Supporting Information for Palladium-Catalyzed Conjugate Reduction of Enones into α,β-Dideuterioketones with Hexamethyldisilane and Deuterium Oxide

Hidehito Otsuka, Eiji Shirakawa, and Tamio Hayashi

General Remarks. All manipulations of oxygen- and moisture-sensitive materials were conducted with a standard Schlenk technique under a nitrogen atmosphere. Nuclear magnetic resonance spectra were taken on a JEOL JNM LA-500 (¹H, 500 MHz; ¹³C, 125 MHz; ³¹P, 202 Hz) spectrometer using tetramethylsilane (¹H and ¹³C) as an internal standard and 85% phosphoric acid (³¹P) as an external standard. Elemental analyses were performed at the Microanalytical Center, Kyoto University. Unless otherwise noted, reagents are commercially available and were used without further purification. Anhydrous DMA was purchased from Aldrich Chemical Co. and was dried further with molecular sieves 3Å, which was evacuated at 260 °C for 12 h prior to use. Deuterium oxide (99.96%-d) was purchased from Cambridge Isotope Laboratories, Inc. Hexamethyldisilane (2) was purchased from Shin-Etsu Chemical Co. and was fractionally distilled. $[PdCl(\eta^3-C_3H_5)]_2$ (3), $PdH(Cl)(PPh_3)_2$ (5),² chloro(benzoylmethyl)bis(triphenylphosphine)palladium (7'),³ deuterio(dimethyl)phenylsilane (**6b-d**), ⁴ 1-phenylnon-2-en-1-one (**1a**), ⁵ 4-methyl-1-phenylpent-2-en-1-one (**1b**), ⁶ 6-chloro-1-phenylhex-2-en-1-one (1c), 7 (E)-3-(4-methoxyphenyl)-1-phenylprop-2-en-1-one (1e), ⁸ (E)-1-phenyl-3-[4-(trifluoromethyl)phenyl]prop-2-en-1-one (1f), ⁹ (E)-1-(4-methoxyphenyl)-3-phenylprop-2-en-1-one (**1g**),⁸ (*E*)-1-[4-(dimethylamino)phenyl]-3-phenylprop-2-en-1-one (1h), ¹⁰ and (E)-3-phenyl-1-[4-(trifluoromethyl)phenyl]prop-2-en-1-one (1i)⁹ were prepared according to the literature methods.

Palladium-Catalyzed Conjugate Reduction of Enones with Hexamethyldisilane and Deuterium Oxide. A General Procedure. To a solution of $[PdCl(\eta^3-C_3H_5)]_2$ (3: 3.7 mg, 0.010 mmol) and PPh₃ (10.5 mg, 0.0400 mmol) in DMA (0.50 mL) were added successively an enone (1: 0.40 mmol), hexamethyldisilane (2: 87.9 mg, 0.600 mmol) and D₂O (72 µL, 4.0 mmol). After stirring at 60 °C for 24 h, the resulting mixture was diluted with diethyl ether (20 mL), washed with water (10 mL x 5) and brine (10 mL), and dried over anhydrous magnesium sulfate. Filtration and evaporation of the solvent followed by PTLC on SiO₂ (hexane–ethyl acetate), column chromatography on SiO₂ (hexane–ethyl acetate) or bulb-to-bulb distillation gave the corresponding α , β -dideuterioketone (4). Deuterium ratios were determined by ¹H NMR. The results are summarized in Table 1.

The spectral data of all the α , β -dideuterioketones obtained here are in excellent agreement with the reported data of the corresponding ketones.



2,3-Dideuterio-1-phenyl-1-nonanone (**4a**).¹¹ A colorless oil. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 15:1). ¹H NMR (500 MHz, CDCl₃) δ 0.88 (t, J = 7.0 Hz, 3 H), 1.18–1.42 (m, 10 H), 1.67–1.77 (m, 0.80 H, β), 2.89–2.98 (m, 0.75 H, α), 7.46 (t, J = 7.8 Hz, 2 H), 7.55 (t, J = 7.5 Hz, 1 H), 7.96 (d, J = 7.1 Hz, 2 H). Anal. Calcd for C₁₅H_{19.55}D_{2.45}O: C, 81.60; H, 11.16. Found: C, 81.76; H, 10.94.



