

Supporting Information of

Hydrogen Bonding Assisted Tautomerization of Pyridine Moieties in the
Coordination Sphere of an Ir(I) Complex

Guoyong Song, Yongxin Li, Shanshan Chen, and Xingwei Li^{*}

Division of Chemistry and Biological Chemistry, School of Physical and Mathematical Sciences,

Nanyang Technological University, Singapore 637616

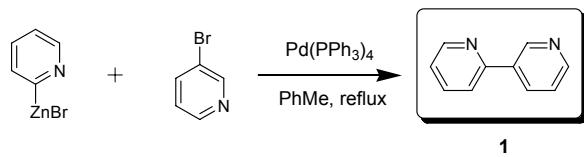
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General procedures

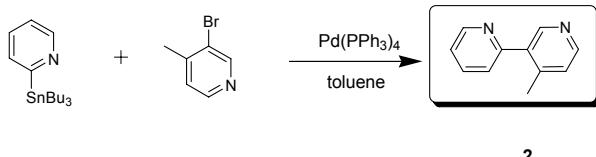
All manipulations were carried out using standard Schlenk techniques or in nitrogen-filled glove-box, except where otherwise noted. All solvent were distilled under N₂ before use and stored glove-box. CDCl₃ was dried by 4Å molecular sieve and CD₂Cl₂, DMSO-d₆ were obtained in sealed ampules from CIL and were used without further purification. Air-sensitive product were stored and weighted in glove-box.

NMR spectra were obtained on a Bruker DPX 300, AMX400 or 500 spectrometer. All spectra were collected at 298K unless otherwise specified. The chemical shift is given as dimensionless δ values and is frequency referenced relative to TMS for ¹H and ¹³C NMR. Assignment of ¹³C signals were facilitated by HMQC and HMBC spectra. Elemental analyses were performed in the Division of Chemistry and Biological Chemistry, Nanyang Technological University. HRMS spectra were obtained in EI or ESI mode on a Finnigan MAT95XP GC/HRMS system (Thermo Electron Corp.). X-ray crystallographic analysis was performed on a Bruker X8 APEX diffractometer.

Synthesis of ligands 1, 2 and 3

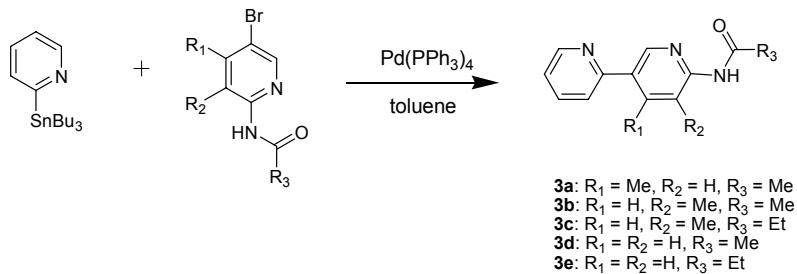


Ligand 1 was synthesized from Negishi coupling. 3-Bromopyridine (300 mg, 1.90 mmol) and Pd(PPh₃)₄ (110 mg, 5 mol %) were dissolved in dry toluene (4 ml), to which was added via syringe a 2-Pyridylzinc bromide solution (0.5 M in THF, 3.8 ml, 1.90 mmol). The mixture was then stirred for 24 hours under reflux. All volatiles were then removed and water was added. The mixture was then extracted with dichloromethane and dried over sodium sulfate. Silica gel chromatography (EtOAc/hexane 1:1) gave 1 as a light-yellow oil (225 mg, 1.44 mmol, 76%). ¹H NMR (500 MHz, CDCl₃): δ 9.19 (d, *J* = 2.2 Hz, 1H), 8.73 (d, *J* = 4.7 Hz, 1H), 8.65 (d, *J* = 4.7 Hz, 1H), 8.33 (d, *J* = 8.0 Hz, 1H), 7.75-7.82 (m, 2H), 7.41 (dd, *J* = 8.0, 2.0 Hz, 1H), 7.29-7.31 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 154.8, 150.1, 149.9, 148.2, 137.1, 133.9, 134.4, 123.6, 122.9, 120.7. HRMS: 156.0722 (calculated 156.0682)

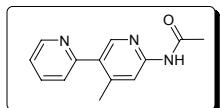


Ligand 2 was synthesized from the Stille coupling reaction. 3-Bromo-4-methylpyridine (200 mg, 1.16 mmol) and Pd(PPh₃)₄ (67 mg, 5 mol %) were dissolved in dry toluene (4 ml), and to which was added 2-(tri-*n*-butylstannyl)pyridine (475 mg, 1.28 mmol). The mixture was stirring for 24 hours under reflux. After removal of toluene under vacuum, the residue was dissolved in

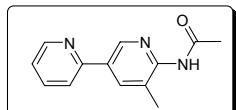
dichloromethane and the palladium black was removed by filtration through Celite. Analytically pure **2** was obtained as an oil after silica gel chromatography using hexanes/ethyl acetate (1:1) as an eluent. Yield: 139 mg (70%). ¹H NMR (500 MHz, CDCl₃): δ 8.71-8.72 (m, 1H), 8.58 (s, 1H), 8.48 (d, *J* = 5.1 Hz, 1H), 7.79 (td, *J* = 7.7, *J* = 1.8 Hz, 1H), 7.41 (d, *J* = 7.8 Hz, 1H), 7.29 (dd, *J* = 7.5, *J* = 4.7 Hz, 1H), 7.20 (d, *J* = 5.0 Hz, 1H), 2.40 (s, 3H, CH₃). ¹³C NMR (125 MHz, CDCl₃): δ 156.8, 149.9, 149.6, 149.1, 145.3, 136.6, 136.3, 125.6, 124.1, 122.3, 19.8. HRMS: 170.0833 (calculated 170.0838).



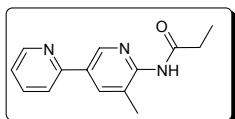
Ligands **3a-e** was synthesized through a directly analogous method for **2**.



