

Amino-Sugar Modular Ligands – Useful Cores for the Formation of Asymmetric Copper 1,4-Addition Catalysts

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Supplementary Data

General

All experiments were carried out under an argon atmosphere using standard Schlenk techniques. Solvents were dried prior to use: THF and toluene were distilled on Na/benzophenone. ¹H and ¹³C NMR spectra were recorded on JEOL EX270, Bruker 400, Varian-Gemini 300 and Varian-Gemini 200 spectrometers. For all other samples δ values were referenced to residual CHCl₃. All *J* values are in Hz. Specific optical rotatory powers [α] were measured with a ADP 440 Polarimeter in dichloromethane (c= 1.0 g/100 mL). Enantioselectivities were determined by chiral GC. GC analyses were performed on a Varian 3900 gas chromatograph using an octakis (6-*O*-methyl-2,3-di-*O*-pentyl)- γ -CD column under the conditions given. Yields were determined by calibration against authentic samples using undecane or dodecane as an internal standard. Thin layer chromatography (TLC) was performed on Merck silica gel 60 F₂₅₄₊₃₆₆ pre-coated plates (0.25 mm) silica. The plates were visualised by the use of a combination of ultraviolet light (254 and 366 nm) and/or aqueous potassium permanganate with heating. Liquid chromatography was by forced flow (flash chromatography) with the solvent systems indicated using silica gel 60 (220-240 mesh) supplied by Fluka. The diorganozinc samples used were commercial products from the following sources: ZnMe₂ (2.0 M toluene solution; Aldrich) and ZnEt₂ (1.0 M hexane solution; Aldrich). All enones were commercially available (Aldrich). The compound 5-methylhex-3-en-2-one was available from Aldrich as a 75-80% mixture with isomeric 5-methylhex-4-en-2-one. This was removed from commercial product by selective MCPBA epoxidation followed by flash chromatography. All other compounds were used as supplied. Ligands **L_H** and **L_I** are commercially available. The ligands **L_A**¹, **L_C**¹ and **L_J**² and the sugars precursors **L_D**³, **L_E**⁴, have been described in the literature -instead of- The ligands **L_A**¹, **L_D**¹ and **L_J**² and the sugars precursors **L_F**³, **L_G**⁴, have been described in the literature.

Preparation of Ligands

Synthesis of **L_B**

A solution of 2-(diphenylphosphino)naphtoic acid (2.6 mmol), 4-dimethylaminopyridine (0.60 mmol) and 1,3-dicyclohexylcarbodiimide (6.1 mmol) in dry dichloromethane (7 mL) was added to a solution of the appropriate amino sugar **L_D** (2.6 mmol) in the same solvent (7 mL). The resulting mixture was stirred at room temperature under inert atmosphere affording a yellow suspension. After 12 hours 2-

(diphenylphosphino)benzoic acid (2.6 mmol) was added and the suspension was stirred for other 12 hours. The residue was removed by filtration. The resulting yellow solution was evaporated under vacuum and the residue was chromatographed on silica gel (4:9 ethyl acetate:hexane) affording the pure product as a white solid (yield: 60-65%).

^1H NMR (200MHz, CDCl_3): δ 6.42 (d, 1H, NH-C2, $^3J_{\text{NH-H2}} = 9.3\text{Hz}$), 5.66 (t, 1H, H3, $^3J_{\text{H3-H4}} = 20.4\text{Hz}$, $^3J_{\text{H3-H2}} = 10.0\text{Hz}$), 5.35 (m, 2H, H1 e H7, $^3J_{\text{H1-H2}} = 3.9\text{Hz}$), 5.01 (dt, 1H, H2, $^3J_{\text{H2-H3}} = 11.1\text{Hz}$), 4.60 (d, 1H, CHHPh, $^2J_{\text{gem}} = 12\text{Hz}$), 4.37 (d, 1H, CHHPh), 4.17 (dd, 1H, H6_{eq}, $^3J_{\text{H6eq-H5}} = 4.4$, $^3J_{\text{H6eq-H6ax}} = 9.8\text{ Hz}$), 3.90 (dt, 1H, H5, $^3J_{\text{H5-H6ax}} = ^3J_{\text{H5-H4}} = 9.3\text{Hz}$), 3.78 (s, 3H, OCH₃) 3.68 (m, 2H, H4, H6_{ax}).

