Supporting Information for

Facile synthesis of mesoporous silica-supported catalyst for Ru-catalyzed transfer hydrogenation of ketones

Guohua Liu^{*}, Mei Yao, Fang Zhang, Yan Gao, Hexing Li^{*}

Department of Chemistry, Shanghai Normal University, Shanghai 200234, P. R. China

Experimental

1. General Methods.

1.1. Materials and general methods.

All experiments, which are sensitive to moisture or air, were carried out under an Ar atmosphere using standard Schlenk techniques. Commercial reagents were used without further purification unless Triblock PEO-PPO-PEO otherwise noted. copolymer P123 [HO(CH₂CH₂O)₂₀(CH₂CH(CH₃)O)₇₀(CH₂-CH₂O)₂₀H], (Trisethoxysilyl)ethyldiphenylphosphine, diphenylshosphine, (1R,2R)-1,2-diphenylethylenediamine [(R,R)-DPEN], (Trisethoxysilyl)ethyldiphenylphosphine and Ruthenium(III) chloride were purchased from Sigma-Aldrich Company Ltd. and used as received. SBA-15 (pore size 7.6 nm) and Ru-SBA-15 (pore sizes 5.7 nm) were synthesized and characterized as described in our previous reports (a) Li, H. X.; Zhang, F.; Wan, Y.; Lu, Y. F. J. Phys. Chem. B: 2006, 110, 22942. b) Wan, Y.; Zhang, F.; Lu, Y. F.; Li, H. X. J. Mol. Catal. A: Chem. 2007, 267, 165. c) Wan, Y.; Chen, J.; Zhang, D. Q.; Li, H. X. J. Mol. Catal. A: Chem. 2006, 258, 89). RuCl₂{PPh₂(CH₂)₂Si(OEt)₃} 1 was synthesized by the literatrue (Kröcher, O.; Köppel, R. A.; Fröba, M.; Baiker, A. J. Catal. 1998, 178, 284-298.)The representative ketones were chosen as model substrates to test the heterogeneous Ruthenium catalysts. The corresponding racemic alcohol were prepared by hydrogenation these ketones by LiAlH₄ in THF at 0

°C. The *ee* values were determined by GC using a Supelco β -Dex 120 chiral column (30 m×0.25 mm (i.d.), 0.25 μ m film).

1.2. Characterization

Ru loadings in the catalysts **5-6** were analyzed by inductively coupled plasma optical emission spectrometer (ICP, Varian VISTA-MPX). The X-ray powder diffraction (XRD) experiments were carried out on a Rigaku D/Max-RB diffractometer with Cu K α radiation. Transmission electron microscopy (TEM) studies were performed on a JEOL JEM2010 electron microscope, operated at an acceleration voltage of 200 kV. Fourier transform infrared (FTIR) spectra were collected with a Nicolet Magna 550 spectrometer by using the KBr method. N2 adsorption isotherms were measured at 77 K with a Quantachrome Nova 4000 analyzer. The samples were measured after being outgassed at 423 K overnight. Pore size distributions were calculated by using the BJH model. The specific surface areas (S_{BET}) of samples were determined from the linear parts of BET plots ($p/p_0 = 0.05-0.95$). Solid-state ²⁹Si MAS NMR, ¹³C CP MAS NMR, and ³¹P CP MAS NMR spectra were recorded at 79.5, 100.6, and 169.3 MHz, respectively, using a Bruker AV-400 spectrometer.

