

Phosphine-Catalyzed Disulfide Metathesis

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Table of contents

General Methods	S2
Analytical data for compound for PCy₃	S2
Preparation of OPCy₃	S2
Analytical data for compounds 1, 2a, 2b, 2c	S2
Analytical data for compounds 2d, 2e 3a, 3b, 3c, 3d, 3e, 4a, 4b	S3
Analytical data for compounds 4c, 5a, 5b, 6a	S4
Reversibility control experiment	S5
GC-MS analysis of reaction mixture (1, 2a, 3a)	S6
GC chromatogram of reaction mixture (1, 2a, 3a)	S6
MS spectrum of 1 and 3a	S7
MS spectrum of 2a	S8

General Methods

Reagents were purchased from Sigma-Aldrich, Merck and Lancaster and used as received. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectra were recorded on a Bruker Avance 400 spectrometer at 400 (100) MHz and/or Bruker Avance DMX 500 at 500 (125) MHz respectively. Chemical shifts are reported as δ values (ppm) with CDCl_3 ($^1\text{H-NMR}$ δ 7.26, $^{13}\text{C-NMR}$ δ 77.16) or DMSO-d_6 ($^1\text{H-NMR}$ δ 2.50, $^{13}\text{C-NMR}$ δ 39.52) as an internal standard. J values are given in Hertz (Hz). $^{31}\text{P-NMR}$ spectra were recorded on a Bruker Avance DMX 500 at 200 MHz. GC-MS analysis were performed on a DB-wax column (J&W Scientific, 30m, 0.25 mm id and 0.15 μm film thickness) connected to a Finnigan SSQ 7000 mass spectrometer (EI, 70 eV, ion source temperature: 150 $^\circ\text{C}$).

Tricyclohexylphosphine (PCy_3).

Air exposed PCy_3 was analyzed by quantitative $^{31}\text{P-NMR}$ (200 MHz, CDCl_3) δ 9.16 (PCy_3 , 55%), δ 48.25 (OPCy_3 , 45%).

Preparation of tricyclohexylphosphine oxide (OPCy_3).

Tricyclohexylphosphine (PCy_3) (20 mg, 0.07 mmol) was dissolved in Toluene (5 ml). The mixture was then heated for 2h at 80 $^\circ\text{C}$ with air bubbling. Solvent evaporation under reduced pressure led to a white powder. $^{31}\text{P-NMR}$ (200 MHz, CDCl_3) δ 48.25 (OPCy_3).

Methyl disulfide 1. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 2.40 (6H, s). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 22.14. MS (EI) m/z 94.1.

Ethyl disulfide 2a. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 1.31 (6H, t, $J = 7.25$ Hz), 2.67 (4H, q, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 14.48, 32.91. MS (EI) m/z 122.1.

Propyl disulfide 2b. $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ 0.99 (6H, t, $J = 7.3$ Hz), 1.70 (4H, sextet, $J = 7.3$ Hz), 2.66 (4H, t, $J = 7.3$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 13.20, 22.58, 41.25.

Phenyl disulfide 2c. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 7.20 (4H, t, $J = 7.25$ Hz), 7.28 (2H, t, $J = 7.56$ Hz), 7.48 (4H, d, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 127.30, 127.66, 129.20, 137.17.

Allyl disulfide 2d. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 3.38 (4H, d, $J = 7.35$ Hz), 5.15 (2H, m), 5.18 (2H, dq, $J = 1.35$ Hz, $J = 16.9$ Hz), 5.80 (2H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 40.87, 118.61, 133.61.

Benzyl disulfide 2e. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 3.74 (4H, s), 7.28 (6H, m), 7.33 (4H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 41.61, 127.29, 128.39, 129.36, 137.29.

Ethyl methyl disulfide 3a. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 1.31 (3H, t, $J = 7.25$ Hz), 2.38 (3H, s), 2.70 (2H, q, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 14.48, 23.49, 32.03. MS (EI) m/z 108.

