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Supporting Information for:

Copper-free click chemistry for the *in situ* crosslinking of photodegradable star polymers

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Department of Chemistry, Columbia University, 3000 Broadway, MC3157, New York, New York 10027, Departments of Chemistry and Molecular and Cell Biology and Howard Hughes Medical Institute, University of California, Berkeley, California 94720, The Molecular Foundry, Materials Sciences Division, Lawrence Berkeley National Laboratory, Berkeley, California 94720, and Department of Chemical Engineering, Columbia University, 500 West 120th Street, New York, New York 10027 General. All reagents were purchased from Aldrich chemical company and were used as supplied unless otherwise noted. Star polymer 1, MOFO, and DIFO were prepared following literature procedures.^{1,2,3} SEC measurements were performed on a Knauer GPC system with a Knauer K-2301 refractive index detector and a Spark Holland Basic Marathon autosampler. Three Polymer Laboratories 5 µm particle size PLgel columns (one 100 Å and two MIXED-D pore types) placed in series were employed for the chromatography. The system was calibrated against linear polystyrene standards ranging in molecular weight from 580-377,400 Da. Experiments were performed at room temperature in THF eluant with a flow rate of 1.0 mL/min. Fourier transform infrared spectroscopy (FTIR) was performed using a Nicolet Nexus 870 FTIR system with a Harrick solution cell consisting of two 13 X 2 mm diameter CaF₂ plates with a 25 µm Teflon spacer. FTIR data was analyzed using OMNIC version 7.1a software from Thermo-Nicolet Corporation. Gel samples were degraded by UV irradiation using a Rayonet RPR-100 reactor from the Southern New England Ultraviolet Company fitted with RPR-3500 lamps (350 nm peak wavelength).



Synthesis of diMOFO. MOFO (46 mg, 0.18 mmol) was dissolved in THF (10 mL) in a dry round-bottom flask and cooled to 0 °C under argon atmosphere. *N*- (3-dimethylaminopropyl)-*N*'-ethylcarbodiimide hydrochloride (33.9 mg, 0.18 mmol) and 1-hydroxybenzotriazole (23.9 mg, 0.18 mmol) were then added and the suspension was stirred at 0 °C for 10 min before the dropwise addition of

ethylene diamine (5.3 mg, 0.088 mmol). Upon addition of ethylene diamine the suspension immediately became cloudier. After stirring at 0 °C for 30 min, the mixture was warmed to room temperature and stirred for 4 h after which time the contents of the flask were filtered, concentrated on a rotary evaporator, diluted with ethyl acetate, and washed with 1.0 M HCl (2 X 10 mL), saturated NaHCO₃ (2 X 10 mL), and brine (1 X 10 mL). The organic layer was then dried over MgSO₄, filtered, concentrated, and purified by column chromatography (75% ethyl acetate:hexanes, $R_f = 0.3$, anisaldehyde stain for TLC plates) to yield **diMOFO** (45.1 mg, 94%) as a white solid. ¹H NMR (300 MHz, CDCl₃): δ 7.56 (d, 4 H, *J* = 8.1 Hz), 7.41 (br, 2 h), 7.34 (d, 4 H, *J* = 8.1 Hz), 3.65 (s, 4 H), 3.04 (d, 4 H, J = 20.6 Hz), 2.21 (m, 4 H), 1.89 (m, 8 H), 1.71 (m, 4 H), 1.34 (m, 4 H); ¹³C NMR (300 MHz, CDCl₃): δ 168.93, 140.10, 132.61, 130.76, 127.12, 104.88, 104.75, 96.72, 94.39, 91.03, 90.63, 48.21, 47.90, 45.02, 44.71, 41.14, 34.28, 29.64, 26.92, 20.73; IR (cm⁻¹) 3307, 2926, 2842, 2224, 1630, 1537, 1504, 1444, 1290; APCI⁺LRMS calcd. for C₃₄H₃₈F₂N₂O₂ [M+H]⁺ 545.3, found 545.0.



Synthesis of diDIFO. DIFO (10 mg, 0.046 mmol) was dissolved in THF (4 mL) in a dry round-bottom flask and cooled to 0 °C under argon atmosphere. *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide hydrochloride (8.8 mg, 0.046 mmol) was added and the suspension was stirred at 0 °C for 10 min before the dropwise addition of ethylene diamine (1.4 mg, 0.023 mmol). As for the synthesis of **diMOFO** above, upon addition of ethylene diamine the suspension

immediately became cloudier. After stirring at 0 °C for 30 min, the mixture was warmed to room temperature and stirred overnight (monitored by TLC, 5% methanol:dichloromethane, $R_f = 0.4$) after which time the contents of the flask were filtered, concentrated on a rotary evaporator, diluted with ethyl acetate, and washed with 1.0 M HCl (2 X 10 mL), saturated NaHCO₃ (2 X 10 mL), and brine (1 X 10 mL). The organic layer was then dried over MgSO₄, filtered, concentrated, and purified by column chromatography (2% methanol in methylene chloride, $R_f = 0.4$, anisaldehyde stain for TLC plates) to yield **diDIFO** (6.2 mg, 59%) as a white solid. ¹H NMR (400 MHz, CDCl₃): δ 7.04 (br, 2 H), 3.93 (d, 4 H, *J* = 3.3 Hz), 3.52 (br, 2 H), 3.48 (s, 4 H), 2.56 (m, 4 H), 2.46 (m, 4 H), 2.26 (m, 4 H), 2.06 (m, 4 H); ¹³C NMR (400 MHz, CDCl₃): δ 170.85, 110.94, 83.63, 81.91, 75.78, 68.39, 39.86, 39.05, 33.82, 30.14, 17.56: δ ; IR (cm⁻¹) 3422, 2917, 2849, 2219, 1668, 1539, 1436, 1129, 1104, 1026, 737; APCI⁺ LRMS calcd. for C₂₂H₂₈F₂N₂O₄ [M]⁺ 460.2, found 460.3.

References:

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