

## Supporting Information

### **Polymer supported *Cinchona*-based bifunctional sulfonamide catalyst: A highly enantioselective, recyclable heterogeneous organocatalyst**

Sung Hun Youk, Sang Ho Oh, Ho Sik Ro, Je Eun Lee, Ji Woong Lee and Choong Eui Song\*

*Department of Chemistry, Institute of Basic Science, Sungkyunkwan University, Suwon 440-746, Korea; E-mail : s1673@skku.edu*

#### **Contents**

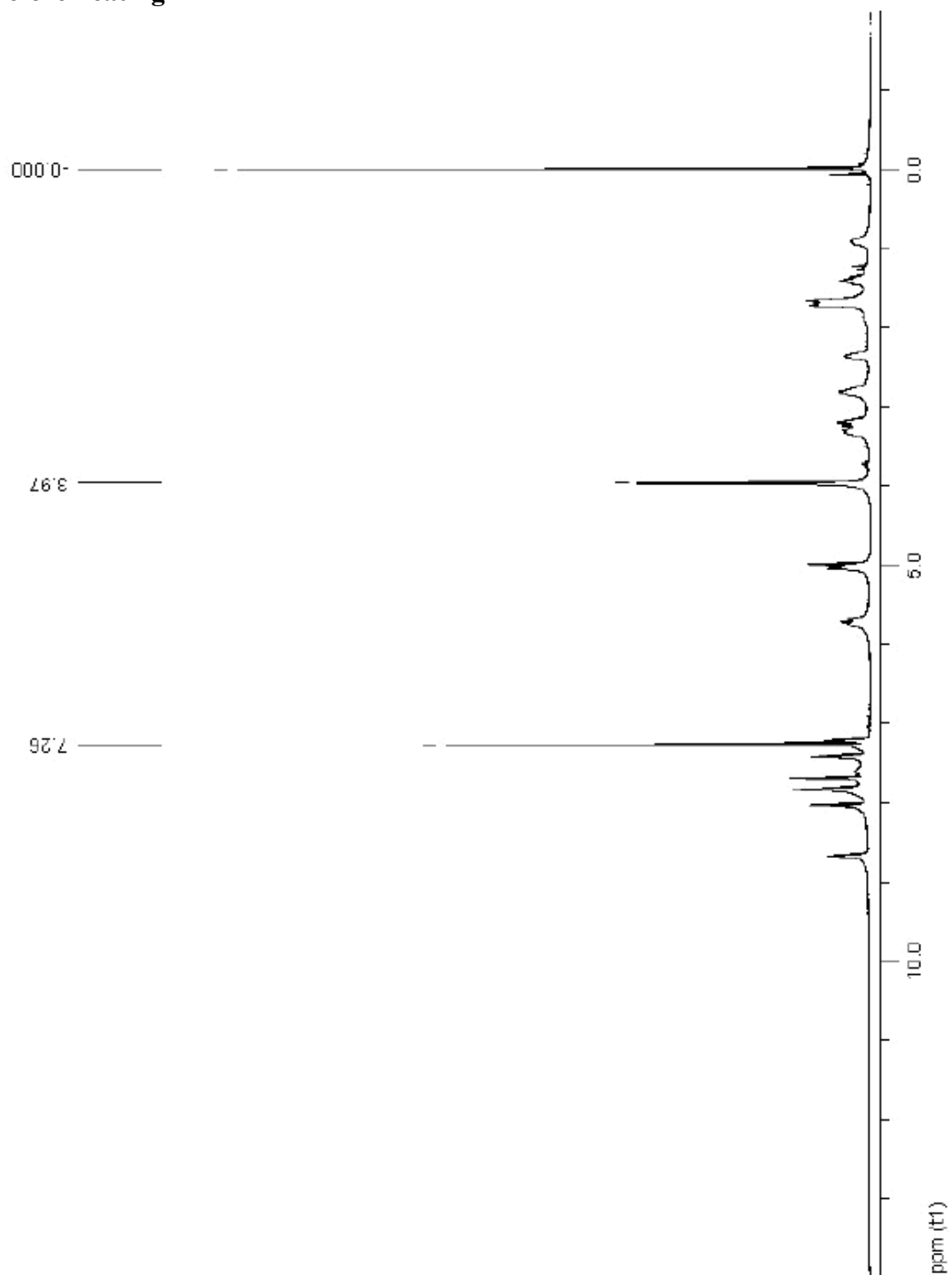
1. General
2.  $^1\text{H}$  NMR spectra of catalyst **I** (before heating and after heating)
3. Preparation and characterization of **III**
4. Preparation and characterization of the polymer supported catalyst **IV**
5. General procedure for methanolysis of prochiral cyclic anhydrides
6.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of **2a-2g**
7. HPLC spectra for Scheme 2
8. HPLC spectra for Table 1
9. References

## 1. General

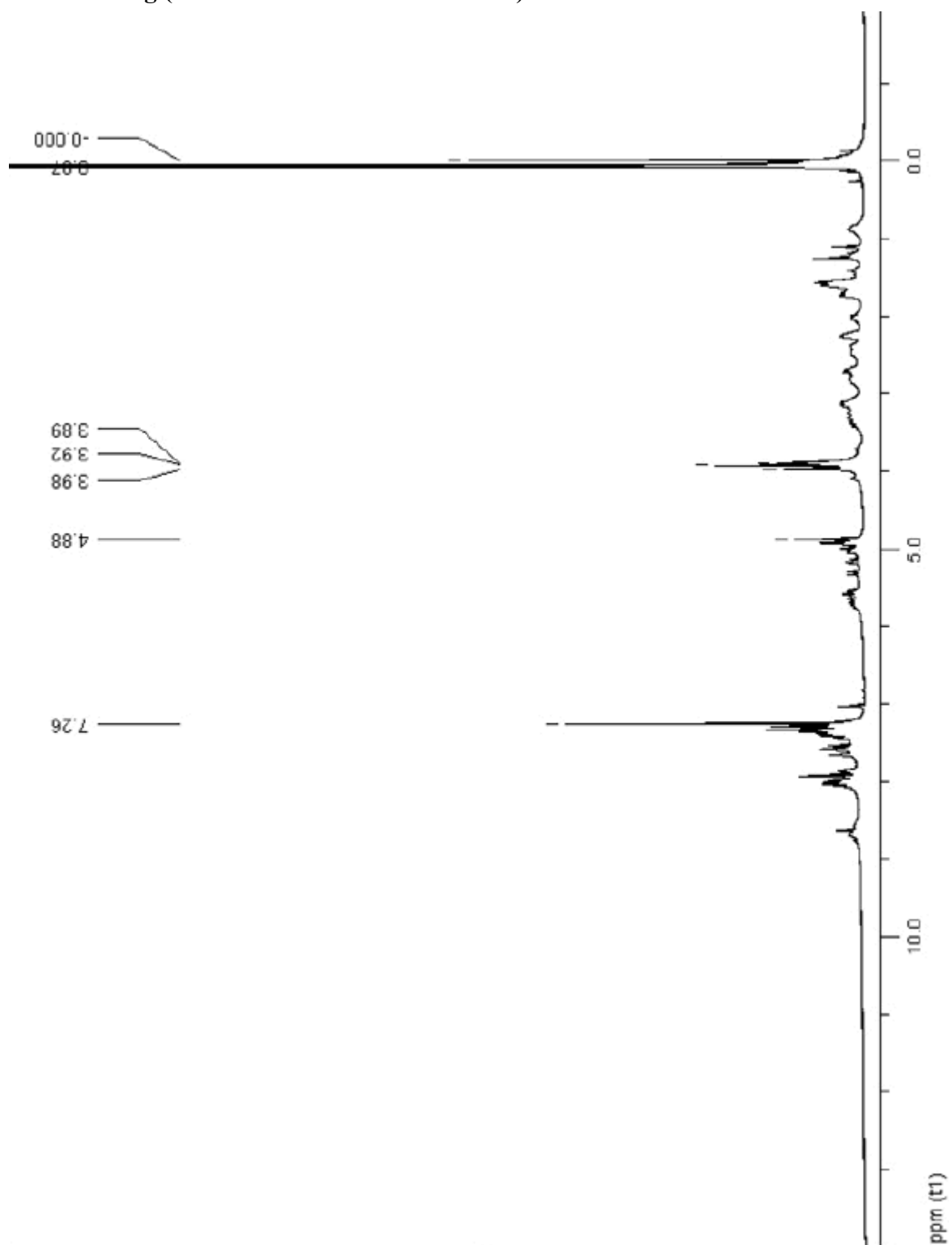
Anhydrides (**1a**, **1b**, **1c**, **1e**) were purchased from Aldrich and used without further purification. Anhydride **1d** was obtained by the hydrogenation of **1a**. Anhydrides **1f** and **1g** were obtained from the corresponding diacids which are commercially available. Quinine was purchased from Aldrich and used without further purification. Thin-layer chromatography was carried out on Merck silica gel 60F plates. HPLC analyses were performed on a Varian star series and a Jasco 1100 series instrument equipped with an isostatic pump using a Hypersil Column (250 × 4.6 mm) or a Kromasil column (250 × 4.6 mm). The <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on Varian 300 or Bruker 400 spectrometers and the <sup>13</sup>C CP/MAS solid-state NMR spectra were recorded on a Bruker DSX 400 spectrometer at KBSI in Taegu (Korea). The IR spectra were obtained using a Bruker vertex 70 spectrometer. The HRMS spectra were recorded on a Jeol JMS-700 Mstation. Elemental analysis was performed with a Fisons EA 1108 CHNS-O. The melting points were determined on a Buchi melting point B-540 apparatus and were uncorrected. The optical rotation was measured on a Perkin Elmer 343 plus polarimeter.

## 2. $^1\text{H}$ NMR spectra of catalyst I (before heating and after heating)

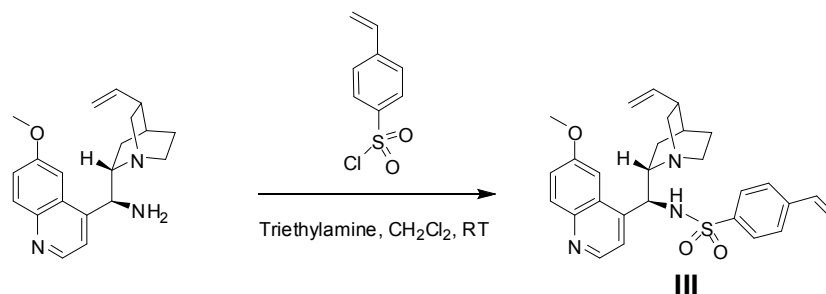
Before heating



**After heating (110°C in chlorobenzene for 12h)**



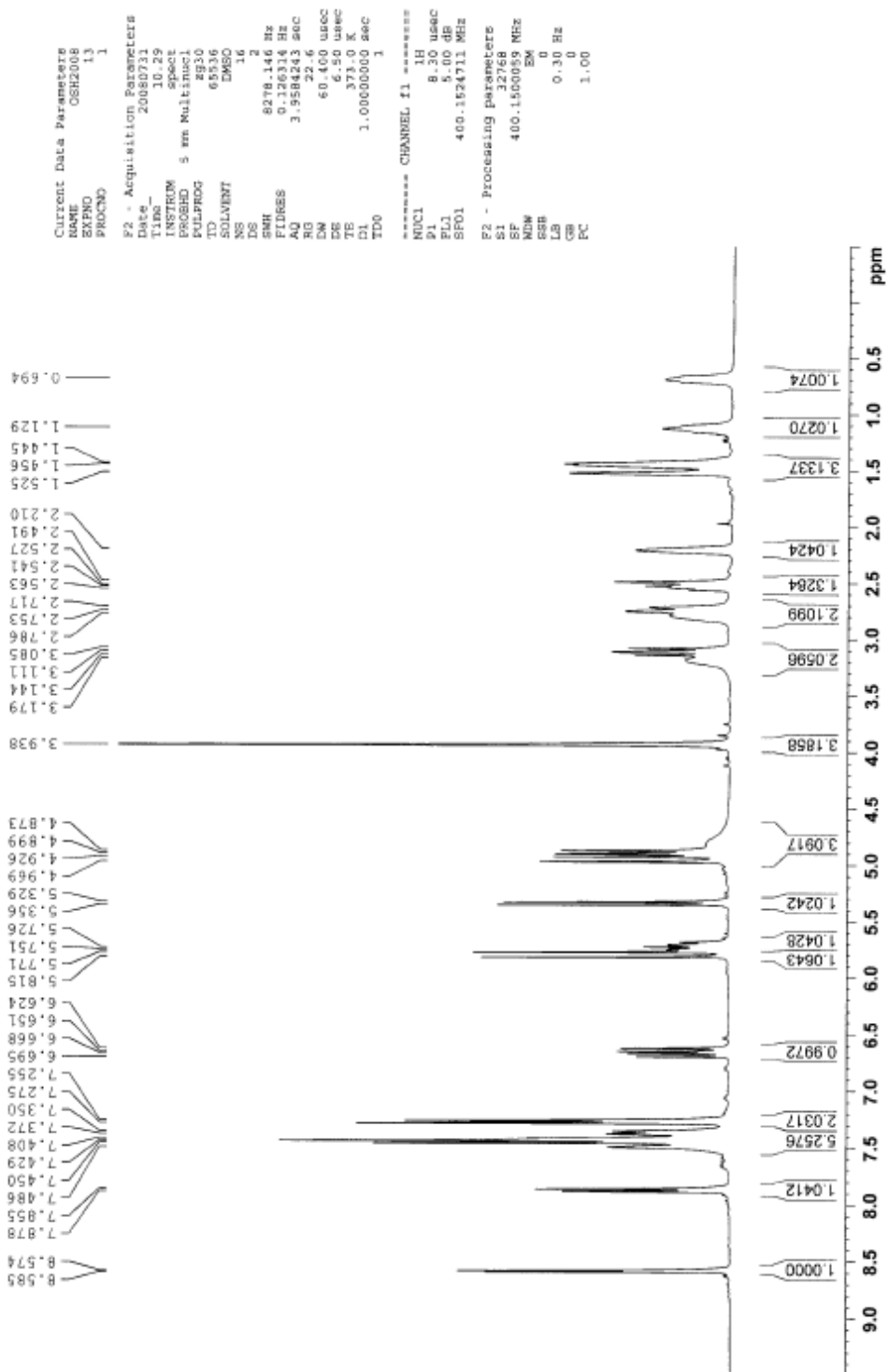
### 3. Preparation and characterization of III