2,3-Dideuterio-4-methyl-1-phenyl-1-pentanone (4b).¹² A colorless oil. Isolated by column chromatography on SiO₂ (hexane:ethyl acetate = 25:1). ¹H NMR (500 MHz, CDCl₃) δ 0.95 (d, J = 6.6 Hz, 6 H), 1.54–1.70 (m, $\beta + \gamma$), 2.88–3.00 (m, α), 7.46 (t, J = 7.7 Hz, 2 H), 7.55 (t, J = 7.4 Hz, 1 H), 7.96 (d, J = 7.2 Hz, 2 H); (500 MHz, DMSO) δ 0.90 (d, J = 7.1 Hz, 6 H), 1.44–1.53 (m, 0.87 H, β), 1.59 (octet, J = 6.7 Hz, 0.78 H, γ), 2.94–3.03 (m, 0.56 H, α), 7.52 (t, J = 7.8 Hz, 2 H), 7.63 (t, J = 7.5 Hz, 1 H), 7.97 (d, J = 7.6 Hz, 2 H). The deuterium ratio was determined by ¹H NMR using DMSO-*d* as a solvent, as the peak of the β protons overlap with that of water in CDCl₃ as a solvent. Anal. Calcd for C₁₂H_{13.40}D_{2.60}O: C, 80.58; H, 10.48. Found: C, 80.35; H, 10.22.



6-Chloro-2,3-dideuterio-1-phenyl-1-hexanone (**4c**).¹³ A colorless oil. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 10:1). ¹H NMR (500 MHz, CDCl₃) δ 1.53 (q, J = 7.5 Hz, 2 H), 1.69–1.79 (m, 0.83 H, β), 1.83 (quint, J = 7.2 Hz, 2 H), 2.93–3.01 (m, 0.44 H, α), 3.55 (t, J = 6.7 Hz, 2 H), 7.46 (t, J = 7.7 Hz, 2 H), 7.56 (t, J = 7.4 Hz, 1 H), 7.96 (d, J = 7.0 Hz, 2 H). Anal. Calcd for C₁₂H_{12.27}D_{2.73}ClO: C, 67.53; H, 8.37. Found: C, 67.68; H, 8.30.



2,3-Dideuterio-1,3-diphenyl-1-propanone (**4d**).¹¹ A white solid. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 5:1). ¹H NMR (500 MHz, CDCl₃) δ 3.01–3.10 (m, 0.98 H, β), 3.23–3.33 (m, 0.79 H, α), 7.21 (t, *J* = 7.3 Hz, 1 H), 7.26 (d, *J* = 7.3 Hz, 2 H), 7.30 (t, *J* = 7.5 Hz, 2 H), 7.45 (t, *J* = 7.5 Hz, 2 H), 7.56 (t, *J* = 7.5 Hz, 1 H), 7.96 (d, *J* = 8.2 Hz, 2 H). Anal. Calcd for C₁₅H_{11.77}D_{2.23}O: C, 84.78; H, 7.70. Found: C, 84.94; H, 7.59.



2,3-Dideuterio-3-(4-methoxyphenyl)-1-phenyl-1-propanone (4e). ¹⁴ A white solid. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 5:1). ¹H NMR (500 MHz, CDCl₃) δ 2.96–3.03 (m, 0.91 H, β), 3.20–3.28 (m, 0.69 H, α), 3.79 (s, 3 H), 6.84 (d, *J* = 8.8 Hz, 2 H), 7.17 (d, *J* = 8.8 Hz, 2 H), 7.45 (t, *J* = 7.4 Hz, 2 H), 7.55 (t, *J* = 7.4 Hz, 1 H), 7.95 (d, *J* = 7.4 Hz, 2 H). Anal. Calcd for C₁₆H_{13.60}D_{2.40}O₂: C, 79.18; H, 7.64. Found: C, 79.09; H, 7.64.