^{13}C NMR (200MHz, CDCl_3): δ 169.6, 166.8, 160.3, 138.7-126.2, 113.8, 101.4, 98.7, 80.0, 70.7, 69.1, 63.6, 55.7, 53.2.

^{31}P NMR (400MHz, CDCl_3): δ -4.95, -13.38.

$[\alpha]_{\text{D}}^{20} = +54.20$ (c=0.60, CH_2Cl_2).

HRMS (ESI) M+Na calcd: 1014.3265 m/z, found: 1014.3319 m/z.

Synthesis of **L_F**

A solution of 2-(diphenylphosphino)benzoic acid (2.6 mmol), 4-dimethylaminopyridine (0.60 mmol) and 1,3-dicyclohexylcarbodiimide (6.1 mmol) in dry dichloromethane (7 mL) was added to a solution of the appropriate amino sugar **L_D** (2.6 mmol) in the same solvent (7 mL). The resulting mixture was stirred for 12 hours at room temperature under inert atmosphere affording a yellow suspension.

The residue was removed by filtration. The resulting yellow solution was evaporated under vacuum, and the residue was chromatographed on silica gel (4:9 ethyl acetate:hexane) affording the pure product as a white solid (yield: 60-65%).

^1H NMR (400MHz, CDCl_3): δ 6.27 (d, 1H, NH-C2, $^3J_{\text{NH-H2}} = 8.0\text{Hz}$), 5.56 (s, 1H, H7), 4.95 (d, 1H, H1, $^3J_{\text{H1-H2}} = 4.0\text{Hz}$), 4.66 (d, 1H, CHHPh, $^2J_{\text{gem}} = 12\text{Hz}$), 4.45 (d, 1H, CHHPh), 4.39(dt, 1H, H2, $^3J_{\text{H2-H3}} = 12\text{Hz}$), 4.22 (dd, 1H, H6_{eq}, $^3J_{\text{H6eq-H5}} = 4.0$, $^3J_{\text{H6eq-H6ax}} = 12\text{Hz}$), 3.93 (t, 1H, H3, $^3J_{\text{H3-H4}} = 20.4\text{Hz}$, $^3J_{\text{H3-H2}} = 12\text{Hz}$), 3.83 (s, 3H, OCH₃), 3.85-3.75(m, 2H, H5 and H6_{ax}), 3.67 (t, 1H, H4, $^3J_{\text{H4-H5}} = 7.9\text{Hz}$).

^{13}C NMR (400MHz, CDCl_3): δ 169.8, 160.2, 136.8-127.7, 113.6, 101.9, 97.2, 81.6, 70.5, 69.8, 68.8, 63.0, 55.3, 54.9.

^{31}P NMR (400MHz, CDCl_3): δ -11.37.

$[\alpha]_{\text{D}}^{20} = +46.85$ (c=0.815, CH_2Cl_2).

HRMS (ESI) M+H calcd: 676.2459 m/z, found: 676.2449 m/z.

Synthesis of **L_G**

A solution of 2-(diphenylphosphino)benzaldehyde (2.6 mmol) in toluene (5 mL) was added to a solution of the amino sugar **L_D** (2.6 mmol) in the same solvent (5 mL). The resulting mixture was stirred for 2 hours at 80°C affording a yellow solution. The volume of the solvent was reduced under vacuum at ca. 1 mL and hexane (5-6 mL) was slowly added to afford the product as a yellow microcrystalline powder, which was washed with hexane and dried under vacuum (yield: 60%).

^1H NMR (200MHz, C_6D_6): δ 9.06 (d, 1H, N=CH), 5.55 (s, 1H, H7), 4.73 (d, 1H, H1, $^3J_{\text{H1-H2}} = 4.0\text{Hz}$), 4.66 (m, 2H, CHHPh, H3), 4.44 (d, 1H, CHHPh, $^2J_{\text{gem}} = 12\text{Hz}$), 4.35-4.30 (m, 2H, H5 and H6_{eq}), 3.74 (t, 1H, H6_{ax}, $^3J_{\text{H6ax-H5}} = 8.0\text{Hz}$), 3.64 (t, 1H, H4, $^3J_{\text{H4-H5}} = 12\text{Hz}$, $^3J_{\text{H4-H3}} = 12\text{Hz}$), 3.47 (dd, 1H, H2, $^3J_{\text{H2-H3}} = 12\text{Hz}$), 3.38 (s, 3H, OCH₃).

