Asymmetric Synthesis of 4*H*-1,3-Oxazines: Enantioselective Reductive Cyclization of *N*-Acylated **β**-Amino Enones with Trichlorosilane Catalyzed by Chiral Lewis Bases

Masaharu Sugiura, Mako Kumahara and Makoto Nakajima

Faculty of Medical and Pharmaceutical Sciences, Kumamoto University, 5-1 Oe-honmachi, Kumamoto 862-0973, Japan

Supplementary Information

Table of Contents

General Methods	S-1
Synthesis of β-Amino Enones	S-2
Synthesis of <i>N</i> -Acylated β-Amino Enones	S-3
NOESY Correlations of Selected Compounds	S-6
The Reaction of <i>N</i> -Acylated β -Amino Enone with HSiCl ₃	S-7
Transformations of 4 <i>H</i> -1,3-Oxazine 2e	S-13
Determination of the Absolute Configuration of 2e	S-15
¹ H and ¹³ C NMR Spectra of 4 <i>H</i> -1,3-Oxazines	S-16
HPLC Traces of Optically Active Compounds	S-22
References	S-31

General Methods

Melting points (mp) are uncorrected. ¹H and ¹³C NMR spectra were measured in CDCl₃ with JEOL JNM-ECX400 spectrometer. Tetramethylsilane (TMS) ($\delta = 0$ ppm) and CDCl₃ ($\delta = 77.0$ ppm) served as internal standards for ¹H and ¹³C NMR, respectively. Infrared spectra were recorded on JEOL JIR-6500W. Mass spectra were measured with JEOL JMS-DX303HF mass spectrometer. Optical rotations were recorded on JASCO P-1010 polarimeter. High-pressure liquid chromatography (HPLC) was performed on JASCO P-980 and UV-1575.

Thin-layer chromatography (TLC) analysis was carried out using Merck silica gel plates. Visualization was accomplished with UV light, phosphomolybdic acid and/or anisaldehyde. Column chromatography was performed using Kanto Chemical Silica Gel 60N (spherical, neutral, 63-210 μ m).

Dry dichloromethane (dehydrated) was purchased from Kanto Chemical and stored over 4Å MS prior to use. All other solvents were purified based on standard procedures.

Trichlorosilane was purchased from Tokyo Kasei Kogyo (TCI) and used without further purification. A dichloromethane solution (ca. 3 M) of this reagent was prepared and stocked in a screw-top test tube with a Teflon packing. (*S*)-BINAP dioxide (BINAPO) and other chiral phosphine oxides were prepared by oxidation of the corresponding phosphines with hydrogen peroxide in acetone.¹ (*R*)-BQNO was prepared according to the literature.² All other chemicals were purified based on standard procedures.

All reactions using trichlorosilane were performed under argon atmosphere using ovenand heating gun-dried glassware equipped with a rubber septum and a magnetic stirring bar. All glassware and syringes used for trichlorosilane were rinsed with ethanol and soaked in aqueous NaOH (ca. 1 M) for several hours and washed as usual.

Synthesis of **β**-Amino Enones

(Z)-3-Amino-1-phenylbut-2-en-1-one

The title compound was prepared according to the literature procedure with a slight modification.³ A solution of 1-phenylbutane-1,3-dione (3.00 g, 18.5 mmol) and ammonium acetate (7.14 g, 92.4 mmol) in dry methanol (30 mL) was refluxed for 2.5 h. After cooling to room temperature, water (15 mL) was added and the mixture was stirred well. Precipitated colorless crystals were collected by filtration and dried at 50 °C under vacuum to give (*Z*)-3-amino-1-phenylbut-2-en-1-one (2.76 g, 93%). Spectroscopic data were consistent with the literature data.³

¹H-NMR (400 MHz, CDCl₃) δ 2.07 (s, 3H), 5.17 (brs, 1H), 5.75 (s, 1H), 7.39-7.48 (m, 3H), 7.86-7.90 (m, 2H), 10.21 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 22.9, 92.3, 127.1, 128.2, 130.8, 140.2, 162.8, 189.5.

Ph NH₂ /Pr

(Z)-3-Amino-4-methyl-1-phenylpent-2-en-1-one

A solution of 4-methyl-1-phenylpentane-1,3-dione⁴ (2.27 g, 12.0 mmol) and ammonium acetate (4.62 g, 60 mmol) in dry methanol (23 mL) was refluxed for 3 h. After cooling to room temperature, water (15 mL) was added and the mixture was extracted with diethyl ether (3×). The combined organic layers were dried over anhydrous Na_2SO_4 , filtered, and

concentrated under vacuum. The residue was purified by silica gel column chromatography (hexane/AcOEt = 4/1) to give the title compound (0.948 g, 42%) and a 79:21 mixture of the title compound and (*Z*)-1-amino-4-methyl-1-phenylpent-1-en-3-one (0.809 g, 36%).

