{c} 0.30\right.$ in $\left.\mathrm{CHCl}_{3}\right) 68 \%$ ee $(\mathrm{S})$ ); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 3.54\left(1 \mathrm{H}\right.$, quin, $\left.J 6.2, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 1.88-$ $1.82(1 \mathrm{H}, \mathrm{m}$, cyclohexyl CH$), 1.80-1.72\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 1.71-1.63(2 \mathrm{H}, \mathrm{m}, \mathrm{CH} 2), 1.49(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.32-0.91(6 \mathrm{H}, \mathrm{m}, 3 \times$ $\left.\mathrm{CH}_{2}\right), 1.15\left(3 \mathrm{H}, \mathrm{d}, J 6.3, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz}^{2} \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 72.2\left(\mathrm{CH}(\mathrm{OH}), 45.1(\right.$ cyclohexyl CH$), 28.7\left(\mathrm{CH}_{2}\right), 28.4\left(\mathrm{CH}_{2}\right)$, $26.5\left(\mathrm{CH}_{2}\right), 26.2\left(\mathrm{CH}_{2}\right), 26.1\left(\mathrm{CH}_{2}\right), 20.4\left(\mathrm{CH}_{3}\right)$.

1-Tetralol 24: Enantiomeric excess and conversion determined by HPLC analysis (Chiralcel OD-H, $250 \times 4.6 \mathrm{~mm}$ column, Hexane/IPA 98:2, $0.5 \mathrm{~mL} / \mathrm{min}$, ketone 15.6 min , S isomer 29.7 min ., R isomer 32.2 min .); 99.4\% ee (R) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{27}-25$ (c 0.114 in $\mathrm{CHCl}_{3}$ ) $99.9 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

2-Chloro-1-phenylethanol 29: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=145^{\circ} \mathrm{C}, \mathrm{P}=10 \mathrm{psi}$, ketone 18.3 min , S isomer 21.6 min ., R isomer 22.4 min .); $[\alpha]_{\mathrm{D}}{ }^{25}+46.4$ (c 1.43 in $\left.\mathrm{C}_{6} \mathrm{H}_{12}\right) 96 \%$ ee (S) (lit. ${ }^{5}[\alpha]_{\mathrm{D}}{ }^{25}-50.4$ (c 1.78 in $\left.\mathrm{C}_{6} \mathrm{H}_{12}\right) 98 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

2-Phenoxy-1-phenylethanol 31: Enantiomeric excess and conversion determined by HPLC analysis (Chiralcel OD-H, $250 \times$ 4.6 mm column, Hexane/IPA, $\mathrm{Et}_{2} \mathrm{NH} 90: 10: 0.1,0.7 \mathrm{~mL} / \mathrm{min}$, ketone 18.3 min , R isomer 17.8 min ., S isomer 31.0 min .); $92 \%$ ee $(\mathrm{S})^{6}$; (Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}$-NMR and ${ }^{13} \mathrm{C}$-NMR data.

2-Hydroxy-1-phenylethanol 30: Enantiomeric excess and conversion determined by HPLC analysis (Chiralcel OD-H, $250 \times$ 4.6 mm column, Hexane/EtOH, $\mathrm{Et}_{2} \mathrm{NH} 95: 5: 0.1,0.5 \mathrm{~mL} / \mathrm{min}$, ketone 18.3 min , R isomer 25.4 min ., S isomer 27.2 min .); $99.5 \%$ ee $(\mathrm{S})^{7}$; Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}$-NMR and ${ }^{13} \mathrm{C}$-NMR data.