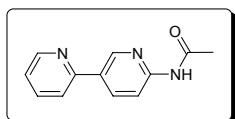
N-(4-methyl-5-(pyridin-2-yl)pyridin-2-yl)acetamide (**3a**), white solid, yield 65%. ¹H NMR (500 MHz, CDCl₃): δ 8.92 (br, 1H, NH), 8.74 (d, *J* = 4.3 Hz, 1H), 8.31 (s, 1H), 8.19 (s, 1H), 7.80 (td, *J* = 7.8, *J* = 1.8 Hz, 1H), 7.41 (d, *J* = 7.8 Hz, 1H), 7.28-7.31 (m, 1H), 2.46 (s, 3H), 2.24 (s, 3H). ¹³C NMR (125 MHz, CDCl₃): δ 168.9, 156.4, 151.1, 149.7, 148.2, 147.9, 136.5, 132.8, 124.2, 122.2, 115.3, 24.8, 20.5. Anal. Calcd for C₁₃H₁₃N₃O (227.3): C, 68.70; H, 5.77; N, 18.49; Found: C, 68.88; H, 5.31; N, 18.19.



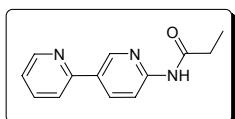
N-(3-methyl-5-(pyridin-2-yl)pyridin-2-yl)acetamide (**3b**), white solid, yield 60%. ¹H NMR (500 MHz, CDCl₃): δ 8.83 (s, 1H), 8.80 (br, 1H, NH), 8.72 (d, *J* = 4.8 Hz, 1H), 8.21 (s, 1H), 7.79 (t, *J* = 6.7, 1H), 7.72 (d, *J* = 7.9 Hz, 1H), 7.28-7.30 (m, 1H), 2.38 (s, 3H), 2.33 (s, 3H). ¹³C NMR (125 MHz, CDCl₃): δ 172.1, 154.4, 150.1, 150.0, 144.0, 138.2, 137.0, 132.8, 127.7, 122.6, 120.4, 24.8, 20.5. Anal. Calcd for C₁₃H₁₃N₃O (227.3): C, 68.70; H, 5.77; N, 18.49; Found: C, 68.56; H, 5.38; N, 18.30.



N-(3-methyl-5-(pyridin-2-yl)pyridin-2-yl)propionamide (**3c**), white solid, yield 67%. ^1H NMR (500 MHz, CDCl_3): δ 9.05 (br, 1H, *NH*), 8.83 (d, J = 2.0 Hz, 1H), 8.70 (d, J = 4.2 Hz, 1H), 8.19 (s, 1H), 7.78 (td, J = 7.8, J = 1.7 Hz, 1H), 7.71 (d, J = 7.9 Hz, 1H), 7.27-7.29 (m, 1H), 2.56 (q, J = 7.4 Hz, 2H, CH_2), 2.37 (s, 3H), 1.26 (t, J = 7.6 Hz, 3H). ^{13}C NMR (125 MHz, CDCl_3): δ 173.4, 154.4, 150.3, 149.9, 143.9, 138.3, 137.1, 132.8, 128.2, 122.7, 120.4, 29.9, 18.4, 9.7. Anal. Calcd $\text{C}_{14}\text{H}_{15}\text{N}_3\text{O}$ (241.3): C, 69.69; H, 6.27; N, 17.41; Found: C, 69.78; H, 6.01; N, 17.50.



N-(5-(pyridin-2-yl)pyridin-2-yl)acetamide (**3d**), white solid, yield 78%. ^1H NMR (500 MHz, CDCl_3): δ 8.94 (s, 1H), 8.93 (br, 1H, *NH*), 8.70 (d, J = 4.7 Hz, 1H), 8.33-8.35 (m, 2H), 7.78 (t, J = 7.9 Hz, 1H), 7.71 (d, J = 7.9 Hz, 1H), 7.26-7.29 (m, 1H), 2.27 (s, 3H). ^{13}C NMR (125 MHz, CDCl_3): δ 168.9, 154.4, 151.9, 150.0, 146.3, 137.0, 136.8, 131.1, 122.5, 120.0, 113.8, 24.8. Anal. Calcd $\text{C}_{12}\text{H}_{11}\text{N}_3\text{O}$ (213.2): C, 67.59; H, 5.20; N, 19.71; Found: C, 67.90; H, 5.31; N, 19.51.



N-(5-(pyridin-2-yl)pyridin-2-yl)propionamide (**3e**), white solid, yield 70%. ^1H NMR (500 MHz, CDCl_3): 8.93 (s, 1H), 8.71 (d, J = 4.5 Hz, 1H), 8.67 (br, 1H, *NH*), 8.32-8.37 (m, 2H), 7.78 (td, J = 7.8, J = 1.6 Hz, 1H), 7.71 (d, J = 8.0 Hz, 1H), 7.26-7.29 (m, 1H), 2.48 (q, J = 7.5 Hz, 2H), 1.28 (t, J = 7.5 Hz, 3H). ^{13}C NMR (125 MHz, CDCl_3): δ 172.6, 154.5, 151.9, 149.9, 146.3, 137.0, 136.8, 131.0, 122.5, 120.1, 113.7, 30.8, 9.4. Anal. Calcd $\text{C}_{13}\text{H}_{13}\text{N}_3\text{O}$ (227.3): C, 68.70; H, 5.77; N, 18.49; Found C, 68.51; H, 5.83; N, 18.64.

Synthesis of iridium complexes **4a-e**

General procedure: To a stirred solution of ligands **3a-e** (0.2mmol) in dry acetone (3ml) was added an acetone solution (2 mL) of $\text{Ir}(\text{COD})_2\text{BF}_4$ (0.2 mmol) at room temperature. After stirred for 1 h, the solution became dark green and dark green to black precipitates formed. The solution was concentrated to ca. 0.5 mL, followed by addition of diethyl ether (5 mL). Filtration gives complexes **4a-e** in nearly quantitative yield.