(Z)-3-Amino-4-methyl-1-phenylpent-2-en-1-one

¹H-NMR (400 MHz, CDCl₃) δ 1.25 (d, *J* = 6.9 Hz, 6H), 2.47 (sept, *J* = 6.9 Hz, 1H), 5.24 (brs, 1H), 5.78 (s, 1H) 7.40-7.46 (m, 3H), 7.87-7.89 (m, 2H), 10.40 (brs, 1H).

(Z)-1-Amino-4-methyl-1-phenylpent-1-en-3-one

¹H-NMR (400 MHz, CDCl₃) δ 1.16 (d, *J* = 6.9 Hz, 6H), 2.61 (sept, *J* = 6.9 Hz, 1H), 5.24 (brs, 1H), 5.47 (s, 1H) 7.40-7.46 (m, 3H), 7.52-7.62 (m, 2H), 9.98 (brs, 1H).



(Z)-3-Amino-1,3-diphenylprop-2-en-1-one

A solution of 1,3-diphenylpropane-1,3-dione (2.00 g, 9.0 mmol) and ammonium formate (2.80 g, 45 mmol) in dry ethanol (20 mL) was refluxed for 21.5 h. After cooling to room temperature, the mixture was concentrated under vacuum. Water was added to the residue and the mixture was extracted with AcOEt (3×). The combined organic layers were washed with saturated aqueous NaHCO₃ (2×) and brine (1×), dried over anhydrous Na₂SO₄, filtered, and concentrated under vacuum to give crude (*Z*)-3-amino-1,3-diphenylprop-2-en-1-one (2.017 g) which was used without further purification. Spectroscopic data were consistent with the literature data.⁵

¹H-NMR (400 MHz, CDCl₃) δ 5.45 (brs, 1H), 6.16 (s, 1H), 7.42-7.55 (m, 6H), 7.60-7.70 (m, 2H), 7.90-8.03 (m, 2H), 10.43 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 91.4, 126.2, 126.9, 128.1, 128.7, 130.4, 130.8, 137.1, 140.1, 163.0, 189.7.

Synthesis of N-Acylated **β**-Amino Enones

N-((Z)-4-Oxo-4-phenylbut-2-en-2-yl)benzamide (1a)

Typical Procedure A: Pyridine (0.44 mL, 2 equiv.) and benzoyl chloride (0.62 mL, 2 equiv.) were added successively to a solution of (*Z*)-3-amino-1-phenylbut-2-en-1-one (436.7 mg, 2.71 mmol) in dichloromethane (5.4 mL) at room temperature. After being stirred for 1 h, water was added and the mixture was extracted with AcOEt (3×). The combined organic layers were dried over anhydrous MgSO₄, filtered, and concentrated under vacuum to give crude

crystals which was purified by recrystallization from ethanol/water (4/1) to afford **1a** as yellowish needles (the 1st crop: 554.4 mg, the 2nd crop: 81.5 mg; total 88%).

mp 108-110 °C; IR (KBr, cm⁻¹) 3059, 1691, 1622, 1616, 1589, 1583, 1269, 1063, 702; ¹H-NMR (400 MHz, CDCl₃) δ 2.69 (s, 3H), 6.19 (s, 1H), 7.48-7.61 (m, 6H), 7.93-7.98 (m, 2H), 8.09-8.13 (m, 2H), 13.82 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 22.8, 102.5, 127.7, 128.0, 128.6, 128.9, 132.4, 132.7, 133.7, 138.7, 158.2, 166.2, 191.9; Anal. Calcd. for C₁₇H₁₅NO₂: C, 76.96; H, 5.70; N, 5.28; Found: C, 76.72; H, 5.83; N, 5.33.



