

Unusual Fragmentation of CH₂Cl₂ by an Ir(III) Centre Bonded to a Doubly Metalated Tp^{Ms} Ligand (Tp^{Ms} = Hydrotris(3-mesitylpyrazol-1-yl)borate)

Jorge A. López,^a Kurt Mereiter,^b Margarita Paneque,^c Manuel L. Poveda,^c Oracio Serrano,^{a,c} Swiatoslaw Trofimenko^d and Ernesto Carmona^{*c}

^a *Universidad de Guanajuato, Instituto de Investigaciones Científicas, Cerro de la Venada s/n, Col. Pueblito de Rocha, Guanajuato, México*

^b *Department of Chemistry, Vienna University of Technology, Getreidemarkt 9/164SC, A-1060 Vienna, Austria*

^c *Instituto de Investigaciones Químicas, Departamento de Química Inorgánica, Universidad de Sevilla, CSIC, Américo Vespucio s/n, Isla de la Cartuja, 41092 Sevilla, Spain. Fax: 34 954460565; Tel: 34 954489558; E-mail: guzman@us.es*

^d *Department of Chemistry and Biochemistry, University of Delaware, Newark, Delaware 19716-2522, USA*

Supplementary Information

Tp^{Ms}Ir(η⁴-CH₂=CHC(Me)=CH₂) (1).

A solution of [IrCl(coe)₂]₂ in CH₂Cl₂ (0.116 g, 0.13 mmol, 10 mL) is treated at 20 °C with an excess of 2-methylbutadiene (0.5 mL) and subsequently with TITp^{Ms} (0.200 g, 0.26 mmol). Following stirring at room temperature for 36 h, and separation of insoluble TlCl by centrifugation, all volatile components were removed *in vacuo*. Extraction with hexane and crystallization at -20 °C gave **1** as a yellow microcrystalline solid (0.170 g, 85% yield). The X-ray structure of **1** has been determined and will be reported elsewhere. Selected spectroscopic data: IR(nujol): ν(B-H) 2455 cm⁻¹. ¹³C{¹H} NMR (CDCl₃, 25 °C): δ 86.7 (Cq), 73.2 (CH), 18.0 (Me), 8.0, 5.6 (CH₂) (isoprene

ligand). $^{11}\text{B}\{^1\text{H}\}$ NMR (CDCl_3 , 25 °C): δ -3.3. Anal. Calcd. For $\text{C}_{41}\text{H}_{48}\text{BN}_6\text{Ir}\cdot 1/2$ CH_2Cl_2 : C, 57.6; H, 5.6; N, 9.6. Found: C, 57.9; H, 5.9; N, 9.0.

$\text{Tp}^{\text{Ms''}}\text{Ir}(\text{NCMe})$ (3**).**

UV photolysis of a solution of 0.25 g (0.3 mmol) of **1** in 10 mL of C_6H_6 for 3 h gave, after removal of the volatiles and washing with hexane, 0.19 g (82% yield) of the dinitrogen adduct **2**. A solution of **2** in NCMe (0.25 g, 0.32 mmol, 2 mL) was heated at 60 °C overnight. The volatiles were removed under vacuum, and the residue washed with hexane affording 0.21 g of the desired compound **3** (94% yield). Crystallization from a mixture of Et_2O :pentane: CH_2Cl_2 afforded crystals suitable for X-ray studies. Selected spectroscopic data: IR(nujol): $\nu(\text{B-H})$ 2465; $\nu(\text{C}\equiv\text{N})$ 2290 cm^{-1} . ^1H NMR (CDCl_3 , 25 °C): δ 3.71, 2.10 (d, 1 H each, $^2J_{\text{HH}} = 11.1$ Hz, IrCH_2), 2.73, 1.95 (d, 1 H each, $^2J_{\text{HH}} = 10.8$ Hz, IrCH_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 25 °C): δ 3.4 (NCMe), -0.3, -1.1 (Ir-CH_2). $^{11}\text{B}\{^1\text{H}\}$ NMR (CDCl_3 , 25 °C): δ -3.1.

$\text{Tp}^{\text{Ms'}}\text{IrCl}(\text{=C(H)pz}^{\text{Ms}})$ (5**).**

The chlorocarbene compound **4** was generated from **3** (0.050 g, 0.065 mmol) dissolved in 3 mL of CH_2Cl_2 by heating at 80 °C for 0.5 h. The resulting solution was reacted with 1 equiv of 3-mesitylpyrazol to give compound **5** as an orange microcrystalline solid, which can be chromatographed on SiO_2 . Crystallization from Et_2O :pentane: CH_2Cl_2 gave crystals suitable for X-ray studies. Selected spectroscopic data: IR(nujol): $\nu(\text{B-H})$ 2485 cm^{-1} . ^1H NMR (CDCl_3 , 25 °C): δ 15.34 (s, 1 H, $\text{Ir=CHpz}^{\text{Ms}}$), 4.43, 2.25 (d, 1 H each, $^2J_{\text{HH}} = 11.7$ Hz, IrCH_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3 , 25 °C): δ 231.5 ($\text{Ir=CHpz}^{\text{Ms}}$), 2.5 (Ir-CH_2). $^{11}\text{B}\{^1\text{H}\}$ NMR (CDCl_3 , 25 °C): δ -3.0.