Methyl propyl disulfide 3b. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 0.98 (3H, t, $J = 7.25$ Hz), 1.70 (2H, sextet, $J = 7.25$ Hz), 2.38 (3H, s), 2.67 (2H, t, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ . 13.20, 22.58, 23.44, 40.42.

Methyl phenyl disulfide 3c. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 2.42 (3H, s), 7.21 (2H, t, $J = 7.25$ Hz), 7.32 (1H, t, $J = 7.25$ Hz), 7.52 (2H, d, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, CDCl_3) δ 22.98, 126.95, 127.67, 129.10, 137.00.

Allyl methyl disulfide 3d. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 2.39 (3H, s), 3.40 (2H, m), 5.15 (1H, m), 5.21 (1H, dq, $J = 1.30$ Hz, $J = 16.95$ Hz), 5.84 (1H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 22.52, 40.21, 118.49, 133.74.

Benzyl methyl disulfide 3e. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 2.17 (3H, s), 3.98 (2H, s), 7.28 (3H, m), 7.32 (2H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 22.13, 41.31, 127.21, 128.36, 129.30, 137.55.

Ethyl propyl disulfide 4a. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 0.94 (3H, t, $J = 7.25$ Hz), 1.25 (3H, t, $J = 7.25$ Hz), 1.64 (2H, sextet, $J = 7.25$ Hz), 2.68 (2H, t, $J = 7.25$ Hz), 2.70 (2H, q, $J = 7.25$ Hz). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 12.70, 14.22, 21.86, 31.74, 39.99.

Allyl ethyl disulfide 4b. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 1.24 (3H, t, $J = 7.30$ Hz), 2.70 (2H, q, $J = 7.30$ Hz), 3.37 (2H, m), 5.13 (1H, m), 5.20 (1H, dq, $J = 1.35$ Hz, $J = 16.95$ Hz), 5.80 (1H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 14.26, 31.56, 41.10, 118.44, 133.68.

Benzyl ethyl disulfide 4c. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 1.16 (3H, t, $J = 7.25$ Hz), 2.49 (2H, q, $J = 7.25$ Hz), 3.95 (2H, s), 7.28 (3H, m), 7.33 (2H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 14.10, 31.21, 42.19, 127.20, 128.33, 129.25, 137.56.

Allyl propyl disulfide 5a. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 0.94 (3H, t, $J = 7.30$ Hz), 1.63 (2H, sextet, $J = 7.15$ Hz), 2.68 (2H, q, $J = 7.15$ Hz), 3.36 (2H, dt, $J = 1$ Hz, $J = 4.85$ Hz), 5.14 (1H, m), 5.21 (1H, dq, $J = 1.30$ Hz, $J = 16.95$ Hz), 5.80 (1H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 12.75, 21.84, 39.79, 41.02, 118.43, 133.68.

Benzyl propyl disulfide 5b. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 0.86 (3H, t, $J = 7.30$ Hz), 1.54 (2H, sextet, $J = 7.30$ Hz), 2.44 (2H, t, $J = 7.1$ Hz), 3.95 (2H, s), 7.28 (3H, m), 7.33 (2H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 12.75, 21.66, 39.28, 42.13, 127.18, 128.32, 129.25, 137.57.

Allyl benzyl disulfide 6a. $^1\text{H-NMR}$ (500 MHz, DMSO- d_6) δ 3.14 (4H, d, $J = 7.35$ Hz), 3.96 (2H, s), 5.10 (4H, m), 5.75 (2H, m), 7.28 (3H, m), 7.34 (2H, m). $^{13}\text{C-NMR}$ (500 MHz, DMSO- d_6) δ . 40.44, 41.97, 118.57, 127.24, 128.36, 133.34, 137.40.

Reversibility control experiment

1 and **2a** (350 mM of each) were mixed together in CDCl_3 and in presence of 5 mol% of PCy_3 . After equilibration of the reaction (**1**, **2a** and **3a** were present in solution with a ratio 1:1:2.2 respectively, Figure S1a-b), **2b** (350 mM) was added. A new equilibration was formed, indicating reversibility of the reaction (Figure S1c-d).

