

**Electronic Supplementary Information (ESI)**

**Supramolecular Single-Walled Carbon Nanotubes (SWCNTs) Network Polymer Made by Hybrids of SWCNTs and Water Soluble Calix[8]arenes**

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**Experimental**

**Materials.** All solvents and reagents were used as supplied except the following. Milli-Q water was used for preparation of aqueous solutions. As-prepared (AP) HiPco SWCNT ( $d \sim 1$  nm) was purchased from Carbon Nanotechnologies, Inc.

**Measurements.** The  $^1\text{H}$  NMR spectra were recorded at 270 MHz with a JEOL-JNM EX270 spectrometers. UV-Vis absorption spectra were recorded with a JASCO V-630 spectrophotometer. Dynamic light scattering (DLS) measurements were carried out on Beckman Coulter N5 Submicron Particle Size Analyzer at 20 °C. Tapping mode atomic force microscopy (TM-AFM) was taken on multimode SPA 400 (SEIKO Instruments). Nanoprobe cantilevers (SI-DF20, SEIKO Instruments) were utilized. The sample was prepared by slow evaporation on substrate overnight at room temperature. Mica was used as a substrate for TM-AFM measurement. Transmission electron microscopy (TEM) image was performed using a JEM-1025 operated at an accelerating voltage of 100 kV. TEM sample was prepared by depositing one drops of desired solution on a 100 mesh copper grid covered with a carbon film and removing the excessive solution with Kimwipes wipers, and drying the TEM grid at room temperature.

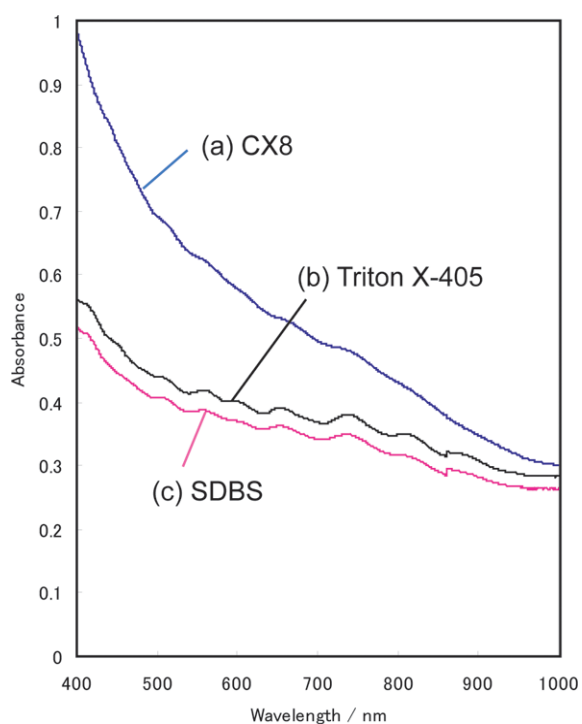
**Purification of SWCNTs.** The purification of HiPco SWCNTs followed literature method.<sup>1</sup>

**Adamantanetrimethylammonium Iodide (AdTriMe), Guest Dimer and Adamantane carbomethylpyridinium bromide (AdPy)** were prepared according to previous papers.<sup>2,3</sup>

**Solubilization of SWCNTs with *p*-Sulfonatocalixarenes.** *p*-Sulfonatocalixarene (0.0269

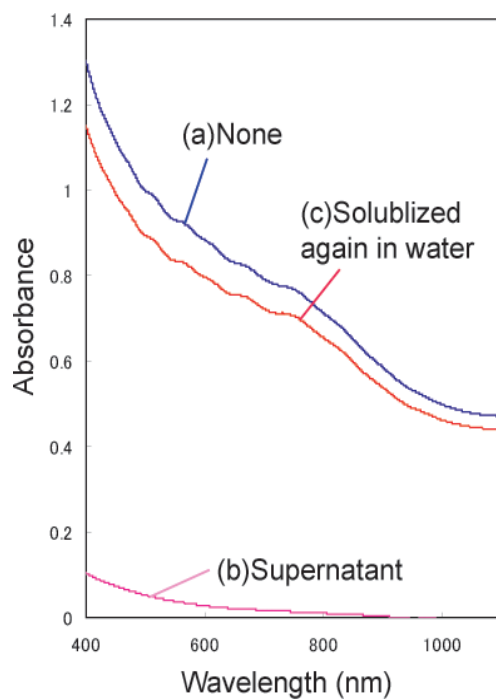
mmol) was dissolved in aqueous solution (5 mL). To the mixture, purified SWCNTs (1 mg) were added and sonicated in low-energy ultrasonic bath (Bransonic 2510) for 3 h, followed by centrifugation.

