

Novel binuclear lanthanide luminescent peptide conjugates: Structural analysis in aqueous solution of potential metallo enzymatic mimics

Célia S. Bonnet, Marc Devocelle and Thorfinnur Gunnlaugsson*

Electronic Supplementary Information

Figure 1. ES MS spectrum of **1**

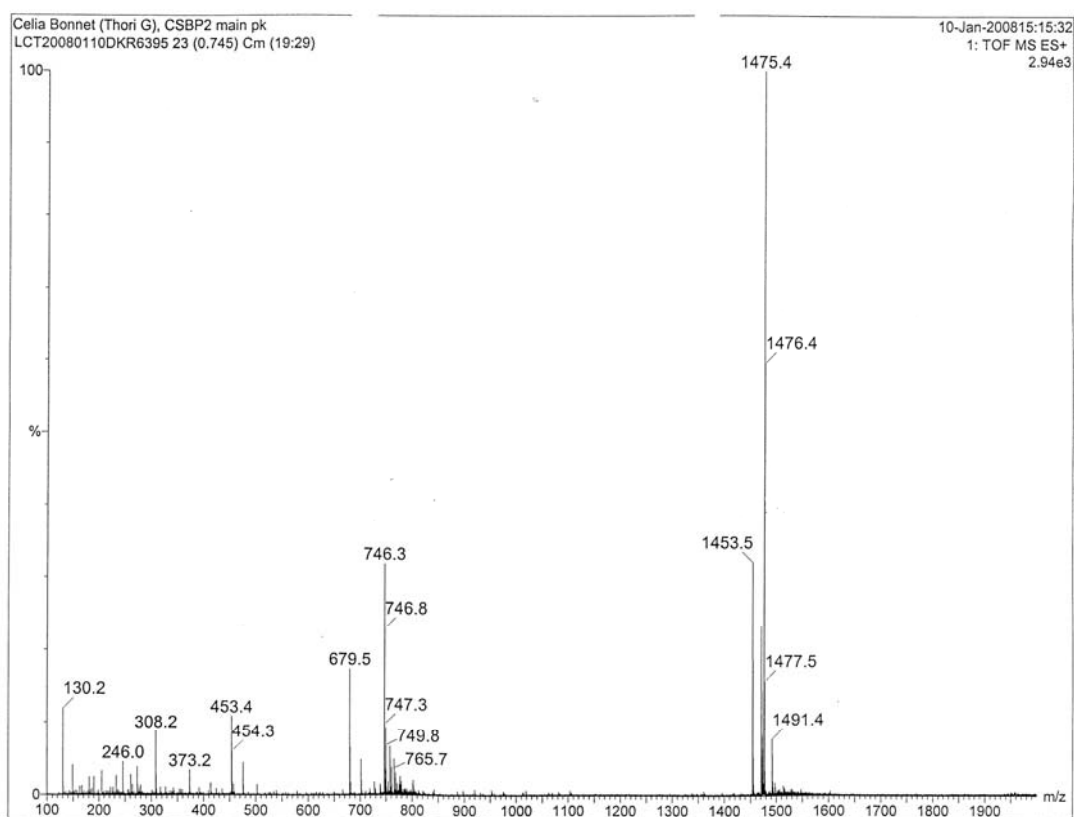


Figure 2. Maldi-Tof spectrum of **2**

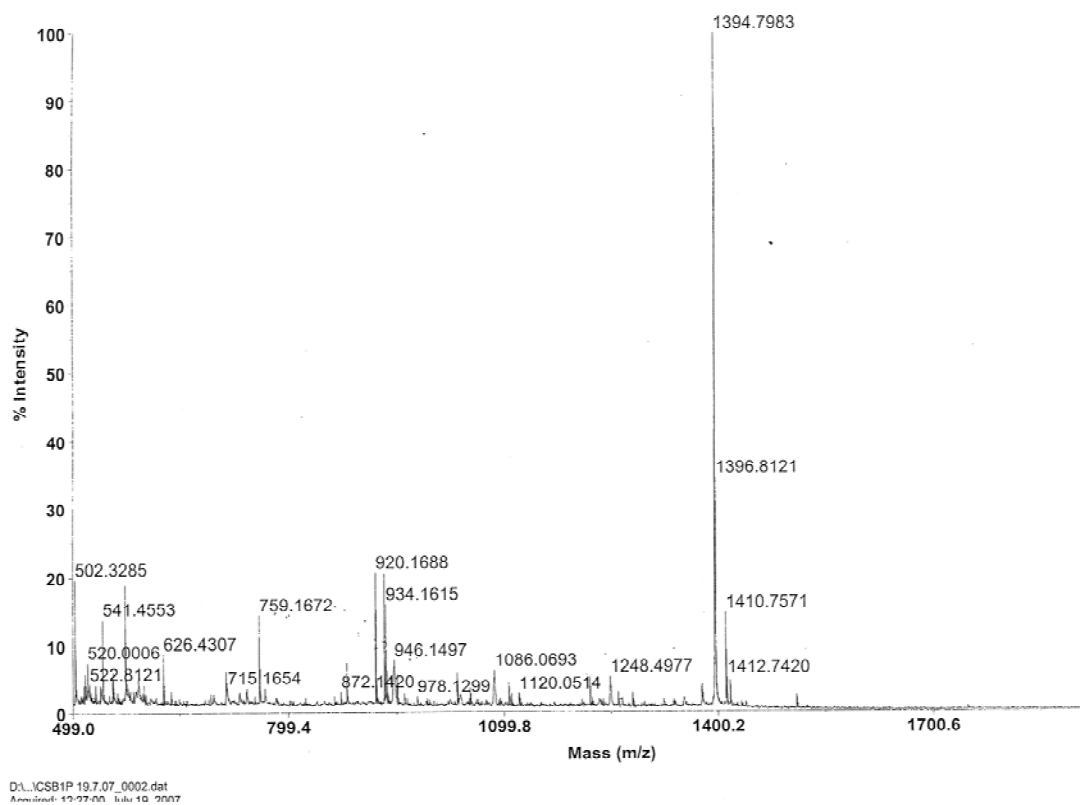


Figure 3. ^1H NMR spectrum (600 MHz) of **1** in a mixture 9:1 $\text{H}_2\text{O}:\text{D}_2\text{O}$

