Experimental

General techniques: All reactions requiring anhydrous conditions were conducted in flame-dried glass apparatus under an atmosphere of N₂. THF was freshly distilled from sodium benzophenone ketyl prior to use. Preparative chromatographic separations were performed on Fluka silica gel 60 (35-70 μ m, 220-440 mesh) and reactions followed by thin layer chromatography (TLC) analysis using Merck aluminum-backed kieselgel 60 plates (2-25 μ m) with fluorescent indicator (254 nm) and visualised with UV, iodine, phosphomolybdic acid, basic KMnO₄ or ninhydrin. All commercially available reagents were purchased from Aldrich and were used as received unless otherwise noted. Commercial NaH (as a 55-60% dispersion in oil) was washed with pentane (2 portions that cover the amount of NaH dispersion used) and the last traces of pentane are removed under high vacuum. "Aqueous half saturated brine" refers to a water:brine, 1:1 (v:v) solution.

Melting points were recorded using open capillary tubes on a Heidolph melting point apparatus and are uncorrected. Infra-red spectra were recorded on a Bruker Alpha-P FT-IR spectrometer using an Opus Wizard Interface. ¹H (300 MHz) and ¹³C NMR (75 MHz) spectra were recorded in Fourier transform mode at the field strength specified on Varian Mercury 300 spectrometer at ambient temperature. Spectra were obtained from the specified deuterated solvents in 5 mm diameter tubes. Chemical shift in ppm is quoted relative to residual solvent signals calibrated as follows: **CDCl3** $\delta_{\rm H}$ (CHCl3) = 7.26 ppm, $\delta_{\rm C}$ = 77.2 ppm; **C**₆**D**₆ $\delta_{\rm H}$ (C₆HD₅) = 7.16 ppm, $\delta_{\rm C}$ = 128.0 ppm. Multiplicities in the ¹H NMR spectra are described as: s = singlet, d = doublet, t = triplet, q = quartet, 'quint' (pseudo-quintet) m = multiplet, br = broad; coupling constants are reported in Hz. Electrospray (ES) mass spectra (MS and HRMS) were obtained with a Micromass LCT spectrometer. Ion mass/charge (*m/z*) ratios are reported as values in atomic mass units.

(Z)-tert-Butyl1,2,5,6,8,9,10,11-octahydro-4H-cyclohepta[d]azocine-3(7H)-
carboxylate 11



To an ice cold solution of the tosylate of alcohol **10** (2.26 g, 5.0 mmol, 1.0 eq.) in THF (70 mL) was added dropwise an ice cold suspension of NaH (0.59 g of a 55% NaH dispersion in oil, corresponding to 0.32 g of NaH, 25 mmol, 5.0 eq.) in THF (210 mL). The resulting mixture was refluxed for 6 h, recooled to 0 °C, carefully quenched with a saturated NH₄Cl aqueous solution (50 mL) and extracted with Et₂O (3 x 100 mL). The combined organic layers were washed with brine (50 mL), dried (Na₂SO₄) and the solvents were removed under vacuum. The residue obtained was purified by flash chromatography (Et₂O:pentane, 1:9) to afford the bicyclic amine **11** as a clear viscous oil (1.23 g, 88%).

Rf (Et₂O:pentane, 1:9) 0.75; IR (neat) 2916, 1689, 1463, 1409, 1364, 1163 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 3.48-3.26 (m, 2H), 3.26-3.09 (m, 2H), 2.29-2.02 (m, 8H), 1.81-1.61 (m, 4H); 1.56-1.38 (m, 13H); ¹³C NMR (CDCl₃, 75 MHz) δ 155.2, 131.7, 130.4, 79.4, 49.2, 48.3, 33.7, 32.5, 32.2, 31.8, 28.6, 28.5, 29.0, 27.3, 26.6; MS (ES) *m*/*z* 280 (M+H)⁺, 265 (100), 224, 180; HRMS (ES) *m*/*z* 280.2242 (calcd. for C₁₇H₃₀NO₂: 280.2277).

