Supplementary Information

Imido Transfer of Sulfonylimino- λ^3 -Bromane Makes Possible the Synthesis of Sulfonylimino- λ^3 -Iodanes

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General Information. IR spectra were recorded on JASCO FT/IR-420 spectrometers. ¹H NMR and ¹³C NMR spectra were obtained on a JEOL JNM-AL300 or JNM-AL400 spectrometers. Chemical shifts (δ) are reported in parts per million (ppm) downfield from internal Me₄Si. Mass spectra (MS) were obtained on a Waters LCT Premier spectrometer. Melting points were determined with a Yanaco micro melting points apparatus and are uncorrected. Dichloromethane was dried over CaH₂ and distilled.

General Procedure for Transimidation of Imino- λ^3 -Bromane 1 Yielding Sulfonylimino- λ^3 -Iodanes 2. A Typical Example (Table 1, Entry 1): [*N*-(Trifluoromethanesulfonyl)imino]phenyl- λ^3 -iodane (2a). To a stirred solution of [*N*-(trifluoromethanesulfonyl)imino](*p*-trifluoromethylphenyl)- λ^3 -bromane (1)¹ (27.8 mg, 0.075 mmol) in dichloromethane (1 mL) was added iodobenzene (18.4 mg, 0.09 mmol) at room temperature under argon and the mixture was stirred for 1 h. After evaporation under an aspirator vacuum, repeated decantation of the crude reaction mixture with dichloromethane gave (trifluoromethanesulfonyl)imino- λ^3 -iodane 2a (25.3 mg, 97%).² White powder; mp 214-217 °C; IR (KBr) 1631, 1284, 1176, 1132, 976, 648, 602 cm⁻¹; ¹H NMR (400 MHz, CD₃CN) δ 8.11 (d, *J* = 7.2 Hz, 2H), 7.71 (t, *J* = 7.2 Hz, 1H), 7.57 (t, *J* = 7.2 Hz, 2H); HRMS (ESI, positive) calcd for C₉H₈F₃IN₂NaO₂S [(M+MeCN+Na)⁺] 414.9201, found 414.9226. Anal. Calcd for C₇H₃F₃INO₂S: C, 23.95; H, 1.44; N, 3.99. Found: C, 23.81; H, 1.62; N, 4.31.

[*N*-(**Trifluoromethanesulfonyl**)**imino**](2-methylphenyl)- λ^3 -**iodane** (2b): white powder; mp 119-121 °C; IR (KBr) 1635, 1284, 1182, 1132, 976, 652, 598 cm⁻¹; ¹H NMR (400 MHz, CD₃CN) δ 8.16 (d, *J* = 7.2 Hz, 1H), 7.62 (t, *J* = 7.2 Hz, 1H), 7.54 (d, *J* = 7.2 Hz, 1H), 7.32 (t, *J* = 7.2 Hz, 1H), 2.69 (s, 3H); ¹³C NMR (75 MHz, CD₃CN) δ 142.7, 138.5, 134.8, 132.2, 129.9, 123.8, 121.8 (q, ¹*J*_{CF} = 324.5 Hz), 25.2; HRMS (ESI, positive) calcd for C₁₀H₁₀F₃IN₂NaO₂S [(M+MeCN+Na)⁺] 428.9358, found 428.9319. Anal. Calcd for C₈H₇F₃INO₂S: C, 26.32; H, 1.93; N, 3.84. Found: C, 26.43; H, 2.14; N, 3.74.

[*N*-(Trifluoromethanesulfonyl)imino](4-methylphenyl)- λ^3 -iodane (2c): white powder; mp 181-184 °C; IR (KBr) 1631, 1282, 1173, 1130, 970, 648, 596 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.96 (d, *J* = 7.4 Hz, 2H), 7.34 (d, *J* = 7.4 Hz, 2H), 2.40 (s, 3H); HRMS (ESI, positive) calcd for C₁₀H₁₀F₃IN₂NaO₂S [(M+MeCN+Na)⁺] 428.9358, found 428.9339. Anal. Calcd for C₈H₇F₃INO₂S: C, 26.32; H, 1.93; N, 3.84. Found: C, 26.39; H, 2.10; N, 3.71.