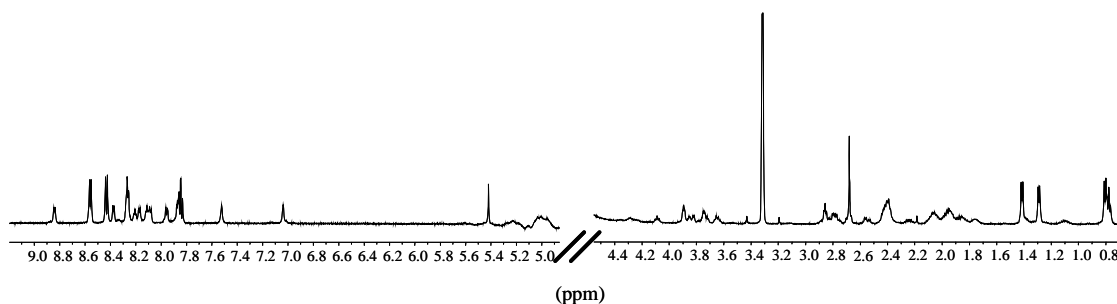


Figure 4. Speciation distribution obtained at pH 7, with $[\mathbf{2}] = 35 \mu\text{M}$ and with the following conditional stability constant : $\log \beta_{11} = 6.8$, $\log \beta_{21} = 11.9$. It should be noted that when $[\text{M}] = 0.8 \cdot [\mathbf{2}]$ (conditions used for the measurements of q-value), no free metal is present and less than 5% of binuclear complex is in solution.

