## Supporting Information

# A Novel Ylide-stabilized Carbene; Formation and Electron Donating Ability of Amino(sulfur-ylide)carbene

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#### **Experimental Section**

**General Remarks.** General chemicals were used as received. All manipulations were carried out under an argon atmosphere unless otherwise noted. Solvents were purified by MBRAUN MB-SPS system. Preparative thin-layer chromatography (PTLC) was performed using Merck silica gel 60 PF<sub>254</sub>. Gel permeation liquid chromatography (GPLC) was performed using LC-918 or LC-908 with JAIGEL 1H+2H columns (Japan Analytical Industry) using chloroform or toluene as solvents, respectively. NMR spectra were recorded by a JEOL AL-400 spectrometer (<sup>1</sup>H, 400 MHz; <sup>13</sup>C, 100 MHz). Chemical shifts are reported in  $\delta$ . <sup>1</sup>H NMR spectra are referenced to residual protons in deuterated solvent; <sup>13</sup>C NMR spectra are referenced to carbon-13 in the deuterated solvent. Low resolution mass spectra were obtained with a JEOL JMS-700P using *m*-nitrobenzyl alcohol as a matrix. High resolution mass spectra were recorded by a JEOL JMS-700P using PEG600 or Ultramark® as internal standards. Infrared spectra were recorded with a JASCO Irtron IRT-30. All melting points were measured with a Yanaco MP-S3 and uncorrected. Elemental analyses were performed by the Microanalytical Laboratory of Department of Chemistry, Faculty of Science, The University of Tokyo.

*N*-Mesitylpyrrole was synthesized according to the literature.<sup>1</sup>

<sup>(1)</sup> Y. Fang, D. Leysen and H. C. J. Ottenheijm, Synth. Commun., 1995, 25, 1857.

## Synthesis of sulfonium salt 2.

To a solution of ditolyl sulfoxide (1.24 g, 5.4 mmol) and *N*-mesitylpyrrole (1.00 g, 5.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 cm<sup>3</sup>) was added a solution of trifluoroactic anhydride (0.76 cm<sup>3</sup>, 5.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5.0 cm<sup>3</sup>) at 0 °C over 30 min. The reaction mixture was gradually warmed up to room temperature and then treated with saturated aq. LiClO<sub>4</sub> (20 cm<sup>3</sup>). The aqueous layer was extracted by CHCl<sub>3</sub> and organic layer was combined and dried over anhydrous CaCl<sub>2</sub>. After the removal of solvent under reduced pressure, the resulting material was subjected to GPLC to give sulfonium perchlorate 1 (602 mg, 22%). A solution of sulfonium salt 1 (250 mg, 0.50 mmol) and NaBPh<sub>4</sub> (206 mg, 0.60 mmol) in MeOH (30 cm<sup>3</sup>) was stirred for 30 min at room temperature. The resulting white solid was collected by suction filtration and recrystallized by THF/methanol to give sulfonium salt 2 as a white solid (300 mg, 83%). mp 135-136 °C;  $\delta$  (400 MHz; CDCl<sub>3</sub>) 1.92 (6H, s, 2,6-CH<sub>3</sub>(Mes)), 2.36 (3H, s, 4-CH<sub>3</sub>(Mes)), 2.41 (6H, s, 4-CH<sub>3</sub>(Tol)), 6.14-6.16 (1H, m, CH(pyrrole)), 6.21-6.22 (1H, m, CH(pyrrole)), 6.69-6.71 (1H, m, *CH*(pyrrole)), 6.78 (4H, t, *J* 7.0, 4-*H*(BPh)), 6.88 (8H, t, *J* 7.4, 3,5-*H*(BPh)), 6.98 (6H, m, 3,5-*H*(Mes), 3,5-*H*(Tol)), 7.27 (4H, m, 2,6-*H*(Tol)), 7.39 (8H, br s, 2,6-*H*(BPh));  $\delta_{\rm C}(100 \text{ MHz}; \text{CDCl}_3)$  17.8 (g, 2,6-CH<sub>3</sub>(Mes)), 21.6 (q, 4-CH<sub>3</sub>(Mes)), 22.0 (q, 4-CH<sub>3</sub>(Tol)), 100.5 (s, 3-C(pyrrole)), 110.3 (d, C(pyrrole)), 122.2 (d, 4-C(BPh)), 124.5 (s, 1-C(Tol)), 125.9 (dq, J 5.8, 3,5-C(BPh)), 127.5 (d, C(pyrrole)), 129.5 (d, 3,5-C(Mes)), 129.7 (d, 2,6-C(Tol)), 130.6 (d, C(pyrrole)), 132.1 (d, 3,5-C(Tol)), 135.0 (s, 2,6-C(Mes)), 135.3 (s, 1-C(Mes)), 136.8 (d, 2,6-C(BPh)), 140.0 (s, 4-C(Mes)), 145.3 (s, 4-C(Tol)), 164.4 (g, J<sub>BC</sub> 98.1, 1-C(BPh)); m/z (FAB<sup>+</sup>) 398.1959 (M<sup>+</sup>-BPh<sub>4</sub>. C<sub>27</sub>H<sub>28</sub>NS requires 398.1942); Anal. Found: C, 83.44; H, 6.76; N, 1.96%. C<sub>51</sub>H<sub>48</sub>BNS·H<sub>2</sub>O requires C, 83.25; H, 6.85; N, 1.90%.