2,3-Dideuterio-1-phenyl-3-[4-(trifluoromethyl)phenyl]-1-propanone (**4f**). ¹⁵ A white solid. Isolated by column chromatography on SiO₂ (hexane:ethyl acetate = 15:1). ¹H NMR (500 MHz, CDCl₃) δ 3.08–3.16 (m, 1.03 H, β), 3.27–3.35 (m, 0.65 H, α), 7.37 (d, *J* = 7.8 Hz, 2 H), 7.46 (t, *J* = 7.8 Hz, 2 H), 7.55 (d, *J* = 7.9 Hz, 2 H), 7.57 (t, *J* = 7.8 Hz, 1 H), 7.95 (d, *J* = 7.9 Hz, 2 H). Anal. Calcd for C₁₆H_{10.68}D_{2.32}F₃O: C, 68.49; H, 5.50. Found: C, 68.53; H, 5.72.



2,3-Dideuterio-1-(4-methoxyphenyl)-3-phenyl-1-propanone (**4g**). ¹⁶ A white solid. Isolated by column chromatography on SiO₂ (hexane:ethyl acetate = 10:1). ¹H NMR (500 MHz, CDCl₃) δ 3.00–3.08 (m, 0.95 H, β), 3.19–3.26 (m, 0.55 H, α), 3.87 (s, 3 H), 6.93 (d, *J* = 8.9 Hz, 2 H), 7.21 (t, *J* = 7.2 Hz, 1 H), 7.25 (d, *J* = 7.2 Hz, 2 H), 7.30 (t, *J* = 7.2 Hz, 2 H), 7.94 (d, *J* = 8.9 Hz, 2 H). Anal. Calcd for C₁₆H_{13.50}D_{2.50}O₂: C, 79.14; H, 7.68. Found: C, 79.06; H, 7.58.



2,3-Dideuterio-1-[4-(dimethylamino)phenyl]-3-phenyl-1-propanone (**4h**). ¹⁷ A white solid. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 5:1). ¹H NMR (500 MHz, CDCl₃) δ 3.00–3.08 (m, β), 3.06 (s, 6 H, N(CH₃)₂), 3.13–3.22 (m, α), 6.66 (d, *J* = 9.1 Hz, 2 H), 7.20 (t, *J* = 7.0 Hz, 1 H), 7.23–7.28 (m, 2 H), 7.29 (t, *J* = 7.7 Hz, 2 H), 7.88 (d, *J* = 9.1 Hz, 2 H); (500 MHz, DMSO) δ 2.83–2.91 (m, 1.04 H, β), 2.99 (s, 6 H), 3.12–3.19 (m, 0.42 H, α), 6.69 (d, *J* = 9.2 Hz, 2 H), 7.12–7.18 (m, 1 H), 7.20–7.29 (m, 4 H), 7.81 (d, *J* = 9.2 Hz, 2 H). The deuterium ratio was determined by ¹H NMR using DMSO-*d* as a solvent, as the peak of the β

protons overlap with that of the dimethylamino group in CDCl_3 as a solvent. Anal. Calcd for $\text{C}_{17}\text{H}_{16.46}\text{D}_{2.54}\text{ON}$: C, 79.79; H, 8.48. Found: C, 79.55; H, 8.21.



2,3-Dideuterio-3-phenyl-1-[4-(trifluoromethyl)phenyl]-1-propanone (**4i**). ¹⁸ A white solid. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 8:1). ¹H NMR (500 MHz, CDCl₃) δ 3.01–3.11 (m, 0.95 H, β), 3.24–3.35 (m, 0.69 H, α), 7.18–7.27 (m, 3 H), 7.31 (t, *J* = 7.4 Hz, 2 H), 7.72 (d, *J* = 8.1 Hz, 2 H), 8.05 (d, *J* = 8.1 Hz, 2 H). Anal. Calcd for C₁₆H_{10.64}D_{2.36}OF₃: C, 68.48; H, 5.51. Found: C, 68.48; H, 5.77.