[*N*-(**Trifluoromethanesulfonyl**)imino](4-methoxyphenyl)- λ^3 -iodane (2d): white powder; mp 115-117 °C; IR (KBr) 1587, 1487, 1456, 1275, 1167, 1124, 987, 658, 598 cm⁻¹; ¹H NMR (400 MHz, acetone- d_6) δ 8.14 (d, *J* = 7.2 Hz, 2H), 7.12 (d, *J* = 7.2 Hz, 2H), 3.93 (s, 3H); HRMS (ESI, positive) calcd for C₁₀H₁₀F₃IN₂NaO₃S [(M+MeCN+Na)⁺] 444.9307, found 444.9269. Anal. Calcd for C₈H₇F₃INO₃S: C, 25.21; H, 1.85; N, 3.68. Found: C, 25.09; H, 1.99; N, 3.63.

[*N*-(Trifluoromethanesulfonyl)imino](4-fluorophenyl)- λ^3 -iodane (2e):² white powder; mp 229-231 °C; IR (KBr) 1630, 1275, 1161, 1124, 987, 650, 604 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆)

δ 8.20-8.10 (m, 2H), 7.46-7.36 (m, 2H); HRMS (ESI, positive) calcd for C₉H₇F₄IN₂NaO₂S [(M+MeCN+Na)⁺] 432.9107, found 432.9099. Anal. Calcd for C₇H₄F₄INO₂S: C, 22.78; H, 1.09; N, 3.80. Found: C, 22.54; H, 1.23; N, 4.17.

[*N*-(Trifluoromethanesulfonyl)imino](4-chlorophenyl)- λ^3 -iodane (2f): pale yellow powder; mp 224-226 °C; IR (KBr) 1624, 1468, 1275, 1213, 1165, 1124, 987, 806, 658, 606 cm⁻¹; ¹H NMR (400 MHz, acetone- d_6) δ 8.22 (d, *J* = 8.7 Hz, 2H), 7.66 (d, *J* = 8.7 Hz, 2H); HRMS (ESI, positive) calcd for C₀H₇ClF₃IN₂NaO₂S [(M+MeCN+Na)⁺] 448.8811, found 448.8831.

[*N*-(**Trifluoromethanesulfonyl**)imino](4-bromophenyl)- λ^3 -iodane (2g): white powder; mp 224-226 °C; IR (KBr) 1620, 1469, 1275, 1163, 1126, 987, 804, 656, 606 cm⁻¹; ¹H NMR (400 MHz, acetone- d_6) δ 8.14 (d, J = 8.0 Hz, 2H), 7.82 (d, J = 8.0 Hz, 2H); HRMS (ESI, positive) calcd for C₉H₇⁷⁹BrF₃IN₂NaO₂S [(M+MeCN+Na)⁺] 492.8306, found 492.8296. Anal. Calcd for C₇H₄BrF₃INO₂S: C, 19.55; H, 0.94; N, 3.26. Found: C, 19.72; H, 1.18; N, 3.18.

[*N*-(Trifluoromethanesulfonyl)imino][4-(trifluoromethyl)phenyl]- λ^3 -iodane (2h): white powder; mp 204-206 °C; IR (KBr) 1626, 1282, 1180, 1132, 970, 650, 600 cm⁻¹; ¹H NMR (400 MHz, acetone-*d*₆) δ 8.42 (d, *J* = 8.0 Hz, 2H), 7.98 (d, *J* = 8.0 Hz, 2H); HRMS (ESI, positive) calcd for C₈H₄F₆INNaO₂S [(M+Na)⁺] 441.8809, found 441.8832; calcd for C₉H₈F₆INNaO₃S [(M+MeOH+Na)⁺] 473.9072, found 473.9057. Anal. Calcd for C₈H₄F₆INO₂S: C, 22.93; H, 0.96; N, 3.34. Found: C, 23.14; H, 1.23; N, 3.51.