Complex 4a: Complex **4a** was synthesized by following the above general procedure using **3a** and Ir(COD)₂BF₄ in acetone. Yield: 98%. Single crystals suitable for X-ray analysis were obtained by slow diffusion of ether into a CH₂Cl₂ solution of **4a** after one day. ¹H NMR (400 MHz, CD₂Cl₂): δ 13.9 (br, 1H, NH of carbene), 10.3 (br, 1H, NH of amide), 8.26 (d, J = 5.2 Hz, 1H), 7.99-8.06 (m, 2H), 7.44 (td, J = 7.8, J = 1.4 Hz, 1H), 7.05 (d, J = 1.4 Hz, 1H), 5.01 (d, J = 2.4 Hz, 2H, COD), 3.85 (t, J = 2.3 Hz, 2H, COD), 2.71 (s, 3H, CH₃), 2.34 (s, 3H, CH₃), 2.33-2.37 (m, 3H, COD), 2.17-2.25 (m, 3H, COD), 2.12 (s, 2H, COD). ¹³C NMR (75 MHz, CD₂Cl₂): δ 180.0 (Ir-C), 175.2, 164.4, 151.8, 150.7, 146.6, 142.2, 141.0, 124.6, 123.5, 112.6, 91.1, 63.9, 32.8, 30.0, 24.1 ppm. Anal. Calcd for C₂₁H₂₅BF₄IrN₃O (614.5): C, 41.05; H, 4.10; N, 6.84; Found: C, 41.21; H, 4.31; N, 6.48.

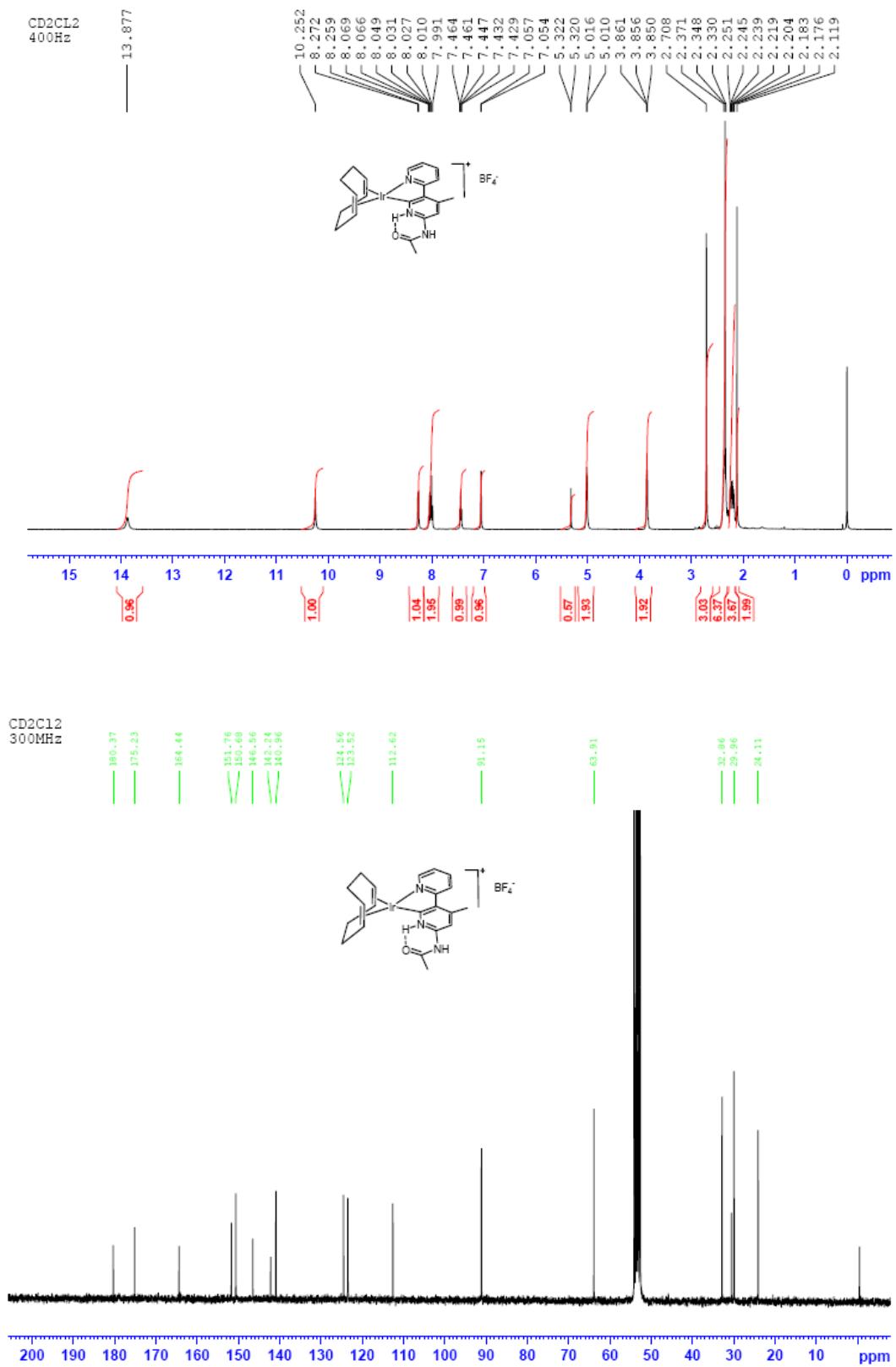
Complex 4b. Complex **4b** was synthesized by following the above general procedure using **3a** and Ir(COD)₂BF₄ in acetone. Yield: 95%. ¹H NMR (300 MHz, DMSO-d₆): δ 14.2 (br, 1H, NH of carbene), 10.8 (br, 1H, NH of amide), 8.72 (s, 1H), 8.66 (d, J = 5.4 Hz, 1H), 8.33 (d, J = 8.0Hz, 1H), 8.14 (t, J = 7.7 Hz, 1H), 7.51 (t, J = 6.3 Hz, 1H), 3.80 (br, 4H, COD), 2.35 (s, 3H, CH₃), 2.34 (s, 3H), 2.27-2.29 (m, 3H, COD), 2.01-2.08 (m, 5H, COD). ¹³C NMR (75 MHz, DMSO-d₆): δ 175.4 (Ir-C), 175.0, 160.1, 150.5, 147.4, 140.4, 138.8, 137.8, 124.9, 121.1, 116.3, 33.0 (br), 24.8, 16.3. Anal. Calcd for C₂₁H₂₅BF₄IrN₃O (614.5): C, 41.05; H, 4.10; N, 6.84; Found: C, 41.43; H, 4.44; N, 6.61.