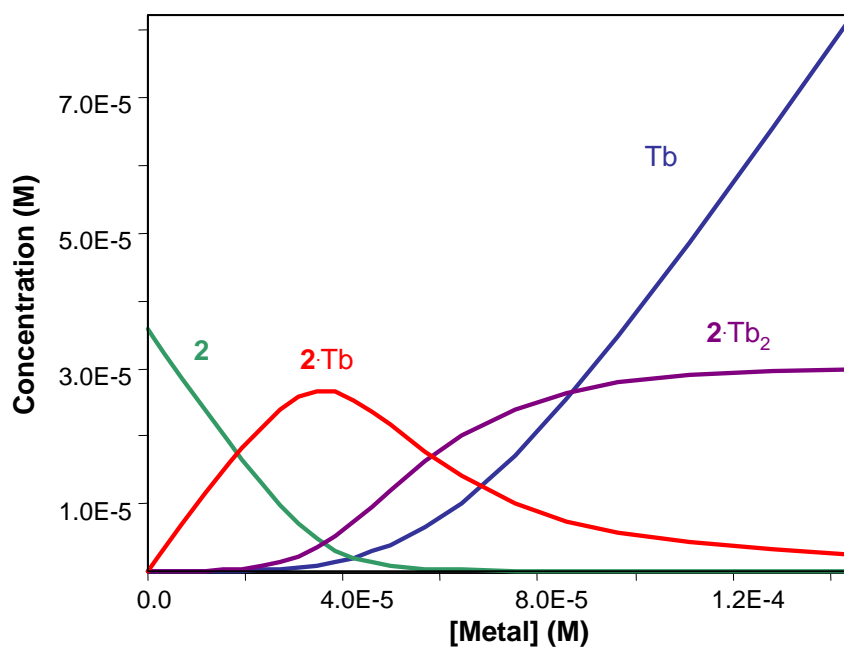


Figure 5. Luminescent decay rates of a Tb^{3+} bound to **2** with increasing H_2O concentration. $[\text{Tb}^{3+}] = 0.8*[\text{P}] = 13.5 \text{ mM}$, $\text{pH} = 7.0$, $[\text{HEPES}] = 10 \text{ mM}$, $[\text{NaCl}] = 0.1 \text{ M}$. The excitation wavelength was 280 nm, and Tb^{3+} sensitized emission was recorded at 545 nm. Same results are obtained by direct excitation on the metal at 368 nm. Extrapolation to an H_2O -free solution provides the rate constant in D_2O .

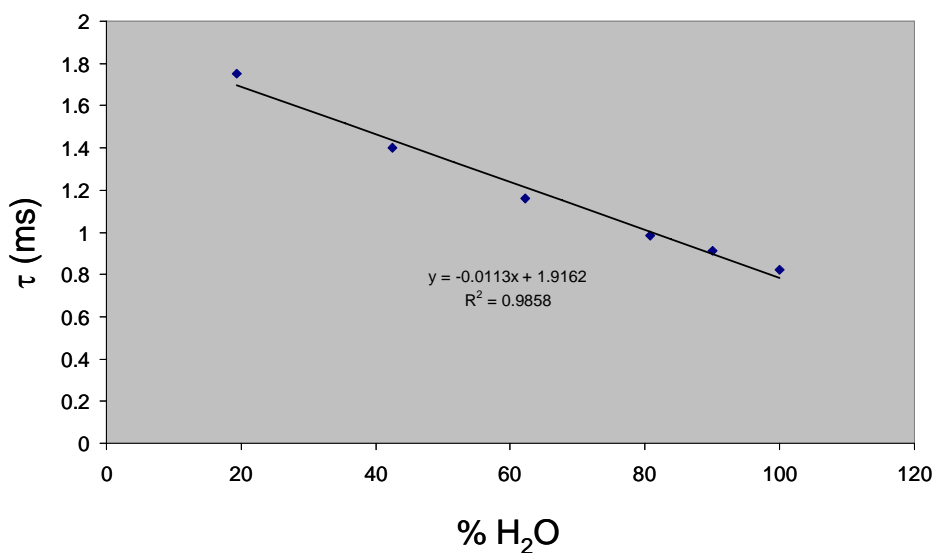


Figure 6. Luminescent decay rates of a Eu^{3+} bound to **2** with increasing H_2O concentration. $[\text{Eu}^{3+}] = 0.8*[\text{P}] = 15.5 \text{ mM}$, $\text{pH} = 7.0$, $[\text{HEPES}] = 10 \text{ mM}$, $[\text{NaCl}] = 0.1 \text{ M}$. The excitation wavelength was 344 nm, and Eu^{3+} sensitized emission was recorded at 616 nm. Extrapolation to an H_2O -free solution provides the rate constant in D_2O .

