Supporting Information

Bifunctional Organocatalyst for Methanolytic Desymmetrization of Cyclic Anhydrides: Increasing Enantioselectivity by Catalyst Dilution

Ho Sik Rho, Sang Ho Oh, Ji Woong Lee, Jin Young Lee, Jik Chin* and Choong Eui Song*

Department of Chemistry, Sungkyunkwan University, Suwon, 440-746 (Korea) R&D center, AmorePacific Corporation, Yongin, 446-729 (Korea) Department of Chemistry, University of Toronto, 80 St. George st. Toronto, ON (Canada)

General. Anhydrides (**1a**, **1b**, **1d**, **1e**) were purchased from Aldrich and used without further purification. Anhydride **1c** was obtained from **1d** by hydrogenation of the double bond. Quinine a¹nd hydroquinine were purchased from Aldrich and used without further purification. Thiourea and urea catalysts (**I-II**) were synthesized by the literature procedure.^[1]

Chromatographic purification of products was carried out by flash chromatography using Merck silica gel 60 (230–400 mesh). Thin-layer chromatography was carried out on Merck silica gel 60F plates. HPLC analyses were performed on a Jasco 1100 Series instrument equipped with an isostatic pump using Hypersil Column ($200 \times 4.6 \text{ mm}$). ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) were recorded on Varian 300 spectrometers using TMS as an internal standard.

General procedure for methanolysis of prochiral cyclic anhydrides

Methanol (202 μ L, 5 mmol) was added dropwise to a stirred solution of anhydrides (0.5 mmol) and catalysts (**I–IV**) (5–10%) in appropriate solvents (2.5–80 mL) at the temperature

^[1] B. Vakulya, S. Varga, A. Csampai, T. Soos, Org. Lett. 2005, 7, 1967-1969.

indicated in table 2. The reaction mixture was stirred at that temperature until the starting material was consumed, as indicated by TLC analysis (1–30h). The reaction was quenched by adding HCl (1N, 3 mL) in one portion. The aqueous phase was extracted with EtOAc (2×100 mL). The organic phase was dried (MgSO₄), filtered, and concentrated *in vacuo* to yield the crude product. Purification by column chromatography (EtOAc/Hexanes) gave hemiester products (**2a–2e**)

The enantiomeric excess (*ee*) of each product was determined by HPLC analysis of a diastereomeric mixture of the corresponding amide-ester prepared from hemiester according to literature procedure^[2] (Scheme 1).

Scheme 1



^[2] Y, M, Song, J. S. Choi, J. W. Yang, H. Han, Tetrahedron Lett. 2004, 45, 3301-3304.

NMR Spectra for 2a



¹H NMR (300 MHz, CDCl₃) δ 1.36-1.62 (m, 4H), 1.72-1.84 (m, 2H), 1.96-2.10 (m, 2H), 2.80-2.92 (m, 2H), 3.68 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 23.76, 23.89, 26.08, 26.38, 42.45, 42.65, 51.87, 174.20, 180.26; The ee value was determined by HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 8.50 min, t(major) = 11.79 min) of the diastereomeric mixture of the corresponding amide-ester prepared as described previously.



¹H NMR (300 MHz, CDCl₃) δ 2.32-2.65 (m, 4H), 3.02-3.12 (m, 2H), 3.69 (s, 3H), 5.68 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 25.68, 25.86, 39.58, 39.70, 52.10, 125.18, 125.30, 173.82, 179.66; The ee value was determined by HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 11.02 min, t(major) = 14.68 min) of the diastereomeric mixture of the corresponding amide-ester prepared as described previously.



¹H NMR (300 MHz, CDCl₃) δ 1.35-1.58 (m, 4H), 1.75-1.83 (m, 2H), 2.57-2.62 (m, 2H), 2.82-3.04 (m, 2H), 3.69 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 23.98, 24.17, 39.96, 40.22, 40.52, 46.83, 51.42, 172.98, 179.01; The ee value was determined by HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 12.04 min, t(major) = 17.99 min) of the diastereomeric mixture of the corresponding amide-ester prepared as described previously.

NMR Spectra for 2d



¹H NMR (300 MHz, CDCl₃) δ 1.34 (bd, J = 9.0 Hz, 1H), 1.50 (dt, J = 9.0 Hz and 1.8 Hz, 1H), 3.01-3.41 (m, 4H), 3.59 (s, 3H), 6.21 (dd, J = 5.0 and 3.0 Hz, 1H), 6.33 (dd, J = 5.0 and 3.0 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 46.23, 46.72, 48.08, 48.37, 48.93, 51.67, 134.46, 135.71, 173.00, 178.15; The ee value was determined by HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 16.56 min, t(major) = 23.25 min) of the diastereomeric mixture of the corresponding amide-ester prepared as described previously.

NMR Spectra for 2e



¹H NMR (300 MHz, CDCl₃) δ 1.42 (bd, J = 9.0 Hz, 1H), 2.13 (bd, J = 9.0 Hz, 1H), 2.62 (m, 2H), 3.09 (m, 2H), 3.65 (s, 3H), 6.22 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 45.49, 45.52, 45.90, 47.47, 47.57, 51.97, 138.00, 138.19, 174.00, 180.13; The ee value was determined by HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 15.23 min, t(major) = 16.20 min) of the diastereomeric mixture of the corresponding amide-ester prepared as described previously.