N-((Z)-4-Methyl-1-oxo-1-phenylpent-2-en-3-yl) benzamide (1b)

According to Typical Procedure A, the reaction of (*Z*)-3-amino-4-methyl-1-phenylpent-2-en-1-one (473 mg, 2.5 mmol), pyridine (0.40 mL, 2 equiv.), and benzoyl chloride (0.56 mL, 2 equiv.) in dichloromethane (5 mL) at rt for 2.5 h gave the crude product. The crude product was diluted with dry dichloromethane (5 mL) and treated with diethylamine (0.45 mL) at rt for 9 h to remove excess benzoyl chloride. The mixture was concentrated under vacuum and the residue was purified by silica gel chromatography (hexane/AcOEt = 15/1) to afford **1b** as yellowish crystals (578 mg, 78%).

mp 80-81 °C; IR (KBr, cm⁻¹) 2964, 1693, 1618, 1591, 1257, 1236, 712, 687; ¹H-NMR (400 MHz, CDCl₃) δ 1.32 (d, *J* = 6.4 Hz, 6H), 4.24 (sept, *J* = 6.4 Hz, 1H), 6.36 (s, 1H) 7.48-7.67 (m, 6H), 7.93-8.02 (m, 2H), 8.12-8.20 (m, 2H), 13.87 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 21.6, 30.0, 97.7, 127.7, 128.0, 128.5, 128.8, 132.36, 132.41, 134.1, 139.1, 165.5, 168.4, 192.4; Anal. Calcd. for C₁₉H₁₉NO₂: C, 77.79; H, 6.53; N, 4.77; Found: C, 77.87; H, 6.45; N, 4.81.

N-((*Z*)-3-Oxo-1,3-diphenylprop-1-en-1-yl)benzamide (1c)

According to Typical Procedure A, the reaction of (*Z*)-3-amino-1,3-diphenylprop-2-en-1-one (970 mg, 4.0 mmol), pyridine (0.65 mL, 2 equiv.), and benzoyl chloride (0.92 mL, 2 equiv.) in dichloromethane (8 mL) at rt for 18.5 h gave the crude product. The crude product was diluted with dry dichloromethane (10 mL) and treated with diethylamine (0.83 mL) at rt for 1.5 h to remove excess benzoyl chloride. The mixture was concentrated under vacuum and

the residue was purified by silica gel chromatography (hexane/AcOEt = 15/1) to afford **1c** as yellowish oil which gradually crystallized on standing (1.133 g, 91%).

mp 118-119 °C; IR (KBr, cm⁻¹) 3057, 1687, 1624, 1583, 1564, 1464, 1290, 1225, 1045, 719; ¹H-NMR (400 MHz, CDCl₃) δ 6.46 (s, 1H), 7.42-7.64 (m, 11H), 7.97-8.04 (m, 2H), 8.08-8.13 (m, 2H), 13.36 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 105.2, 127.3, 127.8, 128.0 (2C), 128.6, 128.8, 129.7, 132.66, 132.69, 133.2, 136.3, 138.4, 157.0, 165.1, 191.9; Anal. Calcd. for C₂₂H₁₇NO₂: C, 80.71; H, 5.23; N, 4.28; Found: C, 80.39; H, 5.35; N, 4.26.

4-Nitro-N-((Z)-4-oxo-4-phenylbut-2-en-2-yl)benzamide (1d)

According to Typical Procedure A, the reaction of (*Z*)-3-amino-1-phenylbut-2-en-1-one (322 mg, 2.0 mmol), pyridine (0.32 mL, 2 equiv.), and *p*-nitrobenzoyl chloride (742 mg, 2 equiv.) in dichloromethane (10 mL) at rt for 1 h gave the crude product which was purified by recrystallization from AcOEt to afford **1d** as orange needles (the 1st crop: 346 mg, 56%). mp 196-197 °C; IR (KBr, cm⁻¹) 3113, 1697, 1626, 1593, 1475, 1346, 1277, 710; ¹H-NMR (400 MHz, CDCl₃) δ 2.67 (s, 3H), 6.27 (s, 1H), 7.48-7.61 (m, 3H), 7.93-7.99 (m, 2H), 8.24-8.30 (m, 2H), 8.35-8.42 (m, 2H), 14.05 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 22.5, 103.4, 124.1, 127.8, 128.7, 129.2, 132.9, 138.3, 139.2, 150.1, 157.4, 164.0, 192.2; Anal. Calcd. for C₁₇H₁₄N₂O₄: C, 65.80; H, 4.55; N, 9.03; Found: C, 65.57; H, 4.57; N, 9.01.

