

Supporting Information

Dendron-Stabilized Palladium Nanoparticles: Effect of Generation and Peripheral Groups on Particle Size and Hydrogenation Activity

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Syntheses of G_n Py-COOMe (n = 1, 2, 3) dendrons

Methyl acrylate (12.9 g, 150 mmol) was added to a methanol solution of 4-picolylamine (5.4 g, 50 mmol). The mixture was stirred for 120 h, then concentrated under reduced pressure. The residue was purified by dialysis (Spectra/Por® CE MWCO: 100) in MeOH, and dried under vacuum at 40 °C for 24 h to give 13.2 g (94 %) of G₁ Py-COOMe dendron as light orange oil. G₂ and G₃ Py-COOMe dendrons were also synthesized from G₁ and G₂ Py-NH₂ dendrons according to the above procedures, respectively.

Syntheses of G_n Py-C_m (n = 1, 2, 3, m = 6, 12) and Py-NH₂ (n = 1, 2) dendrons

To a methanol solution of G₁ Py-COOMe (2.8 g, 10 mmol) was added *n*-hexylamine (15.2 g, 150 mmol) slowly at -75 °C and the mixture was stirred at room temperature for 72 h. The combined solution was concentrated under reduced pressure, and the residue was purified by dialysis

(Spectra/Por® CE MWCO: 100) in MeOH, then dried under vacuum at 40 °C for 24 h to give 4.19 g (95 %) of G₁ Py-C₆ dendron as light orange oil. According to the above procedure, G₂ and G₃ Py-C₆ dendrons were also synthesized from G₂ and G₃ Py-COOMe, respectively. Use of *n*-docecylamine in place of *n*-hexylamine or ethylene diamine could afford G_n Py-C₁₂ or G_n Py-NH₂ dendrons, respectively.

Selected data for G₂ Py-C₆ and G₂ Py-C₁₂:

G₂ Py-C₆: ¹H NMR (270 MHz, *d*₆-DMSO, TMS, 30 °C): δ8.47 (2H, d, aromatic CH of 4-picolylamine), 7.82 (6H, br, NH), 7.30 (2H, d, aromatic CH of 4-picolylamine), 3.58 (2H, s, CH₂ of 4-picolylamine), 3.20 (4H, br, NHCH₂CH₂N), 3.10 (8H, m, CH₂(CH₂)₄CH₃), 2.66 (12H, br, NCH₂CH₂CO), 2.54 (4H, m, NHCH₂CH₂N), 2.31 (12H, br, NCH₂CH₂CO), 1.36 (8H, m, CH₂(CH₂)₃CH₃), 1.23 (24H, m, (CH₂)₃CH₃), 0.87 (12H, m, (CH₂)₅CH₃). ¹³C{¹H} NMR (68 MHz, *d*₆-DMSO, TMS, 30 °C): δ170.6 (CONH), 148.9 (aromatic carbon of 4-picolylamine), 128.3 (aromatic carbon of 4-picolylamine), 123.1 (aromatic carbon of 4-picolylamine), 56.0 (CH₂ of 4-picolylamine), 51.9 (NHCH₂CH₂N), 49.5 (NCH₂CH₂CO), 41.5 (CH₂(CH₂)₄CH₃), 36.8 (NHCH₂CH₂N), 33.2 (NCH₂CH₂CO), 31.1 (CH₂(CH₂)₃CH₃), 28.9 (CH₂(CH₂)₂CH₃), 26.0 (CH₂CH₂CH₃), 21.9 (CH₂CH₃), 13.7 ((CH₂)₅CH₃). IR (KBr): 3079 (ν_{N-H}), 2956, 2929, 2857 (ν_{C-H}), 1645 (ν_{C=O}, amide), 1552 (δ_{N-H}, amide). MS m/z (ESI) 968.1 (M-H⁺).

G₂ Py-C₁₂: ¹H NMR (270 MHz, *d*₆-DMSO, TMS, 30 °C): δ8.46 (2H, d, aromatic CH of 4-picolylamine), 7.80 (6H, br, NH), 7.28 (2H, d, aromatic CH of 4-picolylamine), 3.57 (2H, s, CH₂ of 4-picolylamine), 3.18 (4H, br, NHCH₂CH₂N), 3.01 (8H, m, CH₂(CH₂)₄CH₃), 2.63 (12H, br, NCH₂CH₂CO), 2.52 (4H, m, NHCH₂CH₂N), 2.25 (12H, br, NCH₂CH₂CO), 1.36 (8H, m, CH₂(CH₂)₃CH₃), 1.22 (72H, m, (CH₂)₃CH₃), 0.85 (12H, m, (CH₂)₅CH₃). ¹³C{¹H} NMR (68 MHz,