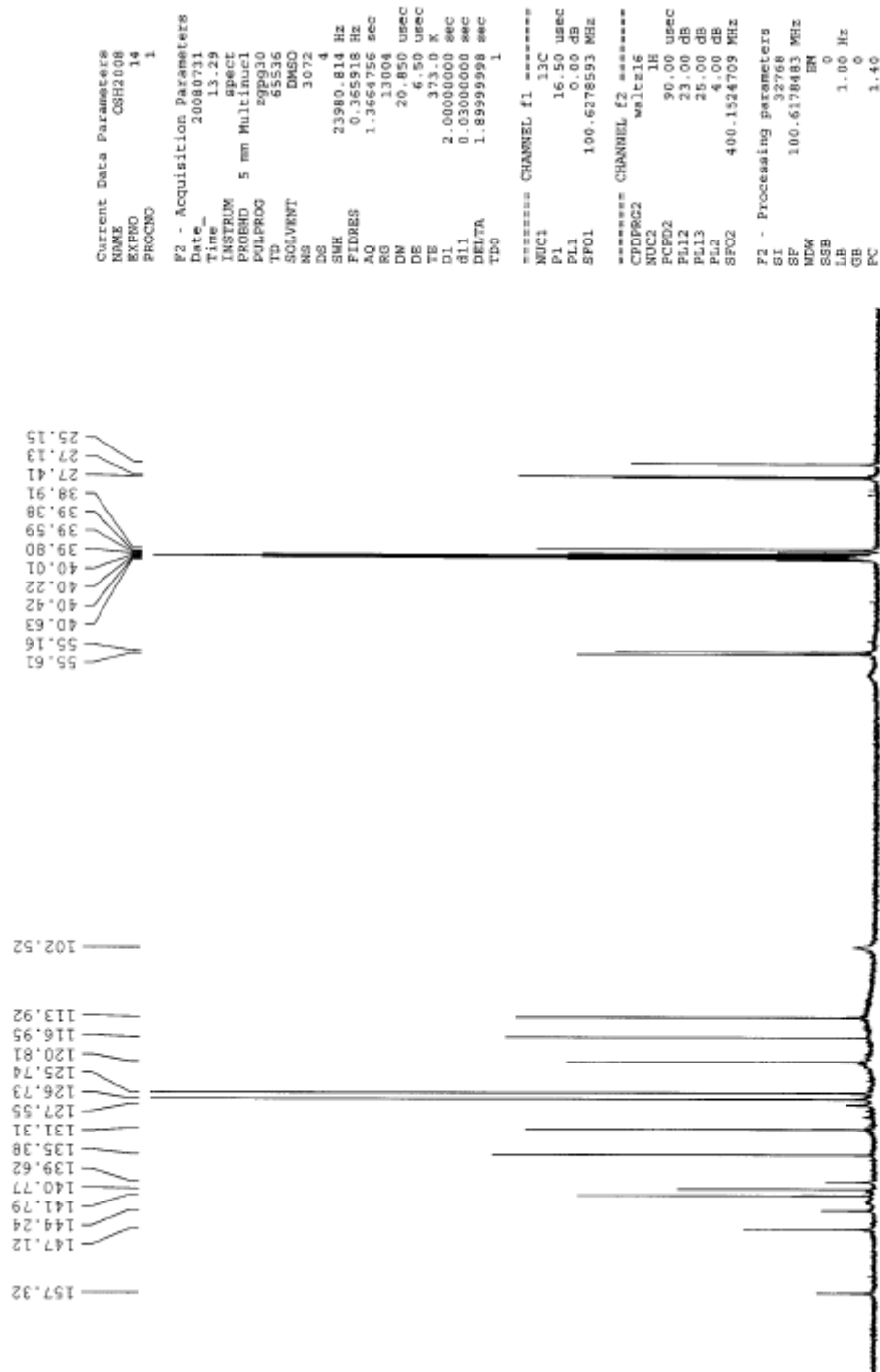
To a solution of 9-amino(9-deoxy)epiquinine<sup>[1]</sup> (9.0 g, 27.8 mmol) and 4-vinylbenzenesulfonyl chloride<sup>[2]</sup> (4.45 g, 21.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (200 mL) was added NEt<sub>3</sub> (3.4 mL, 24.4 mmol) at room temperature. The mixture was stirred at room temperature under an Ar atmosphere for 2h. After the addition of 120 mL of water, the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried with Na<sub>2</sub>SO<sub>4</sub>, filtered and then the filtrate was evaporated *in vacuo*. The residue was purified by column chromatography on a silica gel with EtOAc-MeOH (20 : 1) to afford compound **III** (8.5 g, 79.1% yield) as a white solid. Mp 79.9 °C;  $[\alpha]_D^{20} = +32.19$  ( $c = 1$  in CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, *d*<sub>6</sub>-DMSO, 110 °C)  $\delta$  0.65 – 0.75 (m, 1H), 1.05 – 1.20 (m, 1H), 1.35 – 1.55 (m, 3H), 2.47 – 2.60 (m, 1H, overlapped with DMSO), 2.67 – 2.90 (m, 2H), 3.05 – 3.30 (m, 1H), 3.93 (s, 3H), 4.60 – 5.00 (m, 3H), 5.34 (d,  $J = 10.8$  Hz, 1H), 5.30 – 5.81 (m, 2H), 6.66 (dd,  $J = 17.8$ ,  $J = 10.8$  Hz, 1H), 7.26 (d,  $J = 8$  Hz, 2H), 7.35 – 7.48 (m, 5H), 7.86 (d,  $J = 9.2$  Hz, 1H), 8.58 (d,  $J = 4.4$  Hz, 1H) ppm; <sup>13</sup>C NMR (400 MHz, *d*<sub>6</sub>-DMSO, 110 °C)  $\delta$  25.15, 27.13, 27.41, 55.16, 55.61, 102.52, 113.92, 116.95, 120.81, 125.74, 126.73, 127.55, 131.31, 135.38, 139.62, 140.77, 141.79, 144.24, 147.12, 157.32 ppm; IR(KBr)  $\nu$  3584, 3202, 3074, 2940, 1919, 1620, 1396, 1323, 1228, 1155,

1029  $\text{cm}^{-1}$ ; HRMS (FAB) Calcd for  $[\text{C}_{28}\text{H}_{31}\text{N}_3\text{O}_3\text{S}+\text{H}]^+$  : 490.2164; found: 490.2166; Anal.  
Calcd for  $\text{C}_{28}\text{H}_{31}\text{N}_3\text{O}_3\text{S}$ : C, 68.68; H, 6.38; N, 8.58; S, 6.55. Found: C, 68.60; H, 6.38; N,  
8.39; S, 6.56.

