

Supplementary Information

Entry to new *N,O*-ligands from oxygen-depleted calixarenes

Ella Tzadka (Bukhaltsev),^a Israel Goldberg^a and Arkadi Vigalok^{a*}

School of Chemistry, The Raymond and Beverly Sackler Faculty of Exact Sciences, Tel Aviv University, Tel Aviv 69967, Israel

Email: avigal@post.tau.ac.il

General Information

All operations with air- and moisture-sensitive compounds were performed in a nitrogen-filled Innovative Technology glovebox. All solvents were degassed and stored under high-purity nitrogen over activated 4Å molecular sieves. All deuterated solvents were stored under high-purity nitrogen over 3Å molecular sieves. Commercially available reagents were purchased from Aldrich and used as received. The NMR spectra were recorded on Bruker AC 200MHz and Bruker AMX 400MHz spectrometers. ¹H and ¹³C NMR signals are reported in ppm downfield from TMS. ¹H signals are referenced to the residual proton of a deuterated solvent (7.26 ppm for CDCl₃, 7.15 ppm for C₆D₆). For ¹³C NMR spectra, the following signals were used as a reference: 77.36 ppm for CDCl₃, 128.62 ppm for C₆D₆. t-Butylcalix[4]arene 1,3-di-*O*-triflate was prepared according to ref. 1. Abbreviations: DBU= 1,8-Diazabicyclo[5.4.0]undec-7-ene, dba= Dibenzylideneacetone, DMF= *N,N*-Dimethylformamide.

Synthesis of **1**.

The reaction was carried out under an inert atmosphere of pure nitrogen. To a solution of P(t-Bu)₃H⁺ BF₄⁻ (16mg, 0.055mmol) and Pd₂dba₃ (12.5 mg, 0.014 mmol) in 10 ml of dry DMF, CuI (250 mg, 1.31 mmol), DBU (334 mg, 2.20 mmol), trimethylsilylacetylene (540 mg, 5.5 mmol) and t-butylcalix[4]arene 1,3-di-*O*-triflate (500 mg, 0.55 mmol) were added and the mixture was heated at 100°C for 4 hours. The solvent was evaporated and the resulting crude product was dissolved in CH₂Cl₂ (40ml), washed with saturated aqueous NH₄Cl (20ml) and H₂O (20ml). Drying the CH₂Cl₂ extract over MgSO₄ followed by solvent removal under vacuum gave the crude product. The product was dissolved in 3ml of CH₂Cl₂ and precipitated with 15ml of MeOH giving 350 mg of **1** (79% yield). ¹H NMR (CDCl₃): 0.32 (18H, s, TMS), 0.82 (18H, s, t-Bu), 1.35 (18H, s, t-Bu), 3.57 (4H, d, ²J_{HH}= 13.6 Hz, CH₂), 4.60 (4H, d, ²J_{HH}= 13.6 Hz, CH₂), 5.25 (2H, s, OH), 6.59 (4H, s, Ar-H), 7.12 (4H, s, Ar-H). ¹³C NMR: 0.66 (TMS), 30.95, 32.09 (CMe₃), 34.26, 34.44 (CMe₃), 36.74 (CH₂), 103.03, 103.28 (C≡C), all aromatic (s): 119.41, 124.08, 125.69, 129.00, 141.78, 142.31, 150.97, 151.48. IR (film) 2144 cm⁻¹ (C≡C). M. p. 263.2-265.7 °C.

Synthesis of **2**.

To a solution of **1** (70mg, 0.087mmol) in 2ml THF and 10ml methanol, K₂CO₃ (94mg, 0.68mmol) was added. The whole mixture was stirred under nitrogen for 4 hours and quenched with water. The product was extracted with CH₂Cl₂ and the organic layer dried over MgSO₄. Solvent evaporation afforded pure **2** (60mg, 86% yield).

