

Supporting Information

Evidence of Tetraphenylporphyrin Monoacids by Ion-Transfer Voltammetry at Polarized Liquid|Liquid Interfaces

*Bin Su,^a Fei Li,^a Raheleh P. Nia,^a Claude Gros,^b Jean-Michel Barbe,^b Zdenek Samec^c and Hubert H. Girault^{*a}*

^a Laboratoire d'Electrochimie Physique et Analytique, Ecole Polytechnique Fédérale de Lausanne, Station 6, CH-1015 Lausanne, Switzerland, Fax: +41 21 6933667; Tel: +41 21 6933145; E-mail: hubert.girault@epfl.ch

^b Institut de Chimie Moléculaire de l'Université de Bourgogne, ICMUB-UMR 5260, 21078 Dijon Cedex, France

^c J. Heyrovsky Institute of Physical Chemistry of ASCR, v.v.i, Dolejskova 3, 182 23 Prague 8, Czech Republic

S1. Cyclic voltammetry of 50 μM H_2TPP at various scan rates

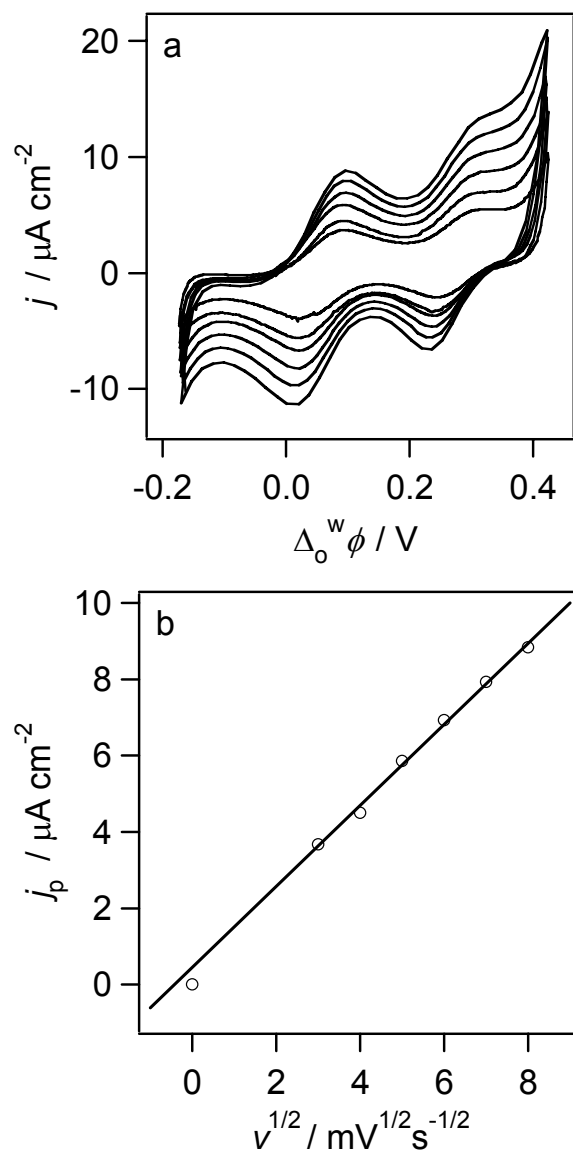


Figure S1. (a) CVs at various scan rates (9, 16, 25, 49 and 64 mV/s from inner to outer) for the interface between 5 mM BTPPATPFB + 50 μM H_2TPP in DCE and 10 mM LiCl + 100 mM HCl in water; (b) The first anodic peak current as a function of the square root of the scan rate.

S2. Cyclic voltammetry at various concentrations of H₂TPP

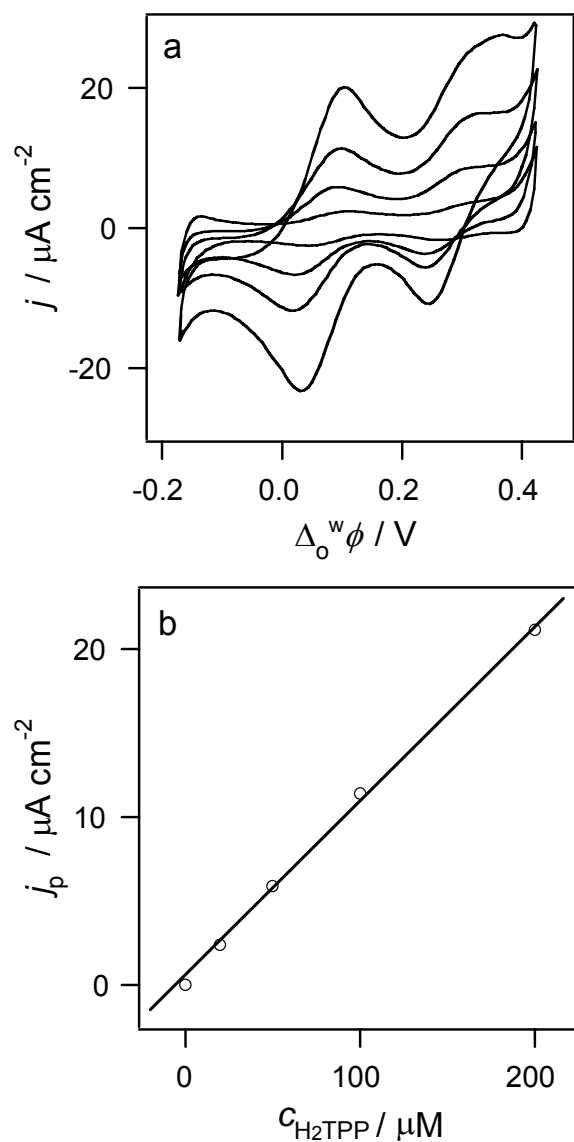


Figure S2. (a) CVs (25 mV/s) at various concentrations of H₂TPP in DCE (20, 50, 100 and 200 μM, from inner to outer): 5 mM BTPPATPFB in DCE and 10 mM LiCl + 100 mM HCl in water; (b) The first anodic peak current as a function of H₂TPP concentration.

S3. Cyclic voltammetry of 50 μM H_2TPP at various pH

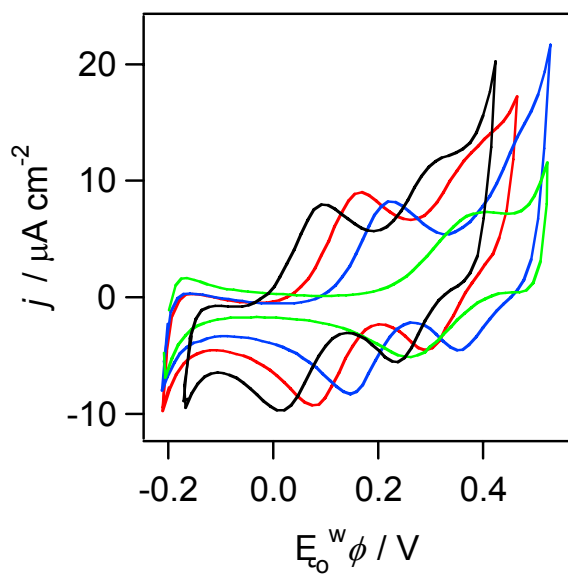


Figure S3. CVs (49 mV/s) at various aqueous pH (1, black; 2, red; 3, blue; 5, green;): 5 mM BTPPATPFB in DCE and 10 mM LiCl in water, and the water phase acidified by HCl.