# <sup>1</sup>H-NMR spectrum of catalyst III (*d*<sub>6</sub>-DMSO, 110 °C)

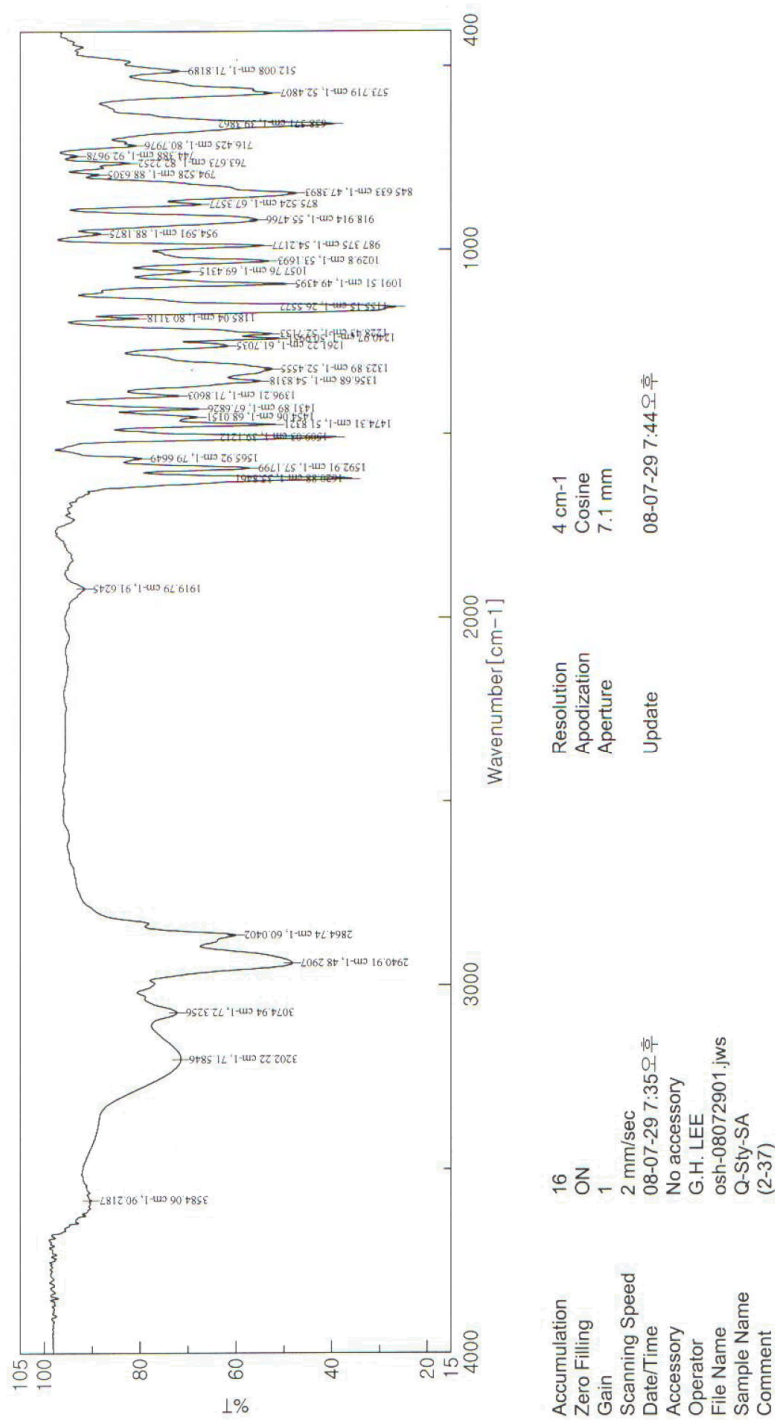


### <sup>13</sup>C-NMR spectrum of catalyst III (*d*<sub>6</sub>-DMSO, 110 °C)

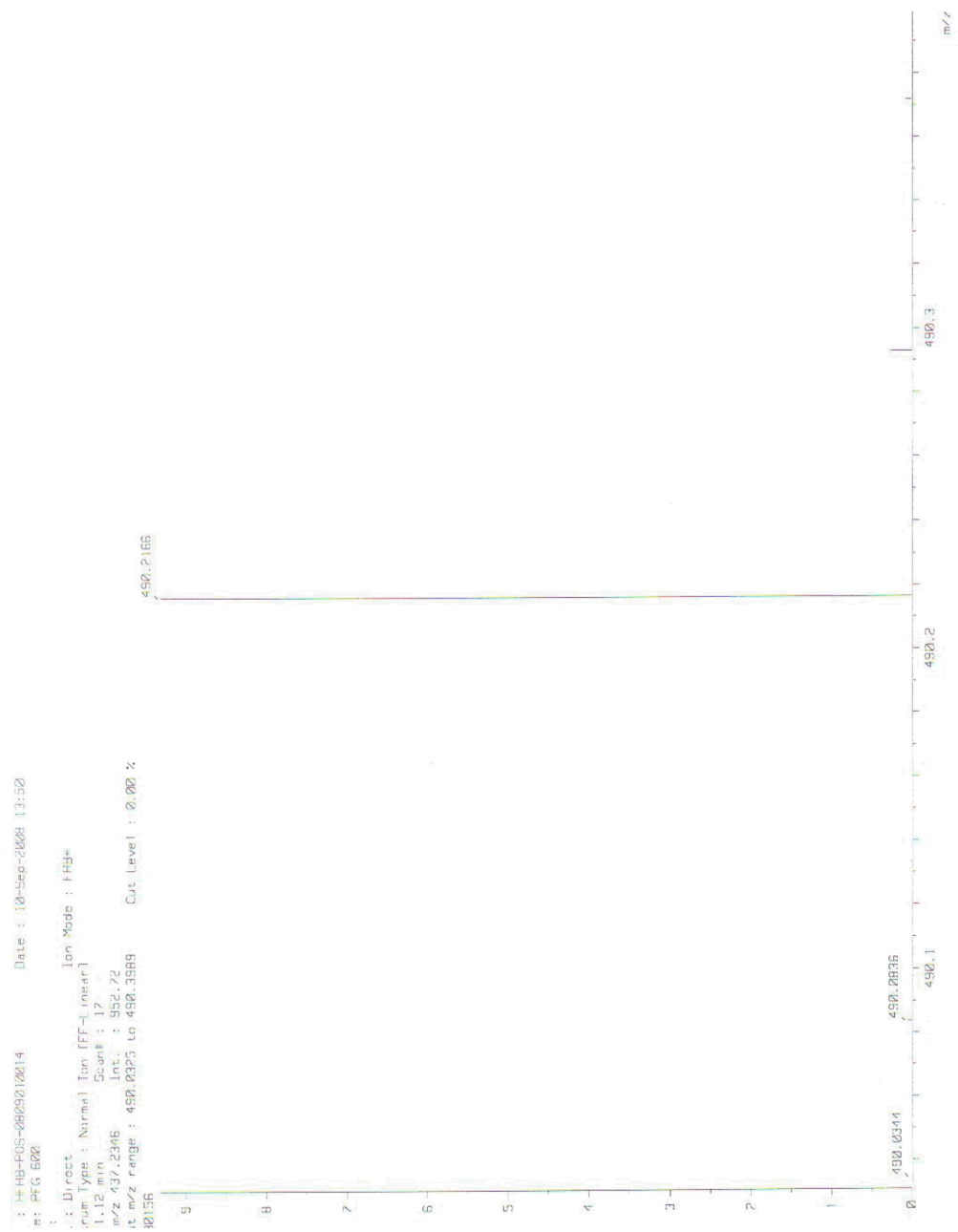




### IR spectrum of catalyst III (KBr)



### HRMS spectrum of catalyst III

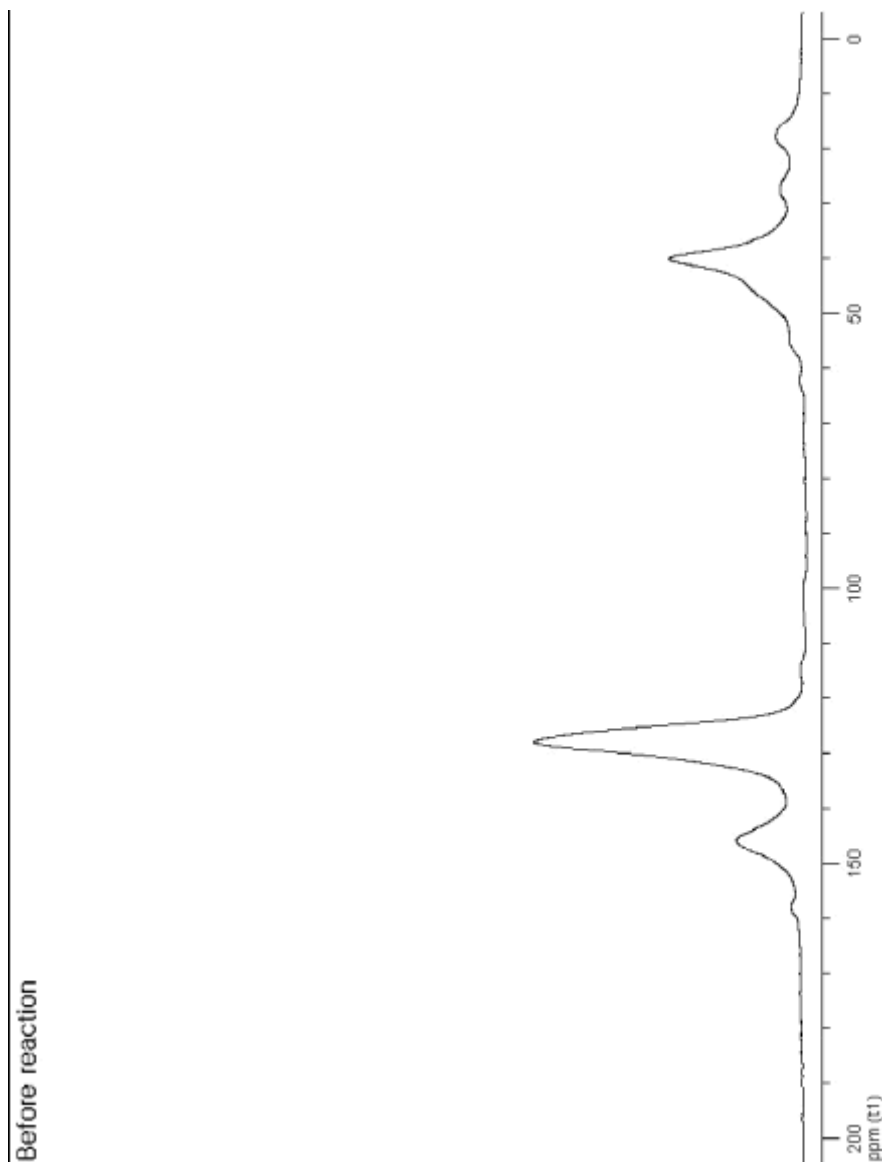


#### 4. Preparation and characterization of catalyst IV

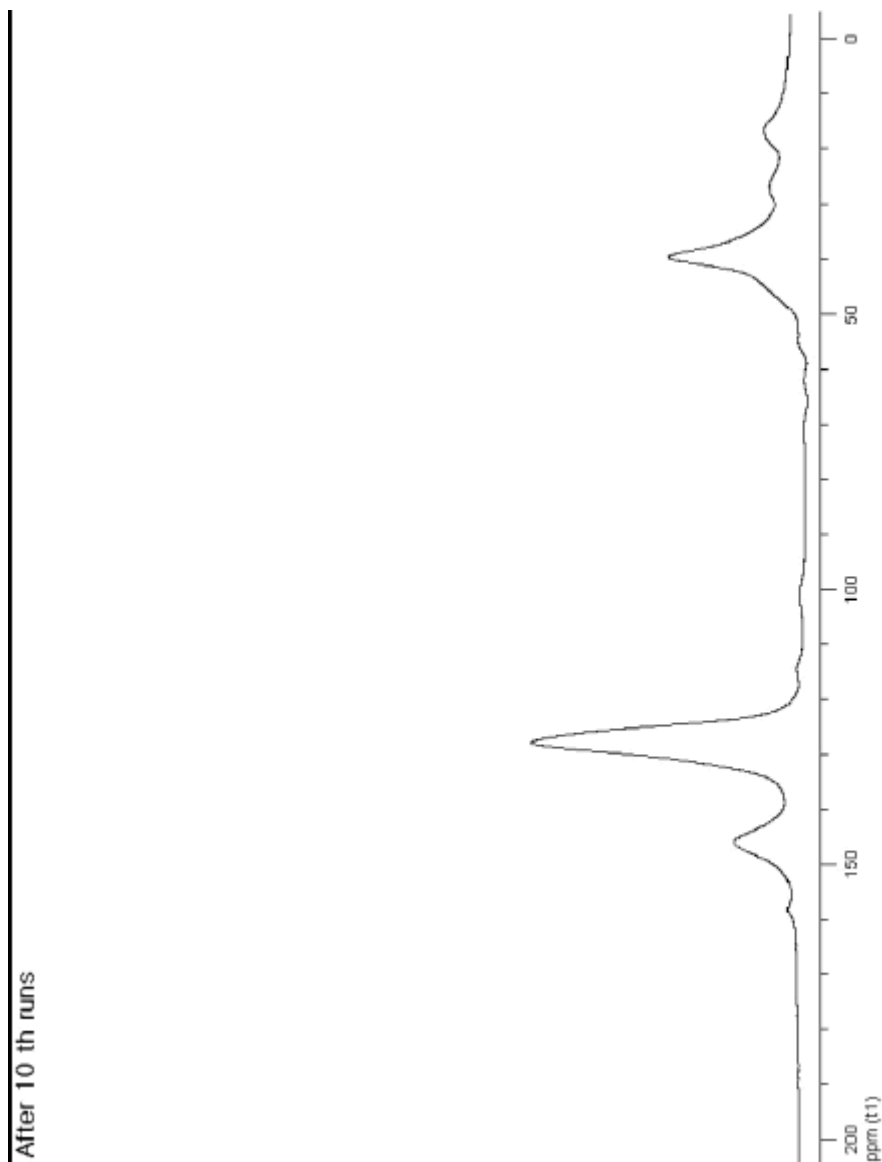
The stabiliser (t-butyl catechol) was first removed from styrene and commercial 1,4-divinylbenzene by washing samples with an excess of 1 per cent aqueous sodium hydroxide and twice with water. To a solution of water (112 mL), acacia gum (5.7 g) and sodium chloride (3.6 g), was added a styrene (67.8 mmol, 7.8 mL), 1,4-divinyl benzene (1.3 mmol, 195  $\mu$ L), azobisisobutyronitrile (1.3 mmol, 213 mg), catalyst **III** (3.4 mmol, 1.66 g) and chlorobenzene (12.6 mL). The reaction mixture was deoxygenated by purging with argon atmosphere for at least 30 min at room temperature. Then, the reaction vessel was heated to 110  $^{\circ}$ C and stirred for 12 hours under Ar atmosphere. After cooling to room temperature, the reaction mixture was filtered and washed with water. The polymeric residue was suspended in 300 mL of water and stirred for at least 1 hour and then filtered. The filtered residue was suspended in 400 mL of methanol and stirred for at least 1 hour and then filtered. After washing successively with methanol, dichloromethane and diethyl ether, the polymer was dried overnight in *vacuo* to afford 6 g of catalyst **IV** as a white solid. IR (KBr)  $\nu$  3399, 3081, 3001, 2920, 1621, 1600, 1356, 1240, 1154, 1028  $\text{cm}^{-1}$ ; Anal. Calcd for : N, 2.16; S, 1.66. Found: N, 2.01; S, 1.31.

