### **Supporting Information**

# Analysis of Carbohydrates and Steroids by Desorption Electrospray Ionization Mass Spectrometry

#### Tiina J. Kauppila,<sup>a</sup> Nari Talaty,<sup>b</sup> Ayanna U. Jackson,<sup>b</sup> Tapio Kotiaho,<sup>a, c</sup> Risto Kostiainen\*<sup>a</sup> and R. Graham Cooks\*<sup>b</sup>

Faculty of Pharmacy, University of Helsinki, 00014 Helsinki, Finland <sup>b</sup> Department of Chemistry, Purdue University, West Lafayette, IN 47907, USA. Email:cooks@purdue.edu <sup>c</sup>Department of Chemistry, University of Helsinki, 00014 Helsinki, Finland

**Abstract**: Additional information is provided including a table of solvent systems studied, tables of the ions observed in the xylose and cortisone spectra with the different spray solvent systems, structures of the compounds studied and graphs illustrating the optimization of the solvent system in the positive and negative ion modes.

## Table S1 Solvent systems studied.

	Solvent	Salt Concentration	Monitored Ion
Posi	itive ion mode		
1	water/methanol/formic acid	0.1%	$[M+H]^+$
2	water/methanol/lithium chloride	100 µM	$[M+Li]^+$
3	water/methanol/sodium chloride	100 µM	$[M+Na]^+$
4	water/methanol/potassium chloride	100 µM	$[M+K]^+$
5	water/methanol/silver nitrate	100 µM	$[M+Ag]^+$
6	water/methanol/ammonium acetate	100 µM	$\left[\mathrm{M+NH_4}\right]^+$
Nea	ative ion mode		
Neg	ative ion mode		
Neg 1	ative ion mode water/methanol/ammonium hydroxide	0.1%	[M-H] <sup>-</sup>
Neg 1 2	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform	0.1% 3/6/1	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup>
Neg 1 2 3	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride	0.1% 3/6/1 10 μM	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup>
Neg 1 2 3 4	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride water/methanol/ammonium bromide	0.1% 3/6/1 10 μM 10 μM	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Br] <sup>-</sup>
Neg 1 2 3 4 5	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride water/methanol/ammonium bromide water/methanol/formic acid	0.1% 3/6/1 10 μM 10 μM 0.01%	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Br] <sup>-</sup> [M+CHOO] <sup>-</sup>
Neg 1 2 3 4 5 6	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride water/methanol/ammonium bromide water/methanol/formic acid water/methanol/acetic acid	0.1% 3/6/1 10 μM 10 μM 0.01% 0.01%	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Br] <sup>-</sup> [M+CHOO] <sup>-</sup> [M+CH3COO] <sup>-</sup>
Neg 1 2 3 4 5 6 7	ative ion mode water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride water/methanol/ammonium bromide water/methanol/formic acid water/methanol/sulfuric acid	0.1% 3/6/1 10 μM 10 μM 0.01% 0.01% 10 μM	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Br] <sup>-</sup> [M+CHOO] <sup>-</sup> [M+CH <sub>3</sub> COO] <sup>-</sup> [M+HSO <sub>4</sub> ] <sup>-</sup> ,
Neg 1 2 3 4 5 6 7	water/methanol/ammonium hydroxide water/methanol/chloroform water/methanol/ammonium chloride water/methanol/ammonium bromide water/methanol/formic acid water/methanol/sulfuric acid	0.1% 3/6/1 10 μM 10 μM 0.01% 0.01% 10 μM	[M-H] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Cl] <sup>-</sup> [M+Br] <sup>-</sup> [M+CHOO] <sup>-</sup> [M+CH <sub>3</sub> COO] <sup>-</sup> [M+HSO <sub>4</sub> ] <sup>-</sup> , [M+SO <sub>4</sub> ] <sup>2-</sup>



Figure S1 Structures of the compounds studied



**Figure S2** Positive ion mode DESI signal-to-noise ratios of the analyte ions using different spray solvents. Solvents: 1.  $H_2O/MeOH/HCOOH$  (0.1%), 2.  $H_2O/MeOH/LiCl$  (100  $\mu$ M), 3.  $H_2O/MeOH/NaCl$  (100  $\mu$ M), 4.  $H_2O/MeOH/KCl$  (100  $\mu$ M), 5.  $H_2O/MeOH/AgNO_3$  (100  $\mu$ M), 6.  $H_2O/MeOH/CH_3COONH_4$  (100  $\mu$ M). DESI analysis of 0.5  $\mu$ L of analyte (10  $\mu$ M – 10 mM depending on ionization efficiency) from a teflon substrate

#### Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2008



**Figure S3** Negative ion mode DESI signal-to-noise ratios of the analyte ions using different spray solvents. Solvents: 1.  $H_2O/MeOH/NH_4OH$  (0.1%), 2.  $H_2O/MeOH/CHCl_3$  (3/6/1), 3.  $H_2O/MeOH/NH_4Cl$  (10  $\mu$ M), 4.  $H_2O/MeOH/NH_4Br$  (10  $\mu$ M), 5.  $H_2O/MeOH/CHOOH$  (0.01%), 6.  $H_2O/MeOH/CH_3COOH$  (0.01%), 7.  $H_2O/MeOH/H_2SO_4$  (10  $\mu$ M), 8.  $H_2O/MeOH/NH_4NO_3$  (100  $\mu$ M). DESI analysis of 0.5  $\mu$ L of analyte (10  $\mu$ M – 10 mM depending on ionization efficiency) from a teflon substrate



**Figure S4** Comparison of **(a)** adduct formation using MeOH/H<sub>2</sub>O/NaCl (10  $\mu$ M) and **(b)** reactive DESI using MeOH/H<sub>2</sub>O/CH<sub>3</sub>COOH (0.05%)/NH<sub>2</sub>OH (5%) in the direct analysis of cortisone (1 mM) in positive ion mode from teflon. Simple salt adduct formation resulted in a higher signal intensity and somewhat better signal to noise ratios than did reaction to form the oxime product. Both spectra have been background subtracted for comparison