

Electronic Supplementary Information

**Mesoporous silica-catalysed continuous chemical fixation of CO₂ with
N,N'-dimethylethylenediamine in supercritical CO₂: Efficient synthesis of
1,3-dimethyl-2-imidazolidinone**

Tsunetake Seki,[†] Yoshiaki Kokubo, Shinichiro Ichikawa, Tomoyuki Suzuki, Yoshihito Kayaki and
Takao Ikariya*

*Department of Applied Chemistry, Tokyo Institute of Technology,
Ookayama, Meguro-ku, Tokyo 152-8552, Japan*

[†]Research Fellow of the Japan Society for the Promotion of Science

N₂ adsorption–desorption isotherms of the silica catalysts used

Experimental: The Q-30 (amorphous silica; CARiACT Q-30; Fuji Silysia) and A-P (amorphous silica; Kanto Chemical) samples were analysed as-received, while the HMS (mesoporous silica with wormhole framework; Aldrich) and MCM-41 (mesoporous silica with hexagonal framework; Aldrich) samples were those converted to larger particles (500–800 µm) by pressurization at 20 MPa and subsequent sieving for the use in the reaction. The sample was activated at 500 °C for 2 h in a stream of dry air (dew point – 40 °C) prior to analysis, which were the same pretreatment conditions employed for the reaction. N₂ adsorption–desorption measurement was performed at – 196 °C with a BELSORP-mini (Bel Japan) using static adsorption procedures.

Note: Both MCM-41 and HMS exhibited the adsorption step centered in the relative pressure (P/P_0) region from 0.3 to 0.6 indicating the presence of framework-confined mesopores. There is, however, a striking difference in the higher relative pressure region of 0.6 to 1.0. A steep increase in the isotherm of HMS in this region indicates that the fraction of “textural” mesoporosity is much greater for HMS than for MCM-41, which is a well-known difference between the two mesoporous silicas: see, P. T. Tanev and T. J. Pinnavaia, *Chem. Mater.*, 1996, **8**, 2068.

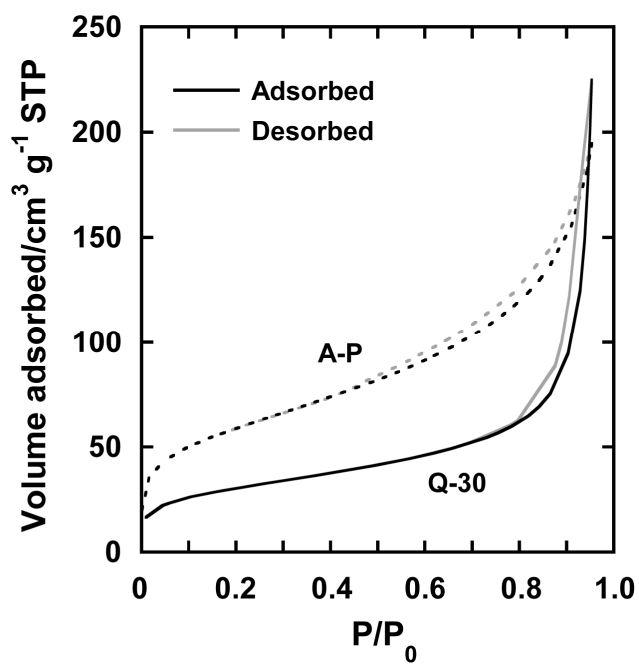


Fig. S1 N₂ adsorption-desorption isotherm of Q-30 and A-P.

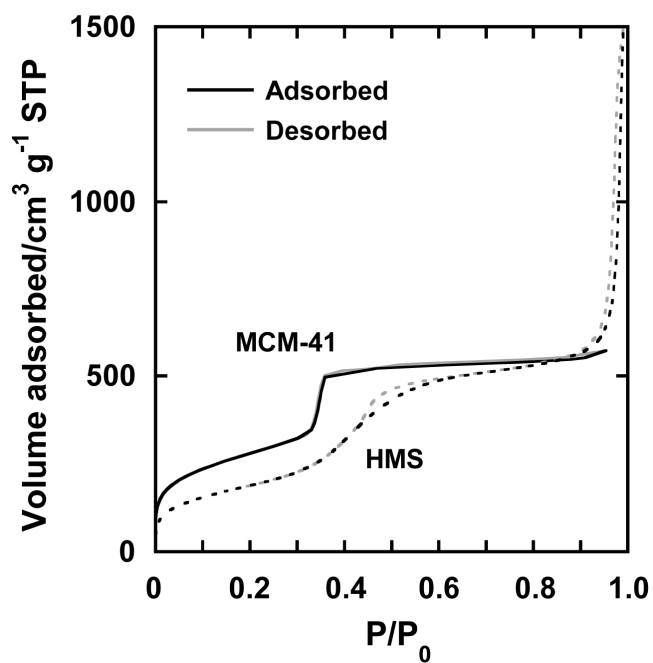


Fig. S2 N₂ adsorption-desorption isotherm of MCM-41 and HMS.