

Effective Gold(III)–CuCl₂-catalyzed addition of alcohols to alkenes

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Experimental details

Catalytic testing was performed in a closed glass reactor (2.5 mL, SUPELCO) equipped with a micro-syringe through which can take a sample to analysis. For each reaction, the gold catalyst (0.10 mmol) and/or a proper amount (without specified, the amount is two equal mole of gold, i.e., 0.20 mmol) of CuCl₂ was added into 1.5 mL of the mixtures of alcohol and alkene (1.21 mmol). Then the reactor was closed tightly and the reactants mixed in the room temperature. The reaction was assumed to start when the reactor was put into the oil bath at 120 °C (unless otherwise specified) and stirred extensively. At a desired reaction time, the reaction product was taken and analyzed with a GC.

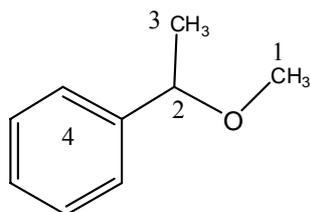
Commercially available reagents including AuCl₃ (99.99%), H₂AuCl₄·3H₂O (99.9+%), AuCl (99.9+%), PdCl₂ (99.9%) were purchased from Sigma-Aldrich. CuCl₂·2H₂O (98%, Panreac), methanol (99.9+%, Merck), ethanol (99.9%, Scharlau), 1-propanol (99.5+%, Sigma-Aldrich), 2-propanol (99.9%, Scharlau), 1-butanol (99.4+%, Sigma-Aldrich), 2-butanol (99%, Aldrich), tert-butyl alcohol (99+%, Aldrich) were used as received without further purification.

GC/MS and $^1\text{H-NMR}$ Analysis

GC/MS analyses were performed on a Agilent 5973N spectrometer equipped with the same column and in the same conditions as GC.

Flash column chromatograph was performed over silica gel (0.04–0.06 mm, Scharlau) using the mixtures of hexane and ethyl acetate (hexane/ethyl acetate = 95: 1) as effluent. $^1\text{H-NMR}$ spectra were recorded in CDCl_3 (or CD_3OD) with TMS as an internal standard at ambient temperature on a Bruker Avance 300 operating at 300 MHz.

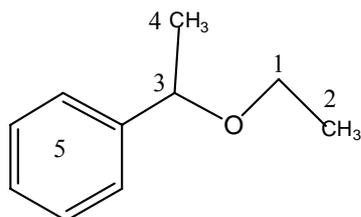
1-methoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.2 (m, 5H, C_6H_5), 4.3 (q, 1H, C2-H), 3.2 (s, 3H, C1-H), 1.5 (d, 3H, C3-H)

MS (80 eV): m/z (%) = 136 (3) (M^+), 121 (100), 105 (29), 77 (20), 51(10), 91(10)

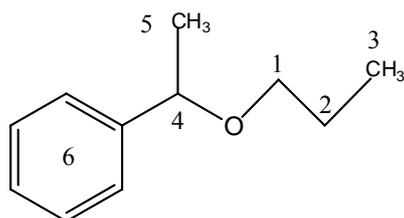
1-ethoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.3 (m, 5H, C_6H_5), 4.2–4.4 (q, 1H, C3-H), 3.1–3.3 (q, 2H, C1-H), 1.5 (d, 3H, C4-H), 1.1 (t, 3H, C2-H)

MS (80 eV): m/z (%) = 150 (2) (M^+), 135 (100), 105 (100), 77 (55), 79(47), 43(29)

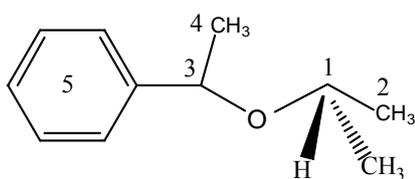
1-propoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.3 (m, 5H, C_6H_5), 4.2–4.4(q, 1H, C4-H), 3.1–3.3 (t, 2H, C1-H), 1.5 (d, 3H, C5-H), 1.2–1.3 (m, 2H, C2-H), 0.8 (t, 3H, C3-H)

MS (80 eV): m/z (%) = 164 (0.5) (M^+), 107 (100), 105 (88), 149 (73), 79(38), 43(29)

1-(2-methyl-ethoxyethyl)-benzene

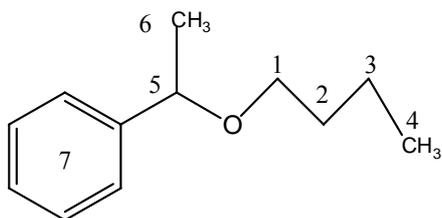


(Contain diastereomeric pairs)

$^1\text{H-NMR}$ δ : 7.2 (m, 5H, C_6H_5), 4.3–4.4(q, 1H, C3-H), 3.2–3.3 (q, 1H, C1-H), 1.5 (d, 3H, C4-H), 1.2 (d, 3H, C2-H)