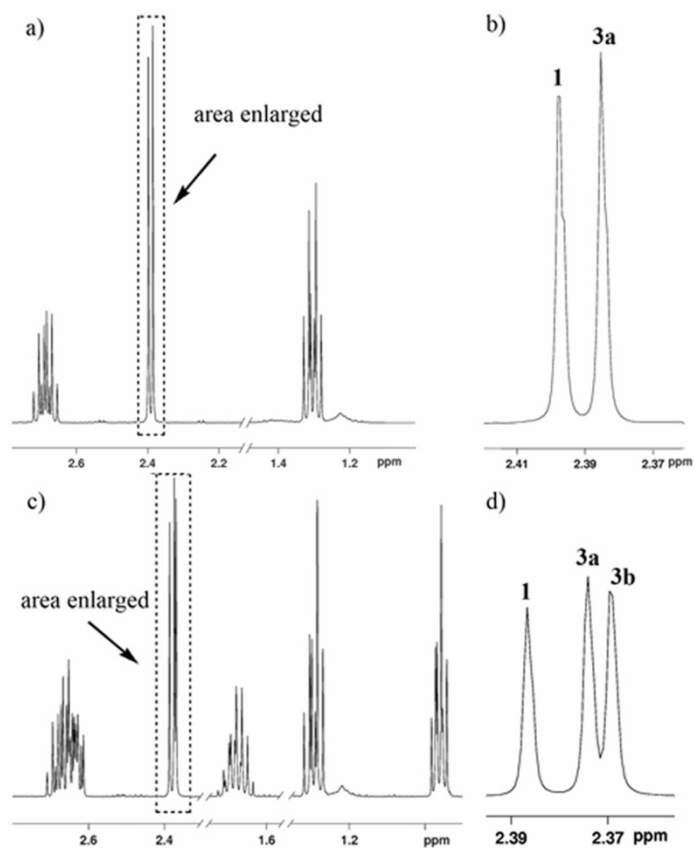


Figure S1: $^1\text{H-NMR}$ spectra of the reaction mixture: a) after equilibration between **1** and **2a**; b) enlarged area of the methyl region of a); c) after equilibration between **1**, **2a** and **2b** (50 min); d) enlarged area of the methyl region. Compound numbering as in Table 1.

GC-MS analysis of **1** and **2a** after reaction.

A GC-MS analysis was performed in order to confirm the $^1\text{H-NMR}$ results and monitor the presence of the two symmetrical disulfides and the unsymmetrical disulfide resulting from the exchange reaction. A mixture of methyl disulfide **1** (0.23 mmol) and ethyl disulfide **2a** (0.23 mmol) in benzene and in presence of 5 mol% of PCy_3 was injected after 11 days of reaction (around 150 ng) for separation on a DB-wax column. The following temperature program was used: 40 °C (1 min), 5 °C/min, 225 °C (15 min). Injection was 220 °C (split closed 30 s), and the transfer was held at 230 °C. The GC chromatogram confirmed the $^1\text{H-NMR}$ results. The retention time of 5.41 min, 6.95 min and 8.60 min (see Figure S1) correspond to **1**, ethyl methyl disulfide **3a** and **2a** respectively. The reaction products were clearly identified by Mass Spectrometry (see Figure S2, Figure S3 and Figure S4).