#### Trapping ASYC by elemental sulfur and methylation of trapping product with MeI.

Sulfonium salt 2 (100 mg, 0.139 mmol) was dissolved in THF (10 cm<sup>3</sup>). The solution was cooled to -40 °C and a solution of MesLi (26 mg, 0.21 mmol) in Et<sub>2</sub>O (ca. 5 cm<sup>3</sup>) was added to the cooled solution. The reaction mixture was stirred for 1 h at -40 °C and a THF solution (ca. 5 cm<sup>3</sup>) of elemental sulfur (18 mg, 0.070 mmol) was added to the reaction mixture. The temperature was gradually warmed up to room temperature. This reaction mixture was monitored by low-resolution mass spectroscopy showing m/z 428 (corresponding to thioamide 3). The reaction mixture was treated with excess amount of MeI. The volatile material was removed under reduced pressure and the resulting yellow solid was purified by PTLC (solvent: CHCl<sub>3</sub>/EtOH = 8:1) to give thioether **4** (95 mg, 90%). mp 124-125 °C;  $\delta$  (400 MHz; CDCl<sub>3</sub>) 1.85 (3H, s, SCH<sub>3</sub>), 1.92 (6H, s, 2,6-CH<sub>3</sub>(Mes)), 2.37 (3H, s, 4-CH<sub>3</sub>(Mes)), 2.41 (6H, s, 4-CH<sub>3</sub>(Tol)), 6.05 (1H, d, J 3.2, CH(pyrrole)), 6.72 (1H, d, J 3.6, CH(pyrrole)), 6.81 (4H, t, J 7.0, 4-*H*(BPh)), 6.94 (8H, t, *J* 7.2, 3,5-*H*(BPh)), 7.00 (2H, s, 3,5-*H*(Mes)), 7.13 (4H, m, 3,5-*H*(Tol)), 7.28 (4H, m, 2,6-H(Tol)), 7.39 (8H, br s, 2,6-H(BPh));  $\delta_{C}(100 \text{ MHz}; \text{ CDCl}_{3})$  17.8 (q, 2,6-CH<sub>3</sub>(Mes)), 19.4 (q, SCH<sub>3</sub>), 21.4 (q, 4-CH<sub>3</sub>(Mes)), 21.8 (q, 4-CH<sub>3</sub>(Tol)), 108.4 (s, 3-C(pyrrole)), 111.2 (d, C(pyrrole)), 121.7 (d, 4-C(BPh)), 123.0 (s, 1-C(Tol)), 125.6 (dq, J<sub>BC</sub> 5.4, 3,5-C(BPh)), 129.2 (d, 3,5-C(Mes)), 129.5 (d, 2,6-C(Tol)), 131.4 (d, C(pyrrole)), 131.7 (s, C(pyrrole)), 132.3 (d, 3,5-C(Tol)), 133.7 (s, 2,6-C(Mes)), 135.1 (s, 1-C(Mes)), 136.4 (d, 2,6-C(BPh)), 140.2 (s, 4-C(Mes)), 145.8 (s, 4-C(Tol)), 164.3 (g,  $J_{BC}$  98.5,

### Attempted direct observation of ASYC

In a 5 mm diameter NMR tube, to the mixture of sulfonium salt **2** (@@ mg, @@ mmol) and MesLi (@@mg, @@ mmol) was added THF- $d_8$  (@@ mmol) in liquid N<sub>2</sub> bath. The NMR tube was flame-sealed and the reaction mixture was warmed to -40 °C. The reaction mixture was observed by <sup>13</sup>C NMR keeping the temperature as -40 °C.