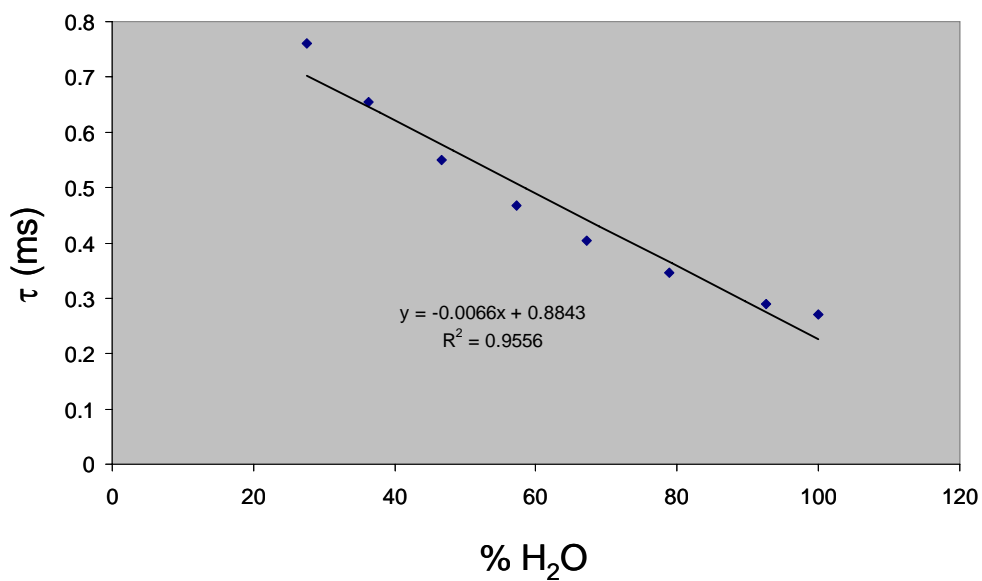


Figure 7. ^1H NMR spectra (600 MHz) of **2**, **2** basified by NaOH and the corresponding Eu(III) complexes in a mixture 9:1 $\text{H}_2\text{O}:\text{D}_2\text{O}$

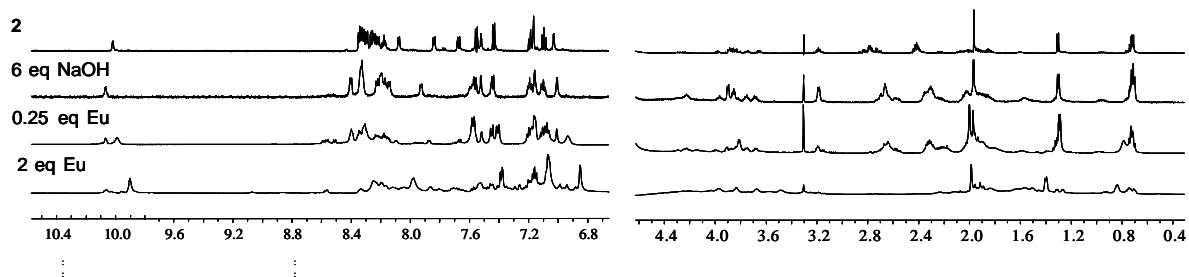


Figure 8. CD-Spectra of **2** (shown as P1) and the changes observed upon addition of one and two equivalents of Tb(III).

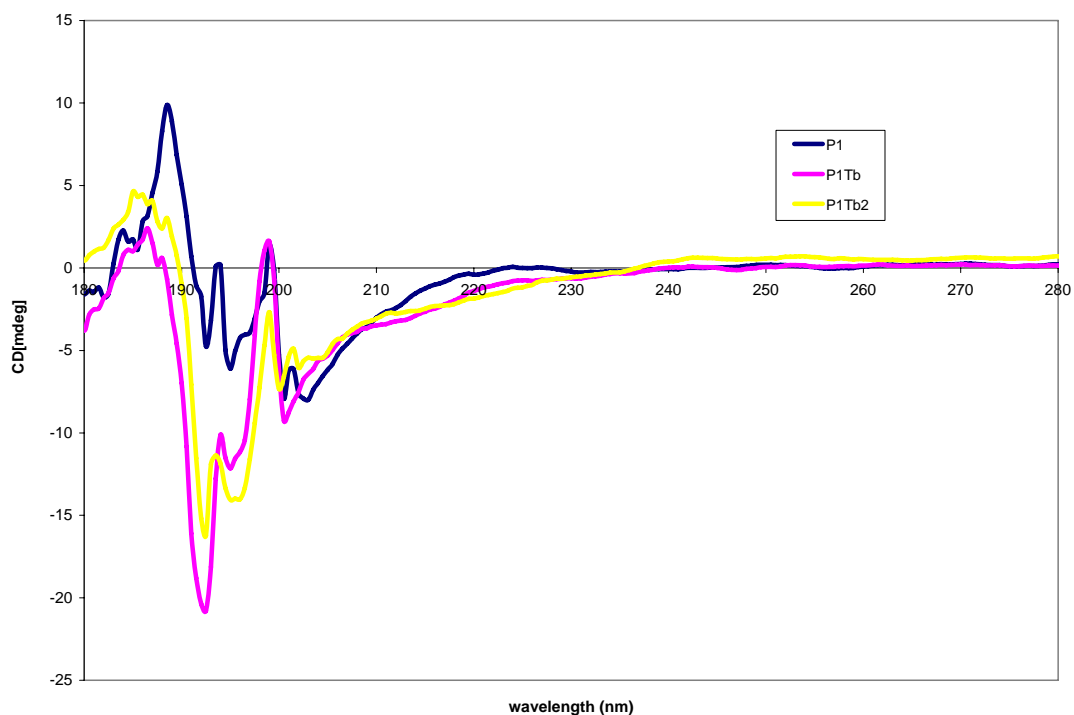


Figure 9. Schematic representation of the mono and di-nuclear complexes of **2**.