tert-Butyl 4, 10-dioxoazacyclotridecane-1-carboxylate 12



A solution of [7,8]-bicyclic amine **11** (0.70 g, 2.5 mmol, 1.0 eq.) in CH_2Cl_2 (45 mL) and MeOH (16.5 mL) was cooled to -78°C. O₃ was then bubbled until a persistent blue color

was seen after which argon was bubbled until the blue coloration disappeared. DMS (0.90 mL, 12.5 mmol, 5.0 eq.) was added and stirring continued at -78° C for further 10 min. The reaction was allowed to warm to room temperature over 2 h, stirred overnight and then partitioned between CH₂Cl₂ (125 mL) and aqueous half saturated brine (125 mL). The layer were separated and the organic layer was dried (Na₂SO₄), the solvent removed under vacuum to afford a colorless oil which upon standing solidified (0.78 g, quantitative). A small sample was purified by flash column chromatography (Et₂O:pentane, 1:1) and recrystallized from EtOAc:Heptane.

Rf (Et₂O:hexane, 1:1) 0.17; mp 80-81°C (EtOAc:Heptane); IR (neat) 2926, 1689, 1475, 1407, 1364, 1232 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 3.40 (t, *J* = 5.8 Hz, 2H), 3.17 (t, *J* = 6.3 Hz, 2H), 2.68 (t, *J* = 5.8 Hz, 2H), 2.41-2.32 (m, 4H), 2.28 (t, *J* = 7.4 Hz, 2H), 1.73-1.52 (m, 6H), 1.4 (s, 9H), 1.24 ('quint', *J* = 6.9 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 212.5, 210.6, 156.3, 80.0, 49.1, 44.2, 42.6, 42.1, 41.8, 38.5, 28.6, 26.8, 23.9, 23.3, 22.3; MS (ES) *m*/*z* 312 (M+H)⁺ (100), 256, 212, 194, 176; HRMS (ES) *m*/*z* 312.4402 (calcd. for C₁₇H₃₀NO4: 312.2175).

Octahydro-1*H*-pyrrolo[1,2]quinolin-7(7a*H*)-one 13



To an ice cold solution of the macrocyclic diketoamine **12** (170 mg) in CH₂Cl₂ (20 mL) was added TFA (5 mL). After stirring at 0°C for 5 mins, the cooling bath was removed and stirring continued at room temperature for further $\frac{1}{2}$ h. The excess of TFA was then removed by repeated co-evaporation with toluene (3 portions of 20 mL). To the residue thus obtained was added CH₂Cl₂ (30 mL) and saturated NaHCO₃ (30 mL) and the resulting biphasic system was vigorously stirred for approx. 5 min.. The phases were separated and the aqueous layer was extracted with CH₂Cl₂ (5 portions of 30 mL). The combined organic layers were dried (Na₂SO₄) and concentrated under reduced pressure. Purification of the obtained residue by flash column chromatography (MeOH: CH₂Cl₂, 1:9) yielded pure **13** (59 mg, 0.3 mmol, 55% from [7,8]-bicyclic amine **11**).

R_f (MeOH: CH₂Cl₂, 1:9) 0.68; IR (neat) 2933, 1702, 1446, 1352, 1172 cm⁻¹; ¹H NMR (C₆D₆, 300 MHz) δ 2.86 (ddd, J = 13.1, 11.5, 3.8 Hz, 1H), 2.74-2.52 (m, 3H), 2.34 (dddd,

J = 13.5, 4.3, 4.2, 2.2 Hz, 1H), 2.22 (dddd, J = 14.3, 11.5, 6.9, 1.0 Hz, 1H), 1.97 (ddd, J = 14.3, 3.7, 2.4 Hz, 1H), 1.93 (br m, 1H), 1.68-1.15 (m, 7H), 1.10-0.90 (m, 4H); ¹³C NMR (C₆D₆, 75 MHz) δ 208.9, 66.6, 52.4, 50.1, 44.6, 38.3, 36.6, 31.4, 24.7, 23.4, 22.5, 21.6; MS (ES) *m*/*z* 194 (M+H)⁺ (100), 143, 128; HRMS (ES) *m*/*z* 194.1483 (calcd. for C₁₂H₂₀NO: 194.1545)











X-Ray structure in Scheme 2

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 1.0 3.93 7.51 <u>}</u> 9:86 2.0 1.08 1.09 3.13 1.00 3.0 4.0 5.0 6.0 C₆D₆, 300 MHz 7.0 ppm (t1)

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