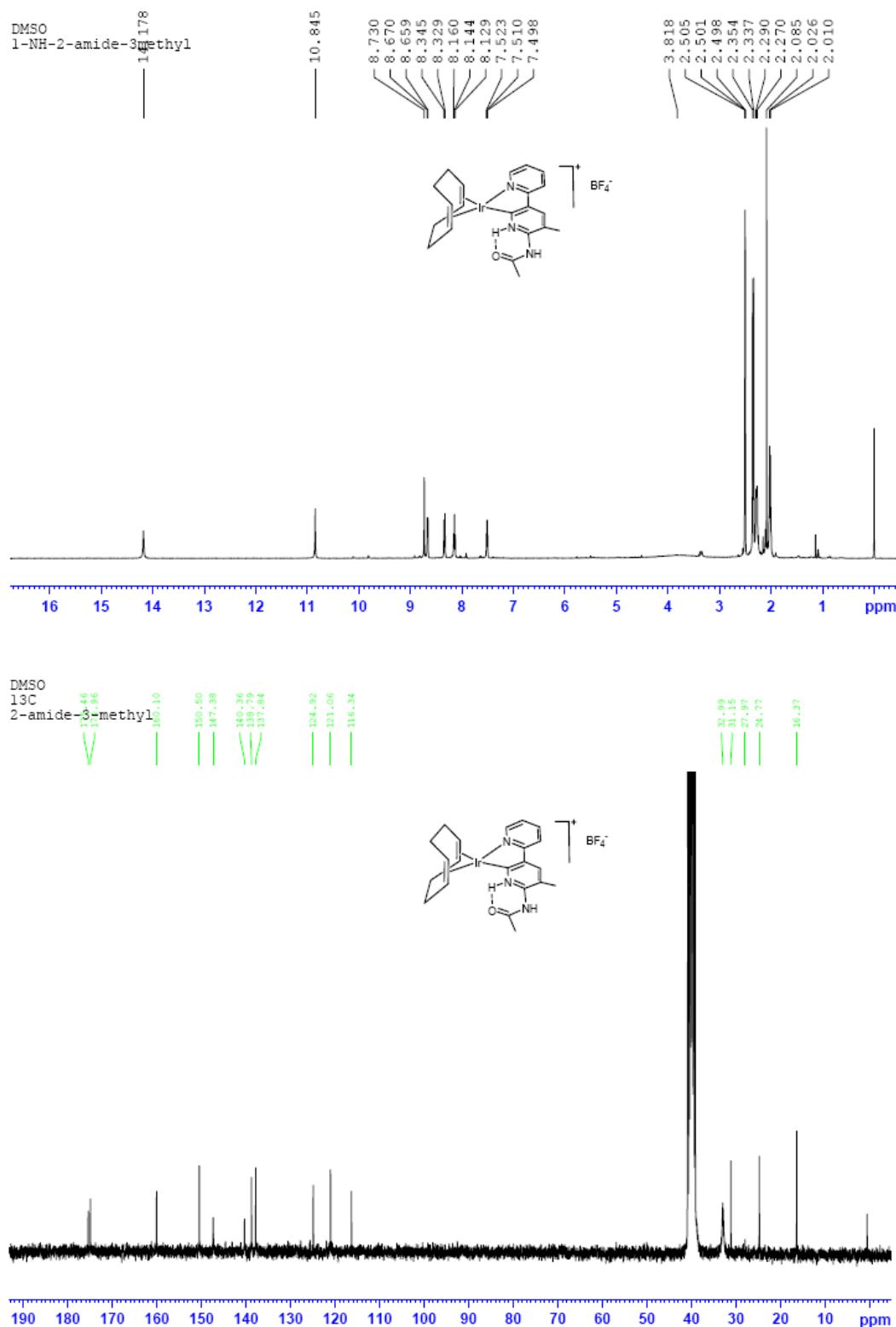
Complex 4c. Complex **4c** was synthesized by following the above general procedure using **3c** and Ir(COD)₂BF₄ in acetone. Yield: 97%. ¹H NMR (400 MHz, CD₂Cl₂): δ 14.5 (br, 1H, NH of carbene), 9.6 (br, 1H, NH of amide), 8.32 (d, J = 4.3 Hz, 1H), 8.06-8.10 (m, 2H), 7.87 (d, J = 8.0 Hz, 1H), 7.48 (t, J = 6.6 Hz, 1H), 4.92 (br, 2H, COD), 3.88 (br, 2H, COD), 2.84 (q, J = 7.4Hz, 2H, CH₂), 2.69 (s, 3H, CH₃), 2.38-2.56 (m, 4H, COD), 2.18-2.24 (m, 4H, COD), 1.25 (t, J = 7.4 Hz, 1H). ¹³C NMR (100 MHz, CD₂Cl₂): δ 179.6, 177.6 (Ir-C), 163.3, 150.2, 147.2, 142.7, 141.1, 137.7, 125.0, 119.9, 118.4, 90.8, 62.6, 33.1, 30.1, 15.9, 8.3. Anal. Calcd for C₂₂H₂₇BF₄IrN₃O (628.5): C, 42.04; H, 4.33; N, 6.69; Found: C, 41.91; H, 4.52 N, 6.60.