#### Synthesis of ASYC-Rh complex 6

To a solution of sulfonium salt **2** (100 mg, 0.139 mmol) in THF (10 cm<sup>3</sup>) was added a solution of MesLi (26 mg, 0.21 mmol) in Et<sub>2</sub>O (ca. 5 cm<sup>3</sup>) at -40 °C. After 1 h stirring at -40 °C, a solution of [Rh(cod)Cl]<sub>2</sub> (41 mg, 0.084 mmol) in THF was added to the mixture at -40 °C. The reaction mixture was gradually warmed up to room temperature for 1.5 h. After the removal of solvents under reduced pressure, the resulting solid was subjected to GPLC to give [(ASYC)Rh(cod)Cl] 5 as a yellow solid (182 mg, 58%). To a solution of **5** (50 mg, 0.078 mmol) in THF (3.0 cm<sup>3</sup>), CO gas was bubbled via gas inlet tube for 1 min. The volatile material was removed under reduced pressure and the resulting pale yellow solid was extracted by hexane. After the removal of solvent, pure  $[(ASYC)Rh(CO)_2Cl]$  6 was obtained as a yellow solid (32.2 mg, 70%). mp 101-102 °C (dec.); δ (400 MHz; CDCl<sub>3</sub>) 2.11 (6H, s, 2,6-CH<sub>3</sub>(Mes)), 2.34 (3H, s, 4-CH<sub>3</sub>(Mes)), 2.42 (6H, s, 4-CH<sub>3</sub>(Tol)), 6.25 (1H, d, J 3.0, CH(pyrrole)), 6.85 (1H, d, J 3.0, *CH*(pyrrole)), 6.93 (2H, s, 3,5-*H*(Mes)), 7.33 (4H, m, 3,5-*H*(Tol)), 7.51 (4H, m, 2,6-*H*(Tol));  $\delta_{C}(126)$ MHz; CDCl<sub>3</sub>) 18.5 (q, 2,6-CH<sub>3</sub>(Mes)), 21.2 (q, 4-CH<sub>3</sub>(Mes)), 21.5 (q, 4-CH<sub>3</sub>(Tol)), 104.7 (s, 3-C(pyrrole)), 108.7 (d, C(pyrrole)), 128.4 (s, 1-C(Tol)), 128.5 (d, 3,5-C(Mes)), 128.8 (d, C(pyrrole)), 129.3 (d, 2,6-C(Tol)), 130.8 (d, 3,5-C(Tol)), 135.6 (s, 2,6-C(Mes)), 137.5 (s, 1-C(Mes)), 140.1 (s, 4-C(Mes)), 142.7 (s, 4-C(Tol)), 172.5 (d, J<sub>CRh</sub> 37.2, 2-C(pyrrole)), 183.8 (d, J<sub>CRh</sub> 78.4, cis-CO), 186.4 (d,  $J_{CRh}$  51.1, trans-CO); m/z (FAB<sup>+</sup>) 556.0815 (M<sup>+</sup>-Cl. C<sub>29</sub>H<sub>27</sub>NO<sub>2</sub>SRh requires 556.0818), 535 (100,  $M^+$ -(CO)<sub>2</sub>), 500 (44,  $M^+$ -(CO)<sub>2</sub>Cl);  $\nu_{max}/cm^{-1}$  2057 (CO) and 1981 (CO).

#### X-ray crystallographic analyses.

Single crystals of **2** and **6** were grown from a CHCl<sub>3</sub> solution for **2** and Et<sub>2</sub>O/THF solution for **6**. The intensity data were collected on Rigaku/MSC Mercury CCD with Mo/K $\alpha$  radiation ( $\lambda = 0.71069$  Å). The structures were solved by the direct method and refined by full-matrix least squares on F<sup>2</sup> using SHELXS97. The structures were refined anisotropically except for hydrogen atoms. Hydrogen atoms were idealized by using the riding models.

#### NMR spectra of new compounds.

<sup>1</sup>H NMR spectrum of sulfonium salt **2** 

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<sup>13</sup>C NMR spectrum of sulfonium salt **2** 



77.795 77.482 77.160 22.037 -21.592 -17.772 -

 $\rm YYu1al400YdataYYatabeYcarben authentic$ Ycarbene precursor BPh4¥Solfonium salt 2–13C,als 2008 4 30 AY exp103–dry–C

<sup>1</sup>H NMR spectrum of thioether **4** 

165, 157 164, 870 164, 889 183, 889 183, 589 184, 584 184, 584 184, 587 184, 587 184, 587 184, 587 183, 587 184, 587184, 587 184

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## <sup>13</sup>C NMR spectrum of thioether 4

YYulal400YdataYYatabeYcarben authenticYtrapped with sulfurYTrapping of ASYC by elemental sulfur and Mel-13C.als Trapping of ASYC by elemental sulfur and Mel-13C.a. DFILE COMNT DATIM OBNUC EXMOD OBFRQ OBSET OBFIN POINT ¥¥u1al400¥data¥Yatabe¥carben Trapping of ASYC by elemental Thu Jul 17 09:52:01 2008 13C YIIW Y Y BCM ICM 100.40 MHz 125.00 KHz 10500.00 Hz 32768 27173.90 Hz 11029 1.2059 sec 1.7940 sec 6.00 usec H POINT FREQU SCANS ACQTM PD PW1 IRNUC CTEMP SLVNT EXREF BF RGAIN 1H1H 23.5 c CDCL3 77.16 ppm 0.10 Hz 24 120 145 140 125 faipeilean fi shiirbaa idi looka Marisia salla talkaha shiirbaharada arabiya baraharat iyoo hareen arkeeda 23 22 21 20 19 18 17 di di 165 164 PPM 0 50 200 150 \* 100 44 Ż 165.055 1645.083 1645.083 1645.083 1645.083 1645.073 1145.765 1145.765 1145.765 1145.765 1145.765 1145.765 1145.765 1145.765 1145.765 1135.665 1135.765 1135.765 1135.765 1135.765 1135.765 1135.755 1135 21.83121.37719.40517.822

<sup>1</sup>H NMR spectrum of Rh complex **6** 

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# <sup>13</sup>C NMR spectrum of Rh complex 6