### UV-Vis Spectra of SWCNTs Suspended in CX8, Triton X-405 and SDBS



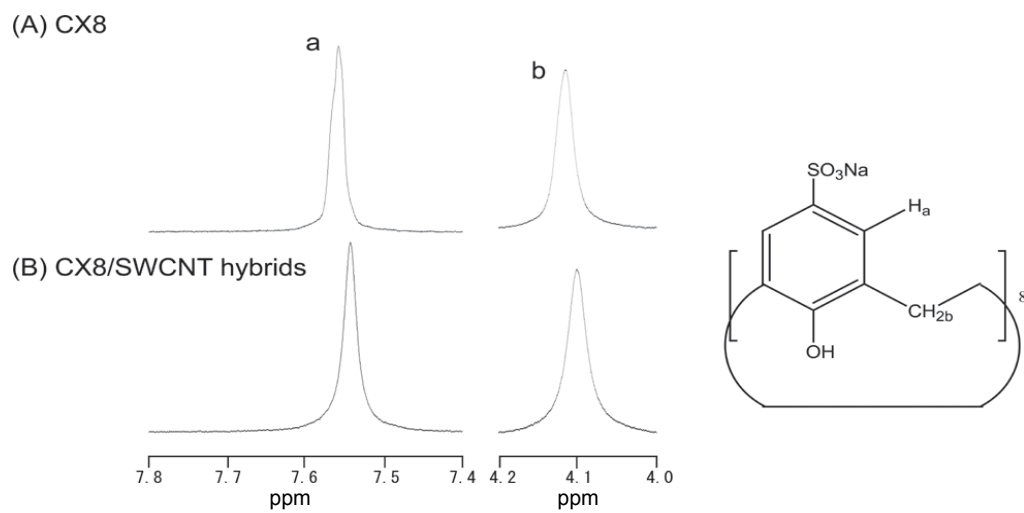
**Figure 1S.** UV-Vis spectra of the aqueous supernatant (5 mL) containing SWCNTs with (a) **CX8**, (b) **Triton X-405** and (c) sodium dodecylbenzene sulfonate (**SDBS**). Applied concentration of **CX8**, **Triton X-405** and **SDBS** was 5.38 mM. Solubility of SWCNTs with **CX8**, **Triton X-405** and **SDBS** was  $2.94 \times 10^{-2}$ ,  $1.87 \times 10^{-2}$  and  $1.74 \times 10^{-2}$  mg/mL, respectively.

**UV-Vis Spectra of CX8/SWCNT hybrids upon Addition of Salts**



**Figure 2S.** UV-Vis spectra of (a) **CX8/SWCNT hybrids**, (b) the supernatant of **CX8/SWCNT hybrids** with KCl salts (150 eq. to **CX8**) and (c) re-dissolved **CX8/SWCNT hybrids** in water.

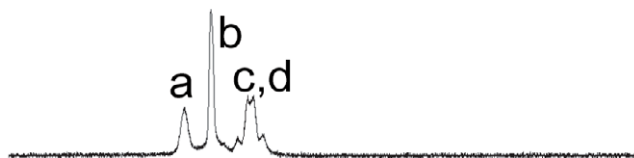
**$^1\text{H}$  NMR Spectra of CX8 and CX8/SWCNT hybrids**



**Figure 3S.**  $^1\text{H}$  NMR spectra of (A) CX8 (2.66 mM), (B) CX8/SWCNT hybrids (Concentration of CX8 was 2.66 mM) in  $\text{D}_2\text{O}$  at 25 °C.