Complex 4d. Complex **4d** was synthesized by following the above general procedure using **3d** and Ir(COD)₂BF₄ in acetone. Yield 96%. ¹H NMR (400 MHz, DMSO-d₆): δ 13.8 (br, 1H, NH of carbene), 12.0 (br, 1H, NH of amide), 8.74 (d, J = 8.6 Hz, 1H), 8.64 (d, J = 5.1 Hz, 1H), 8.30 (d, J = 7.9 Hz, 1H), 8.14 (t, J = 7.6 Hz, 1H), 7.52 (t, J = 6.2 Hz, 1H), 7.04 (d, J = 8.5 Hz, 1H), 3.78 (br, 4H, COD), 2.27 (s, 3H, CH₃), 2.25-2.28 (m, 4H, COD), 2.03-2.08 (m, 4H, COD) ppm. ¹³C NMR (100 MHz, DMSO-d₆): δ 178.7 (Ir-C), 174.2, 160.0, 150.4, 149.0, 140.0, 137.9, 137.4, 124.9, 121.0, 106.4, 32.9, 24.7. Anal. Calcd for C₂₀H₂₃BF₄IrN₃O (600.4): C, 40.01; H, 3.86; N, 7.00; Found: C, 40.22; H, 3.91; N, 6.85.

Complex 4e. Complex **4e** was synthesized by following the above general procedure using **3e** and Ir(COD)₂BF₄ in acetone. Yield: 98%. ¹H NMR (300 MHz, CD₂Cl₂): δ 14.2 (br, 1H, NH of carbene), 10.4 (br, 1H, NH of amide), 8.21-8.25 (m, 2H), 8.07 (td, J = 8.0, J = 1.5 Hz, 1H), 7.80 (d,

$J = 8.0$ Hz, 1H), 7.48 (t, $J = 6.0$ Hz, 1H), 7.30 (dd, $J = 8.7, J = 2.0$ Hz, 1H), 5.09 (br, 2H, COD), 3.93 (br, 2H, COD), 2.70 (q, $J = 7.4$ Hz, 1H), 2.36-2.41 (m, 4H, COD), 2.19-2.28 (m, 4H, COD), 1.24 (t, $J = 7.4$ Hz, 3H). ^{13}C NMR (75 MHz, DMSO- d_6): δ 178.6 (Ir-C), 177.6, 160.0, 150.4, 149.0, 139.9, 137.8, 137.4, 124.8, 121.0, 106.4, 33.0, 31.2, 8.9. Anal. Calcd for $\text{C}_{21}\text{H}_{25}\text{BF}_4\text{IrN}_3\text{O}$ (614.5): C, 41.05; H, 4.10; N, 6.84; Found 41.24; H, 4.01; N, 6.99.





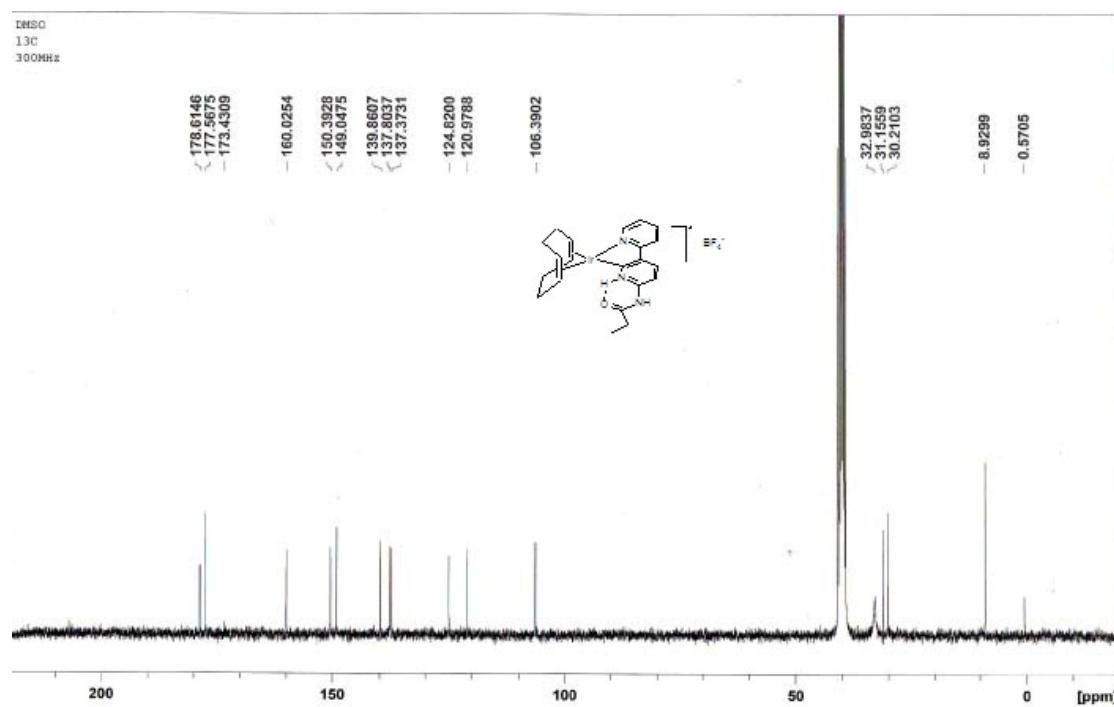
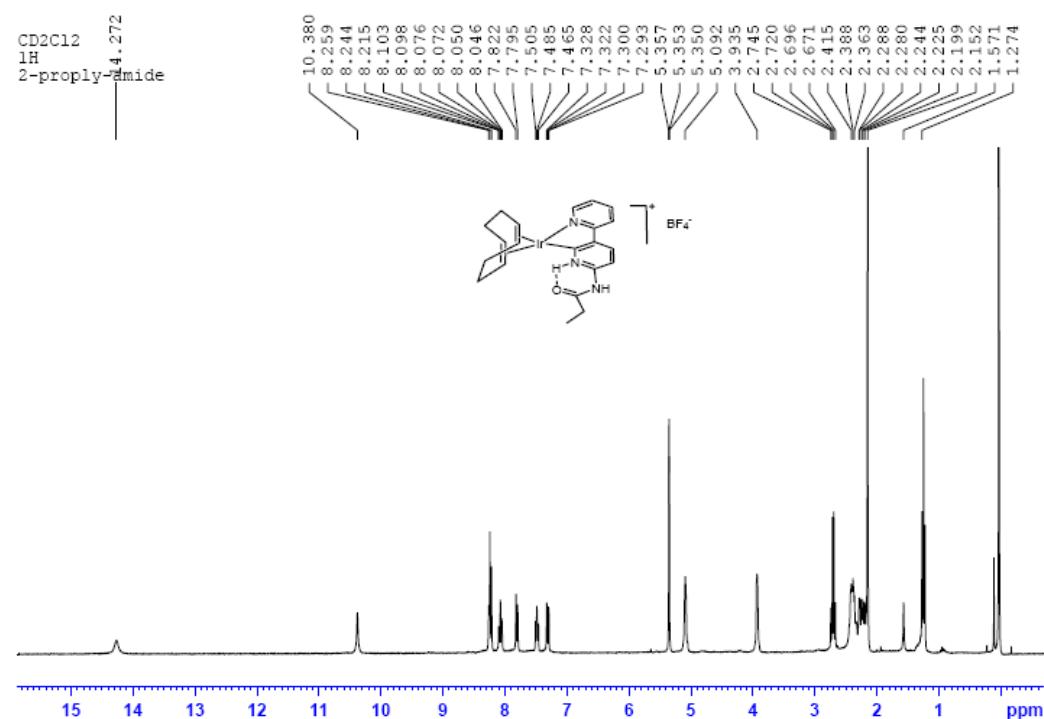