Tabel 1. Entry 1 (Catalyst I – 10 mol%)



Table 1. Entry 2 (Catalyst I – 5 mol%)



Table 1. Entry 3 (Catalyst II – 10 mol%)



Table 1. Entry 4 (Catalyst III – 10 mol%)



Table 1. Entry 5 (Catalysy IV – 10 mol%)



Table S1. Supporting Data for Figure 2: Solvent Effect on Enantioselectivity ^[a]

Entry	Solvent	Temperature	Time (h)	Yield (%) ^[b]	% <i>ee</i> ^[c]
1	Dioxane (40 mL)	RT	10	85	97
2	THF (40 mL)	RT	10	84	95
3	Et ₂ O (40 mL)	RT	10	80	94
4	CH ₂ Cl ₂ (40 mL)	RT	10	80	75
5	Toluene (40 mL)	RT	10	75	74
6	Methyl cyclohexane	RT	10	79	43
	(40 mL)				
7	MeOH $(40 \text{ mL})^{[d]}$	RT	3	82	31

[a] Unless otherwise indicated, reactions were carried out with **1a** (0.5 mmol), 10 equiv of MeOH (5 mmol) and catalysts **I** (10 mol%) in solvent (40 mL) at RT. [b] Isolated yields after chromatographic purification. [c] Determined by HPLC (see below HPLC profiles). [d] MeOH was used as solvent and reagent.

Tuble Sat Supporting Dut in English at Concentration Enterto on Englisher to the	Table S2. Supporting	Data for Figure 2	: Concentration Effect on	Enantioselectivity ^[a]
--	----------------------	--------------------------	---------------------------	-----------------------------------

- and set supporting 2 and for a gard 2 content atom 2 more on 2 manufactor at						
Entry	Solvent	Temperature	Time (h)	Yield (%) ^[b]	% <i>ee</i> ^[c]	
8	THF (2.5 mL)	RT	0.6	82	82	
9	THF (5 mL)	RT	1	84	87	
10	THF (20 mL)	RT	7	83	93	
11	THF (80 mL)	RT	15	81	96	

[a] Unless otherwise indicated, reactions were carried out with **1a** (0.5 mmol), 10 equiv of MeOH (5 mmol) and catalysts **I** (10 mol%) in solvent (THF) at RT. [b] Isolated yields after chromatographic purification. [c] Determined by HPLC (see below HPLC profiles).

Table S3. Temperature Effect on Enantioselectivity [a]

Entry	Solvent	Temperature	Time (h)	Yield (%) ^[b]	% <i>ee</i> ^[c]
1	THF (2.5 mL)	0 °C	5.5	81	81
2	THF (2.5 mL)	-20 °C	11	80	77
3	THF (40 mL)	0 °C	18	82	95
4	THF (40 mL)	-20 °C	30	78	95

[a] Unless otherwise indicated, reactions were carried out with 1a (0.5 mmol), 10 equiv of MeOH (5 mmol) and catalysts I (10 mol%) in solvent (THF). [b] Isolated yields after chromatographic purification. [c] Determined by HPLC (see below HPLC profiles).



Table S1. Entry 1 (Dioxane 40 mL, RT)



Table S1. Entry 2 (THF 40 mL, RT)



Table S1. Entry 3 (Diethyl Ether 40 mL, RT)



Table S1. Entry 4 (Dichloromethane 40 mL, RT)



Table S1. Entry 5 (Tolene 40 mL, RT)



Table S1. Entry 6 (Methylcyclohexane 40 mL, RT)



Table S1. Entry 7 (Methanol 40 mL, RT)



Table S2. Entry 1 (THF 2.5 mL, RT)



Table S2. Entry 2 (THF 5 mL, RT)



Table S2. Entry 3 (THF 20 mL, RT)



Table S2. Entry 4 (THF 80 mL, RT)



Table S3. Entry 1 (THF 2.5 mL, 0 °C)



Table S3. Entry 2 (THF 2.5 mL, -20 °C)



Table S3. Entry 3 (THF 40 mL, 0 °C)



Table S3. Entry 4 (THF 40 mL, -20 °C)



HPLC Spectra for Table 2

Table 2. Entry 1





Table 2. Entry 2



Table 2. Entry 4

¹H NMR Spectroscopic Data for Self-Association of Catalyst I

N-H Chemical shift : 9.3 ppm

Catalyst I (67.5 mM in *d*₈-Toluene, RT)

N-H Chemical shift : 10.6 ppm

Catalyst I (139 mM in d_8 -Toluene, RT)

N-H Chemical shift : 11.0 ppm

Catalyst I (212 mM in d₈-Toluene, RT)

N-H Chemical shift : 11.1 ppm

I-ent-3d

Table S-4. Transition state energy for I-3 and I-ent-3

	E for I-3	E for I-ent3	ΔE	
Substrate	(in hartree)	(in hartree)	(in hartrre)	ΔE (kcal)
1 a	-2704.595525	-2704.593124	0.002401	1.505615
1b	-2703.361477	-2703.357996	0.003481	2.182462
1c	-2742.676366	-2742.673837	0.002528	1.585244
1d	-2741.43547	-2741.43385	0.001621	1.016242
1e	-2741.439809	-2741.437039	0.002769	1.736288