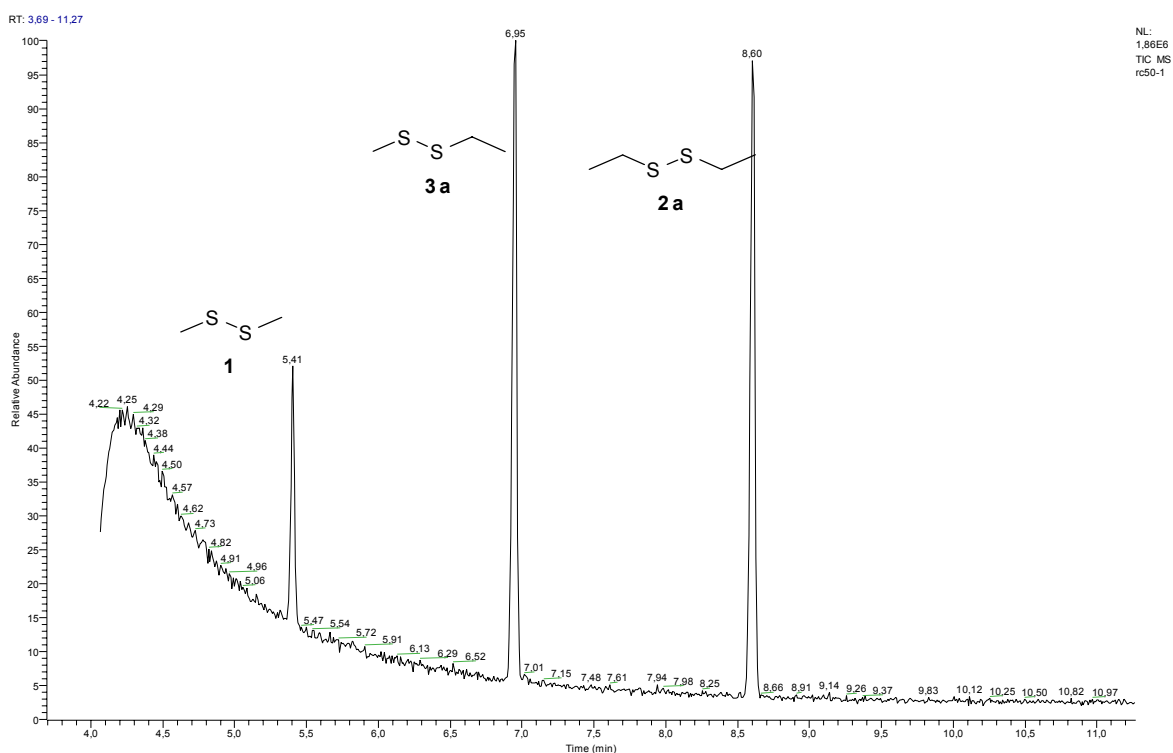


Figure S2: GC chromatogram of a mixture of **1** (0.23 mmol), **2a** (0.23 mmol) and PCy_3 (5 mol%) in benzene after 11 days. The three picks with a retention time (RT) of 5.41 min, 6.95 min and 8.60 min correspond to **1**, **3a** and **2a** respectively.

rc50-1 #155-161 RT: 5.37-5.42 AV: 7 SB: 99 4.83-5.17, 5.52-6.00 NL: 1.48E5
T: + c EI Q1MS [30.00-300.00]