**$^{13}\text{C}$  CP/MAS solid-state NMR (400 MHz) spectrum of catalyst IV**

**Before reaction**

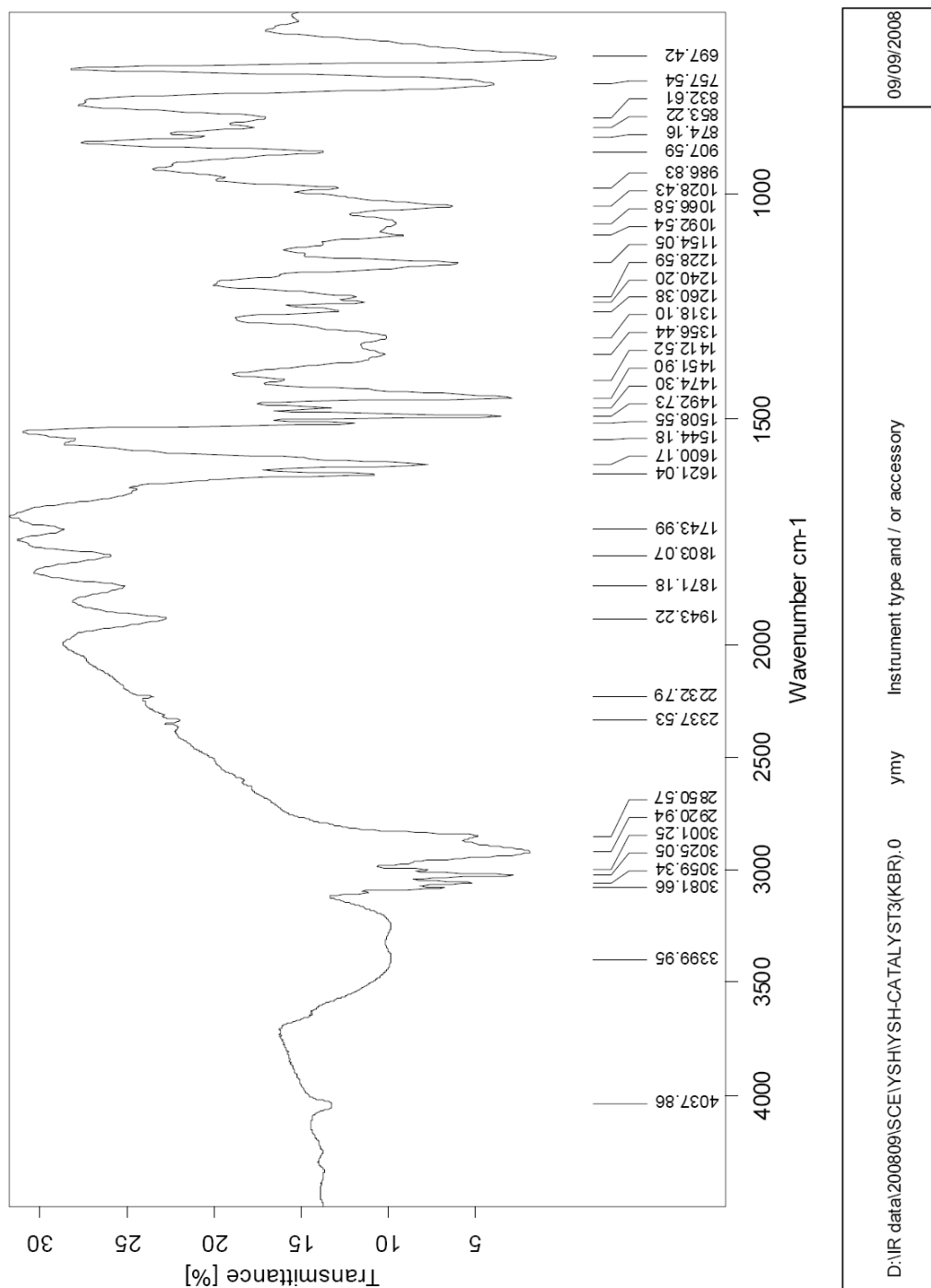


### After 10 runs

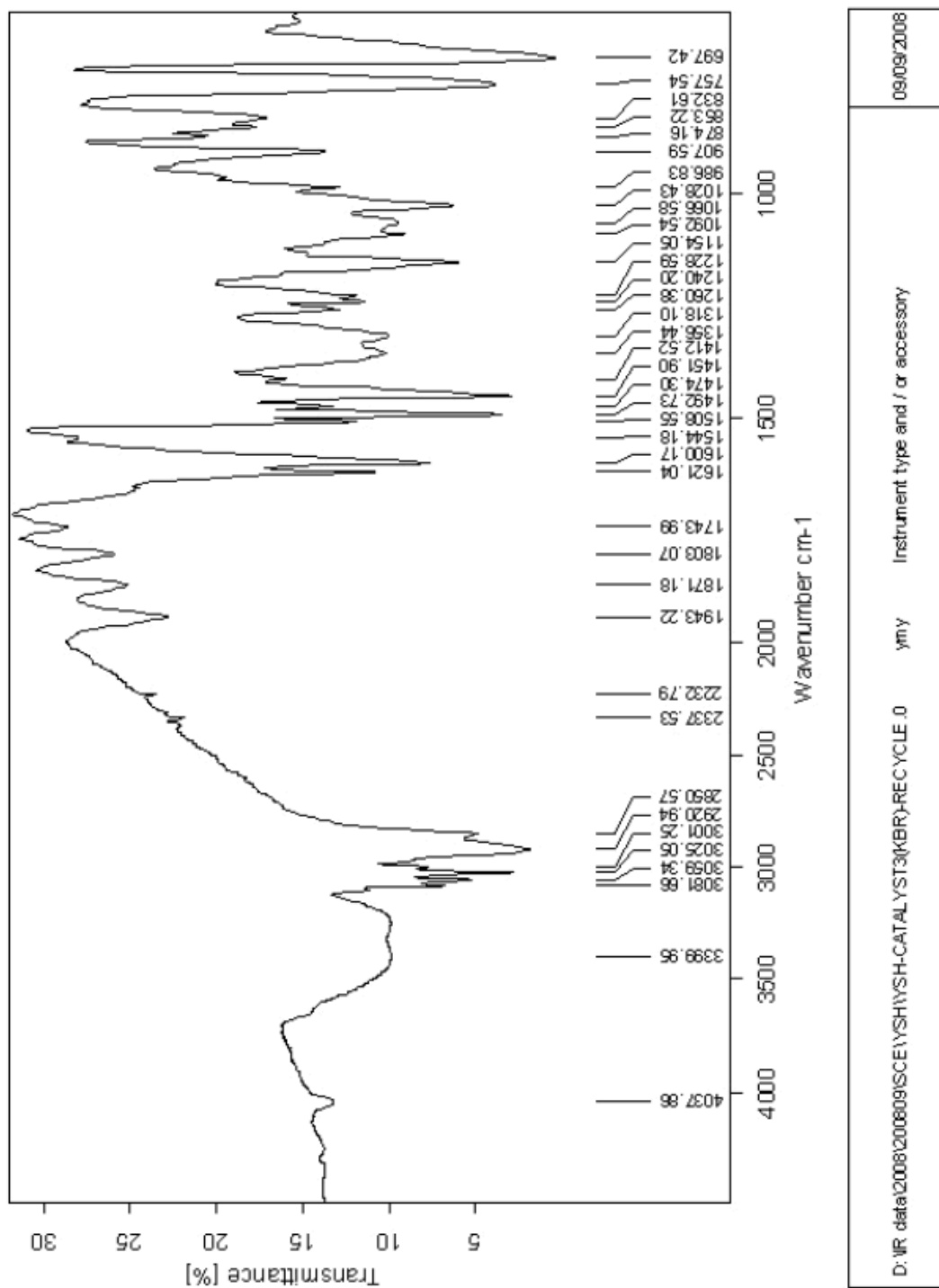


## IR spectrum of catalyst IV (KBr)

Before reaction



After 10 runs

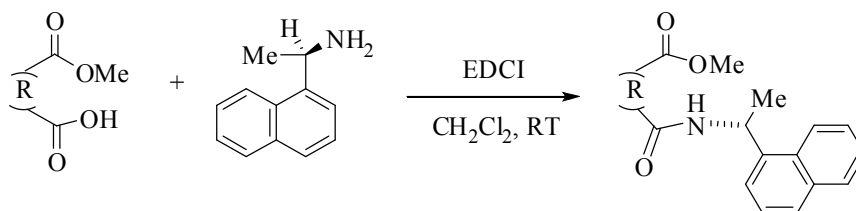


## 5. General procedure for methanolysis of prochiral cyclic anhydrides

Methanol (202  $\mu\text{L}$ , 5 mmol) was added dropwise to a stirred solution of anhydrides (0.5 mmol) and catalysts (10 mol %) in appropriate solvents (10 mL) at the temperature indicated in Scheme 2, Table 1. The reaction mixture was stirred at that temperature until the starting material was consumed, as indicated by TLC analysis. After filtration of resin, the filtrate was concentrated *in vacuo* to yield the pure hemiester products (**2a–2g**) in nearly quantitative yields.

The enantiomeric excess (*ee*) of each product was determined by HPLC analysis of a diastereomeric mixture of the corresponding amide-ester prepared from the hemiester according to the literature procedure<sup>[3]</sup> (Scheme 1).

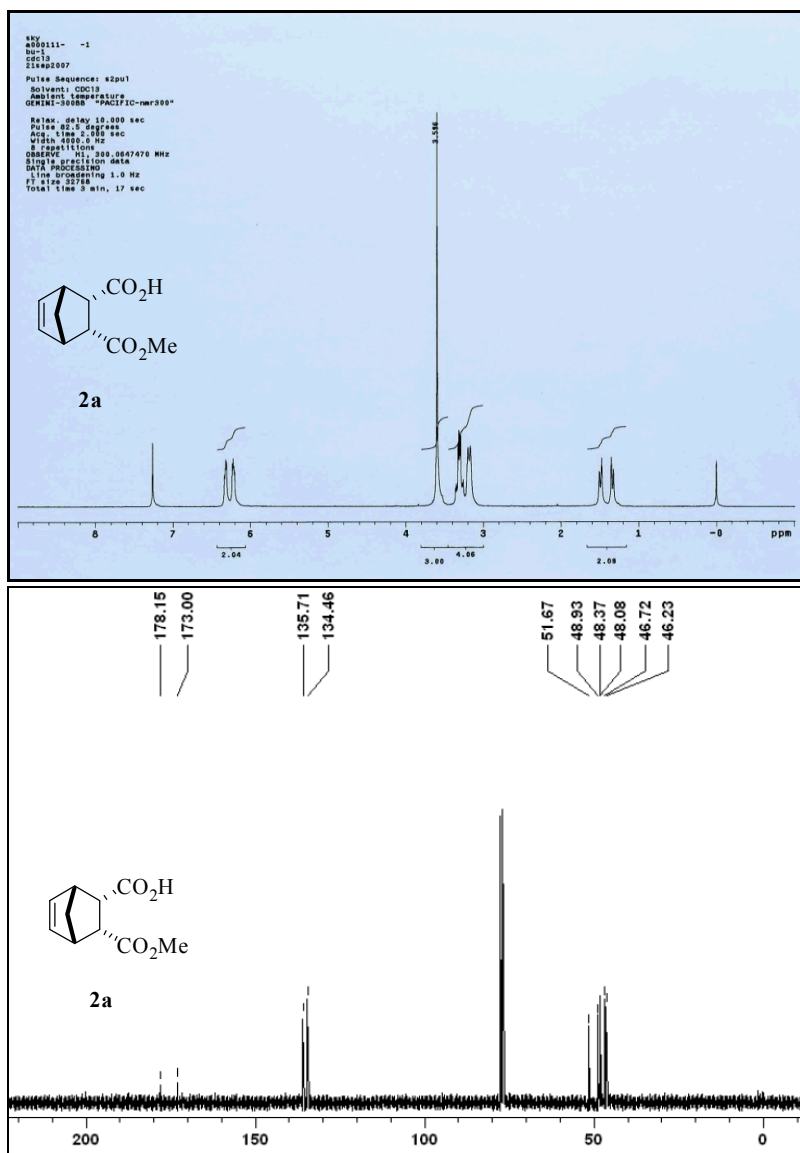
Scheme 1





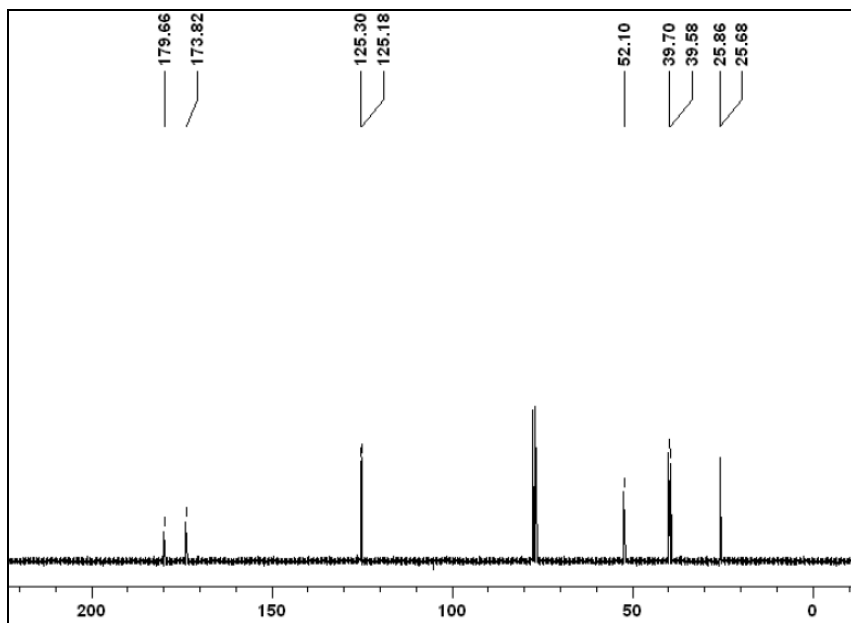
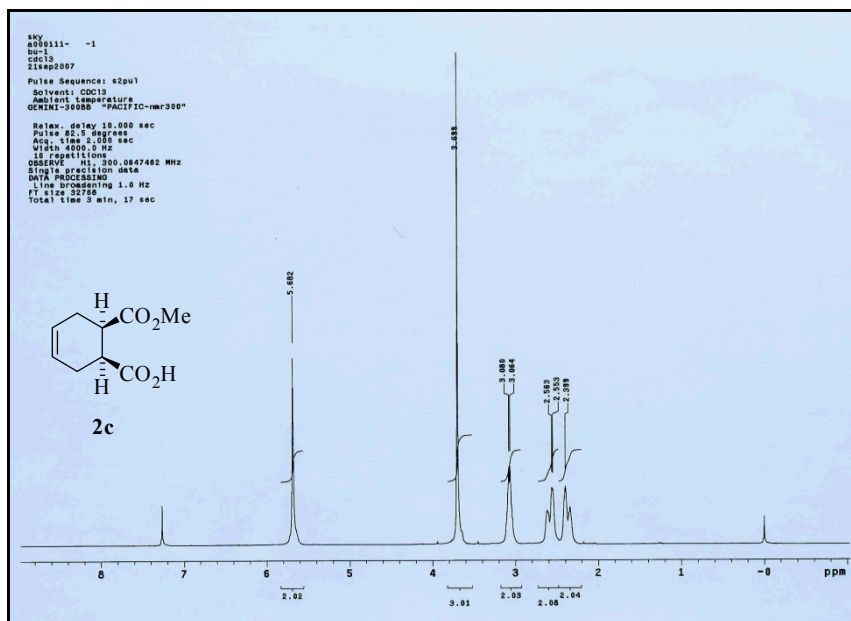
## 6. $^1\text{H}$ and $^{13}\text{C}$ NMR spectra of 2a-2g