2. Synthetic procedures.



2.1. Preparations of the mesoporous silica-supported Ruthenium chiral catalysts

2.1.1. Synthesis of catalyst 6 by postgrafting method. The typical procedures as follows: Under argon

atmosphere, to a stirred solution of RuCl₂{PPh₂(CH₂)₂Si(OEt)₃} 1 (0.20 g, 0.15 mmol) in 2 mL dry CH_2Cl_2 was added dropwise (1R,2R)-1,2-diphenylethylenediamine (0.033 g, 0.15 mmol) in 2 mL dry CH₂Cl₂ at room temperature. The resulting mixture was stirred and refluxed for 24 h.. After evaporation of most of the solvent, the residues were washed thoroughly with hexane to afford RuCl₂{PPh₂(CH₂)₂Si(OEt)₃}₂-DPEN 4 (0.13 g, 0.11 mmol, 73.3%) as a red solid, which is not purified and used directly in following reaction. Then pure siliceous support [SBA-15 (poresize of 7.6), 1.0 g] was dehydrated at 125 °C under 0.01 Torr for 4 h before the addition of the fresh 4 in dry toluene (25 mL). The resulting mixture was stirred and refluxed for 24 h under Argon atmosphere, during which time the various Ru were grafted onto the supports. After being cooled, filtrated, and washed thoroughly with toluene and C_2H_5OH : CH_2Cl_2 (v:v = 1:1), the solid was dried at 60 °C under reduced pressure overnight to afford the mesoporous catalyst 6 (1.08 g, 72.7% relative to 4) in the form of a red powder (Scheme 1). **ICP** analyses show that the Ru loadings amount in the catalyst **6** is about 6.80 mg per gram catalyst 6 (theoretic amounts 7.18 mg relative to 6). IR 3450, 2960, 2920, 2870, 1650, 1440, 1050, 959, 806, 692, 545, 468 cm⁻¹ (IR spectrum was shown in Fig.S1); Elemental analysis (%): C 3.39, H 2.55, N 0.19; d_{pore} : 6.3 nm; S_{BET} : 431 m²/g. ²⁹Si MAS/NMR (79.5 MHz): Q⁴ (δ = -116 ppm), Q^3 (δ = -106 ppm), and Q^2 (δ = -97 ppm); ¹³C CP/MAS (100.6 MHz): 129, 72, 63, 22 ppm; ³¹P CP/MAS (169.3 MHz): 71 ppm (NMR spectrum was shown in Fig.S2); The TEM images was shown in Fig.S3.

2.1.2. Synthesis of the catalyst 5 by the postmodification method through reaction of Ru-SBA-15 with (1R,2R)-1,2-DPEN. The typical procedures as follows: Under argon atmosphere, 1.0 g of Ru-SBA-15 (Ru, 62 µmol) prepared by reported method was added into a stirred solution of (1R, 2R)-DPEN 2 (0.0132 g, 62 µmol) in 25 mL of toluene and the mixture was reflux for 10 h. After being cooled, filtrated, and washed thoroughly with toluene and C₂H₅OH : CH₂Cl₂ (v:v = 1:1), the solid was dried at 60 °C under reduced pressure overnight to afford the mesoporous catalyst 5 (1.01 g, 75.7% relative to 2) in the form of a light red powder (Scheme 1). **ICP** analyses show that the Ru loadings amount in the catalyst 5 is 6.26 mg per gram catalyst 5 (theoretic amounts 6.26 mg relative to 5). IR 3430, 2970,

2930, 1650, 1442, 1050, 958, 795, 692, 553, 463 cm⁻¹ (IR spectrum was shown in Fig.S1); Elemental analysis (%): C 3.08, H 2.14, N 0.17; d_{pore}: 4.8 nm; S_{BET}: 371 m²/g. ²⁹Si MAS/NMR (79.5 MHz): Q⁴ (δ = -112 ppm), Q³ (δ = -104 ppm), and Q² (δ = -96 ppm); ¹³C CP/MAS (100.6 MHz): 128, 69, 73, 65, 23 ppm; ³¹P CP/MAS (169.3 MHz): 69, 19 ppm (NMR spectrum was shown in Fig.S2); The TEM images was shown in Fig.S3.

3. General Procedures for Asymmetric Transfer Hydrogenation of ketones. Under argon atmosphere, the catalyst **6** (0.074 g, 5.03 µmol based on Ru from ICP) and potassium *tert*-butoxide (11.20 mg, 0.10 mol) in a test tube were stirred in dry CH₂Cl₂ (5 mL) for 1 h at room temperatute. Then anhydrous 2-propanol (1 mL, 0.013 mol) and the ketone (0.5 mmol) were added with a syringe and mixture was stirred at 50 °C for 24-48 h. After the completion of the reaction, dry CH₂Cl₂ (1 mL) was added and the mixture was stirred for 1 min, then the reactor was centrifuged (2000 r/min) for 1-2 min. The solution was fast passed through a short column (silica gel, eluent: Et₂O). The catalytic activity and enantiomeric excess were determined by chiral GC using a Supelco β -Dex 120 chiral column (30 m×0.25 mm (i.d.), 0.25 µm film) as shown in Fig. 4S.

Figure S1. FTIR spectra of Catalysts 5-6,.



Figure S2. NMR spectra of the catalyst **5** and **6**: (a) 29 Si MAS NMR, (b) 13 C CP MAS NMR, and (c) 31 P CP MAS NMR





Figure S3. The TEM images of the catalysts 5-6 viewed along [001] directions.

