

Supporting Information for:

Copper(I) Mediated Oligomerisation of a Phosphaalkyne

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S1. Removal of the CuI matrix from 2

S2. ^{31}P -NMR spectrum of the resulting mixture

S1. Removal of the CuI from 2

A solution of NaCN (800 mg, 16.3 mmol) in 20 ml H₂O was layered with 7 ml Et₂O and the mixture was thoroughly degassed by bubbling dinitrogen through it for 30 minutes. Solid 2·x CH₃CN (100 mg, 0.1 mmol) was added and the mixture was vigorously stirred for 5 hours. After that time all the solid was dissolved and the Et₂O phase showed a pale yellow colour. Separation of the organic phase and removal of all volatiles yielded a yellow, oily residue (ca. 30 mg, 70 % based on C₂₀H₂₇P₅) from which ³¹P{¹H}-NMR data was recorded in C₆D₆ (see S2). On storage of the mixture at -30 °C for one month a few crystals of **6** were obtained, which were suitable for X-Ray crystallography (see S3). Yield: 30 mg (73%)

S2. ^{31}P -NMR spectrum of the resulting mixture

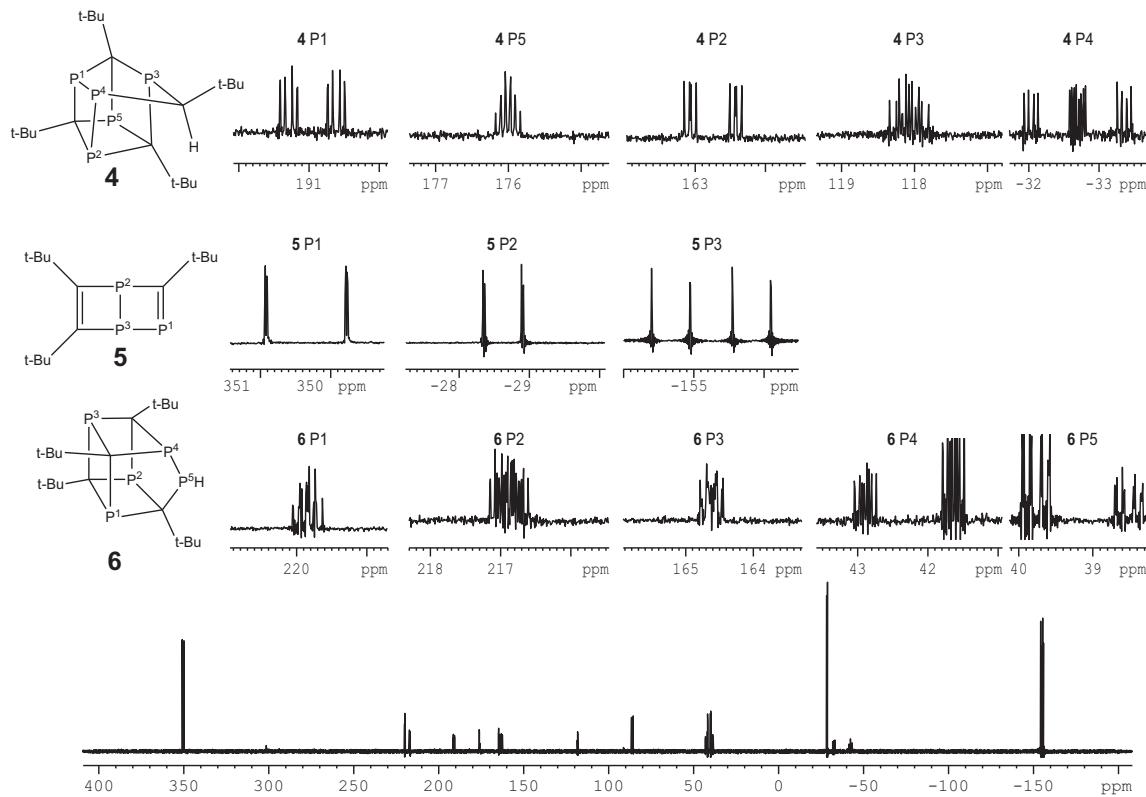
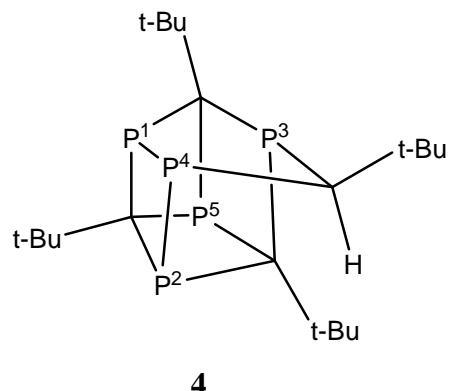
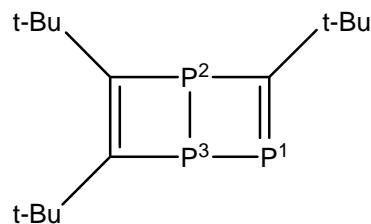


