# Amino-substituted rylene dicarboximides and their quinoidal charge delocalization after deprotonation

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#### **Supporting information**

General: The solvents and chemicals used were of commercial grade. *N*-(2, 6-diisopropylphenyl)-4-bromo-naphthalene 1,8-dicarboximide (6), *N*-(2, 6-diisopropylphenyl)-9-bromo-perylene 3, 4dicarboximide (7), *N*-(2, 6-diisopropylphenyl)-1,6,9,14-tetrakis(4-(1,1,3,3-tetramethylbutyl) phenoxy)-11-bromo-terrylene 3,4-dicarboximide (8) and *N*-(2, 6-diisopropylphenyl)-1,6-bis(4-(1,1,3,3-tetramethylbutyl)phenoxy)-13-bromo-quarterrylene 3, 4-dicarboximide (9) were synthesized as previously described <sup>1-3</sup>. Column chromatography was performed on silica gel (Geduran Si<sub>60</sub>, Merck). <sup>1</sup>H, <sup>13</sup>C, H, H COSY and NOE NMR were recorded on Bruker DPX 250, DRX 500 and Avance 700 NMR spectrometers. Infrared spectra were obtained on a Nicolet FT IR320. FD mass spectra were recorded with a VG-instruments ZAB 2-SE-FPD instrument. MALDI-TOF mass spectra were recorded on a Bruker MALDI-TOF spectrometer. UV/Vis/NIR spectra were recorded in 1-cm quartz cuvette on a Perkin-Elmer Lambda 900 spectrophotometer. Elemental analyses were performed by the Department of Chemistry and Pharmacy of the University of Mainz.

#### General procedure for Buchwald condition.

The bromo-RI (0.18 mmol), 4-aminobenzonitrile (0.36 mmol), tris-(dibenzylideneacetone)-dipalladium(0) (17 mg, 0.018 mmol), tris-*tert*-butylphosphine (18 mg, 0.09mmol), sodium-*tert*-butoxide (67 mg, 0.69 mmol) and dry toluene (10 mL) were stirred at 80°C under argon overnight. After cooling, the mixture was evaporated in vacuo and purified by a column chromatography on silica gel using dichloromethane as an eluent.

#### *N*-(2,6-diisopropylphenyl)-4-( (*p*-cyanophenyl)amino)naphthalene 1,8-dicarboximide (2)

Using **6**, the general procedure afforded the title compound (81 mg, 95%). <sup>1</sup>H NMR (700MHz, acetone- $d_6$ , 50°C):  $\delta = 8.90(s, N-H, 1H)$ , 8.76 (d, J=8.4 Hz, 1H), 8.63(d, J=7.2 Hz, 1H), 8.53(d, J=8.1 Hz, 1H), 7.87(t,  $J_1=7.6$  Hz,  $J_2=8.1$  Hz, 1H), 7.80(d, J=8.1 Hz, 1H), 7.74(d, J=8.3 Hz, 2H), 7.54 (d, J=8.2 Hz, 2H), 7.42(t, J=7.8 Hz, 1H), 7.33(d, J=7.8 Hz, 2H), 2.64 (sep, J=6.8 Hz, 2H), 1.13(d, J=6.8 Hz, 12H); H,H COSY NMR (700MHz, acetone- $d_6$ , 50°C): coupling of  $\delta=(8.76, 8.61, 7.87)$ , (8.53, 7.80), (7.74, 7.54), (7.42, 7.33); <sup>13</sup>C-NMR (62.5 MHz, acetone- $d_6$ , 25°C):  $\delta = 165.0$ , 164.3, 147.4, 146.8, 145.8, 134.6, 133.7, 132.7, 132.5, 131.3, 129.8, 129.7, 127.1, 125.4, 124.5, 124.0, 120.0, 119.7, 116.6, 114.4, 105.4, 30.6, 24.2 ppm. IR: v =3315, 3068, 2962, 2929, 2869, 2219, 1700, 1648, 1575, 1508, 1463, 1357, 1309, 1236, 1164, 1056, 1010, 917, 831, 777 cm<sup>-1</sup>; UV-Vis (acetone)  $\lambda_{max}$ , nm ( $\epsilon$ ): 431 (21300); MS (FD): m/z, 473.4 (100%), M+; Found: C, 78.81; H, 5.87; N, 8.47 %.