Catalyst 5:

Catalyst 6:







季	氨戊重	分析结果		分狙結果		祥晶信息				
)号	组分名	8	保留时间()	分钟)	峰高(微伏)	峰面积(微伏*秒)	浓度	半峰宽(秒)	分离度	理论塔板
1			21.06	7	25800	161786.9	99.7531	6.27	0.00	224941.08
2			21.80	7	47	400.4	0.2469	8.52	3.53	130590.26
			急计:		Σ=25847	Σ=162187.3	Σ=100.0000			

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参;	飲役置	5	}祈结果	分割	植结果		样晶信息					
3	组分:	名	保留时间(约	分钟)	峰高(微(٤)	峰面积(微伏*秒)	浓	度	半峰宽(秒)	分离度	理论塔板
L			15.60	0	812		3767.6	82.0	923	4.64	0.00	225295.67
2			21.25	3	171		736.8	16.0	539	4.31	44.59	484933.07
3			21.66	0	7		85.1	1.85	i38	12.15	1.74	63299.40
			算计:		Σ=99)	S=4589.5	Σ=100	.0000			



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参	數设置	5	析結果	分	狙结果	1	洋品信息					
)号	组分	名	保留时间(外	分钟)	峰高(微仇	大) [峰面积(微伏*秒)	浓度	半峰宽(秒)	分离度	理论塔板	
1			24.94	7	6934		32436.3	99.5275	4.68	0.00	566832.50	
2			25.44	7	56		154.0	0.4725	2.75	4.75	170655	
			总计:		Σ=699	0	Σ=32590.3	Σ=100.0000				

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参	敗设置	5	计析结果	分	祖结果	3	样晶信息						
D号	组分	名	保留时间(约	分钟)	峰高(微伏	t)	峰面积(微伏*秒)	浓	度	半峰宽(秒)	分离度	理论塔板	
1			33.20	7	1536		41653.8	43.8	3945	27.12	0.00	29884.65	
2			33.80	7	1170		53241.4	56.1	1055	45.51	0.58	11000.24	
			总计:		Σ=2706	5	Σ=94895.2	Σ=100	0.0000				



参数设置			分析结果	分	粗结果		样晶信息					
ID号	组 分	名	保留时间(;	分钟)	峰高(微伏	0	峰面积(微伏*秒)	浓度	半峰宽(秒)	分离度	理论塔板	
1			32.98	0	9611		66252.2	99.9177	6.89	0.00	456207.27	
2			33.62	0	36		54.6	0.0823	1.52	5.37	979354	
			急计:		Σ=9647	7	Σ=66306.8	Σ=100.0000				

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	AL AL AL			/5 AL	130	TT HHIH /SP					
)号	组分	名	保留时间(5	}钟)	峰高(微伏)	峰面积(微伏*秒)	浓度	半峰宽(秒)	分寓度	理论塔板	İ
1			24.447	7	1920	22434.9	99.1750	11.68	0.00	87240.88	
2			25.073	3	26	186.6	0.8250	7.18	2.35	243177.90	
			息计:		Σ=1946	Σ=22621.5	Σ=100.0000				

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Recycling-experiment 2:



号	1组分名	保留时间(分钟)	峰高(微伏)	峰面积(微伏*秒)	浓度	半峰宽(秒)	分寓度	理论塔板
1		25.080	4638	22929.3	99.8754	4.94	0.00	512929.94
2		26.140	21	28.6	0.1246	1.36	11.87	733889
		急计:	Σ=4659	Σ=22957.9	Σ=100.0000			



Recycling-experiment 3:

_	1111 000 - 2 T 1 1 1 1 7 7 7 1 1			100.000		
	25.167	14522	71900.0	99.7929	4.95	0
	25.867	57	149.2	0.2071	2.62	6
	急计:	Σ=14579	Σ=72049.2	Σ=100.0000		

.53

194633...

Recycling-experiment 4:



Recycling-experiment 5:



Recycling-experiment 6:



Recycling-experiment 7:



李	観衣査	分析结末	分組	結果	祥晶信息				
号	11 分名	保留时间(分钟)	峰高(微伏)	峰面积(微伏*秒)	浓度	半峰宽(秒)	分离度	理论塔板
1		25.16	7	14522	71900.0	99.1963	4.95	0.00	514956.84
2		26.12	7	42	582.5	0.8037	13.87	3.60	70718.73
		急计:		Σ=14564	Σ=72482.6	Σ=100.0000			





Data were obtained at S/C = 500 using 8 as substrate.



参	啟设置	5	祈结果	分:	組結果 🛛	样品信息					
号	组分	名	保留时间(约	(钟)	峰高(微伏) 峰面积(微(伏*秒) 浓	度	半峰宽(秒)	分离度	理论塔板
1			25.01	3	9918	46901	.5 98.	6084	4.73	0.00	557625.79
2			26.02	7	41	661.	9 1.3	3916	16.14	3.43	51801.40
			急计:		Σ=9959	Σ=4756	53.4 Σ=10	0.0000			