3,4-Dideuterio-3-phenyl-2-butanone (**4j**).¹⁹ A colorless oil. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 8:1). ¹H NMR (500 MHz, CDCl₃) δ 2.08–2.16 [m, 1.57 H, C(=O)C<u>H</u>₃], 2.70–2.78 (m, 0.44 H, α), 2.84–2.91 (m, 1.02 H, β), 7.16–7.22 (m, 3 H), 7.28 (t, J = 7.5 Hz, 2 H). Anal. Calcd for C₁₀H_{8.03}D_{3.97}O: C, 78.92; H, 10.57. Found: C, 78.88; H, 10.34.



3,4-Dideuterio-2-nonanone (**4k**).²⁰ A colorless oil. Isolated by column chromatography on SiO₂ (hexane:ethyl acetate = 10:1) followed by bulb-to-bulb distillation (110 °C/12 mmHg). ¹H NMR (500 MHz, CDCl₃) δ 0.87 (t, *J* = 6.9 Hz, 3 H), 1.20–1.35 (m, 8 H), 1.50–1.60 (m, β), 2.08–2.14 [m, C(=O)C<u>H</u>₃], 2.34–2.43 (m, α); (500 MHz, DMSO) δ 0.90 (t, *J* = 7.0 Hz, 3 H), 1.17–1.37 (m, 8 H), 1.40–1.51 (m, 0.94 H, β), 2.05–2.12 [m, 2.80 H, C(=O)C<u>H</u>₃], 2.37–2.45 (m, 0.95 H, α). The deuterium ratio was determined by ¹H NMR using DMSO-*d* as a solvent, as the peak of the β protons overlap with that of water in CDCl₃ as a solvent. Anal. Calcd for C₉H_{15.68}D_{2.32}O: C, 74.77; H, 14.16. Found: C, 74.98; H, 14.04.



1,2,4,5-Tetradeuterio-1,5-diphenyl-3-pentanone (**4**).²¹ A white solid. Isolated by PTLC on SiO₂ (hexane:ethyl acetate = 5:1). ¹H NMR (500 MHz, CDCl₃) δ 2.62–2.72 (m, 0.78 H, α), 2.82–2.90 (m, 2.0 H, β), 7.15 (d, J = 7.4 Hz, 4 H), 7.18 (tt, J = 7.6, 1.9 Hz, 2 H), 7.27 (t, J = 7.5 Hz, 4 H). Anal. Calcd for C₁₇H_{12,78}D_{5,22}O: C, 83.83; H, 9.60. Found: C, 84.02; H, 9.33.

The Assignment of the Peaks of α - and β -Methylenes in ¹H NMR.

4d-*h*, 4e-*h*, 4f-*h*, 4g-*h*, 4h-*h* and 4i-*h*: On treatment of 1,3-diphenyl-1-propane (4d-*h*: 20.3 mg, 0.965 mmol) with NaH (11.1 mg, 56% in oil, 0.259 mmol) and D₂O (1.0 mL) in THF (0.50 mL) at room temperature for 18 h, the triplet at 3.31 ppm (J = 7.7 Hz) had disappeared, whereas the triplet at 3.07 ppm (J = 7.7 Hz) had changed to a broad singlet. Consequently, the triplets at 3.31 ppm and 3.07 ppm were assigned to the α - and β -methylenes, respectively. The chemical shifts of the α - and β -methylenes of the other chalcon derivatives (4e-*h*, 4f-*h*, 4g-*h*, 4h-*h* and 4i-*h*) were assigned based on those of 4d-*h*. 4a-*h*, 4b-*h*, 4c-*h*, 4j-*h*, 4k-*h* and 4l-*h*: The chemical shifts of the α - and β -methylenes of 4j-*h* are assigned according to the literature. Those of 4a-*h*, 4b-*h* and 4c-*h* were assigned based on those of 4k-*h*, whereas the assignment was done for 4l-*h* based on 4j-*h*.