## Reduction of ketones using catalyst 11 in Water

1-Phenylethanol 14: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 $50 \mathrm{~m}, \mathrm{~T}=115{ }^{0} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $9.7 \mathrm{~min}, \mathrm{R}$ isomer 14.2 min ., S isomer 15.1 min .); $96 \%$ ee $(\mathrm{R})\left(\mathrm{lit} .{ }^{1}[\alpha]_{\mathrm{D}}{ }^{26}+45.4\right.$ (c 0.50 in $\mathrm{CHCl}_{3}$ ) $98 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(3'-Trifluoromethylphenyl)ethanol 16: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=130^{\circ} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone 8.5 min , R isomer 14.8 min ., S isomer 15.5 min .); $96 \%$ ee (R) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{26}+27.1$ (c 1.60 in $\mathrm{CH}_{3} \mathrm{OH}$ ) $96 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(4'-Methylphenyl)ethanol 33: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=125{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone 11.9 min , R isomer 14.0 min ., S isomer 14.6 min .); $[\alpha]_{\mathrm{D}}{ }^{22}+56.8$ (c 0.41 in $\left.\mathrm{CHCl}_{3}\right) 94 \%$ ee (R) (lit. ${ }^{8}[\alpha]_{\mathrm{D}}{ }^{25}+53.2$ (c 0.236 in $\mathrm{CHCl}_{3}$ ) $96 \%$ ee (R)); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.27$ ( 2 H , d (overlapping with the $\mathrm{CHCl}_{3}$ peak, $\left.J 7.9, \mathrm{Ar} H\right), 7.16(2 \mathrm{H}, \mathrm{d}, J 8.1, \mathrm{Ar} H), 4.87\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 2.34(3 \mathrm{H}, \mathrm{s}, \mathrm{ArCH} 3), 1.78(1 \mathrm{H}, \mathrm{br}$ $\mathrm{s}, \mathrm{OH}), 1.48\left(3 \mathrm{H}, \mathrm{d}, J 6.6, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 142.6,137.0,129.0,125.1(\mathrm{Ar}-\mathrm{C}), 70.1(\mathrm{CH}), 24.9\left(\mathrm{CH}_{3}\right), 20.9$ $\left(\mathrm{ArCH}_{3}\right)$.

1-(4'-Bromophenyl)ethanol 34: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=150{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone 11.6 min , R isomer 17.9 min ., S isomer 18.6 min .) ; $[\alpha]_{\mathrm{D}}{ }^{22}+33.4$ (c 0.40 in
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$\mathrm{CHCl}_{3}$ ) $93 \%$ ee (R) (lit. ${ }^{8}[\alpha]_{\mathrm{D}}{ }^{26}+32.8\left(\mathrm{c} 1.600\right.$ in $\left.\mathrm{CHCl}_{3}\right) 80 \%$ ee (R)); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.48-7.44(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$,
7.25-7.21 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H), 4.84\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 2.03(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.45\left(3 \mathrm{H}, \mathrm{d}, J 6.5^{`}, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(100.6 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 144.8(\mathrm{ArC}-\mathrm{Br}), 131.6,127.2,121.2(\mathrm{Ar}-\mathrm{C}), 69.8(\mathrm{CH}), 25.3\left(\mathrm{CH}_{3}\right)$.

1-(3'-Methoxyphenyl)ethanol 17: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin-$\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=140^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $12.9 \mathrm{~min}, \mathrm{R}$ isomer 18.1 min ., S isomer 18.7 min .); $[\alpha]_{\mathrm{D}}{ }^{22}+32.9$ (c 0.75 in $\mathrm{MeOH}) 97 \%$ ee (R) (lit. ${ }^{5}[\alpha]_{\mathrm{D}}{ }^{22}-34.9(\mathrm{c} 0.849 \mathrm{in} \mathrm{MeOH})>99 \%$ ee $\left.(\mathrm{S})\right) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 7.26\left(1 \mathrm{H}, \mathrm{dd}, J^{1}=J^{2} 8.0\right.$, $\mathrm{Ar} H)$, 6.96-6.92 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar} H$ ), 6.83-6.79 (1H, m, $\mathrm{Ar} H), 4.86\left(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{3}\right), 3.81\left(3 \mathrm{H}, \mathrm{s}, \mathrm{ArOCH}_{3}\right), 1.94(1 \mathrm{H}, \mathrm{br}$ $\mathrm{s}, \mathrm{OH}), 1.48\left(3 \mathrm{H}, \mathrm{d}, J 6.5, \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) / \mathrm{ppm}: 159.8$ ( $\mathrm{Ar} C-\mathrm{OMe}$ ), 147.6, 129.6, 117.7, 112.9, 110.9 (Ar-C), $70.4(\mathrm{CH}), 55.2\left(\mathrm{OCH}_{3}\right), 25.2\left(\mathrm{CH}_{3}\right)$.