### NMR Spectra of 2a



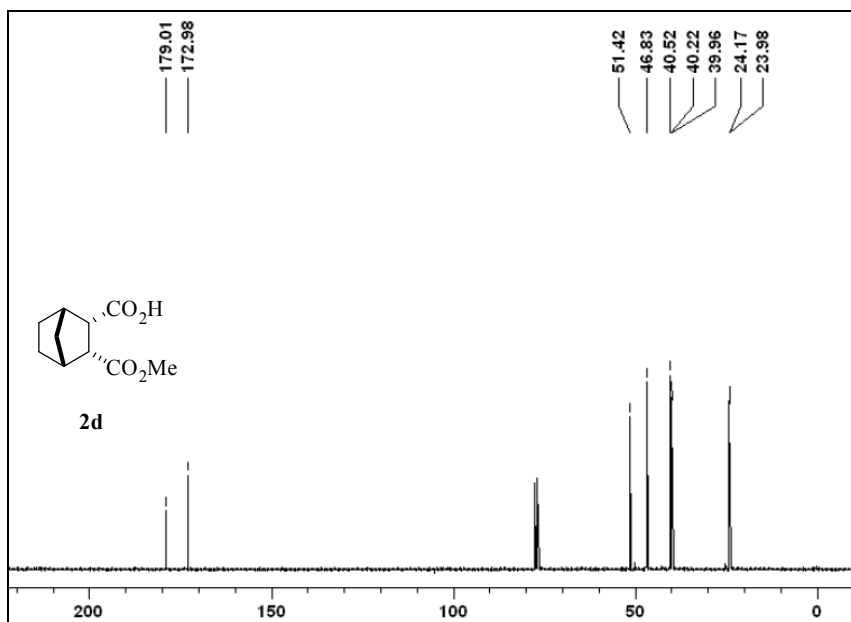
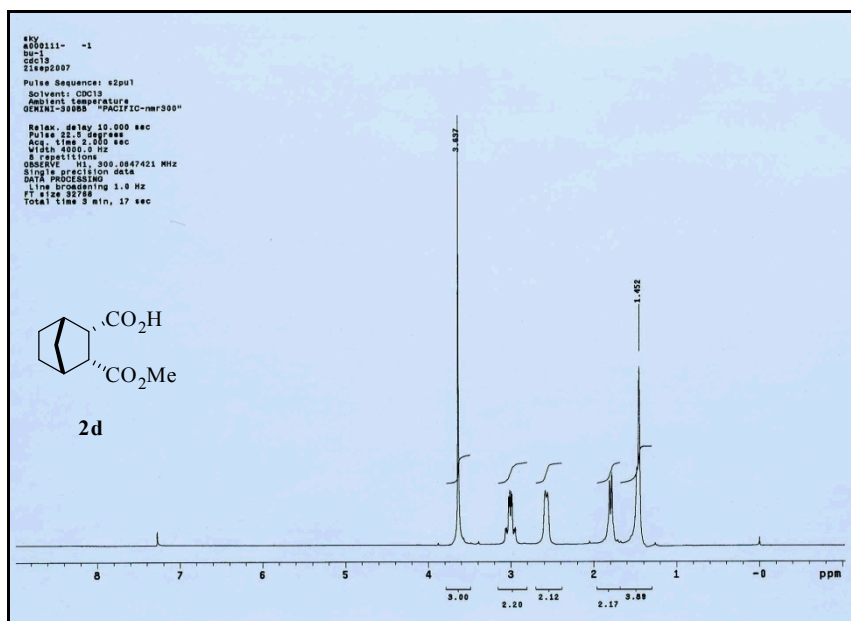


## NMR Spectra of 2c



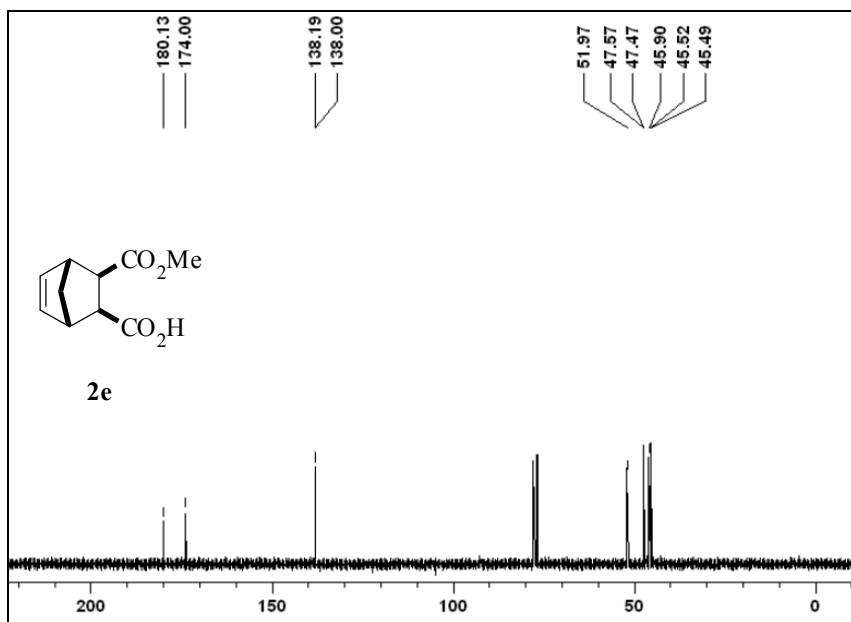
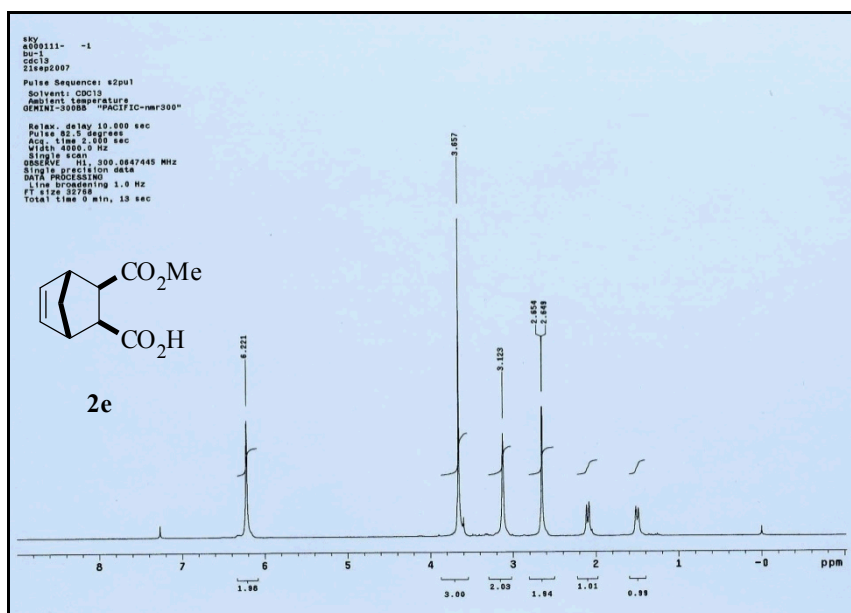
$^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  2.32-2.65 (m, 4H), 3.02-3.12 (m, 2H), 3.69 (s, 3H), 5.68 (m, 2H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  25.68, 25.86, 39.58, 39.70, 52.10, 125.18, 125.30, 173.82, 179.66.

## NMR spectra of 2d



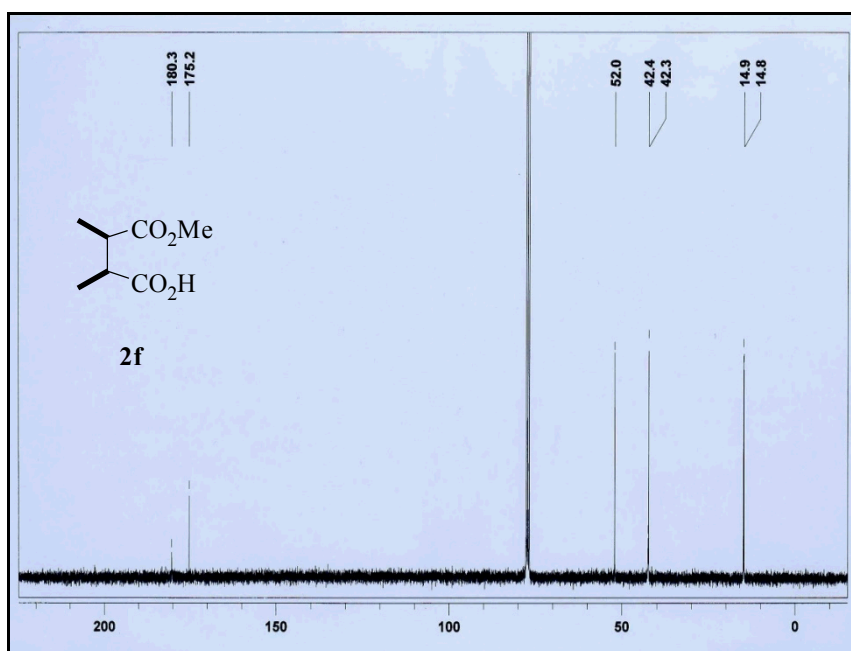
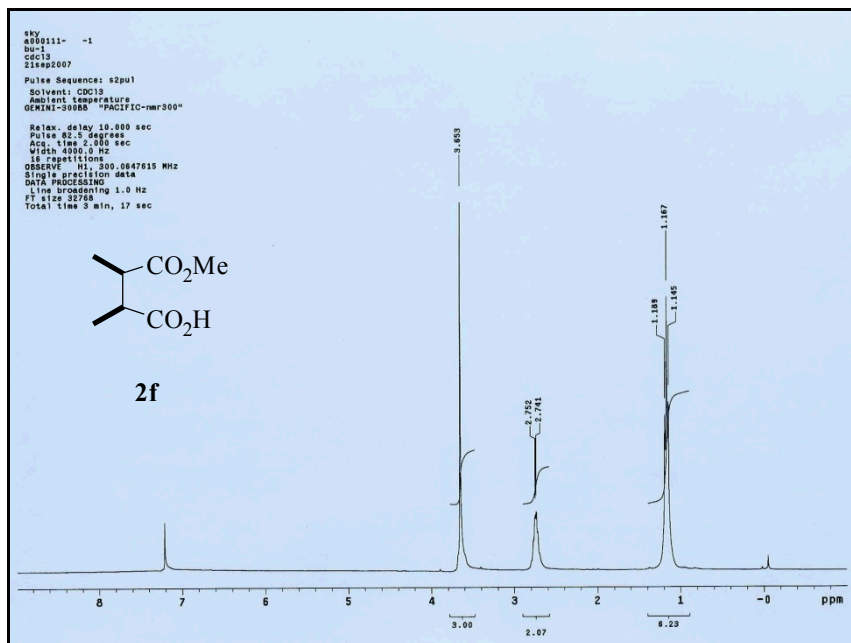
<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 1.35-1.58 (m, 4H), 1.75-1.83 (m, 2H), 2.57-2.62 (m, 2H), 2.82-3.04 (m, 2H), 3.69 (s, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 23.98, 24.17, 39.96, 40.22, 40.52, 46.83, 51.42, 172.98, 179.01.

## NMR Spectra of 2e



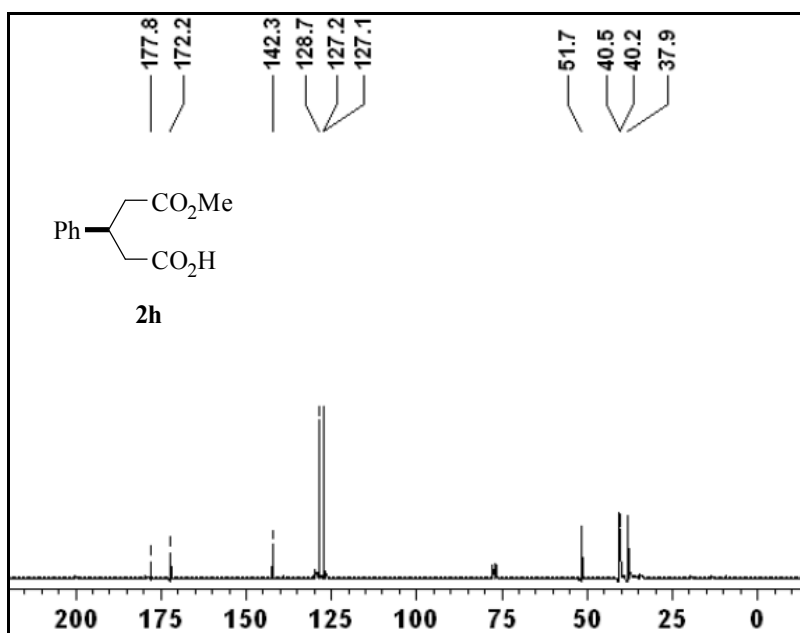
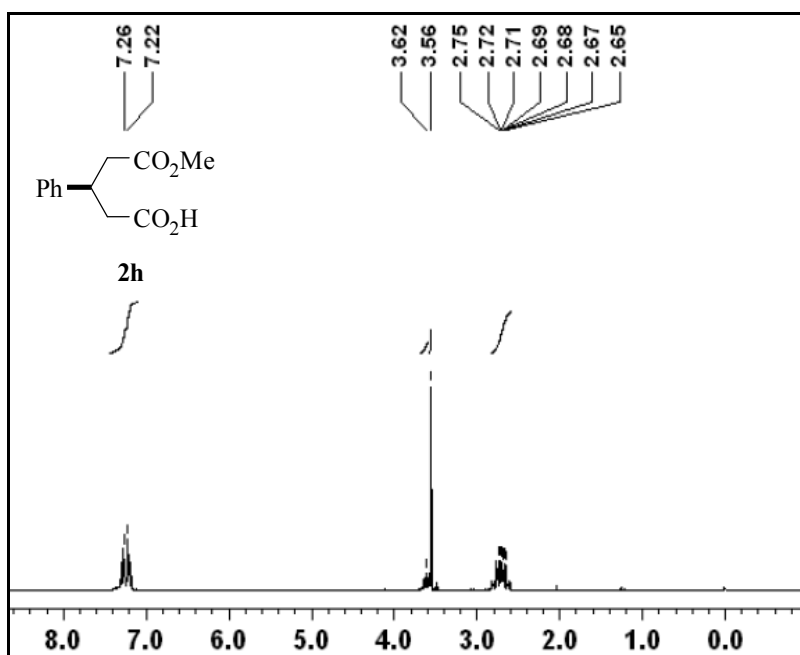
$^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.42 (bd,  $J = 9.0$  Hz, 1H), 2.13 (bd,  $J = 9.0$  Hz, 1H), 2.62 (m, 2H), 3.09 (m, 2H), 3.65 (s, 3H), 6.22 (m, 2H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  45.49, 45.52, 45.90, 47.47, 47.57, 51.97, 138.00, 138.19, 174.00, 180.13