[*N*-(**Trifluoromethanesulfonyl**)imino](4-cyanophenyl)- λ^3 -iodane (2i): white powder; mp 166-168 °C; IR (KBr) 2227, 1624, 1579, 1475, 1392, 1282, 1176, 976, 818, 648, 594 cm⁻¹; ¹H NMR (400 MHz, acetone-*d*₆) δ 8.39 (d, *J* = 8.1 Hz, 2H), 8.04 (d, *J* = 8.1 Hz, 2H); HRMS (ESI, positive) calcd for C₁₀H₇F₃IN₃NaO₂S [(M+MeCN+Na)⁺] 439.9154, found 439.9146. Anal. Calcd for C₈H₄F₃IN₂O₂S: C, 25.55; H, 1.07; N, 7.45. Found: C, 25.83; H, 1.28; N, 7.47.

[*N*-(Trifluoromethanesulfonyl)imino][4-(ethoxycarbonyl)phenyl]- λ^3 -iodane (2j): white powder; mp 220-221 °C; IR (KBr) 2985, 1714, 1635, 1587, 1394, 1369, 1286, 1207, 1174, 1133, 1009, 978, 847, 754, 646, 596 cm⁻¹; ¹H NMR (400 MHz, acetone-*d*₆) δ 8.32 (d, *J* = 7.6 Hz, 2H), 8.17 (d, *J* = 7.6 Hz, 2H), 4.41 (q, *J* = 7.2 Hz, 2H), 1.40 (t, *J* = 7.2 Hz, 3H); HRMS (ESI, positive) calcd for C₁₁H₁₃F₃INNaO₅S [(M+MeOH+Na)⁺] 477.9409, found 477.9453. Anal. Calcd for C₁₀H₉F₃INO₄S: C, 28.38; H, 2.14; N, 3.31. Found: C, 28.37; H, 2.28; N, 3.24.

[*N*-(**Trifluoromethanesulfonyl**)**imino**][**2**-(**acetyl**)**phenyl**]- λ^3 -**iodane** (**2k**)**:** colorless prisms (recrystallized from dichloromethane–hexane); mp 101-102 °C; IR (nujol) 1631, 1576, 1309, 1292, 1173, 1119, 953, 781, 642, 602 cm⁻¹; ¹H NMR (400 MHz, CD₃CN) δ 8.49 (d, *J* = 8.1 Hz, 1H), 8.37 (d, *J* = 8.1 Hz, 1H), 8.15 (m, 1H), 7.92 (m, 1H), 2.88 (s, 3H); ¹³C NMR (75 MHz, CD₂Cl₂) δ 202.6, 138.2, 134.2, 131.9, 131.7, 127.3, 121.5 (q, ¹*J*_{CF} = 324.5 Hz), 117.6, 25.7. Anal. Calcd for C₉H₇F₃INO₃S·1/2H₂O: C, 26.88; H, 2.01; N, 3.48. Found: C, 26.95; H, 1.96; N, 3.69.

[*N*-(Trifluoromethanesulfonyl)imino](2,2,2-trifluoroethyl)- λ^3 -iodane (3): white powder;

mp 161 °C; IR (KBr) 1626, 1282, 1182, 1132, 974, 756, 650, 600 cm⁻¹; ¹H NMR (400 MHz, acetone- d_6) δ 5.06 (q, ³ $J_{\rm HF}$ = 10.9 Hz, 2H); HRMS (ESI, positive) calcd for C₅H₅F₆IN₂NaO₂S [(M+MeCN+Na)⁺] 420.8918, found 420.8926.

1-(Trifluoromethanesulfonamido)-1,2-benziodoxol-3(1*H***)-one (6): white powder; mp 194 °C; IR (KBr) 3095, 1666, 1630, 1568, 1371, 1292, 1232, 1200, 1124, 926, 831, 746 cm⁻¹; ¹H NMR (400 MHz, CD₃CN) \delta 8.16 (dd, J = 8.1, 1.5 Hz, 1H), 8.08-7.99 (m, 2H), 7.82 (ddd, J = 9.1, 6.6, 1.5 Hz, 1H); HRMS (ESI, positive) calcd for C₁₀H₈F₃IN₂NaO₄S [(M+MeCN+Na)⁺] 458.9099, found 458.9091. Anal. Calcd for C₈H₅F₃INO₄S: C, 24.32; H, 1.28; N, 3.55. Found: C, 24.32; H, 1.50; N, 3.88.**