1-(2'-Naphthyl)ethanol 35: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$ $236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=155^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $32.4 \mathrm{~min}, \mathrm{R}$ isomer 40.1 min ., S isomer 41.0 min .); $91 \%$ ee ( R ) (lit. ${ }^{1}[\alpha]_{\mathrm{D}}{ }^{25}+41.2$ (c 0.50 in EtOH) $95 \%$ ee (R)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-Phenylpropan-1-ol 22: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M$1950 \mathrm{~m}, \mathrm{~T}=115{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone 14.1 min , R isomer 21.5 min ., S isomer 22.6 min .); $96 \%$ ee ( R ). Refer to results of reductions in $\mathrm{HCO}_{2} \mathrm{H}: \mathrm{Et}_{3} \mathrm{~N}$ for $[\alpha]_{\mathrm{D}},{ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-(2'-Trifluoromethylphenyl)ethanol 20: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}-1950 \mathrm{~m}, \mathrm{~T}=120{ }^{\circ} \mathrm{C}, \mathrm{P}=10 \mathrm{psi}$, ketone $10.7 \mathrm{~min}, \mathrm{R}$ isomer 17.0 min ., S isomer 18.0 min .); $[\alpha]_{\mathrm{D}}{ }^{24}+22.9$ (c 0.46 in MeOH ) $51 \%$ ee (R) (lit. ${ }^{5}[\alpha]_{\mathrm{D}}{ }^{22}-45.5$ (c 0.661 in MeOH) $97 \%$ ee (S)); Refer to ref ${ }^{1}$ for ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

1-Cyclohexylethanol 32: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta-236 \mathrm{M}$ $1950 \mathrm{~m}, \mathrm{~T}=92{ }^{\circ} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone $25.0 \mathrm{~min}, \mathrm{R}$ isomer 40.5 min ., S isomer 41.1 min .); $84 \%$ ee ( S ). Refer to results of reductions in $\mathrm{HCO}_{2} \mathrm{H}: \mathrm{Et}_{3} \mathrm{~N}$ for $[\alpha]_{\mathrm{D}},{ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

2-Furylethanol 25: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 50m, $\mathrm{T}=85{ }^{\circ} \mathrm{C}, \mathrm{P}=15 \mathrm{psi}$, ketone $10.5 \mathrm{~min}, \mathrm{R}$ isomer 15.0 min ., S isomer 15.9.$) ; 98 \%$ ee (R). Refer to results of reductions in $\mathrm{HCO}_{2} \mathrm{H}: \mathrm{Et}_{3} \mathrm{~N}$ for $[\alpha]_{\mathrm{D}},{ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}$-NMR data.

2-Thienylethanol 26: Enantiomeric excess and conversion determined by GC analysis (Chrompac cyclodextrin- $\beta$-236M-19 $50 \mathrm{~m}, \mathrm{~T}=125{ }^{0} \mathrm{C}, \mathrm{P}=9 \mathrm{psi}$, ketone 13.2 min , R isomer 15.6 min ., S isomer 16.3 min .); $97 \%$ ee ( R ). Refer to results of reductions in $\mathrm{HCO}_{2} \mathrm{H}: \mathrm{Et}_{3} \mathrm{~N}$ for $[\alpha]_{\mathrm{D}},{ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ data.

## References

1) 
2) 
3) 
4) 
5) 
6) Nakamu, K., Matsuda, T.J. Org. Chem. 1998, 63, 8957.
7) 
8) 1801. 
1) Xu, Y.; Alcock, N. W.; Clarkson, G. J.; Docherty, G.; Woodward, G.; Wills, M. Org. Lett. 2004, 6, 4105.
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Proton NMR spectrum of Catalyst 11


CCDC 603515 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif .

Solid state structure of DM3 with and without hydrogens




## Experimental data for dm3

The asymmetric unit contains the complex. The hydrogen on the amine was located.
The angle between mean planes through the phenyl ring C18-C23 and the cyclopentenyl ring C24-C28 is 65.10 ( 0.10 ) degrees.
There are no other major factors in the crystal packing with no $\pi$ stacking and only a few long (over 3A) CH-pi interactions.