^{13}C NMR (200MHz, C_6D_6): δ 162.5, 138.5-127.5, 113.5, 102.1, 99.7, 82.3, 75.3, 72.9, 69.4, 69.2, 64.3, 62.7, 54.6.

^{31}P NMR (400MHz, CDCl_3): δ -11.62.

$[\alpha]_{\text{D}}^{20} = +37.97$ ($c = 0.52$, CH_2Cl_2).

HRMS (ESI) $\text{M}+\text{H}$ calcd: 660.2510 m/z , found: 660.2495 m/z .

Preparation of racemic ZnEt_2 addition products

A solution of the copper-catalyst precursor CuTC (1mol%, 0.015 mmol) and the corresponding ligand (2 equiv., 0.03 mmol) in 5 mL of dry diethyl ether was stirred for 12 hours at room temperature. The alkylating organometallic reagent ZnEt_2 (1.2 equiv., 3.6 mmol) was added dropwise and then the substrate (3 mmol).

After 12 hours the reaction was quenched with 2M HCl (2 mL). The organic layer was filtered twice through a plug of silica and dried.

Preparation of racemic ZnMe_2 addition products

To a suspension of CuI (5 mmol) in diethyl ether 25 ml at -10°C was added MeLi (2 eq., of a 1.6M solution in hexane, 10 mmol), dropwise. The cloudy yellow mixture was allowed to warm at 0°C over a 30 min period, affording a clear solution. The flask was cooled at -78°C and a solution of enone (3.2 mmol) in diethyl ether was added dropwise. The mixture was kept at 0°C and became again yellow. The reaction was quenched with 2M HCl and the product was extract with diethyl ether.

Table S1 describes the columns and conditions used to determine the ee value from the catalytic reactions.

Table S1. Enantioselectivity determinations of products.

R^1	R^2	R^3	Column ^[a]	Programme	Retention times/mins
C_5H_{11}	CH_3	CH_3	6-Me-2,3-pe- γ -CD	60 $^\circ\text{C}$ isothermal	<i>S</i> 16.2 <i>R</i> 17.5
C_5H_{11}	CH_3	CH_2CH_3	6-Me-2,3-pe- γ -CD	60 $^\circ\text{C}$ isothermal	<i>S</i> 30.0 <i>R</i> 31.1
C_6H_{13}	CH_3	CH_3	6-Me-2,3-pe- γ -CD	60 $^\circ\text{C}$ isothermal	<i>S</i> 33.9 <i>R</i> 35.9
C_6H_{13}	CH_3	CH_2CH_3	6-Me-2,3-pe- γ -CD	60 $^\circ\text{C}$ isothermal	<i>S</i> 64.0 <i>R</i> 65.9
Pr^1	CH_3	CH_3	6-Me-2,3-pe- γ -CD	45 $^\circ\text{C}$ isothermal	<i>S</i> 8.75 <i>R</i> 9.65
Pr^1	CH_3	CH_2CH_3	6-Me-2,3-pe- γ -CD	50 $^\circ\text{C}$ isothermal	<i>S</i> 12.4 <i>R</i> 12.9
CH_3	CH_2CH_3	CH_2CH_3	6-Me-2,3-pe- γ -CD	45 $^\circ\text{C}$ isothermal	<i>S</i> 9.87 <i>R</i> 11.0

[a] 6-Me-2,3-pe- γ -CD is 25 m octakis(6-*O*-methyl-2,3-di-*O*-pentyl)- γ -cyclodextrin 0.25 μm internal diameter (60% in OV1701, w/w).

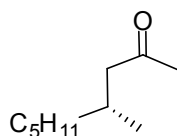
Catalytic Reaction: General procedure for additions to enones

A solution of the copper-catalyst precursor $\text{Cu}(\text{OTf})_2$ (1mol%, 0.003 mmol) and the corresponding ligand (2.5 equiv., 0.0075 mmol) in 2 mL of dry toluene was stirred for 30 minutes at room temperature. The alkylating organometallic reagent ZnR_2 (2 equiv., 0.24 mmol) was added dropwise and then the substrate (0.12 mmol).

After 12 hours the reaction was quenched with 2M HCl (2 mL). Undecane or dodecane (10 μL) was then added as internal standard and the organic layer filtered twice through a plug of silica. Yields and enantiomeric excesses were measured by GC using a octakis (6-*O*-methyl-2,3-di-*O*-pentyl)- γ -CD column.