[*N*-(Trifluoromethanesulfonyl)imino](2-carbamoylphenyl)- λ^3 -iodane (8a): colorless prisms (recrystallized from methanol–water); mp 87-93 °C; IR (KBr) 3384, 3228, 2744, 1664, 1587, 1545, 1441, 1319, 1300, 1211, 1151, 1130, 966, 741, 607 cm⁻¹; ¹H NMR (400 MHz, acetone-*d*₆) δ 8.43 (d, *J* = 7.2 Hz, 1H), 8.34 (d, *J* = 7.2 Hz, 1H), 8.10 (br t, *J* = 7.2 Hz, 1H), 7.87 (br t, *J* = 7.2 Hz, 1H); HRMS (ESI, positive) calcd for C₁₀H₉F₃IN₃NaO₃S [(M+MeCN+Na)⁺] 457.9259, found 457.9278. Anal. Calcd for C₈H₆F₃IN₂O₃S: C, 24.38; H, 1.53; N, 7.11. Found: C, 24.24; H, 1.70; N, 7.40.

[*N*-(Trifluoromethanesulfonyl)imino](2-carbamoyl-4-methylphenyl)- λ^3 -iodane (8b): pale yellow powder; mp 135-139 °C; IR (KBr) 3352, 3205, 1660, 1558, 1452, 1387, 1294, 1180, 1122, 958, 652, 606 cm⁻¹; ¹H NMR (400 MHz, acetone- d_6) δ 8.25 (s, 1H), 8.17 (d, *J* = 9.1 Hz, 1H), 7.92 (d, *J* = 9.1 Hz, 1H), 2.56 (s, 3H); HRMS (ESI, positive) calcd for C₉H₈F₃IN₂NaO₃S [(M+Na)⁺] 430.9150, found 430.9159.

References

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- L. M. Yagupolskii, V. I. Popov, N. V. Pavlenko, I. I. Maletina, A. A. Mironova, R. Yu. Gavrilova and V. V. Orda, *Zh. Org. Khim. USSR*, 1986, 22, 2169.



Fig. Time-course for transimidation to PhI with the λ^3 -bromane **1** (0.05 M) in MeCN- d_3 at 23 °C under argon: decrease (%) of **1** was evaluated by ¹H NMR. Concentration of PhI: (Δ) 0.05 M, (\circ) 0.125 M, (\bullet) 0.25 M.

X-ray Data for 2k

A. Crystal Data: Empirical Formula	$(C_9H_7]$	F_3 INO ₃ S), Formula Weight (393.12)	
Crystal Color, Habit	colorl	less, prism	
Crystal Dimensions	0.30 2	X 0.30 X 0.20 mm	
Crystal System	ortho	rhombic	
Lattice Type	Primi	tive	
Indexing Images	3 osci	illations @ 90.0 seconds	
Detector Position	127.4	0 mm	
Pixel Size	0.100	mm	
Lattice Parameters	a = 10	0.1845(3) Å, b = 19.9169(5) Å, c = 5.9157(2) Å	
	V = 1	199.96(6) Å ³	
Space Group	Pna2 ₁	(No. 33)	
Z value	4		
D _{calc}	2.176	g/cm ³	
F ₀₀₀	752.0	0	
μ(ΜοΚα)	28.81	4 cm ⁻¹	
B. Intensity Measurements			
Diffractometer	Rigak	tu RAXIS-RAPID	
Radiation	MoK	α ($\lambda = 0.71075$ Å), graphite monochromated	
Detector Aperture	280 n	nm x 256 mm	
Data Images	41 ex	posures	
$ω$ oscillation Range (χ =45.0, $φ$ =	=0.0)	130.0 - 190.0°	
Exposure Rate		180.0 sec./o	
$ω$ oscillation Range (χ =45.0, $φ$ =	=180.0) 0.0 - 160.0°	
Exposure Rate		180.0 sec./o	
Detector Position	127.4	0 mm	
Pixel Size	0.100	mm	
$2\theta_{max}$	54.8°		
No. of Reflections Measured	Total	: 10881, Unique: 10360 ($R_{int} = 0.023$)	
Corrections Lorentz-polarization, Absorption (transmission factors: 0.8930 - 0.9459)			
C. Structure Solution and Refinemen	nt		
Structure Solution	Direc	t Methods (SIR92)	
Refinement	Full-r	natrix least-squares on F	
Function Minimized	Σw (IFoI - IFcI) ²	
Least Squares Weights	1		
$2\theta_{max}$ cutoff	54.8°		
Anomalous Dispersion		All non-hydrogen atoms	
No. Observations (All reflection	ns)	2720	
No. Variables		165	
Reflection/Parameter Ratio		16.48	
Residuals: R (All reflections)		0.0157	
Residuals: R1 (I> $2.00\sigma(I)$)		0.0154	
Residuals: wR2 (All reflections)		0.0379	
Goodness of Fit Indicator		1.080	
Max Shift/Error in Final Cycle		0.002	
Maximum peak in Final Diff. N	lap	$0.63 e^{-1}/A^{-3}$	
Minimum peak in Final Diff. M	lap	$0 e^{-A^{3}}$	