## NMR Spectra of 2f



$^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ):  $\delta$  1.02 – 1.21 (m, 6H), 2.71 – 2.79 (m, 2H), 3.66(s, 3H);  $^{13}\text{C}$   
NMR (75 MHz,  $\text{CDCl}_3$ ):  $\delta$  14.8, 14.9, 42.3, 42.4, 52.0, 175.2, 180.3

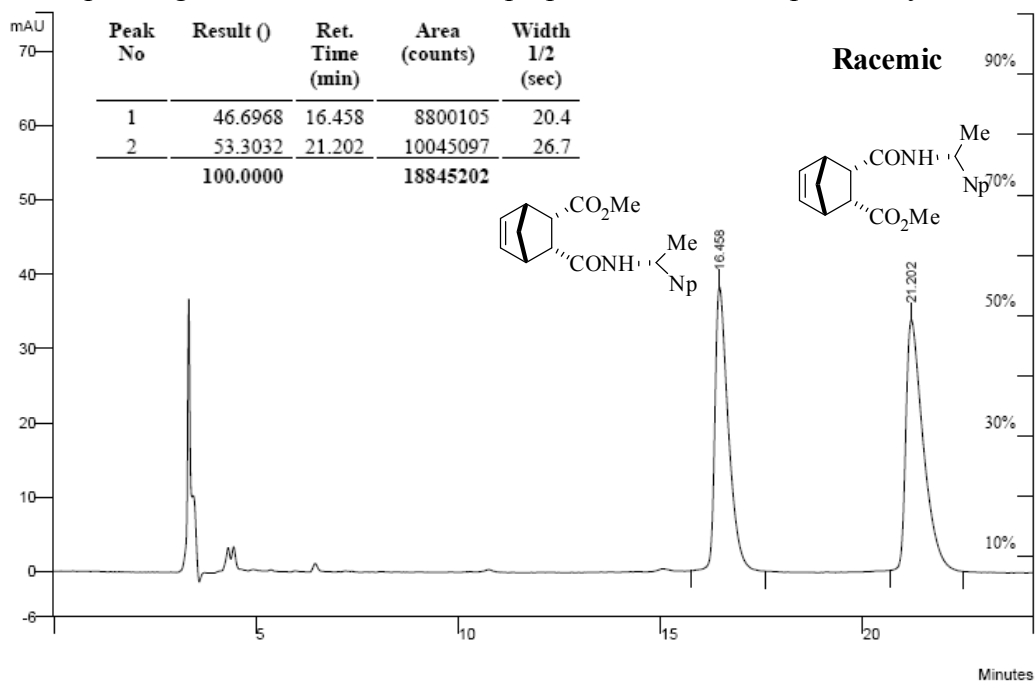
## NMR Spectra of 2g



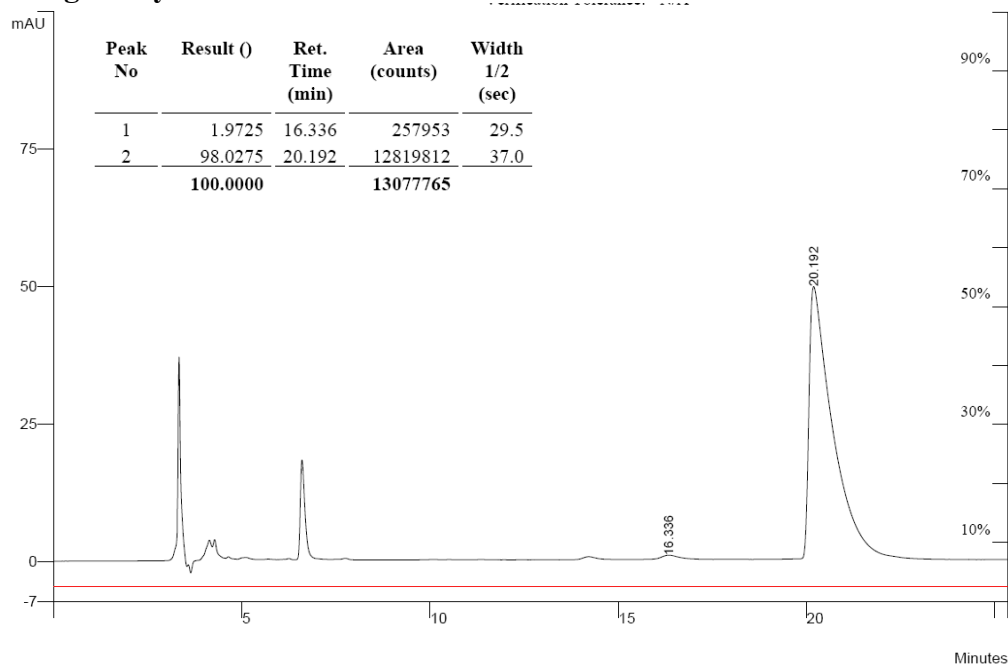
<sup>1</sup>H NMR (300MHz, CDCl<sub>3</sub>): δ 2.6 – 2.8 (m, 4H), 3.56 (s, 3H), 3.58 – 3.67 (m, 1H), 7.10–7.35 (m, 5H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 37.9, 40.2, 40.5, 51.7, 127.1, 127.2, 128.7, 142.3, 172.2, 177.8.

## 7. HPLC spectra for Scheme 2.

The ee value was determined by the HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 16.74 min, t(major) = 21.05 min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

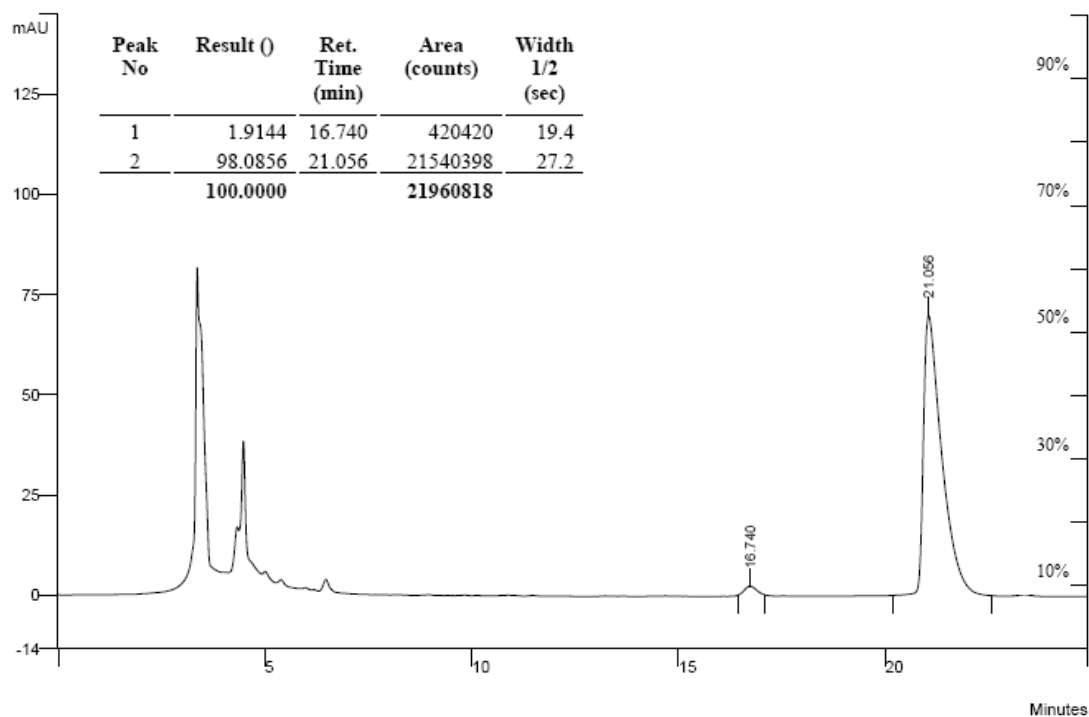


### Using catalyst III

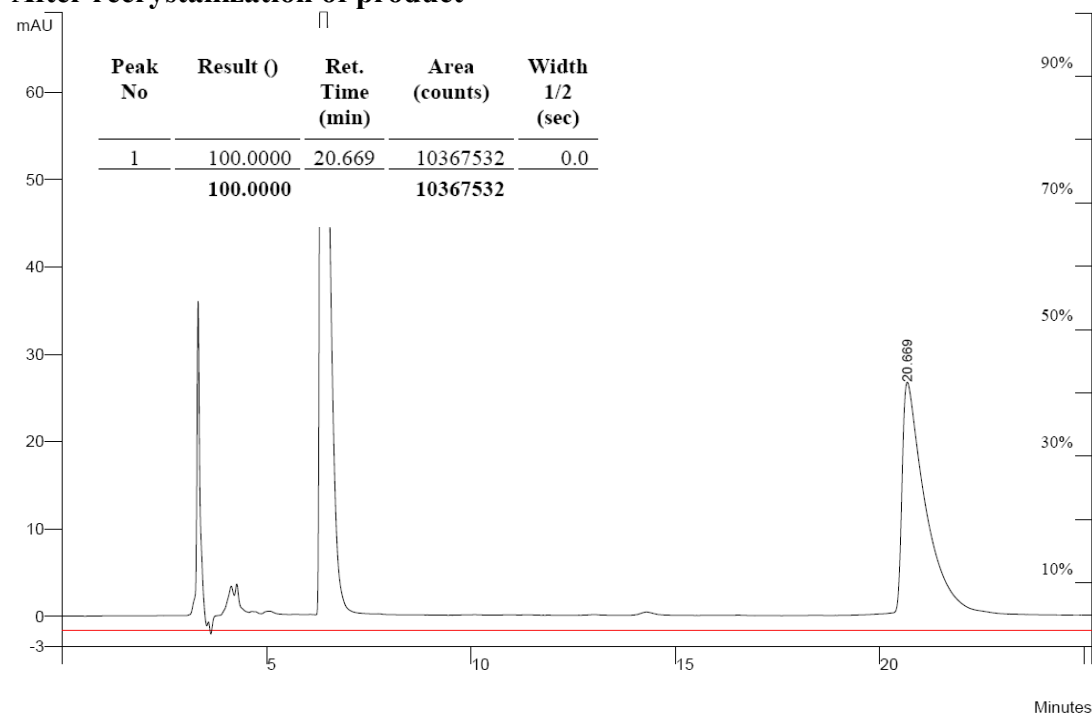




### Using catalyst IV



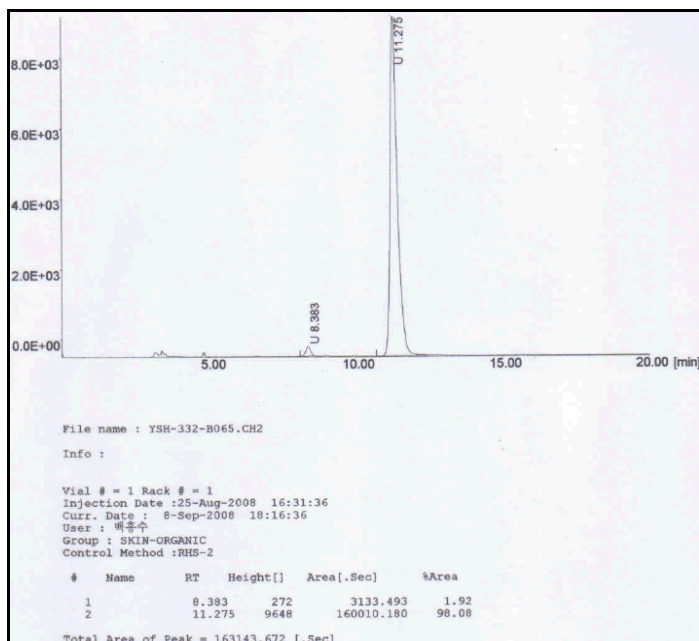
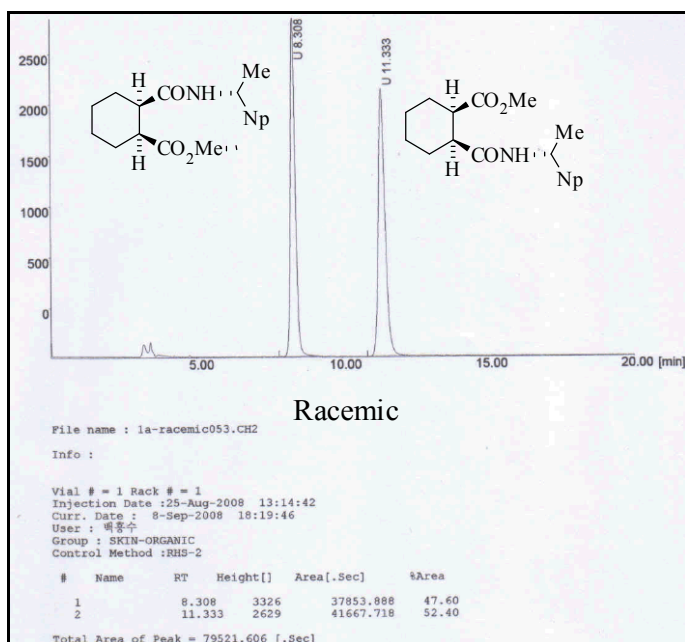
### After recrystallization of product