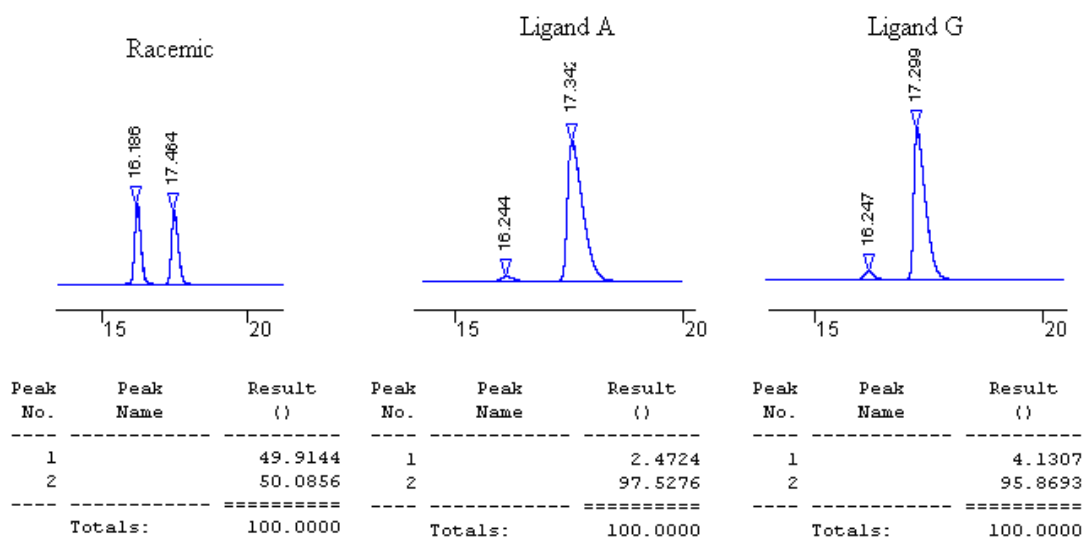
Chiral GC traces for individual compounds

(*R*)-4-methylnonan-2-one

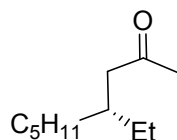


Literature compound,⁵ authenticated by comparison against an existing sample and by ^1H NMR (270MHz, CDCl_3): δ 2.38 (dd, 1H, $\text{CHH}=\text{COMe}$, $J=15.8, 5.7$), 2.22 (dd, 1H, $\text{CHH}=\text{COMe}$), 2.11 (s, 3H, MeCO), 2.0-1.9 (m, 1H, nPentCHMe), 1.4-1.1 (m, 8H, $(\text{CH}_2)_4$), 0.9-0.88 (m, 6H, $(\text{CH}_2)_4\text{Me}$, MeCH); ^{13}C NMR (400MHz, CDCl_3): δ 209.3 ($\text{C}=\text{O}$), 51.3, 36.9, 32.0, 30.4, 39.3, 26.6, 22.6, 19.8, 14.1.

Comparative GC chromatograms of racemic, 95% ee (L_A) and 92% ee (L_G) materials below.