4-Methoxy-N-((Z)-4-oxo-4-phenylbut-2-en-2-yl)benzamide (1e)

According to Typical Procedure A, the reaction of (*Z*)-3-amino-1-phenylbut-2-en-1-one (1.612 g, 10.0 mmol), pyridine (1.62 mL, 2 equiv.), and *p*-methoxybenzoyl chloride (2.71 mL, 2 equiv.) in dichloromethane (20 mL) at rt for 4 h gave the crude product which was purified by recrystallization from ethanol to afford **1e** as orange needles (the 1st crop: 2.1267 g, 72%). mp 127-128 °C; IR (KBr, cm⁻¹) 3001, 2970, 2943, 2841, 1687, 1606, 1498, 1477, 1273, 1246, 1178, 1061, 1032, 847, 758; ¹H-NMR (400 MHz, CDCl₃) δ 2.67 (s, 3H), 3.89 (s, 3H), 6.15 (s, 1H), 6.99-7.05 (m, 2H), 7.45-7.60 (m, 3H), 7.92-7.98 (m, 2H), 8.06-8.12 (m, 2H), 13.82 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 22.7, 55.4, 102.1, 114.1, 126.0, 127.6, 128.6, 130.1, 132.3, 138.8, 158.6, 163.2, 165.7, 191.8; Anal. Calcd. for C₁₈H₁₇NO₃: C, 73.20; H, 5.80; N, 4.74; Found: C, 72.94; H, 5.82; N, 4.65.



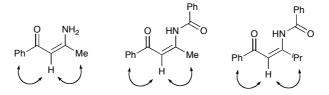
N-((Z)-4-Oxo-4-phenylbut-2-en-2-yl)acetamide (1f)

According to Typical Procedure A, the reaction of (*Z*)-3-amino-1-phenylbut-2-en-1-one (806 mg, 5 mmol), pyridine (0.81 mL, 2 equiv.) and acetyl chloride (0.71 mL, 2 equiv.) in dichloromethane (10 mL) at rt for 20 h gave the crude product which was purified by recrystallization from ethanol/water (4/1) to afford **1f** as yellowish prisms (the 1st crop: 512 mg, the 2nd crop: 171 mg; total 67%).

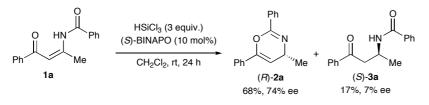
mp 98-99 °C; IR (KBr, cm⁻¹) 3072, 1711, 1616, 1585, 1560, 1502, 1473, 1439, 1308, 1261, 1200, 1086, 1065, 1003, 858, 777, 681; ¹H-NMR (400 MHz, CDCl₃) δ 2.22 (s, 3H), 2.52 (s, 3H), 6.04 (s, 1H), 7.41-7.58 (m, 3H), 7.88-7.93 (m, 2H), 12.81 (brs, 1H); ¹³C-NMR (100 MHz, CDCl₃) δ 22.5, 25.4, 101.5, 127.6, 128.5, 132.4, 138.6, 157.5, 169.8, 191.4; Anal. Calcd. for C₁₂H₁₃NO₂: C, 70.92; H, 6.45; N, 6.89; Found: C, 70.66; H, 6.50; N, 6.83.

NOESY Correlations of Selected Compounds

The geometry of selected β -amino enones was determined to be Z by observation of the NOESY correlations as shown below. That of other enones was determined by analogy.



The Reaction of N-Acylated *β*-Amino Enone with HSiCl₃



Typical Procedure B: To a solution of (*S*)-BINAPO (16.4 mg, 10 mol%) and **1a** (66.2 mg, 0.25 mmol) in dry dichloromethane (1 mL) was added dropwise trichlorosilane (ca. 3 M CH₂Cl₂ solution, 3 equiv.) at 0 °C. The reaction was stirred at rt for 24 h and quenched with water (3 mL) and dichloromethane (5 mL). The mixture was stirred for 1 h, filtered through a Celite pad with dichloromethane and extracted with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, evaporated, and purified by silica gel column chromatography (hexane/AcOEt = $20/1 \sim 2/1$) to give 4H-1,3-oxazine **2a** [42.3 mg, 68%, 74% ee (*R*)] and keto amide **3a** [11.5 mg, 17%, 7% ee (*S*)].

The sense of the enantioselectivity of 2a was found to be opposite to that of 3a by transforming 2a to 3a with hydrobromic acid in ethanol. Their absolute configurations were tentatively assigned by analogy.