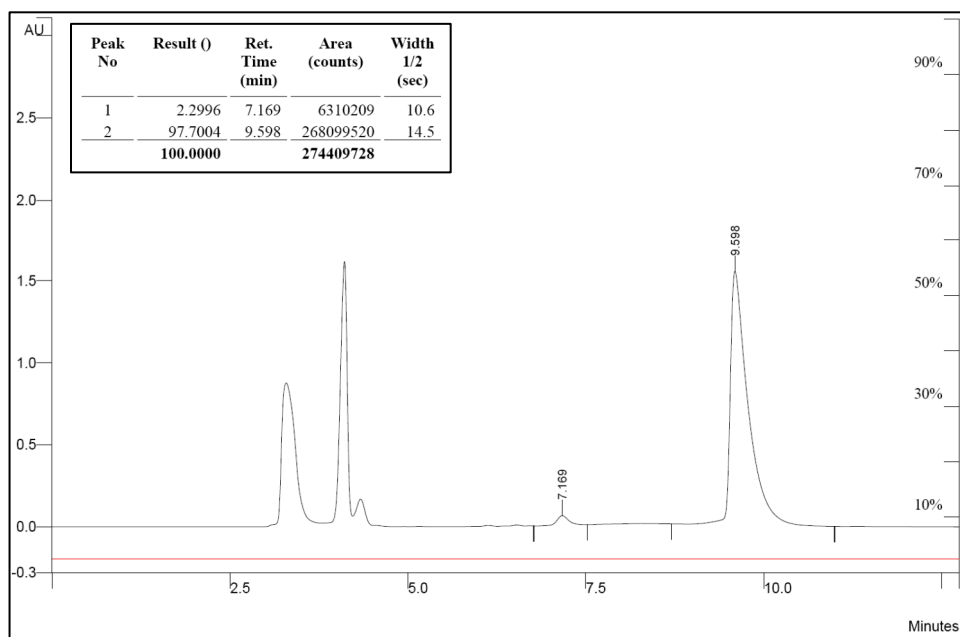
## 8. HPLC spectra for Table 1.

### Table 1 - Entry 1.

The ee value was determined by the HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 8.30 min, t(major) = 11.33 min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

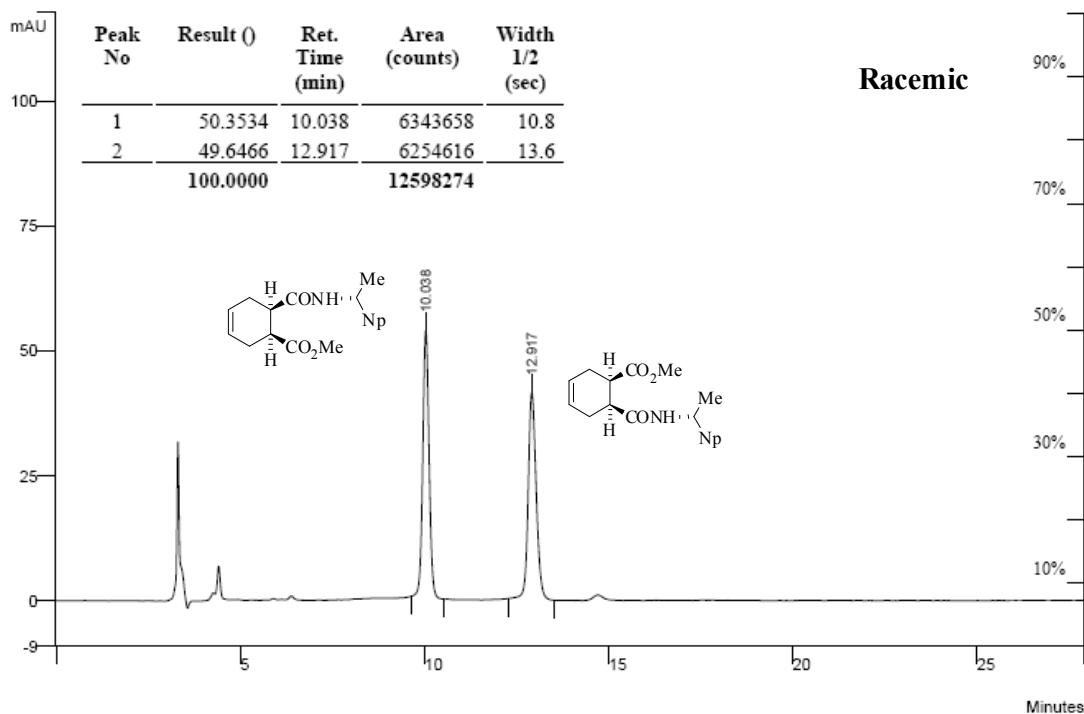


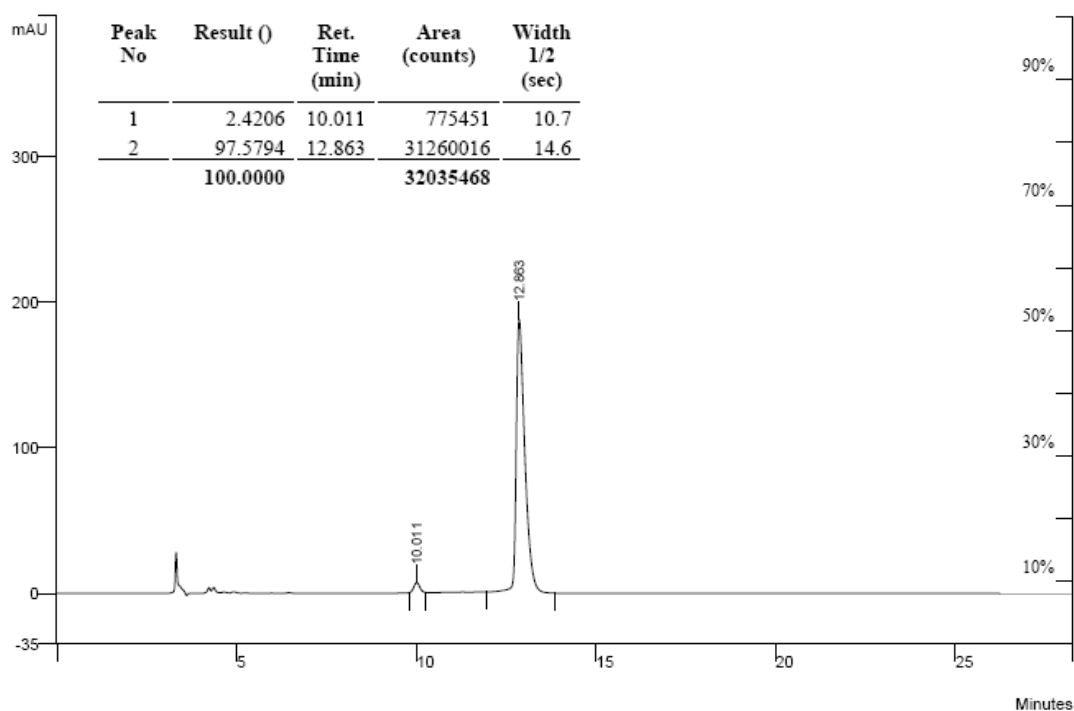
**Table 1 - Entry 2.**



**Table 1 - Entry 3.**

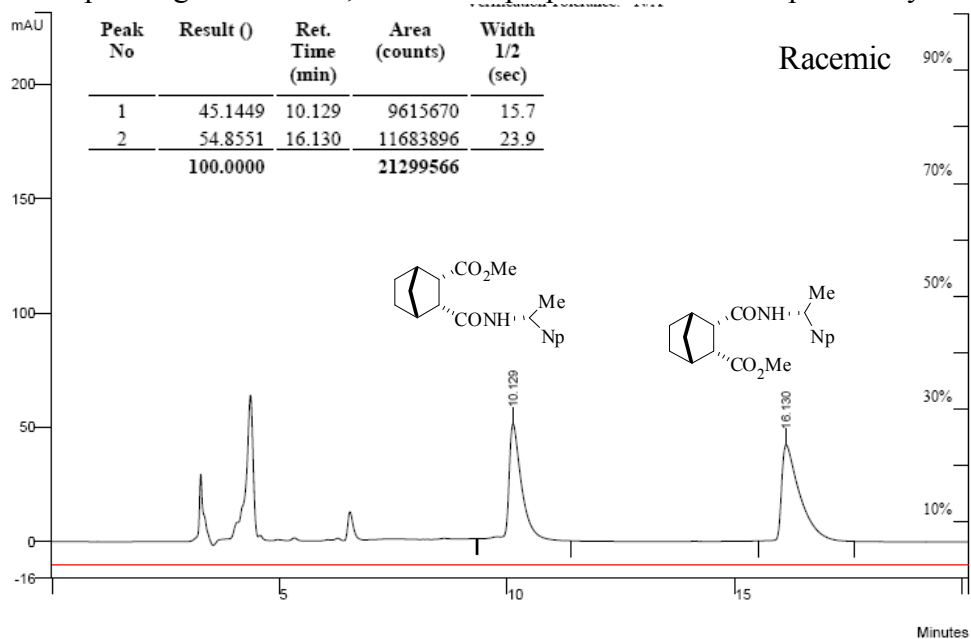
The ee value was determined by the HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 10.01 min, t(major) = 12.86 min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

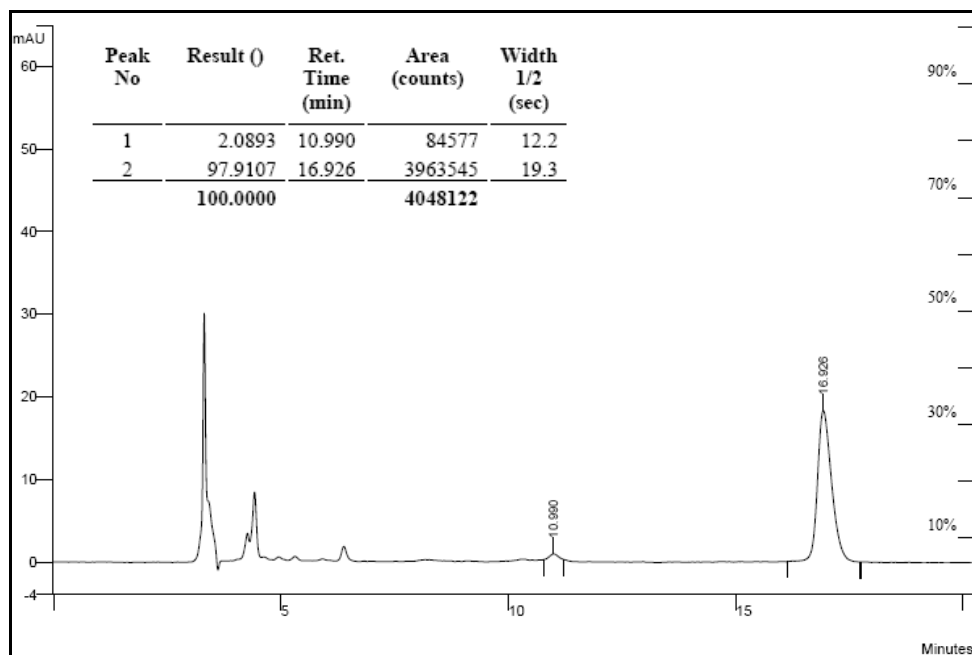




**Table 1 - Entry 4.**

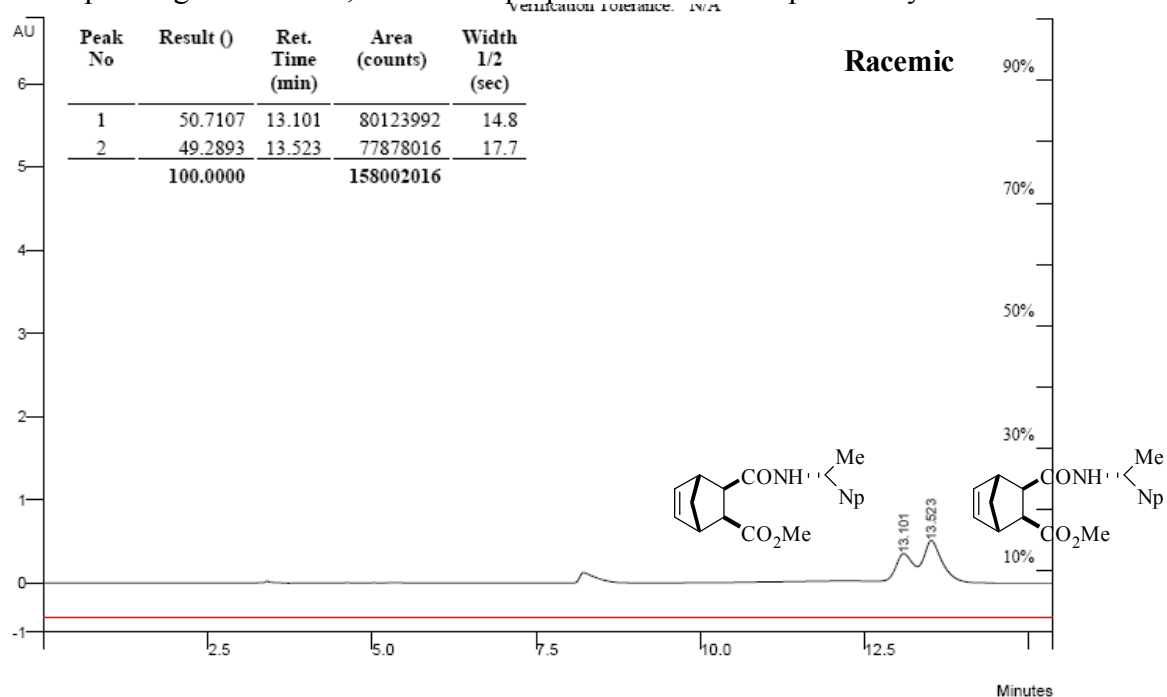
The ee value was determined by the HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 10.99 min, t(major) = 16.92 min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