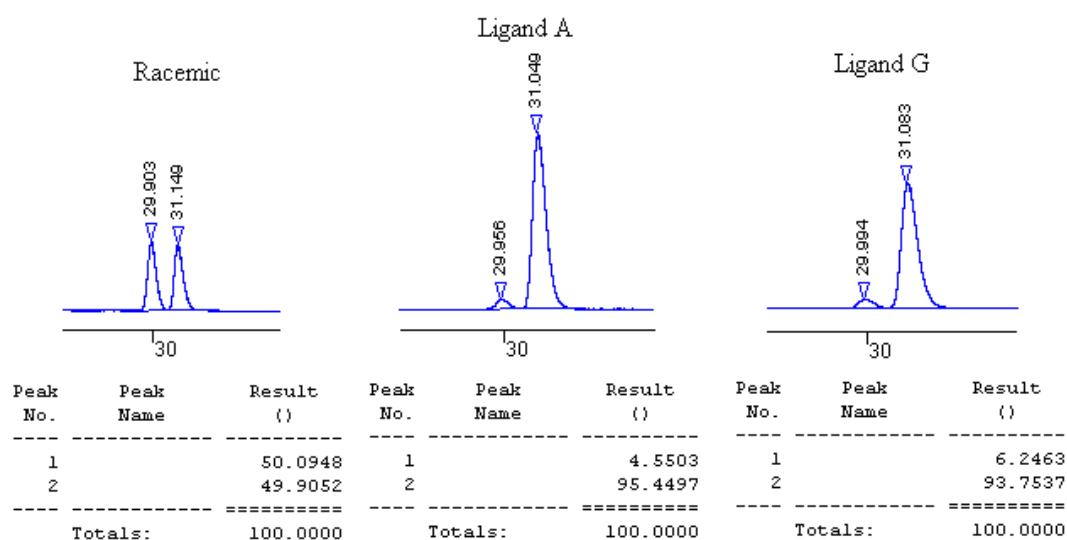


(*R*)-4-ethylnonan-2-one

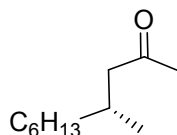


Literature compound,⁶ authenticated by comparison against an existing sample and by ¹H NMR (270MHz, CDCl₃): δ 2.33 (m, 2H, CH₂=COMe, J=7.0Hz), 2.11 (s, 3H, MeCO), 1.83 (m, 1H, nPentCH₂Et), 1.31-1.16 (m, 10H, (CH₂)₄, CH₂ of Et), 0.88-0.80 (m, 6H, (CH₂)₄Me, CH₃ of Et); ¹³C NMR (400MHz, CDCl₃): δ 209.3 (C=O), 48.4, 35.4, 33.5, 32.1, 30.3, 26.4, 26.4, 22.6, 14.0

Comparative GC chromatograms of racemic, 91% ee (**L_A**) and 87% ee (**L_G**) materials below.

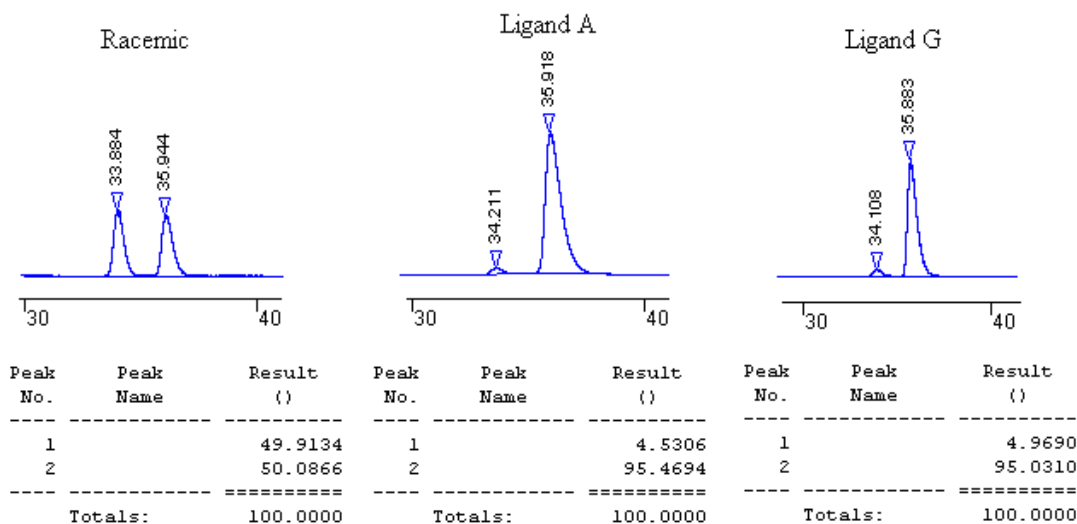


(R)-4-methyldecan-2-one

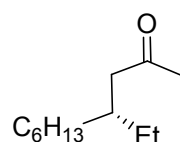


Literature compound,^{5b,6} authenticated by comparison against an existing sample and by ¹H NMR (270MHz, CDCl₃): δ 2.40 (dd, 1H, CHH=COMe, J=16, 5.7), 2.22 (dd, 1H, CHH=COMe), 2.13 (s, 3H; MeCO), 1.89 (m, 1H; nHexCHMe), 1.1-1.35 (m, 10H, (CH₂)₅), overlapped by 0.89 (m, 6H, MeCH, (CH₂)₅Me); ¹³C NMR (400MHz, CDCl₃): δ 209.6 (C=O), 51.7, 37.3, 32.3, 30.8, 29.8, 29.7, 27.3, 23.0, 20.2, 14.5.

Comparative GC chromatograms of racemic, 91% ee (**L_A**) and 90% ee (**L_G**) materials below.