Table 1. Crystal data and structure refinement for complex **4a**.

Empirical formula	$C_{21}H_{25}BF_4IrN_3O$		
Formula weight	614.45		
Temperature	173(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	$P2_1/n$		
Unit cell dimensions	$a = 7.7588(2)$ Å	$\alpha = 90^\circ$.	
	$b = 16.9174(5)$ Å	$\beta = 103.7470(10)^\circ$.	
	$c = 15.9163(5)$ Å	$\gamma = 90^\circ$.	
Volume	$2029.31(10)$ Å ³		
Z	4		
Density (calculated)	2.011 Mg/m ³		
Absorption coefficient	6.635 mm ⁻¹		
F(000)	1192		
Crystal size	0.30 x 0.04 x 0.04 mm ³		
Theta range for data collection	2.41 to 30.54°.		
Index ranges	-11≤h≤11, -24≤k≤24, -20≤l≤22		
Reflections collected	32967		
Independent reflections	6170 [R(int) = 0.0256]		
Completeness to theta = 30.54°	99.2 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7772 and 0.2408		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	6170 / 77 / 343		
Goodness-of-fit on F ²	1.061		
Final R indices [I>2sigma(I)]	R1 = 0.0179, wR2 = 0.0411		
R indices (all data)	R1 = 0.0246, wR2 = 0.0451		
Largest diff. peak and hole	1.377 and -0.610 e·Å ⁻³		

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
Ir(1)	7498(1)	10208(1)	2914(1)	18(1)
B(1)	1336(5)	7364(2)	2333(2)	35(1)
C(1)	8340(30)	10538(8)	1731(10)	38(4)
C(2)	6801(17)	10959(9)	1745(8)	26(2)
C(1A)	8470(20)	10579(7)	1762(9)	36(3)
C(2A)	6717(18)	10834(9)	1670(8)	37(3)
C(3)	5336(11)	10557(5)	1062(5)	31(2)
C(4)	4289(12)	9961(6)	1492(5)	32(2)
C(3A)	4759(11)	10614(4)	1264(5)	32(2)
C(4A)	4527(11)	9718(5)	1314(4)	29(2)
C(5)	5529(3)	9437(2)	2197(2)	29(1)
C(6)	7214(4)	9108(2)	2251(2)	27(1)
C(7A)	8376(13)	9068(7)	1587(6)	31(2)
C(8A)	8368(13)	9863(4)	1126(5)	30(2)
C(7)	7852(15)	9107(6)	1420(7)	31(2)
C(8)	9088(12)	9802(5)	1359(6)	33(2)
C(9)	9243(3)	11810(1)	3446(2)	25(1)
C(10)	9948(3)	12423(1)	3997(2)	29(1)
C(11)	9927(3)	12361(1)	4851(2)	30(1)
C(12)	9229(3)	11687(1)	5146(2)	26(1)
C(13)	8538(3)	11080(1)	4575(1)	19(1)
C(14)	7787(3)	10325(1)	4798(1)	19(1)
C(15)	7296(3)	9812(1)	4070(1)	18(1)
C(16)	7519(3)	10096(1)	5607(1)	20(1)
C(17)	7979(3)	10578(2)	6418(2)	26(1)
C(18)	6793(3)	9350(1)	5681(1)	22(1)
C(19)	6373(3)	8848(1)	4981(1)	20(1)
C(20)	5424(3)	7524(1)	4408(2)	27(1)
C(21)	4674(5)	6766(2)	4663(2)	41(1)
F(1)	2435(3)	6708(1)	2479(1)	54(1)
F(2)	-103(2)	7195(1)	1626(1)	38(1)
F(3)	680(3)	7530(1)	3049(1)	54(1)
F(4)	2229(3)	8016(1)	2119(1)	57(1)

N(1)	8559(2)	11151(1)	3715(1)	20(1)
N(2)	6635(2)	9089(1)	4212(1)	19(1)
N(3)	5653(3)	8107(1)	5033(1)	24(1)
O(1)	5808(3)	7626(1)	3718(1)	34(1)

Table 3. Bond lengths [\AA] and angles [$^\circ$] for **4a**.