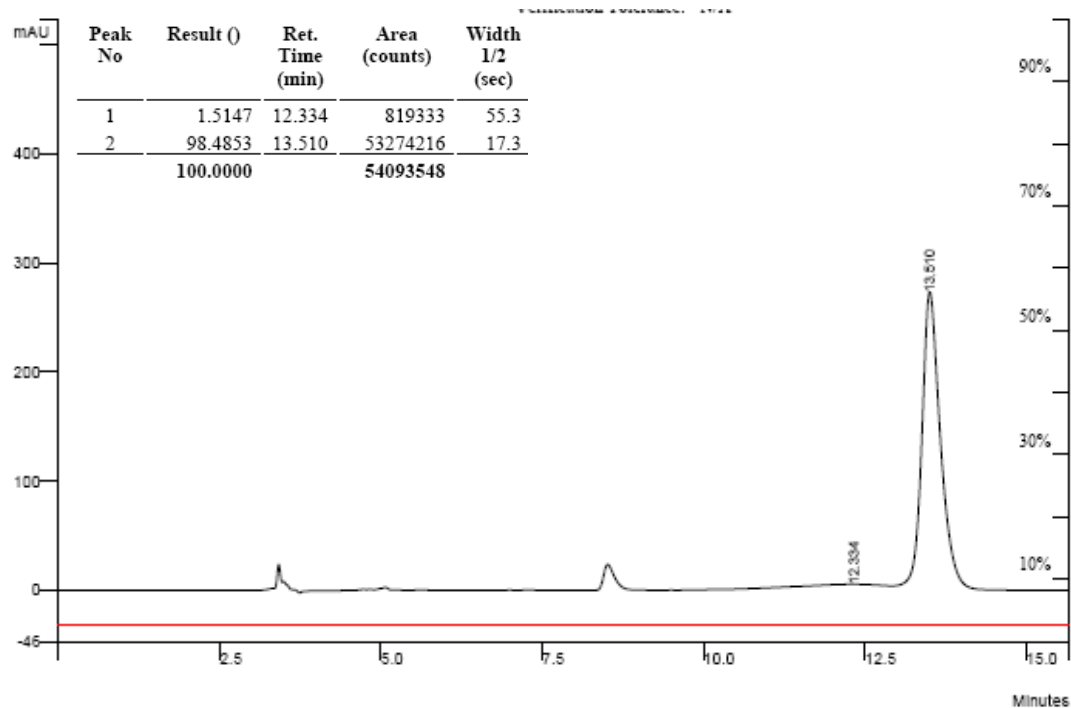




### Table 1 - Entry 5.

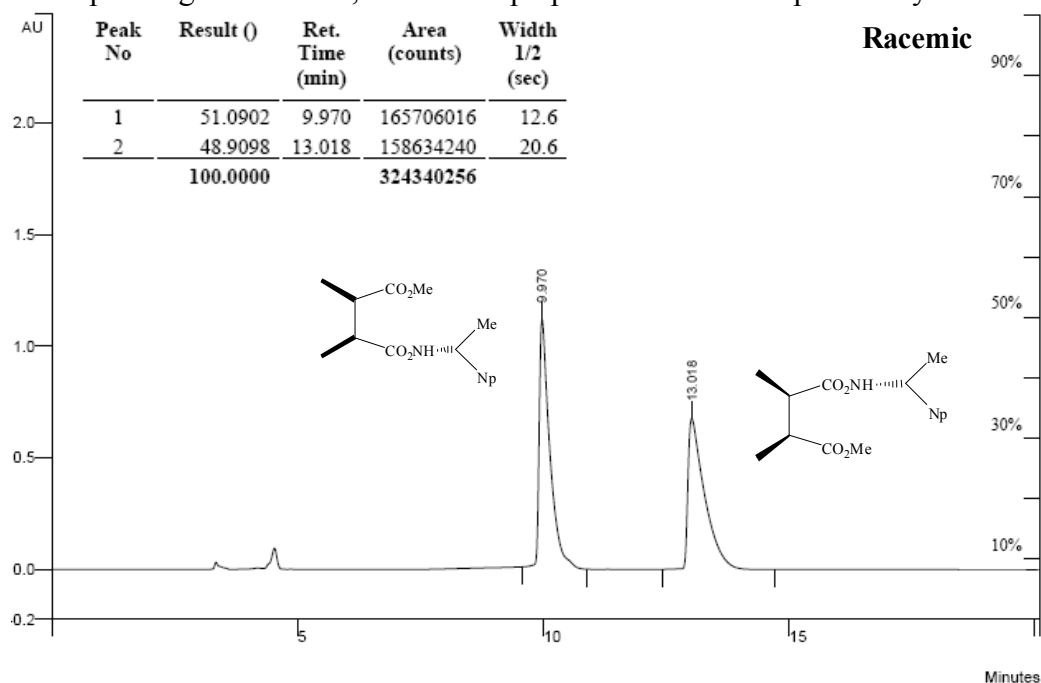
The ee value was determined by the HPLC analysis (Kromasil, 40 : 1, Hexanes : IPA, 1 mL/min, t(minor) = 13.73 min, t(major) = 15.17 min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

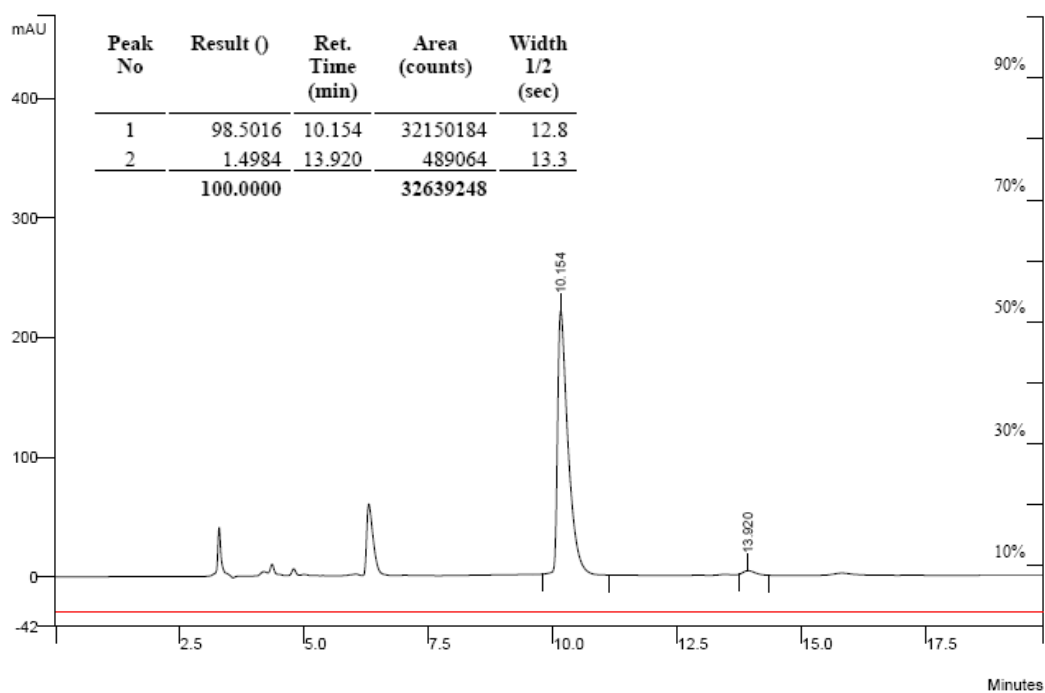




**Table 1 - Entry 6.**

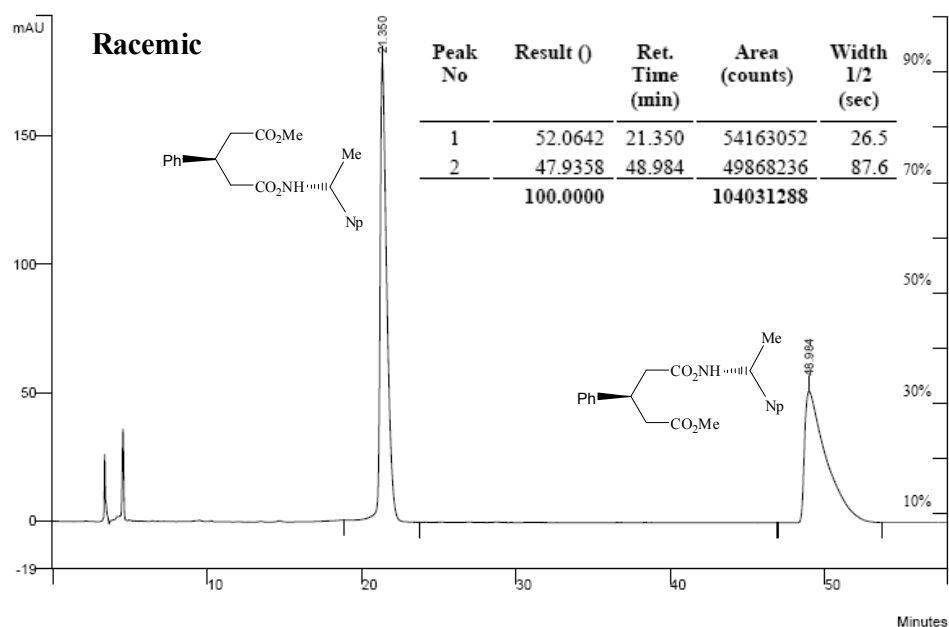
The ee value was determined by the HPLC analysis (Hypersil, 40 : 1, Hexanes : IPA, 1 mL/min,  $t(\text{major}) = 10.15$  min,  $t(\text{minor}) = 13.92$  min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.

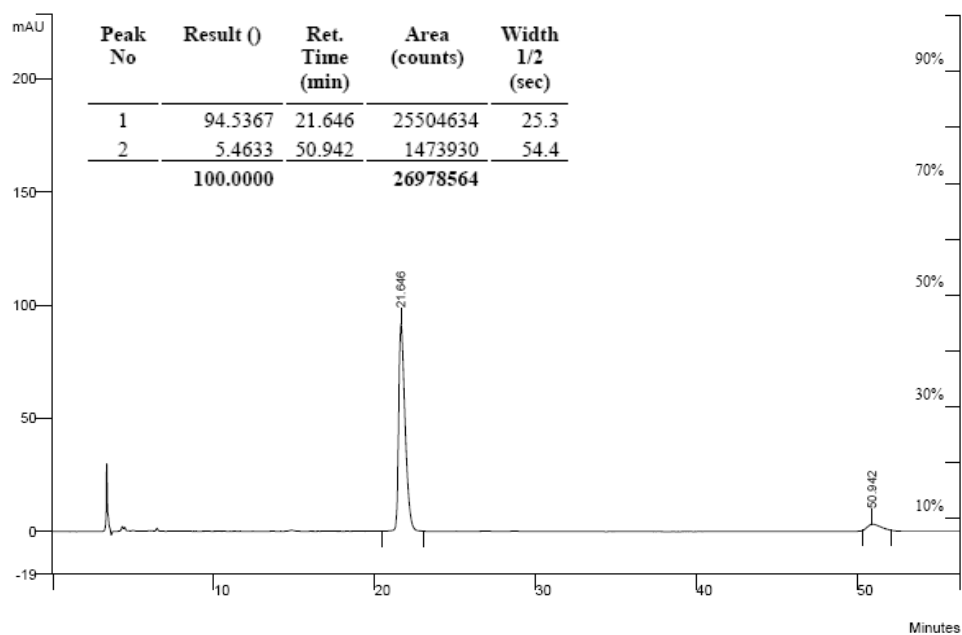




**Table 1 - Entry 7.**

The ee value was determined by the HPLC analysis (Hypersil, 20 : 1, Hexanes : IPA, 1 mL/min,  $t(\text{major}) = 21.64$  min,  $t(\text{minor}) = 50.94$  min) of the diastereomeric mixture of the corresponding amide-ester, which was prepared as described previously.





## 9. References.

1. B. Vakulya, S. Varga, A. Csampai and T. Soos, *Org. Lett.*, 2005, **7**, 1967.
2. D. J. Berrisford, P. A. Lovell, N. R. Suliman and A. Whiting, *Chem. Commun.*, 2005, 5904.
3. Y. M. Song, J. S. Choi, J. W. Yang and H. Han, *Tetrahedron Lett.*, 2004, **45**, 3301.