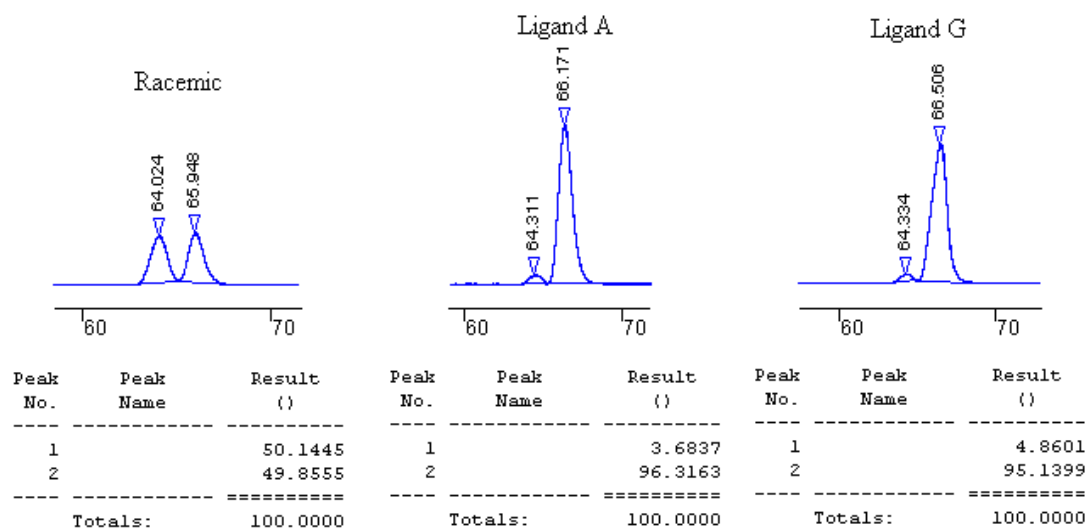


(R)-4-ethyldecan-2-one

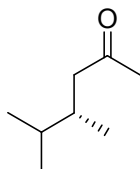


Literature compound,⁷ authenticated by comparison against an existing sample and by ¹H NMR (270MHz, CDCl₃): δ 2.33 (m, 2H, CH₂=COMe, J=7.0Hz), 2.14 (s, 3H, MeCO), 1.83 (m, 1H, nHexCH₂Et), 1.31-1.16 (m, 12H, (CH₂)₅, CH₂ of Et), 0.88-0.80 (m, 6H, (CH₂)₄Me, CH₃ of Et); ¹³C NMR (400MHz; CDCl₃): δ 209.7 (C=O), 48.4, 35.4, 33.5, 32.0, 30.4, 29.6, 26.6, 26.4, 22.7, 14.0, 10.83.

Comparative GC chromatograms of racemic, 93% ee (**L_A**) and 90% ee (**L_G**) materials below.

