

# Supplementary Material (ESI) for Chemical Communications  
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## **Molecular switch based on a cucurbit[6]uril containing bistable [3]rotaxane**

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## GENERAL

All reagents and solvents were of the commercial reagent grade and used without further purification except where noted.

Nuclear Magnetic Resonance (NMR) spectra were recorded on Bruker Avance DPX-400 MHz nuclear magnetic resonance spectrometer. In all cases, samples were dissolved in D<sub>2</sub>O with DSS (3-(trimethyl silyl)-1-propanesulfonic acid), sodium salt as an external standard. Mass spectra were recorded in a reflection positive mode on a Bruker Reflex II MALDI-TOF MS spectrometer and the samples were measured in 2,4,6-Trihydroxyacetophenone (5% sol. in Methanol) as matrix. They were dissolved in H<sub>2</sub>O (2mg/ml), 1:1 mixed with the matrix and 0.5 µl of this mixture was add on the target.

## SYNTHESIS AND CHARACTERIZATION OF [3]ROTAXANE

CB6 (342 mg, 0.342 mmol) was dissolved in 6 M HCl (5 mL) and the resulting solution was stirred for 30 min. Alkyne **1** (60 mg, 0.342 mmol) and subsequently azide **2** (60 mg, 0.171 mmol) were added under vigorous stirring at room temperature. The resulting solution was stirred at 25 °C for 48 h. The solvent was removed under reduced pressure to obtain a colorless film, which was redissolved in water (5 mL) and precipitated into acetone (50 ml). White precipitate was collected by filtration and redissolved in water; the solution was filtered through 0.45 µm membrane filtrate. Crystallization was carried out by diffusing acetone vapor into the filtrate. The crystals were collected by filtration and dried in a vacuum oven at rt for 12 h.

(430 mg, 95 %).

M.p.: > 300 °C.

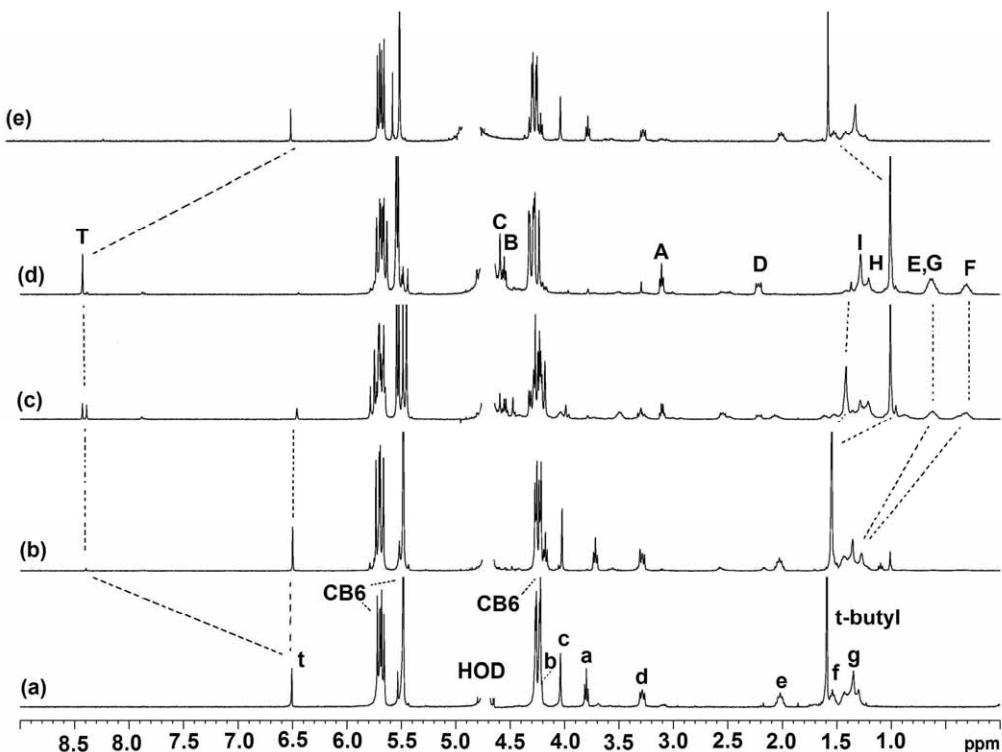
Elemental Analysis for C<sub>102</sub>H<sub>136</sub>Cl<sub>4</sub>N<sub>58</sub>O<sub>24</sub>.22 H<sub>2</sub>O

Calc.: C, 39.80; H, 6.06; N, 25.86; Found: C, 39.55; H, 5.86; N, 26.24.

IR (KBr, pellet, cm<sup>-1</sup>): 3443 (m), 2927 (w), 2353 (vw), 1735 (vs), 1473 (vs).

<sup>1</sup>H NMR (400 MHz, D<sub>2</sub>O): δ 1.3–1.5 (m, 8H, f + g), 1.6 (s, 9H, *t*-butyl), 2.1 (m, 2H, e), 3.3 (t, 2H, <sup>3</sup>J<sub>HH</sub> = 8.1 Hz, d), 3.8 (t, 2H, <sup>3</sup>J<sub>HH</sub> = 6.3 Hz, a), 4.1 (s, 2H, c), 4.15 (t, 2H, b overlapped with CB6), 4.2 (d, 12H, <sup>2</sup>J<sub>HH</sub> = 15.6 Hz, CB6), 5.45 (s, 12H, CB6), 5.94 (dd, 12H, <sup>2</sup>J<sub>HH</sub> = 15.6 Hz, CB6), 6.5 (s, 1H, t).

$^{13}\text{C}$  NMR (100 MHz,  $\text{D}_2\text{O}$ ):  $\delta$  25.2, 25.5, 28.7, 39.9, 41.8, 47.6, 49.1, 51.3 (CB6), 51.5 (CB6), 56.4, 58.6, 70.2 (CB6), 120.6 (triazole, = CH), 139.0 (triazole, = CR), 156.2 (CB6), 156.6 (CB6).



**FIGURE S1:**  $^1\text{H}$  NMR (400 MHz,  $\text{D}_2\text{O}, 25^\circ\text{C}$ ) spectra of; before the addition of base or acid (a) State I, after the addition of (b) 1 equiv of NaOH, (c) 3 equiv of NaOH, (d) 5 equiv of NaOH, (e) after addition of excess HCl and heating the solution to  $60^\circ\text{C}$  for 3 min. and cooling to rt.