^1H NMR: 0.92 (18H, s, tBu), 1.35 (18H, s, tBu), 3.59 (6H, d, $^2J_{\text{HH}} = 13.7$ Hz, CH_2 , overlapped with 2H of $\text{HC}\equiv\text{C}$), 4.49 (4H, d, $^2J_{\text{HH}} = 13.7$ Hz, CH_2), 5.00 (2H, s, OH), 6.77 (4H, s, Ar-H), 7.15 (4H, s, Ar-H). ^{13}C NMR: 31.11, 32.15 (CMe_3), 34.33, 34.62 (CMe_3), 37.32 (CH_2), 81.87, 85.91 ($\text{C}\equiv\text{C}$), all aromatic (s): 118.20, 124.69, 126.41, 128.63, 142.49, 142.63, 151.29, 151.48. MS-ESI: Measured (Calc.) 664.4270 (664.4280), $\text{C}_{48}\text{H}_{56}\text{O}_2$. IR (film) 2097 cm^{-1} ($\text{C}\equiv\text{C}$).

Synthesis of **3**.

To a solution of **2** (50mg, 0.075mmol) and benzyl azide (30mg, 0.165mmol) in 5ml of dry DMF, $\text{CuSO}_4\cdot 5\text{H}_2\text{O}$ (19mg, 0.075mmol) and sodium ascorbate (15mg, 0.075mmol) were added. The mixture was stirred under nitrogen for 18 hours, the solution was filtered through celite. CH_2Cl_2 and brine were added and the organic product was extracted and dried over MgSO_4 . CH_2Cl_2 was evaporated and the product was re-dissolved in 1ml of DCM. Precipitation with 10ml of MeOH gave a white powder of **3** (50mg, 72% yield).

^1H NMR: 0.95 (18H, s, tBu), 1.31 (18H, s, tBu), 3.37 (4H, d, $^2J_{\text{HH}} = 13.7$ Hz, CH_2), 3.96 (4H, d, $^2J_{\text{HH}} = 13.7$ Hz, CH_2), 5.66 (4H, s, $\text{CH}_2\text{-Ph}$), 6.82 (4H, s, Ar-H), 6.95 (4H, s, Ar-H), 7.36 (m, 10H, Ar-H) 7.66 (2H, s, $\text{HC}=\text{C}$), 8.69 (2H, s-broad, OH). ^{13}C NMR: 31.01, 31.96 (CMe_3), 33.93, 34.15 (CMe_3), 36.35 (CH_2), 54.47 ($\text{CH}_2\text{-Ph}$), 124.66, 124.96 ($\text{C}=\text{C}$), all aromatic (s): 125.42, 125.52, 128.18, 128.73, 129.03, 129.27, 135.08, 139.13, 140.86, 145.68, 150.11, 151.90. MS-ESI: Measured (Calc.) 930.5556 (930.5560), $\text{C}_{62}\text{H}_{70}\text{N}_6\text{O}_2$.

Synthesis of **4**.

Ligand **3** (10mg, 0.011mmol) was dissolved in about 3ml of dry toluene and $12\mu\text{L}$ of a TiCl_4 solution in toluene (1.0M, 0.012mmol) was added. The resulting red mixture was heated at $100\text{ }^\circ\text{C}$ for 18 hrs and the solvent was evaporated giving the crude product. After precipitation from toluene/pentane (0.5ml of toluene – 3ml of pentane), 8mg of pure red product was obtained (71% yield).

^1H NMR: 0.93 (18H, s, t-Bu), 1.28 (18H, s, t-Bu), 3.21 (4H, d, $^2J_{\text{HH}} = 13.8$ Hz, CH_2), 3.63 (4H, d, $^2J_{\text{HH}} = 13.8$ Hz, CH_2), 5.77 (4H, s, $\text{CH}_2\text{-Ph}$), 6.57 (4H, s, Ar-H), 6.90 (4H, s, Ar-H), 7.24 (2H, s, $\text{HC}=\text{C}$), 7.36-7.46 (m, 10H, Ar-H). ^{13}C NMR: 31.09, 31.64 (CMe_3), 34.03, 34.38 (CMe_3), 37.96 (CH_2), 56.08 ($\text{CH}_2\text{-Ph}$), 122.97, 123.23 ($\text{C}=\text{C}$), all aromatic (s): 125.26, 126.59, 128.38, 129.10, 129.42, 129.60, 132.97, 133.11, 140.81, 143.08, 146.11, 152.29. FAB-MS: Measured (Calc.) 976(976); $\text{C}_{62}\text{H}_{68}\text{Cl}_2\text{N}_6\text{O}_2\text{Ti}$.

