Supporting Information

A 3D Chiral Nanoporous Coordination Framework Consisting of Homochiral Nanotubes Assembled from Octuple Helices

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Institute of Functional Materials, Department of Chemistry and Key Laboratory of Polyoxometalates Science of Ministry of Education, College of Chemistry, Northeast Normal University, Changchun 130024, China. Email: <u>zmsu@nenu.edu.cn</u> <u>wangenbo@public.cc.jl.cn</u> **X-ray structure analysis:** $C_{31}H_{38}N_3O_8Cd$: $M_r = 693.04$, hexagonal, space group $P6_122$, a = b = 26.712(4) Å, c = 15.919(3) Å, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$, V = 9837(3) Å³, Z = 12, $\rho_{calcd} = 1.404$ mg m⁻³, Flack parameter = -0.01(3), final R1 = 0.0384 for 9677 independent reflections [$I > 2\sigma(I)$]. The data were collected on a Bruker Apex CCD diffractometer at 133(2) K with graphite-monochromated Mo_{Ka} radiation ($\lambda = 0.71073$ Å).

The data were collected on a Bruker Apex CCD diffractometer at 133(2) K with graphite-monochromated Mo_{Ka} radiation ($\lambda = 0.71073$ Å). The structures were solved by direct method and refined by full-matrix least-squares methods with SHELXL.^[1] The anion and DMA molecules were highly disordered and could not be modeled properly, thus the SQUEEZE routine, a part of the PLATON package of crystallographic software, was applied to calculate the solvent disorder area and remove its contribution to the overall intensity data. The refinements of the guest-free structure on the squeezed data gave the final R1 = 0.0384 and wR2 = 0.0798 (R1 =0.0463 and wR2 = 0.1322 without employing PLATON/SQUEEZE). The final formula was calculated from the SQUEEZE results combined with the following characterization techniques. Despite the disorder, identification of the guest molecules is readily accomplished by ¹H NMR spectroscopy. The ¹H NMR spectrum of **1** recorded in D₂O exhibits a set of well-resolved proton signals. The signals at 2.88, 2.73 and 1.91 ppm attributed to three CH₃ groups of DMA,^[2] while the signal at 2.54 ppm ascribes to the CH₃ group of dimethylamine. Furthermore, NMR analysis reveals that the relative molar ratio of dimethylamine and DMA is approximately 1:2. The IR

spectrum of **1**, as expected, exhibits a sharp band at 1619 cm⁻¹ corresponding to $\tilde{v}_{C=O}$ stretching frequency, which is indicative of DMA molecule. According to the previous literature, the peaks of 1540 and 1398 cm⁻¹ are attributed to the asymmetric and symmetric stretching vibrations of the carboxylate group of bpdc ligand.^[3] The peak at ca. 3419 cm⁻¹ is attributed to the N-H absorption vibration of dimethylamine. Further information supporting the formula of **1** is obtained by thermogravimetric analysis, elemental analysis and the consideration of charge balance.

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S1-1. Section of the X-ray crystal structure of **1**, showing the coordination environments of Cd^{2+} ions and their connectivity with bpdc anions. Color code: C black; O red; Cd green.



SI-2. Coordination modes of the bpdc ligands in the structure of **1**. bpdc^A: chelating bis(bidentate), bpdc^B: bridging bis(bidentate). Color code: C black; O red; Cd green.

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SI-3. An angle of 123.92° formed between the planes defined by the two C-O-Cd-O-C chelate rings at one metal center. Color code: C black; O red; Cd green.



SI-4. Perspective view of the trigonal channel that is surrounded by bpdc^B ligands. Color code: C black; O red; Cd green.



SI-5. The ¹H NMR spectrum of **1** recorded in D_2O .



SI-6. FT-IR spectrum of as-synthesized **1**.

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SI-7. TG curve of 1.