d_6 -DMSO, TMS, 30 °C): δ 170.9 (CONH), 149.0 (aromatic carbon of 4-picolylamine), 130.0 (aromatic carbon of 4-picolylamine), 123.2 (aromatic carbon of 4-picolylamine), 56.4 (CH₂ of 4-picolylamine), 51.8 (NHCH₂CH₂N), 49.6 (NCH₂CH₂CO), 38.3 (CH₂(CH₂)₁₀CH₃), 36.7 (NHCH₂CH₂N), 33.3 (NCH₂CH₂CO), 32.2 (CH₂(CH₂)₉CH₃), 31.3 (CH₂CH₂CH₃), 29.0 ((CH₂)₄(CH₂)₃CH₃), 28.9 (CH₂(CH₂)₇CH₃), 28.8 (CH₂(CH₂)₂CH₃), 26.4 (CH₂(CH₂)₈CH₃), 22.1 (CH₂CH₃), 13.9 ((CH₂)₁₁CH₃). IR (KBr): 3055 ($\nu_{\text{N-H}}$), 2924, 2853 ($\nu_{\text{C-H}}$), 1644 ($\nu_{\text{C=O}}$, amide), 1556 ($\delta_{\text{N-H}}$, amide). MS m/z (ESI) 1294.5 (M-H⁺).

General procedure for preparation of the dendritic Pd(0) nanoparticles: Typical procedures for the preparation of the dendritic Pd nanoparticles [G_n Py-C₆ Pd(0) (n = 1, 2, 3)] are as follows. To a CH₂Cl₂ solution (6 mL) of G_n Py-C₆ (0.164 mmol) was added [PdCl(C₃H₅)₂] (0.010 mmol), and stirred vigorously at room temperature for 1 h. The solution was treated with LiB(C₃H₅)₃H (0.020 mmol) and stirred at room temperature for 1 h.

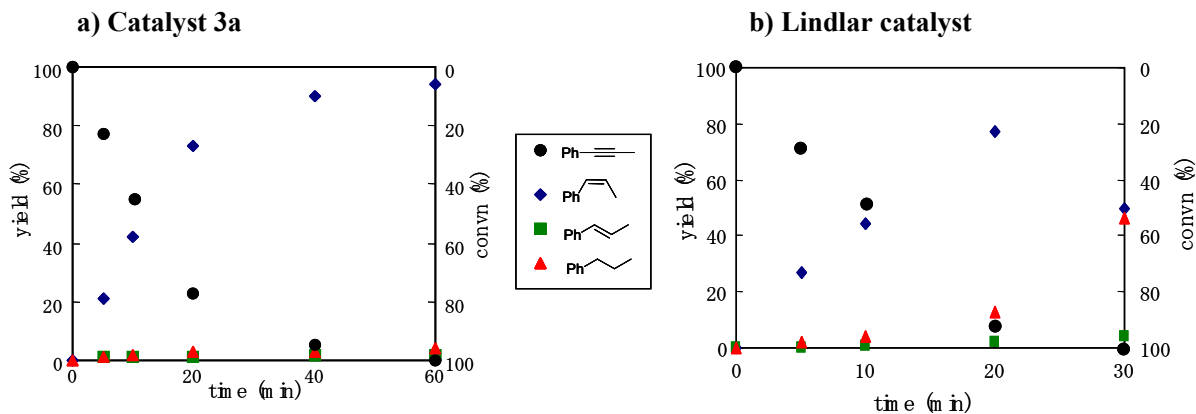
Hydrogenation of olefins and acetylenes catalysed by the dendritic Pd nanoparticles: A side-armed flask attached to a gas burette and a manometer was evacuated and filled with molecular hydrogen, followed by addition of 3 mL of Pd solutions (Pd: 5.0 μ mol) and stirred for half an hour at 25 °C. Hydrogen uptake was measured just after the addition of substrates. The products were determined by GC and GC-MS using an internal standard method.

Competitive hydrogenation of acetylenes catalysed by the dendritic Pd nanoparticles:

G₂ Py-C₆ Pd(0) (Pd : 10 μ mol) in CH₂Cl₂ (3 mL) was stirred at -25 °C for 0.5 h under hydrogen atmosphere. To this Pd solution added a CH₂Cl₂ solution (1 mL) of an equimolar mixture of 3-phenyl-2-propyn-1-ol (0.5 mmol) and 1-phenyl-1-propyne (0.5 mmol). The products were determined by GC-MS using an internal standard method.

Timecourse of the Hydrogenation in Scheme 1

Hydrogenation of 1-Phenyl-1-propyne



Hydrogenation of 3-Phenyl-2-propyn-1-ol