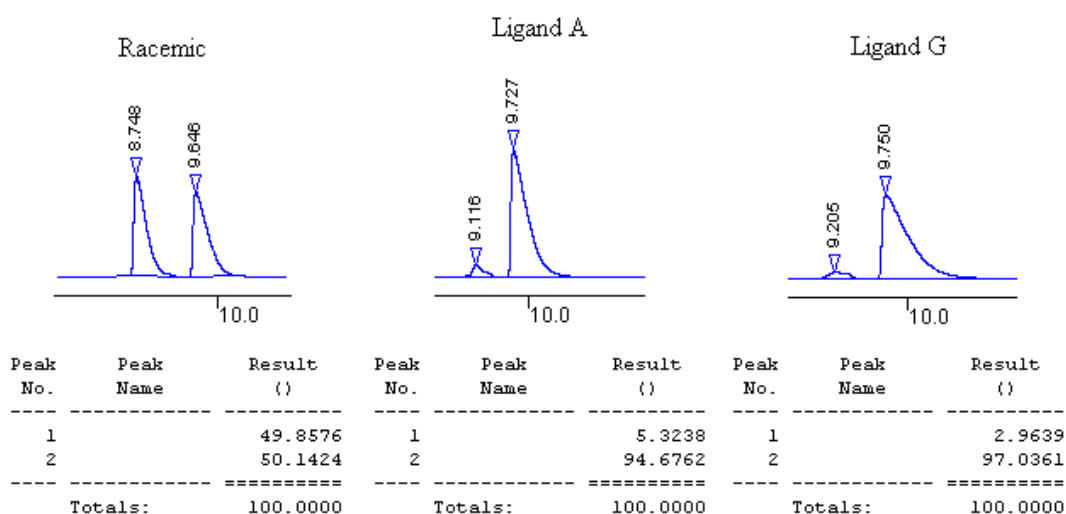


(R)-4-methyl-5-methylhexan-2-one

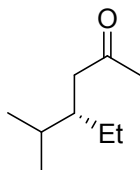


Literature compound,⁸ authenticated by comparison against an existing sample and by ¹H NMR (270MHz, CDCl₃): δ 2.20 (dd, 1H, CHH=COMe, J=16, 5.4), 2.18 (dd, 1H, CHH=COMe), 2.15 (s, 3H; MeCO), 1.90 (m, 1H; CH(Me)₂), overlapped by 0.86 (m, 6H, Me₂CH, Me), ¹³C NMR (400MHz, CDCl₃) δ= 209.5 (C=O), 48.5, 34.75, 32.2, 30.4, 19.8, 18.4, 16.0.

Comparative GC chromatograms of racemic, 89% ee (**L_A**) and 94% ee (**L_G**) materials below.

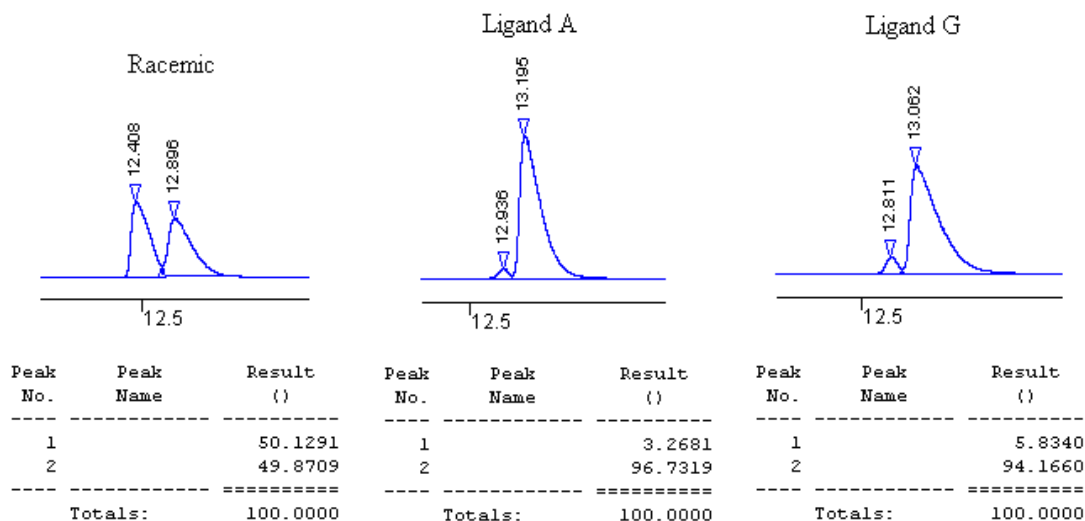


(R)-4-ethyl-5-methylhexan-2-one

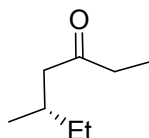


Literature compound,⁹ authenticated by comparison against an existing sample and by ¹H NMR (270 MHz, CDCl₃): δ 2.40 (dd, 1H, CHH=COMe, J= 16.0Hz, 5.6Hz), 2.25 (dd, 1H, CHH=COMe, J=16Hz, J=7.6Hz), 2.15 (s, 3H; MeCO), 1.80-1.65 (m, 2H, CHMe₂ and CHEt), 1.38-1.30 (m, 1H, CHH of Et), 1.16-1.24 (m, 1H, CHH of Et), 0.90 (t, 3H, CH₃ of Et, J= 7.4Hz), 0.86 (t, 3H, MeCHMe, J= 7.4Hz), 0.82 (t, 3H, MeCHMe, J= 7.0Hz); ¹³C (270MHz; CDCl₃) 209.6 (CO), 45.1, 41.1, 30.3, 29.3, 23.9, 19.5, 18.5, 11.7.

Comparative GC chromatograms of racemic, 93% ee (**L_A**) and 88% ee (**L_G**) materials below.