The Reaction of a **a**-Palladioacetophenone with Deuterio(dimethyl)phenylsilane. A solution of chloro(benzoylmethyl)bis(triphenylphosphine)palladium (**7**': 7.9 mg, 0.010 mmol), 1,8-bis(dimethylamino)naphthalene (**10**: 6.5 mg, 0.030 mmol) and *p*-dimethoxybenzene as an internal standard (2.6 mg) in DMF- d_7 (0.60 mL) was stirred at room temperature. To the resulting mixture were added a solution of deuterio(dimethyl)phenylsilane (**6b**-*d*: 2.1 mg, 15 mmol) in DMF- d_7 (20 µL). After the mixture was stirred at room temperature for 1 h, 1-(dimethylphenylsiloxy)-1-phenylethene **9a**, α -deuterioacetophenone (**11**) and acetophenone (**12**) were determined by ¹H NMR.

References

- 1 W. T. Dent, R. Long and A. J. Wilkinson, J. Chem. Soc., 1964, 1585–1588.
- 2 K. Kudo, M. Hidai, T. Murayama and Y. Uchida, J. Chem. Soc., Chem. Commun., 1970, 1701–1702.
- 3 P. Veya, C. Floriani, A. Chiesi-Villa and C. Rizzoli, *Organometallics*, 1993, **12**, 4899–4907.
- 4 A. Mori, A. Fujita, H. Kajiro, Y. Nishihara and T. Hiyama, *Tetrahedron*, 1999, **55**, 4573–4582.
- 5 K. Saigo, K. Kawata and T. Mukaiyama, *Chem. Lett.*, 1976, 771–772.
- 6 J. M. Chong, L. Shen and N. J. Taylor, J. Am. Chem. Soc., 2000, **122**, 1822–1823.
- 7 R. A. Bunce and J. C. Allison, Synth. Commun., 1999, 29, 2175–2186.
- 8 R. M. Kellogg, J. W. Nieuwenhuijzen, K. Pouwer, T. R. Vries, Q. B. Broxterman, R. F. P. Grimbergen, B. Kaptein, R. M. La Crois, E. de Wever, K. Zwaagstra and A. C. van der Laan, *Synthesis*, 2003, 1626–1638.
- 9 G. Dannhardt, W. Kiefer, G. Krämer, S. Maehrlein, U. Nowe and B. Fiebich, *Eur. J. Med. Chem.*, 2000, **35**, 499–510.
- 10 M. Matsui, A. Oji, K. Hiramatsu, K. Shibata and H. Muramatsu, J. Chem. Soc., Perkin Trans. 2, 1992, 201–206.
- 11 D. Wang and Z. Zhang, Org. Lett., 2003, 5, 4645–4648.
- 12 C. S. Cho, J. Mol. Cat. A, 2005, 240, 55-60.
- 13 T. Shimada and Y. Yamamoto, J. Am. Chem. Soc., 2002, 124, 12670-12671.
- 14 M. Hofmann, N. Hampel, T. Kanzian and H. Mayr, Angew. Chem. Int. Ed., 2004, 43, 5402–5405.

- 15 E. Díez-Barra, A. de la Hoz, A. Loupy, A. Martínez-González, V. Martínez-Merino, S. Merino, R. Paugam, P. Sánchez-Verdú, J. Sansoulet and J. Torres, *Tetrahedron*, 1997, 53, 3659–3668.
- 16 L. J. Gooßen and K. Ghosh, Eur. J. Org. Chem., 2002, 3254–3267.
- 17 H. Tatamidani, K. Yokota, F. Kakiuchi and N. Chatani, J. Org. Chem., 2004, 69, 5615–5621.
- 18 M. S. Kwon, N. Kim, S. H. Seo, I. S. Park, R. K. Cheedrala and J. Park, Angew. Chem. Int. Ed., 2005, 44, 6913–6915.
- 19 J. Kubota, T, Ido, M. Kuroboshi, H. Tanaka, T. Uchida and K. Shimamura, *Tetrahedron*, 2006, **62**, 4769–4773.
- 20 A. Arase, M. Hoshi and Y. Masuda, Bull. Chem. Soc. Jpn., 1984, 57, 209–213.
- 21 J. -F. Margathe, M. Shipman and S. C. Smith, Org. Lett., 2005, 7, 4987–4990.
- 22 E. Keinan and N. Greenspoon, J. Am. Chem. Soc., 1986, 108, 7314–7325.