## Crystal Data

C29 H36 Cl N2 O2 Rh S, M = 615.02, Orthorhombic, space group P2(1)2(1)2(1)
$\mathrm{a}=10.0963(12), \mathrm{b}=13.7861(16), \mathrm{c}=19.923(2) \mathrm{A}$,
alpha $=90$ deg., beta $=90$ deg., gamma $=90$ deg.,
$\mathbf{U}=\mathbf{2 7 7 3 . 1}(6) \mathrm{A}^{\wedge} \mathbf{3}$ (by least squares refinement on 7222 reflection positions),
$T=180(2) \mathrm{K}$, lambda $=0.71073 \mathrm{~A}, \mathrm{Z}=4$,
$\mathrm{D}(\mathrm{cal})=1.473 \mathrm{Mg} / \mathrm{m}^{\wedge} 3, \mathrm{~F}(000)=1272$.
$\mathbf{m u}($ MoK-alpha $)=0.816 \mathrm{~mm}^{\wedge}-1$.
Crystal character:orange block.
Crystal dimensions $0.40 \times 0.30 \times 0.08 \mathrm{~mm}$,
Data Collection and Processing.
Siemens SMART (Siemens, 1994) three-circle system with CCD area detector.
The crystal was held at 180(2)
K with the Oxford Cryosystem Cryostream Cooler (Cosier \& Glazer, 1986).
Maximum theta was 29.21 deg.
The hkl ranges were -13/ 13, -18/ 18, -27/ 26.
27747 reflections measured, 6969 unique [ $R($ int $)=0.0312]$.
Absorption correction by Semi-empirical from equivalents;
minimum and maximum transmission factors: $0.6990 ; 0.9376$.
no crystal decay

## Structure Analysis and Refinement.

Systematic absences indicated space group $\mathbf{P 2 ( 1 ) 2 ( 1 ) 2 ( 1 )}$ and shown to be correct by successful refinement.

The structure was solved by direct methods using SHELXS (Sheldrick, 1990) (TREF) with additional light atoms found by Fourier methods.

Hydrogen atoms were added at calculated positions and refined using a riding model with freely rotating methyl groups except the N16-H16 which was located in a Fourier map. Anisotropic displacement parameters were used for all non-H atoms;
$\mathbf{H}$-atoms were given isotropic displacement parameters equal to $\mathbf{1 . 2}$ (or $\mathbf{1 . 5}$ for methyl and NH hydrogen atoms) times the equivalent isotropic displacement parameter of the atom to which the $\mathbf{H}$-atom is attached.
The absolute structure of the individual crystal chosen was checked by refinement of a delta-f" multiplier.
Absolute structure parameter $x=0.03(2)$.
The weighting scheme was calc $w=1 /\left[\backslash s^{\wedge} 2^{\wedge}\left(\mathrm{Fo}^{\wedge} 2^{\wedge}\right)+(0.0362 \mathrm{P})^{\wedge} \mathbf{2}^{\wedge}+1.0007 \mathrm{P}\right]$ where $\mathrm{P}=\left(\mathrm{Fo}^{\wedge} 2^{\wedge}+2 \mathrm{Fc}^{\wedge} 2^{\wedge}\right) / 3$. Goodness-of-fit on $\mathrm{F}^{\wedge} 2$ was 1.141,
R1[for 6772 reflections with
$\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})]=0.0294, \mathrm{wR} 2=0.0718$.
Data / restraints / parameters 6969/ 0/ 327.
Largest difference Fourier peak and hole 0.623 and $\mathbf{- 0 . 7 3 5}$ e. A^-3 .
Refinement used SHELXL 96 (Sheldrick, 1996).
We thank EPSRC and Siemens Analytical Instruments for grants in support of the diffractometer.

Additional material available from the Cambridge Crystallographic Data Centre comprises $\mathbf{H}$-atom coordinates, thermal parameters and the remaining bond lengths and angles.

## References

[ALCOCK, N.W. \& MARKS, P.J. (1994), J. Appl. Cryst. 27, 200-200.]
COSIER, J. \& GLAZER, A. M. (1986), J. Appl. Cryst. 19, 105-107.
SHELDRICK, G.M. (1990), Acta Cryst. A46, 467-473
SHELDRICK, G.M. (1993), Acta Cryst. D49, 18-23
SHELDRICK,G.M.(1996), SHELX-96 (beta-test) (including SHELXS and SHELXL)
SIEMENS (1994), SMART User's manual, Siemens Industrial Automation Inc,
Madison, Wis. USA.