Synthesis of **5**.

To a solution of **4** (5mg, 0.005mmol) in 3 ml of dry CH₂Cl₂ catechol (0.5 mg, 0.005 mmol) or (S)-BINOL (1.4mg, 0.005mmol) was added and the mixture stirred at room temperature overnight. The solvent was evaporated to give the crude product. After precipitation from CH₂Cl₂/pentane (0.5ml of CH₂Cl₂ – 5ml of pentane), pure **5** (**a** or **b**) was obtained as red solid in 70-75% yield.

5a: ¹H NMR (CDCl₃): 0.99 (18H, s, t-Bu), 1.26 (18H, s, t-Bu), 3.37 (4H, d, ²J_{HH}= 15.0 Hz, CH₂), 3.69 (4H, d, ²J_{HH}= 15.0 Hz, CH₂), 5.67 (4H, s, CH₂-Ph), 6.15 (4H, s, Ar-H), 6.58-6.86 (8H, m, Ar-H), 7.40 (m, 10H, Ar-H), 7.78 (2H, s). ¹³C NMR δ, 31.70, 32.25 (CMe₃), 34.18, 35.32 (CMe₃), 39.94 (CH₂), 57.06 (CH₂-Ph), 124.45, 125.43 (C=C), all aromatic (s):116.26 121.91, 126.61, 127.38, 128.28, 128.87, 129.55, 130.29, 130.39, 133.25, 136.08, 143.5, 144.31, 152.04, 154.55.

5b: ¹H NMR (CDCl₃): 0.93 (18H, s, t-Bu), 1.28 (18H, s, t-Bu), 3.21 (4H, d, ²J_{HH}= 13.8 Hz, CH₂), 3.62 (4H, d, ²J_{HH}= 13.8 Hz, CH₂), 5.22 (2H, s, Ar-BINOL), 5.69 (4H, s, CH₂-Ph), 6.56 (4H, s, Ar-H), 6.89 (4H, s, Ar-H), 7.12-7.44 (m, 16H, Ar-H), 7.87-7.98 (6H, m, Ar-H, overlapped with 2H of the triazole ring). ¹³C NMR δ, 31.62, 32.17 (CMe₃), 34.57, 34.91 (CMe₃), 38.47 (CH₂), 56.53 (CH₂-Ph), 123.50, 123.79 (C=C), all aromatic (s):118.55, 123.61, 124.63, 124.97, 125.76, 127.13, 128.07, 129.07, 129.60, 129.72, 129.93, 130.10, 131.97, 133.51, 133.60, 134.00, 134.13, 141.32, 143.65, 146.61, 152.83, 153.48; FAB-MS: Measured (Calc.) 1262 (1262); C₈₂H₈₀N₆O₄Ti.

Trimethylsilyl addition of benzaldehyde.²

General procedure: In a vial, catalyst **4** (12.2mg, 0.0125mmol, 1 mol%) dissolved in 2 ml of dry CH₂Cl₂ under nitrogen and benzaldehyde (133 mg, 1.25 mmol) and trimethylsilylcyanide (137mg, 1.38 mmol) were added with syringe via rubber septum. The solution was stirred for 8 hours and the solvent evaporated. To the crude product, about 2ml of methanol were added to remove the catalyst. The combined solvent was evaporated giving 240mg (92% yield) of the product.

[1]- S. Chowdhury, J. N. Bridson, P. E. Georghiou, *J. Org. Chem.* 2000, **65**, 3299–3302.

[2]- Y. N. Belokon', S. Caveda-Cepas, B. Green, N. S. Ikonnikov, V. N. Khrustalev, V. S. Larichev, M. A. Moscalenko, M. North, C. Orizu, V. I. Tararov, M. Tasinazzo, G. I. Timofeeva, L. V. Yashkina, *J. Am. Chem. Soc.* 1999, **121**, 3968-3973.