The NMR data of **2'** (deprotonation by NaOH). <sup>1</sup>H NMR (500MHz, acetone-*d*<sub>6</sub>, 25°C): δ = 8.83(d, *J*=7.9 Hz, 1H), 8.27(d, *J*=7.4 Hz, 1H), 7.82(d, *J*=9.1 Hz, 1H), 7.47(d, *J*=8.5 Hz, 2H), 7.29(t, *J*=7.7 Hz, 1H), 7.24(t, *J*<sub>1</sub>=7.7 Hz, *J*<sub>2</sub>=7.6 Hz, 1H), 7.21(d, *J*=7.7 Hz, 2H), 7.08(d, *J*=8.5 Hz, 2H), 6.52 (d, *J*=9.1 Hz, 1H), 2.64 (sep, *J*=6.8 Hz, 2H), 1.13(d, *J*=6.8 Hz, 12H).

#### N-(2, 6-diisopropylphenyl)-9-( (p-cyanophenyl)amino)perylene 3,4-dicarboximide (3a)

Using 7, the general procedure afforded the title compound (90 mg, 84%). <sup>1</sup>H NMR (700MHz, acetone- $d_6$ , 50°C):  $\delta = 8.78(d, J=7.5$  Hz, 1H), 8.72(d, J=8.0 Hz, 1H), 8.70(d, J=8.3 Hz, 1H), 8.63(d, J=8.0 Hz, 1H), 8.60(d, J=8.0 Hz, 1H), 8.58(d, J=8.0 Hz, 1H), 8.49(s, 1N-H), 8.35(d, J=8.4 Hz, 1H), 7.76-7.74(m, 2H), 7.66(d, J=8.5 Hz, 2H), 7.44(t, J=7.9 Hz, 1H), 7.36-7.33(m, 4H), 2.83 (sep, J=6.8 Hz, 2H), 1.16(d, J=6.9 Hz, 12H); H,H COSY NMR (700MHz, acetone- $d_6$ , 50°C): coupling of  $\delta = (8.78, 8.35, 7.76)$ , (8.70, 7.76), (8.72, 8.60), (8.63, 8.58), (7.66, 7.36), (7.44,7.36), (2.83, 1.16); NOE NMR (700MHz, acetone- $d_6$ , 50°C): coupling of  $\delta = (8.78, 8.72)$ , (8.70, 8.63), (8.49, 7.36), (8.49, 8.35), (7.76, 7.36); <sup>13</sup>C-NMR (62.5 MHz, CD<sub>2</sub>Cl<sub>2</sub> 25°C):  $\delta = 164.6, 164.4, 148.0, 146.5, 139.8, 137.8, 134.3, 132.3, 132.12, 132.0, 130.2, 129.7, 128.6, 127.4, 125.7, 125.2, 125.0, 124.7, 124.4, 121.5, 120.8, 120.5, 120.0, 119.7, 118.7, 117.2, 103.6, 29.5, 24.1 ppm. IR: v = 3328, 3064, 2962, 2929, 2867, 2217, 1689, 1644, 1563, 1504, 1351, 1288,$ 

1172, 1058, 971, 910, 804, 750 cm<sup>-1</sup>; UV-Vis (acetone)  $\lambda_{max}$ , nm ( $\epsilon$ ): 550(34800); MS (FD): m/z, 598.0 (100%), M<sup>+</sup>; Found: C, 81.75; H, 5.42; N, 6.98 %. Calc. for C<sub>41</sub>H<sub>31</sub>N<sub>3</sub>O<sub>2</sub>: C, 82.39; H, 5.23; N, 7.03 %.

The NMR data of **3a'** (deprotonation by NaOH). <sup>1</sup>H NMR (700MHz, acetone- $d_6$ , 25°C):  $\delta$ =8.81 (d, J=7.7 Hz, 1H), 8.65(d, J=7.5 Hz, 1H), 8.29(d, J=8.3 Hz, 1H), 8.18(d, J=8.4 Hz, 1H), 8.09(d, J=8.8 Hz, 1H), 8.06(d, J=9.7 Hz, 1H), 7.63(d, J=8.6 Hz, 1H), 7.57(d, J=8.4 Hz, 2H), 7.45(t, J=7.7 Hz, 1H), 7.32(t, J=7.6 Hz, 1H), 7.25(d, J=7.8 Hz, 2H), 7.10(d, J=8.4 Hz, 2H), 6.60(d, J=9.4 Hz, 1H), 2.83 (sep, J=6.8 Hz, 2H), 1.16(d, J=6.9Hz, 12H); H,H COSY NMR (700MHz, acetone- $d_6$ , 25°C): coupling of  $\delta$ =(8.81, 8.65, 7.45), (8.29, 8.18), (8.09, 7.63), (8.06, 6.60), (7.57, 7.10), (7.32, 7.25), (2.83, 1.16); NOE NMR (700MHz, acetone- $d_6$ , 25°C): coupling of  $\delta$ =(8.65, 8.18), (8.06, 7.63), (7.10, 6.60).