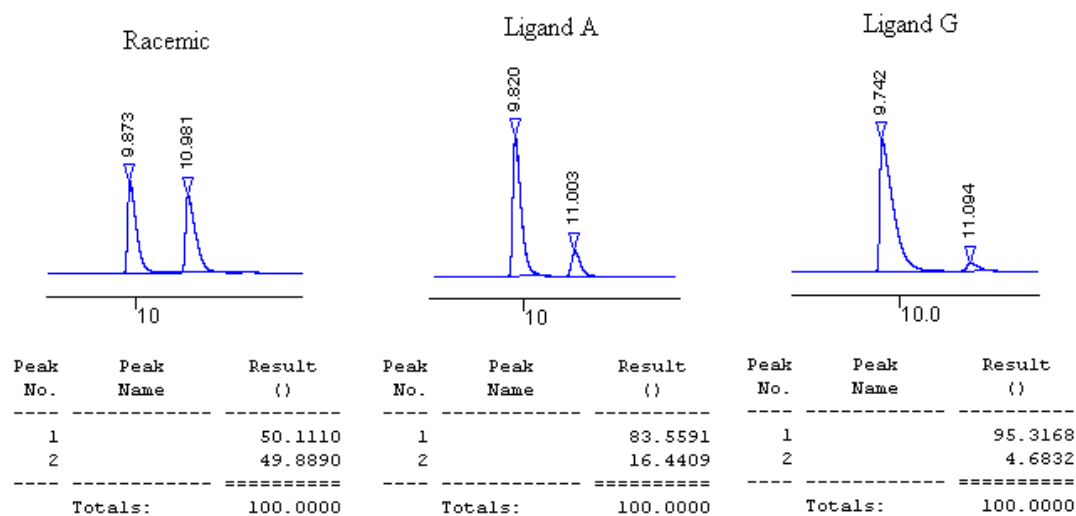


(S)-5-methyl-heptan-3-one

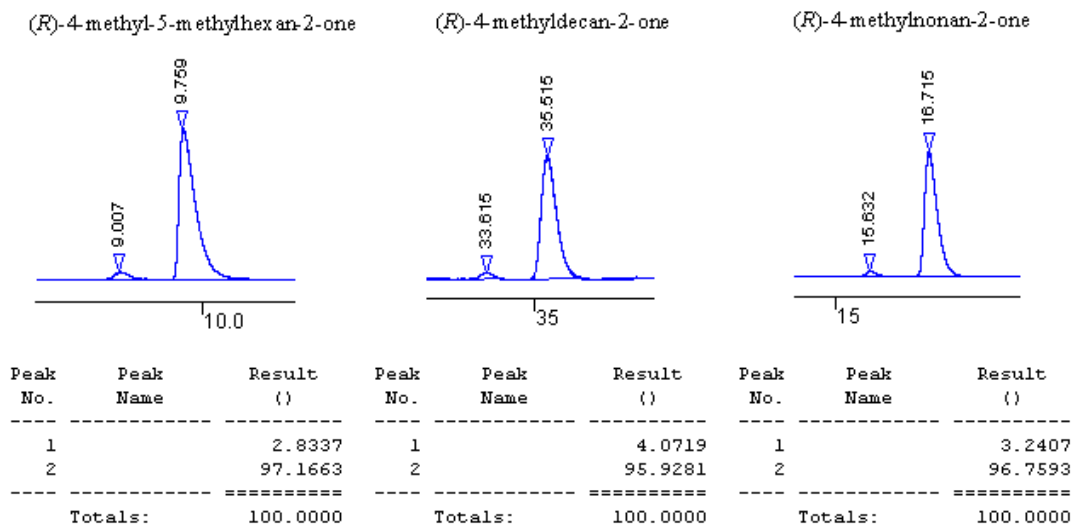


Literature compound,¹⁰ authenticated by comparison against an existing sample and ¹H NMR (270MHz, CDCl₃): δ 2.43-2.33(m, 1H, CH₃CH₂COCHH), 2.18 (dd, 1H, COCHHCH J=16Hz, J=7.6Hz), 1.80-1.65 (m, 1H, MeCH₂Et), 1.29-1.16 (m, 4H, CHCH₂CH₃, COCH₂Me), 1.02 (m, 3H, COCH₂CH₃), 0.88 (m, 6H, CHMe, CHCH₂CH₃); ¹³C (400MHz; CDCl₃) 211.9 (CO), 49.6, 36.5, 31.0, 29.7, 19.5, 11.3, 7.8.

Comparative GC chromatograms of racemic, 67% ee (**L_A**) and 91% ee (**L_G**) materials below. The reversal of the stereochemistry is due to the use of ZnEt₂ (vs ZnMe₂) and the found selectivity was confirmed by polarimetry vs. the literature.



ZnMe₂ 1,4 addition to enones with ligand L₁



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