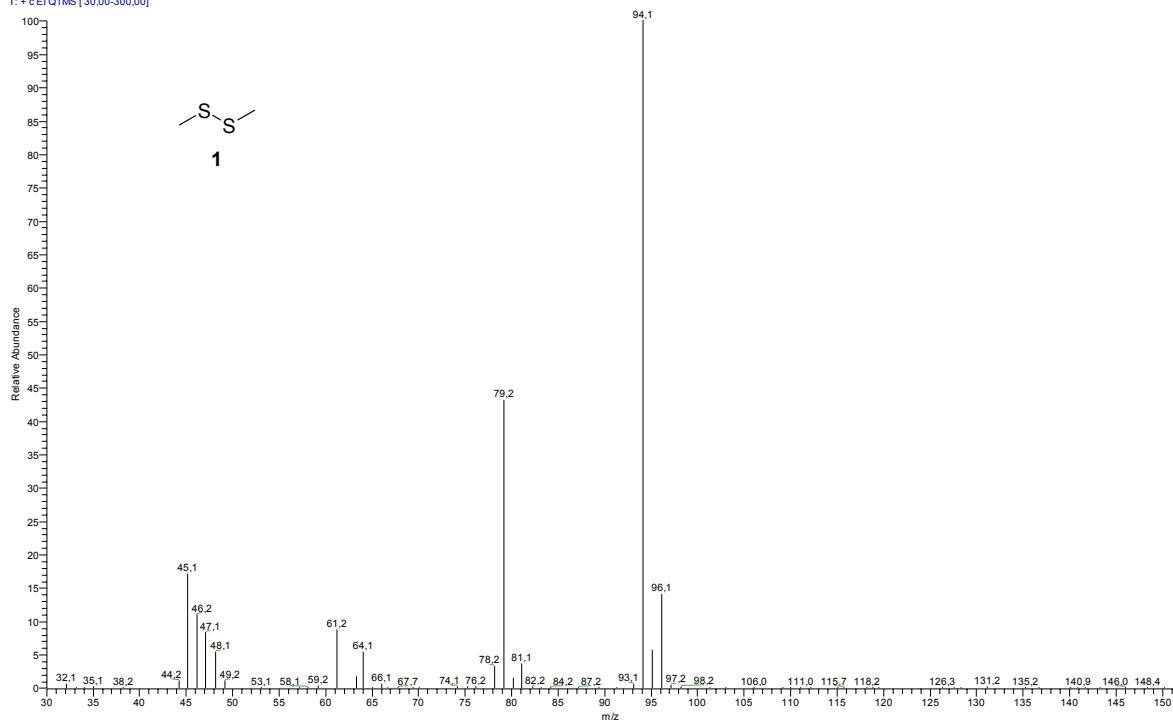


Figure S3: MS spectrum of methyl disulfide **1** (retention time: 5.41 min)

rc50-1 #336-344 RT: 6.91-6.98 AV: 9 SB: 99 4.83-5.17, 5.52-6.00 NL: 4.09E5
T: + c EI Q1MS [30.00-300.00]