#### N-(2, 6-diisopropylphenyl)-9-( (p-octylphenyl)amino)perylene 3,4-dicarboximide (3b)

Using 7 and *p*-octylaniline instead of 4-aminobenzonitrile, the general procedure afforded the title compound (95 mg, 77%). <sup>1</sup>H NMR (700MHz, acetone- $d_6$ , 50°C):  $\delta = 8.75$  (d, J=7.3 Hz, 1H), 8.63 (d, J=8.1 Hz, 1H), 8.55 (d, J=7.9 Hz, 1H), 8.54 (d, J=8.6 Hz, 1H), 8.50 (d, J=8.1 Hz, 1H), 8.48 (d, J=8.5 Hz, 1H), 8.42 (d, J=8.2 Hz, 1H), 8.12 (s, 1N-H), 7.73 (t, J=8.2 Hz, 1H), 7.45 (t, J=7.9 Hz, 1H), 7.38 (d, J=8.5 Hz, 1H), 7.34 (d, J=7.9 Hz, 2H), 7.34 (d, J=8.4 Hz, 2H), 7.29 (d, J=8.4 Hz, 2H), 2.87-2.70 (sep, J=6.8 Hz, 2H), 2.67-2.61 (t, J=7.7 Hz, 2H), 1.68-1.62 (t, J=7.4 Hz, 2H), 1.34-1.30 (m, 10H), 1.18 (d, J=6.9 Hz, 12H), 0.92-0.87ppm (t, J=7.1Hz, 3H); H,H COSY NMR (700MHz, acetone- $d_6$ , 50°C): coupling of  $\delta=(7.73, 8.75, 8.48)$ , (7.38, 8.54), (7.29, 7.34), (7.45, 7.34), (8.50, 8.42), (8.55, 8.63), (2.87, 1.18), (2.67, 1.68), (1.62, 1.34), (1.30, 0.92); <sup>13</sup>C-NMR (62.5 MHz, CD<sub>2</sub>Cl<sub>2</sub> 25°C):  $\delta = 164.71$ , 164.55, 146.52, 144.34, 139.16, 138.96, 138.71, 138.17, 132.27, 131.88, 131.07, 129.97, 129.76, 129.16, 129.56, 129.16, 126.68, 126.47, 126.47, 125.11, 125.02, 124.35, 123.61, 121.84, 120.92, 120.83, 119.88, 118.69, 118.48, 111.08, 35.74, 32.30, 32.06, 29.89, 29.73, 29.69, 29.45, 24.14, 24.10, 23.08, 14.27 ppm. IR: v = 3372, 2964, 2924, 2854, 1690, 1650, 1566, 1514, 1354, 1284, 804,750 cm-1; UV-Vis (acetone)  $\lambda_{max}$ , nm ( $\epsilon$ ):

596 (35233); MS (FD): m/z 683.5 (100%), M<sup>+</sup>; Found: C, 83.39; H, 7.06; N, 3.96 %. Calc. for C<sub>48</sub>H<sub>48</sub>N<sub>2</sub>O<sub>2</sub>: C, 84.17; H, 7.06; N, 4.09 %.

The NMR data of **3b'** (deprotonation by NaOH): <sup>1</sup>H NMR (250MHz, acetone- $d_6$ , 25 °C):  $\delta$  = 8.83 (d, J=7.6 Hz, 1H), 8.67 (d, J=8.5 Hz, 1H), 8.25 (d, J=8.5 Hz, 1H), 8.11 (d, J=8.7 Hz, 1H), 7.99 (d, J=9.0 Hz, 1H), 7.94 (d, J=10.1 Hz, 1H), 7.48 (d, J=9.0 Hz, 1H), 7.46 (t, J=7.7 Hz, 1H), 7.32 (t, J=7.9 Hz, 1H), 7.25 (d, J=7.9 Hz, 2H), 7.16 (d, J=8.4 Hz, 2H), 6.89 (d, J=8.4 Hz, 2H), 6.52 (d, J=10.0 Hz, 1H), 2.77-2.66 (sep, J=6.9 Hz, 4H), 2.63-2.56 (t, J=7.4 Hz, 2H), 1.64-1.59 (t, J=7.4 Hz, 2H), 1.34-1.30 (m, 10H), 1.13-1.10 (d, J=6.9 Hz, 12H), 0.92-0.86 ppm (t, J=7.0 Hz, 3H); H,H COSY NMR (700MHz, acetone- $d_6$ , 25°C): coupling of  $\delta$ =(7.50, 8.83, 8.67), (6.87, 7.12), (6.52, 6.48), (7.35, 7.25), (7.99, 7.48), (8.11, 8.25), (2.77, 1.13), (2.63, 1.64), (1.64, 1.34), (1.30, 0.92).