Fig. 1: $^{31}\text{P}\{^1\text{H}\}$ -NMR data (161 MHz; C_6D_6 ; H_3PO_4) and assignment for the identified products



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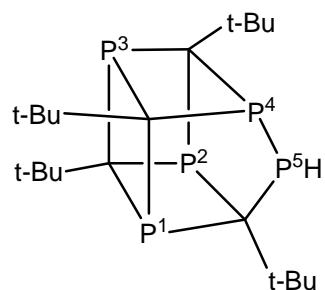
δ_{P} 190.9 (1P, ddd, $^1\text{J}(\text{P}^1, \text{P}^4) = 109.8$ Hz, $^2\text{J}(\text{P}^1, \text{P}^3) = 27.7$ Hz, $^2\text{J}(\text{P}^1, \text{P}^2) = 11.9$ Hz, P^1), 176.0 (1 P, dd, $^2\text{J}(\text{P}^5, \text{P}^3) = 21.2$ Hz, $^2\text{J}(\text{P}^5, \text{P}^4) = 10.8$ Hz, P^5), 162.7 (1 P, ddd, $^1\text{J}(\text{P}^2, \text{P}^4) = 105.3$ Hz, $^2\text{J}(\text{P}^2, \text{P}^3) = 15.2$ Hz, $^2\text{J}(\text{P}^2, \text{P}^1) = 11.9$ Hz, P^2), 118.1 (1 P, dddd, $^2\text{J}(\text{P}^3, \text{P}^1) = 27.7$ Hz, $^2\text{J}(\text{P}^3, \text{P}^4) = 22.2$ Hz, $^2\text{J}(\text{P}^3, \text{P}^5) = 21.2$ Hz, $^2\text{J}(\text{P}^3, \text{P}^2) = 15.3$ Hz, P^3), -32.7 (1 P, dddd, $^1\text{J}(\text{P}^4, \text{P}^1) = 109.8$ Hz, $^1\text{J}(\text{P}^4, \text{P}^2) = 105.3$ Hz, $^2\text{J}(\text{P}^4, \text{P}^3) = 22.2$ Hz, $^2\text{J}(\text{P}^4, \text{P}^5) = 10.8$ Hz, P^4).



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δ_P 350.8 (1P, dd, $^1J(P^1, P^3) = 186.2$ Hz, $^2J(P^1, P^2) = 3.4$ Hz, P^1 , -28.6 (1 P, dd, $J(P^2, P^3) = 89.7$ Hz, $J(P^2, P^1) = 3.4$ Hz, P^2), -155.3 (1 P, dd, $^1J(P^3, P^1) = 186.2$ Hz, $J(P^3, P^2) = 89.7$ Hz, P^3).

See also: P. Binger, S. Leininger, K. Gather and U. Bergsträßer, *Chem. Ber.*, 1997, **130**, 1491.



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δ_P 219.8 (1P, dddd, $^1J(P^1, P^2) = 20.8$ Hz, $^2J(P^1, P^4) = 17.1$ Hz, $^2J(P^1, P^5) = 16.3$ Hz, $^2J(P^1, P^3) = 13.9$ Hz, P^1), 216.9 (1 P, dddd, $^2J(P^2, P^5) = 41.4$ Hz, $^2J(P^2, P^1) = 20.8$ Hz, $^2J(P^2, P^3) = 15.5$ Hz, $^2J(P^2, P^4) = 9.9$ Hz, P^2), 164.6 (1 P, dddd, $^2J(P^3, P^4) = 22.4$ Hz, $^2J(P^3, P^2) = 15.4$ Hz, $^2J(P^3, P^1) = 13.9$ Hz, $^2J(P^3, P^5) = 2.7$ Hz, P^3), 42.1 (1 P, dddd, $^1J(P^4, P^5) = 202.9$ Hz, $^2J(P^4, P^3) = 22.4$ Hz, $^2J(P^4, P^1) = 17.1$ Hz, $^2J(P^4, P^2) = 9.9$ Hz, P^4), 39.3 (1 P, dddd, $^1J(P^5, P^4) = 202.9$ Hz, $^2J(P^5, P^2) = 41.4$ Hz, $^2J(P^5, P^1) = 16.3$ Hz, $^2J(P^5, P^3) = 2.7$ Hz, P^5). The signal for P^5 splits into a doublet when measured without 1H -decoupling ($^1J(P^5, H) = 177.0$ Hz)