**$^1\text{H}$  NMR Spectra of CX8/SWCNT hybrids with AdPy**

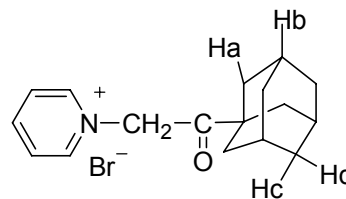
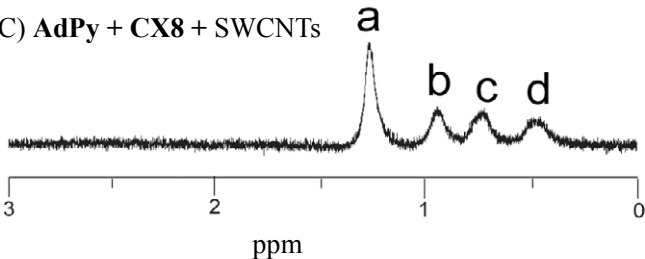
(A) AdPy



(B) AdPy + CX8

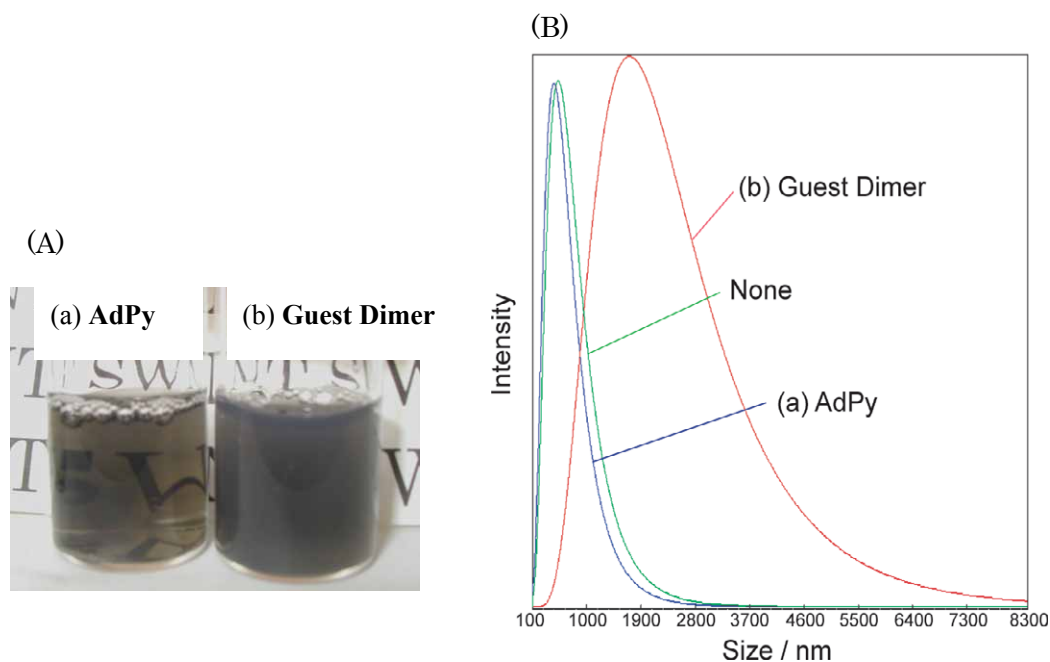


(C) AdPy + CX8 + SWCNTs



**Figure 4S.**  $^1\text{H}$  NMR spectra of (A) AdPy (2.66 mM), (B) AdPy (2.66 mM) with CX8 (2.66 mM) and (C) a supernatant containing SWCNTs solubilized by CX8 (2.66 mM) with AdPy (2.66 mM) in  $\text{D}_2\text{O}$  at 25  $^\circ\text{C}$ . Signals of adamantyl group of AdPy were largely shifted toward downfield with CX8.

### Dynamic Light Scattering (DLS) Measurements



**Figure 5S.** (A) Photographs of **CX8/SWCNT hybrids** upon addition of (a) **AdPy** and (b) **Guest Dimer**. Black suspensions were observed upon addition of **Guest Dimer**. (B) Size distributions obtained by dynamic light scattering: **CX8/SWCNT hybrids** by adding (a) **AdPy** and (b) **Guest Dimer**.

### References

- 1) W. Zhou, Y. H. Ooi, R. Russo, P. Papanek, D. E. Luzzi, J. E. Fischer, M. J. Bronikowski, P. A. Willis and R. E. Smally, *Chem. Phys. Lett.*, 2001, **350**, 6.
- 2) S. Shinkai, K. Arai, T. Matsuda and O. Manabe, *Bull. Chem. Soc. Jpn.*, 1989, **62**, 3856.
- 3) K. Ohga, Y. Takashima, H. Takahashi, Y. Kawaguchi, H. Yamaguchi and A. Harada, *Macromolecules*, 2005, **38**, 5897.