Ir(1)-C(15)	1.999(2)
Ir(1)-N(1)	2.0860(19)
Ir(1)-C(5)	2.123(2)
Ir(1)-C(6)	2.125(2)
Ir(1)-C(2A)	2.200(10)
Ir(1)-C(1)	2.205(11)
Ir(1)-C(2)	2.212(10)
Ir(1)-C(1A)	2.232(9)
B(1)-F(3)	1.383(3)
B(1)-F(1)	1.386(4)
B(1)-F(4)	1.387(4)
B(1)-F(2)	1.413(4)
C(1)-C(2)	1.392(11)
C(1)-C(8)	1.551(11)
C(2)-C(3)	1.532(11)
C(1A)-C(2A)	1.401(11)
C(1A)-C(8A)	1.569(11)
C(2A)-C(3A)	1.548(11)
C(3)-C(4)	1.552(12)
C(4)-C(5)	1.567(8)
C(3A)-C(4A)	1.531(11)
C(4A)-C(5)	1.511(7)
C(5)-C(6)	1.406(4)
C(6)-C(7)	1.518(10)
C(6)-C(7A)	1.544(8)
C(7A)-C(8A)	1.532(10)
C(7)-C(8)	1.534(10)
C(9)-N(1)	1.348(3)
C(9)-C(10)	1.385(3)
C(10)-C(11)	1.367(4)
C(11)-C(12)	1.391(4)
C(12)-C(13)	1.390(3)
C(13)-N(1)	1.378(3)
C(13)-C(14)	1.483(3)
C(14)-C(16)	1.407(3)
C(14)-C(15)	1.425(3)

C(15)-N(2)	1.366(3)
C(16)-C(18)	1.397(3)
C(16)-C(17)	1.497(3)
C(18)-C(19)	1.377(3)
C(19)-N(2)	1.350(3)
C(19)-N(3)	1.383(3)
C(20)-O(1)	1.216(3)
C(20)-N(3)	1.383(3)
C(20)-C(21)	1.503(3)

C(15)-Ir(1)-N(1)	78.44(8)
C(15)-Ir(1)-C(5)	94.90(9)
N(1)-Ir(1)-C(5)	158.03(9)
C(15)-Ir(1)-C(6)	98.20(9)
N(1)-Ir(1)-C(6)	162.35(9)
C(5)-Ir(1)-C(6)	38.64(10)
C(15)-Ir(1)-C(2A)	157.6(5)
N(1)-Ir(1)-C(2A)	99.6(4)
C(5)-Ir(1)-C(2A)	78.6(4)
C(6)-Ir(1)-C(2A)	90.0(5)
C(15)-Ir(1)-C(1)	167.0(5)
N(1)-Ir(1)-C(1)	100.0(5)
C(5)-Ir(1)-C(1)	91.0(5)
C(6)-Ir(1)-C(1)	79.3(4)
C(2A)-Ir(1)-C(1)	35.3(9)
C(15)-Ir(1)-C(2)	155.5(5)
N(1)-Ir(1)-C(2)	93.2(4)
C(5)-Ir(1)-C(2)	84.2(3)
C(6)-Ir(1)-C(2)	96.3(4)
C(2A)-Ir(1)-C(2)	6.3(8)
C(1)-Ir(1)-C(2)	36.7(3)
C(15)-Ir(1)-C(1A)	165.0(5)
N(1)-Ir(1)-C(1A)	97.2(4)
C(5)-Ir(1)-C(1A)	94.1(5)
C(6)-Ir(1)-C(1A)	81.6(3)
C(2A)-Ir(1)-C(1A)	36.8(3)
C(1)-Ir(1)-C(1A)	3.1(10)
C(2)-Ir(1)-C(1A)	37.8(8)

F(3)-B(1)-F(1)	110.9(3)
F(3)-B(1)-F(4)	110.2(2)
F(1)-B(1)-F(4)	110.8(3)
F(3)-B(1)-F(2)	108.8(3)
F(1)-B(1)-F(2)	108.0(2)
F(4)-B(1)-F(2)	108.1(3)
C(2)-C(1)-C(8)	144.8(18)
C(2)-C(1)-Ir(1)	71.9(6)
C(8)-C(1)-Ir(1)	109.6(8)
C(1)-C(2)-C(3)	104.8(16)
C(1)-C(2)-Ir(1)	71.4(6)
C(3)-C(2)-Ir(1)	109.5(7)
C(2A)-C(1A)-C(8A)	105.7(16)
C(2A)-C(1A)-Ir(1)	70.3(6)
C(8A)-C(1A)-Ir(1)	109.8(7)
C(1A)-C(2A)-C(3A)	143.3(16)
C(1A)-C(2A)-Ir(1)	72.8(6)
C(3A)-C(2A)-Ir(1)	107.0(7)
C(2)-C(3)-C(4)	110.7(8)
C(3)-C(4)-C(5)	112.7(6)
C(4A)-C(3A)-C(2A)	109.4(8)
C(5)-C(4A)-C(3A)	108.6(6)
C(6)-C(5)-C(4A)	116.5(4)
C(6)-C(5)-C(4)	132.2(4)
C(4A)-C(5)-C(4)	20.7(3)
C(6)-C(5)-Ir(1)	70.77(14)
C(4A)-C(5)-Ir(1)	116.9(3)
C(4)-C(5)-Ir(1)	106.2(4)
C(5)-C(6)-C(7)	116.0(5)
C(5)-C(6)-C(7A)	131.4(4)
C(7)-C(6)-C(7A)	16.3(6)
C(5)-C(6)-Ir(1)	70.59(15)
C(7)-C(6)-Ir(1)	114.6(4)
C(7A)-C(6)-Ir(1)	111.7(5)
C(8A)-C(7A)-C(6)	110.7(7)
C(7A)-C(8A)-C(1A)	112.0(8)
C(6)-C(7)-C(8)	113.1(8)
C(7)-C(8)-C(1)	107.1(10)