## *N*-(2, 6-diisopropylphenyl)-1,6,9,14-tetrakis(4-(1,1,3,3-tetramethylbutyl)phenoxy)-11-( (*p*-cyanophenyl)amino)terrylene 3,4-dicarboximide (4)

Using **8**, the general procedure afforded the title compound (28 mg, 55%). <sup>1</sup>H NMR (250MHz, acetone- $d_6$ , 25°C):  $\delta$  =9.50-9.45 (dd, 2H), 9.22-9.13 (dd, 2H) 8.35(d, J=8.9Hz, 1H), 8.17(d, J=8.9Hz, 1H), 7.68-7.52(m, 10H), 7.46(d, J=7.2Hz, 2H), 7.38-7.28(m, 7H), 7.39-7.11(m, 6H), 2.92(sep, J=6.8Hz, 2H), 1.95-1.89 (q, 8H), 1.55-1.49 (q, 24H), 1.23-1.19 (q, 12H), 1.01(d, J=6.9Hz, 12H), 0.86-0.83ppm (q, 24H); IR: v = 3326, 3054, 2952, 2923, 2865, 2221, 1698, 1594, 1502, 1365, 1303, 1211, 1170, 1052, 1014, 958, 829, 727 cm<sup>-1</sup>; UV-Vis (acetone)  $\lambda_{max}$ , nm ( $\epsilon$ ): 668(41900); MS (FD): m/z, 1539.3 (100%), M<sup>+</sup>.

### *N*-(2, 6-diisopropylphenyl)-1,6-bis(4-(1,1,3,3-tetramethylbutyl)phenoxy)-13-((*p*-cyanophenyl) amino)quarterrylene 3,4-dicarboximide (5)

Using **9**, the general procedure afforded the title compound (25 mg, 53%). IR: v =3359, 3054, 2954, 2925, 2865, 2211, 1693, 1577, 1500, 1384, 1317, 1265, 1209, 1170, 1064, 1014, 806, 744 cm<sup>-1</sup>; UV-Vis (acetone)  $\lambda_{max}$ , nm ( $\epsilon$ ): 751(51200); MS (FD): m/z, 1255.3 (100%), M<sup>+</sup>.



Fig. S1. <sup>1</sup>H NMR spectra of **2** and **2'** (after deprotonation by NaOH): in acetone-d<sub>6</sub>.

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Proton Items	Ha	$H_{\mathfrak{b}}$	H <sub>c</sub>	H <sub>d</sub>	He	$H_{f}$	Hj	$H_l$	H <sub>m</sub>	H <sub>n</sub>		
$\delta_1$ , ppm	7.74	7.54	7.80	8.53	7.42	7.33	8.91	8.76	7.87	8.63		
$J_{ m 1,~Hz}$	8.3	8.3	8.1	8.1	7.8	7.8	/	8.4		7.2		
$\delta_2$ , ppm	7.47	7.08	6.52	7.82	7.29	7.21		8.27	7.24	8.83		
$J_{2,~\mathrm{Hz}}$	8.5	8.5	9.1	9.1	7.7	7.7	$  \wedge  $	7.4		7.9		
$\delta_1$ - $\delta_2$ , ppm	0.27	0.44	1.28	0.71	0.13	0.12	$/ \setminus$	0.49	0.63	-0.20		

Table S1. Chemical shifts and H,H coupling constants of **2** before or after deprotonation in acetone-d6.

 $\delta_1$ ,  $\delta_2$ ,  $J_1$  and  $J_2$  denote the chemical shift and H,H coupling constants before and after deprotonation, respectively.