MS (80 eV): m/z (%) = 164 (0.3) (M^+), 107 (100), 105 (92), 149 (44), 79(38), 77(30)

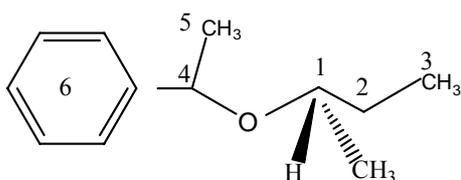
1-butoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.3 (m, 5H, C_6H_5), 4.3–4.4 (q, 1H, C5-H), 3.2–3.5 (t, 2H, C1-H), 1.5 (d, 3H, C6-H), 1.2 (m, 2H, CH_2 (2-3)), 0.8 (t, 3H, C4-H)

MS (80 eV): m/z (%) = 178 (Tr) (M^+), 107 (100), 105 (88), 163 (74), 79 (28), 77 (21)

1-(2-methyl-propoxyethyl)-benzene

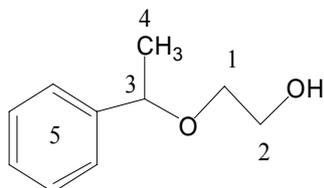


(Contain diastereomeric pairs)

$^1\text{H-NMR}$ δ : 7.3 (m, 5H, C_6H_5), 4.3–4.4 (q, 1H, C4-H), 3.0 (m, 1H, C1-H), 1.5 (d, 3H, C5-H), 1.2 (d, 3H, C1- CH_3), 1.3–1.4 (q, 2H, C2-H), 0.8 (t, 3H, C3-H)

MS (80 eV): m/z (%) = 178 (Tr) (M^+), 105 (100), 107 (36), 163 (23), 106 (16), 79(13)

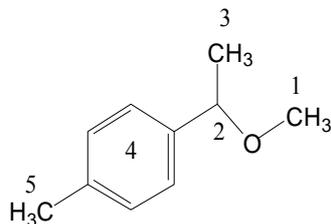
2-(1-phenylethoxy)-ethanol



$^1\text{H-NMR}$ δ : 7.2 (m, 5H, C_6H_5), 4.3–4.4 (q, 1H, C3-H), 3.4–3.5 (t, 2H, C1-H), 3.2 (s, 1H, C2-OH), 3.7 (t, 2H, C2-H), 1.5 (d, 3H, C4-H)

MS (80 eV): m/z (%) = 166 (13) (M^+), 105 (100), 151 (41), 79 (25), 77 (23), 121 (21)

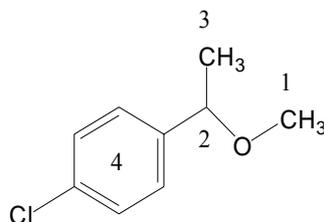
4-methyl-1-methoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.1 (m, 5H, C₆H₅), 4.3 (q, 1H, C2-H), 3.2 (s, 3H, C1-H), 1.5 (d, 3H, C3-H), 2.4 (s, 3H, C5-H)

MS (80 eV): m/z (%) = 150 (8) (M⁺), 135 (100), 91 (60), 119 (41), 136 (15), 117 (13)

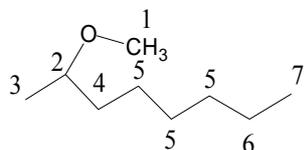
4-chloro-1-methoxyethylbenzene



$^1\text{H-NMR}$ δ : 7.1–7.2 (m, 5H, C₆H₅), 4.3 (q, 1H, C2-H), 3.2 (s, 3H, C1-H), 1.5 (d, 3H, C3-H)

MS (80 eV): m/z (%) = 170 (5) (M⁺), 155 (100), 157 (33), 139 (26), 91 (25), 103 (24)

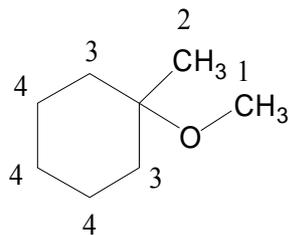
2-methoxyoctane



$^1\text{H-NMR}$ δ : 3.2 (s, 3H, C1-H), 3.0 (m, 1H, C2-H), 1.4 (q, 2H, C4-H), 1.29 (m, 2H, C5-H), 1.33 (m, 2H, C6-H), 1.2 (d, 3H, C3-H), 1.0 (t, 3H, C7-H)

MS (80 eV): m/z (%) = 144 (Tr) (M⁺), 59 (100), 55 (23), 60 (8), 58 (8), 112 (1)

1-methyl-1-methoxycyclohexane



¹H-NMR δ: 3.1 (s, 3H, C1-H), 1.5–1.6 (t, 2H, C3-H), 1.2–1.4 (t, 2H, C4-H),
1.1 (s, 3H, C1-H),

MS (80 eV): m/z (%) = 128 (48) (M⁺), 85 (100), 72 (92), 55 (88), 113 (66), 81 (39)