X-ray Data for 8a

A. Crystal Data: Empirical Formula	$(C_8H_6F_3IN_2O_3S)$, Formula Weight (394.11)		
Crystal Color, Habit	colorless, prism		
Crystal Dimensions	0.20 X 0.20 X 0.20 mm		
Crystal System	orthorhombic		
Lattice Type	Primitive		
Indexing Images	3 oscillations @ 180.0 seconds		
Detector Position	127.40 mm		
Pixel Size	0.100 mm		
Lattice Parameters	a = 11.1853(4) Å, $b = 14.2294(4)$ Å, $c = 14.5837(4)$ Å		
	$V = 2321.14(12) Å^3$		
Space Group	P2 ₁ 2 ₁ 2 ₁ (No. 19)		
Z value	8		
D _{calc}	2.255 g/cm^3		
F ₀₀₀	1504.00		
μ(ΜοΚα)	29.818 cm ⁻¹		
B. Intensity Measurements			
Diffractometer	Rigaku RAXIS-RAPID		
Radiation	MoK α ($\lambda = 0.71075$ Å), graphite monochromated		
Detector Aperture	280 mm x 256 mm		
Data Images	44 exposures		
ω oscillation Range (χ =45.0, ϕ =	=130.0) 130.0 - 190.0°		
Exposure Rate	150.0 sec./º		
$ω$ oscillation Range (χ =45.0, ϕ =	=310.0) 0.0 - 160.0°		
Exposure Rate	150.0 sec./º		
Detector Position	127.40 mm		
Pixel Size	0.100 mm		
$2\theta_{max}$	54.9°		
No. of Reflections Measured	Total: 22608, Unique: 22562 (R _{int} = 0.030)		
Corrections Lorentz-pol	arization, Absorption (transmission factors: 0.536 - 0.551)		
C. Structure Solution and Refinemen	t		
Structure Solution	Direct Methods (SIR92)		
Refinement	Full-matrix least-squares on F		
Function Minimized	$\Sigma \text{ w} (\text{IFoI} - \text{IFcI})^2$		
Least Squares Weights	1		
$2\theta_{max}$ cutoff	54.9°		
Anomalous Dispersion	All non-hydrogen atoms		
No. Observations (All reflection	ns) 5281		
No. Variables	326		
Reflection/Parameter Ratio	16.20		
Residuals: R (All reflections)	0.0223		
Residuals: R1 (I>2.00σ(I))	0.0215		
Residuals: wR2 (All reflections)) 0.0520		
Goodness of Fit Indicator	1.087		
Max Shift/Error in Final Cycle	0.002		
Maximum peak in Final Diff. M	Iap 1.96 $e^{-}/Å^{3}$		
Minimum peak in Final Diff. M	$ap = 0 e^{-1}/A^{3}$		