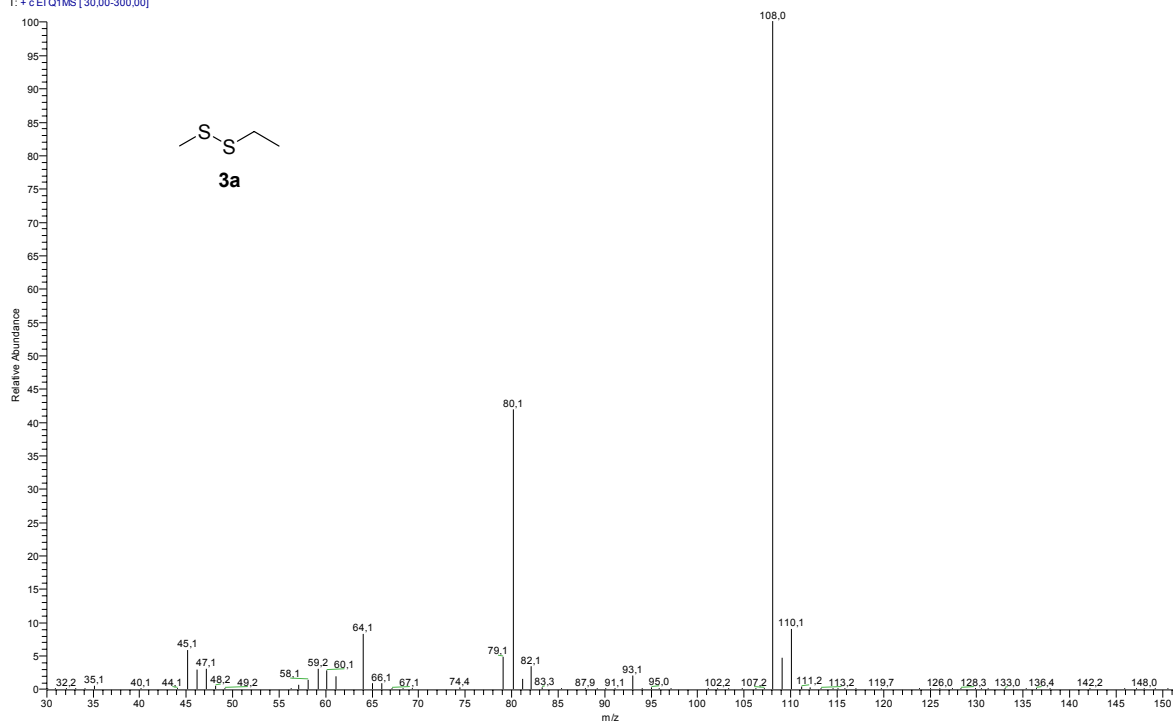


Figure S4: MS spectrum of ethyl methyl disulfide **3a** (retention time: 6.95 min)

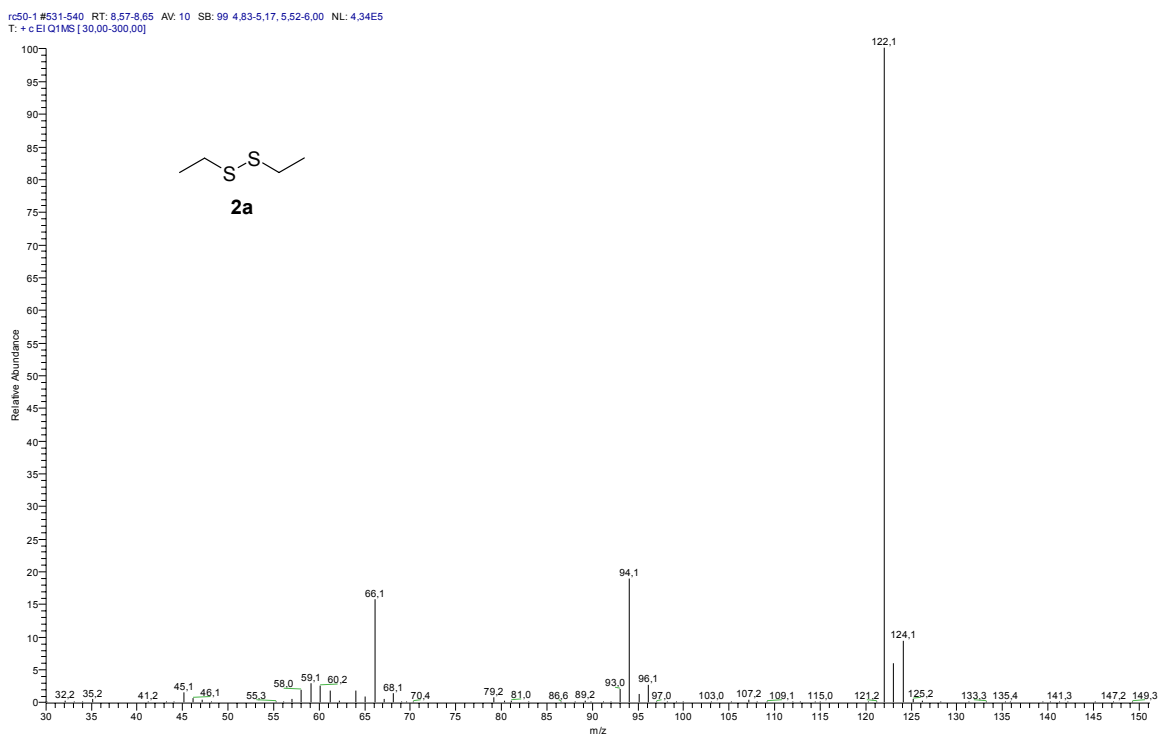


Figure S5: MS spectrum of ethyl disulfide **2a** (retention time: 8.60 min)