N(1)-C(9)-C(10)	123.0(2)
C(11)-C(10)-C(9)	118.4(2)
C(10)-C(11)-C(12)	119.8(2)
C(13)-C(12)-C(11)	120.3(2)
N(1)-C(13)-C(12)	119.4(2)
N(1)-C(13)-C(14)	114.22(19)
C(12)-C(13)-C(14)	126.3(2)
C(16)-C(14)-C(15)	120.5(2)
C(16)-C(14)-C(13)	127.5(2)
C(15)-C(14)-C(13)	112.03(19)
N(2)-C(15)-C(14)	116.66(19)
N(2)-C(15)-Ir(1)	124.57(16)
C(14)-C(15)-Ir(1)	118.70(15)
C(18)-C(16)-C(14)	118.5(2)
C(18)-C(16)-C(17)	116.0(2)
C(14)-C(16)-C(17)	125.6(2)
C(19)-C(18)-C(16)	120.7(2)
N(2)-C(19)-C(18)	119.3(2)
N(2)-C(19)-N(3)	118.5(2)
C(18)-C(19)-N(3)	122.1(2)
O(1)-C(20)-N(3)	122.1(2)
O(1)-C(20)-C(21)	124.0(2)
N(3)-C(20)-C(21)	113.9(2)
C(9)-N(1)-C(13)	119.1(2)
C(9)-N(1)-Ir(1)	124.50(16)
C(13)-N(1)-Ir(1)	116.40(14)
C(19)-N(2)-C(15)	124.30(19)
C(20)-N(3)-C(19)	126.04(19)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
Ir(1)	23(1)	18(1)	12(1)	0(1)	5(1)	0(1)
B(1)	45(2)	37(2)	26(2)	-12(1)	12(1)	-6(1)
C(1)	69(6)	24(5)	26(6)	-7(4)	18(5)	-20(5)
C(2)	55(4)	10(4)	9(3)	0(3)	-3(3)	-6(3)
C(1A)	69(6)	26(5)	27(5)	-1(4)	37(5)	-3(5)
C(2A)	75(5)	14(5)	17(4)	-5(3)	-1(3)	-11(3)
C(3)	31(4)	34(3)	24(3)	7(3)	0(2)	4(3)
C(4)	35(4)	44(5)	15(3)	-3(3)	0(3)	-4(3)
C(3A)	37(4)	33(3)	22(3)	-2(2)	-4(2)	11(3)
C(4A)	32(3)	39(4)	14(3)	-4(2)	0(2)	-1(3)
C(5)	31(1)	36(1)	18(1)	2(1)	2(1)	-11(1)
C(6)	41(1)	26(1)	18(1)	-7(1)	13(1)	-9(1)
C(7A)	38(4)	38(4)	20(4)	-9(3)	15(3)	0(3)
C(8A)	39(4)	26(3)	30(4)	-4(3)	22(3)	4(3)
C(7)	52(6)	16(3)	31(5)	-1(3)	17(4)	1(4)
C(8)	34(4)	41(4)	28(4)	-3(3)	17(3)	0(3)
C(9)	30(1)	22(1)	24(1)	2(1)	7(1)	0(1)
C(10)	30(1)	18(1)	37(2)	2(1)	6(1)	1(1)
C(11)	32(1)	23(1)	35(1)	-8(1)	4(1)	-1(1)
C(12)	30(1)	26(1)	21(1)	-6(1)	5(1)	0(1)
C(13)	19(1)	21(1)	17(1)	-1(1)	4(1)	3(1)
C(14)	19(1)	25(1)	11(1)	-5(1)	0(1)	1(1)
C(15)	19(1)	20(1)	17(1)	0(1)	5(1)	2(1)
C(16)	17(1)	26(1)	15(1)	0(1)	3(1)	4(1)
C(17)	30(1)	34(1)	15(1)	-6(1)	6(1)	-1(1)
C(18)	24(1)	28(1)	14(1)	2(1)	6(1)	3(1)
C(19)	20(1)	24(1)	17(1)	4(1)	5(1)	3(1)
C(20)	32(1)	25(1)	26(1)	1(1)	10(1)	-2(1)
C(21)	63(2)	29(1)	37(2)	-1(1)	24(1)	-12(1)
F(1)	56(1)	52(1)	48(1)	-11(1)	1(1)	9(1)
F(2)	40(1)	50(1)	27(1)	-10(1)	11(1)	-3(1)
F(3)	87(2)	50(1)	32(1)	-17(1)	28(1)	-11(1)
F(4)	60(1)	49(1)	68(1)	-11(1)	29(1)	-19(1)

N(1)	23(1)	19(1)	20(1)	0(1)	6(1)	2(1)
N(2)	23(1)	21(1)	14(1)	0(1)	6(1)	-1(1)
N(3)	31(1)	24(1)	18(1)	2(1)	10(1)	-2(1)
O(1)	50(1)	30(1)	24(1)	-4(1)	16(1)	-9(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**.

	x	y	z	U(eq)
H(1)	9348	10916	1914	46
H(2)	6837	11548	1799	32
H(1A)	9452	10974	1827	43
H(2A)	6755	11413	1798	45
H(3A)	5871	10276	641	37
H(3B)	4514	10963	744	37
H(4A)	3575	9614	1040	39
H(4B)	3459	10257	1762	39
H(3A1)	3973	10886	1579	39
H(3A2)	4428	10787	653	39
H(4A1)	3253	9586	1222	35
H(4A2)	4992	9453	859	35
H(5)	4830(40)	9204(17)	2573(17)	35
H(6)	7520(40)	8670(14)	2687(16)	33
H(7A1)	9608	8929	1887	37
H(7A2)	7919	8651	1156	37
H(8A1)	7271	9906	660	36
H(8A2)	9392	9885	856	36
H(7A)	6810	9129	921	38
H(7B)	8487	8606	1381	38
H(8A)	9122	9895	749	39
H(8B)	10307	9688	1697	39
H(9)	9243	11856	2851	31
H(10)	10433	12877	3787	35
H(11)	10388	12776	5242	36
H(12)	9224	11640	5740	31
H(17A)	9266	10660	6588	39
H(17B)	7593	10300	6881	39
H(17C)	7380	11092	6315	39
H(18)	6588	9188	6221	26
H(21A)	3617	6619	4218	61
H(21B)	4350	6840	5217	61
H(21C)	5565	6347	4721	61

H(2N)	6362	8758	3774	23
H(3N)	5309	7996	5508	28

Table 6. Hydrogen bonds for **4a** [\AA and $^\circ$].

D-H...A	d(D-H)	d(H...A)	d(D...A)	\angle (DHA)
N(2)-H(2N)...O(1)	0.88	1.96	2.629(3)	131.8
N(3)-H(3N)...F(2) ^{#1}	0.88	1.91	2.782(2)	171.7

Symmetry transformations used to generate equivalent atoms:

#1 x+1/2,-y+3/2,z+1/2