Fig. S2. <sup>1</sup>H NMR spectra of **3a** and **3a'** (after deprotonation by NaOH) in acetone-d<sub>6</sub>

Proton Items	Ha	H <sub>b</sub>	H <sub>c</sub>	H <sub>d</sub>	He	$\mathrm{H}_{\mathrm{f}}$	Hg	$\mathrm{H}_{\mathrm{h}}$	H <sub>i</sub>	Hj	$H_k$	Hı	H <sub>m</sub>	H <sub>n</sub>
$\delta_1$ , ppm	7.66	7.35	7.75	8.70	7.44	7.35	8.62	8.58	8.60	8.49	8.72	8.78	7.75	8.35
$J_{ m 1,~Hz}$	8.5	8.5	8.3	8.3	7.9	7.9	8.0	8.0	8.0	$\setminus$ /	8.0	7.5		8.4
$\delta_2$ , ppm	7.57	7.10	6.60	8.06	7.32	7.25	7.63	8.09	8.29	$\bigvee$	8.18	8.65	7.45	8.81
$J_{2,~ m Hz}$	8.4	8.4	9.4	9.7	7.6	7.8	8.6	8.8	8.3	$\land$	8.4	7.5		7.7
$\delta_1$ - $\delta_2$ , ppm	0.09	0.25	1.15	0.64	0.12	0.10	0.99	0.49	0.11	$/ \setminus$	0.54	0.13	0.30	-0.46

Table S1. Chemical shifts and H,H coupling constants of **3a** before or after deprotonation in acetone-d6.

 $\delta_1$ ,  $\delta_2$ ,  $J_1$  and  $J_2$  denote the chemical shift and H,H coupling constants before and after deprotonation, respectively.



Fig. S3. <sup>1</sup>H NMR spectra of **3b** and **3b'** (after deprotonation by NaOH) in acetone-d<sub>6</sub>.

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Proton Items	Ha	H <sub>b</sub>	H <sub>c</sub>	H <sub>d</sub>	He	$H_{\rm f}$	Hg	H <sub>h</sub>	H <sub>i</sub>	Hj	H <sub>k</sub>	Hı	H <sub>m</sub>	H <sub>n</sub>
$\delta_1$ , ppm	7.29	7.34	7.38	8.54	7.44	7.34	8.50	8.42	8.55	8.12	8.63	8.75	7.71	8.48
$J_{ m 1,~Hz}$	8.4	8.4	8.5	8.5	7.9	7.9	8.2	8.2	8.1	/	8.1	7.3		8.5
$\delta_2$ , ppm	6.87	7.16	6.52	7.94	7.35	7.25	7.48	7.99	8.25		8.11	8.67	7.46	8.83
$J_{2,~ m Hz}$	8.2	8.2	10.0	10.0	7.5	7.5	9.0	9.0	8.5	$  \wedge  $	8.5	8.5		7.6
$\delta_1$ - $\delta_2$ , ppm	0.42	0.18	0.86	0.60	0.09	0.09	1.02	0.43	0.30	$/ \setminus$	0.50	0.08	0.25	-0.35

Table S1. Chemical shifts and H,H coupling constants of **3b** and **3b'** (after deprotonation by NaOH) in acetone- $d_6$ .

 $\delta_1$ ,  $\delta_2$ ,  $J_1$  and  $J_2$  denote the chemical shifts and H,H coupling constants before and after deprotonation, respectively.

For comparison, the absorption spectrum of the radical cation of **3b**, generated by chemical oxidation with  $SbCl_5$  following the previously reported procedure <sup>4</sup>, was measured. It shows a sharp absorption peak at 710 nm and a very broad and weak band above 1000 nm in the NIR region.



Fig. S4. The absorption spectra of the radical cation of **3b** generated by chemical oxidization upon more and more SbCl<sub>5</sub> addition. (SbCl<sub>5</sub> stock solution in dichloromethane is 0.001M)



Fig. S5. The absorption change of **3b** deprotonated by BuONa in acetone after 3 days



Figure S6. Absorption switches of 3b in acetone for 5 cycles

### Refernces

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 $^1\mathrm{H}$  NMR (250 MHz, 25 °C) spectrum of 4 in acetone-d<sub>6</sub>





MALDI-TOF-MS spectrum of 5



H, H COSY NMR (700MHz, 50°C) spectrum of 2 in acetone-d6.







NOE NMR (700MHz, 50°C) spectrum of **3a** in acetone-d<sub>6</sub>









NOE NMR (700MHz, 25°C) spectrum of **3a'** (deprotonated by NaOH) in acetone-d<sub>6</sub>



H, H COSY NMR (700MHz, 50°C) spectrum of **3b** in acetone-d<sub>6</sub>





H, H COSY NMR (700MHz, 25°C) spectrum of **3b'** (deprotonated by NaOH) in acetone-d<sub>6</sub>