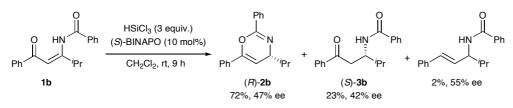
(R)-4-Methyl-2,6-diphenyl-4H-1,3-oxazine (2a)

Viscous oil; TLC: $R_f 0.68$ (hexane/AcOEt = 3/1); $[\alpha]_{D}^{28} + 8.8$ (c 0.600, CHCl₃) for 72% ee (*R*); IR (neat, cm⁻¹) 3061, 2966, 1686, 1647, 1350, 1238, 1146, 1063, 760, 690; ¹H-NMR (400 MHz, CDCl₃) δ 1.45 (d, *J* = 6.9 Hz, 3H), 4.38 (dq, *J* = 6.9, 3.2 Hz, 1H), 5.52 (d, *J* = 3.2 Hz, 1H), 7.33-7.51 (m, 6H), 7.64-7.70 (m, 2H), 8.05-8.12 (m, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 25.0, 48.1, 102.3, 124.1, 127.3, 128.2, 128.5, 128.7, 130.8, 132.5, 133.0, 146.7, 152.0; HRMS (FAB) calcd. for C₁₇H₁₆NO (M+H⁺) 250.1232, found 250.1250; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 19/1, flow rate 1.0 mL/min, UV detection at 254 nm): *t_R* = 4.6 min (*R*, major), 6.8 min (*S*, minor).

(S)-N-(4-Oxo-4-phenylbutan-2-yl)benzamide (3a)

Spectroscopic data were consistent with the literature data.⁶

Yellowish solid; TLC: R_f 0.09 (hexane/AcOEt = 3/1); ¹H-NMR (400 MHz, CDCl₃) δ 1.42 (d, J = 6.9 Hz, 3H), 3.21 (dd, J = 17.0, 5.9 Hz, 1H), 3.48 (dd, J = 17.0, 4.1 Hz, 1H), 4.70 (dddq, J = 7.8, 5.9, 4.1, 6.9 Hz, 1H), 7.06 (brd, J = 7.8 Hz, 1H), 7.41-7.55 (m, 5H), 7.60 (apparent t, J = 7.8 Hz, 1H), 7.78 (apparent d, J = 7.8 Hz, 2H), 7.99 (apparent d, J = 7.8 Hz, 2H); HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 9/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 15.8$ min (S, major), 17.2 min (R, minor).



According to Typical Procedure B, the reaction of **1b** (73.6 mg, 0.25 mmol) and trichlorosilane (ca. 3 M CH₂Cl₂ solution, 2 equiv.) with (*S*)-BINAPO (16.6 mg, 10 mol%) in dry dichloromethane (1 mL) at rt for 9 h gave 4H-1,3-oxazine **2b** [49.8 mg, 72%, 47% ee (*R*)], keto amide **3b** [17.1 mg, 23%, 42% ee (*S*)] and an allylic amide (1.3 mg, 2%, 55% ee).

The sense of the enantioselectivity of **2b** was found to be same as that of **3b** by transforming **2b** to **3b** with hydrobromic acid in ethanol. Their absolute configurations were tentatively assigned by analogy.

(R)-4-Isopropyl-2,6-diphenyl-4H-1,3-oxazine (2b)

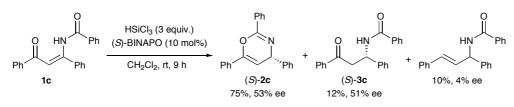
Viscous oil, TLC: $R_f 0.71$ (hexane/AcOEt = 3/1); $[\alpha]_{D}^{21} - 1.2$ (c 1.115, CHCl₃) for 47% ee (*R*); IR (neat, cm⁻¹) 2962, 1691, 1495, 1448, 1261, 762, 692; ¹H-NMR (400 MHz, CDCl₃) δ 1.03 (d, *J* = 6.8 Hz, 3H), 1.04 (d, *J* = 6.8 Hz, 3H), 2.00 (d sept, *J* = 4.1, 6.8 Hz, 1H), 4.20 (dd, *J* = 4.1, 3.7 Hz, 1H), 5.47 (d, *J* = 3.7 Hz, 1H), 7.36-7.50 (m, 6H), 7.65-7.72 (m, 2H), 8.06-8.11 (m, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 17.97, 18.05, 35.2, 57.9, 99.0, 124.1, 127.3, 128.2, 128.5, 128.7, 130.8, 132.6, 133.1, 147.8, 152.1; HRMS (FAB) calcd. for C₁₉H₂₀NO (M+H⁺) 278.1545, found 278.1581; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2propanol = 49/1, flow rate 1.0 mL/min, UV detection at 254 nm): *t_R* = 5.0 min (*R*, major), 7.0 min (*S*, minor).

(S)-N-(4-Methyl-1-oxo-1-phenylpentan-3-yl)benzamide (3b)

Yellowish solid; mp 129-131 °C; R_f 0.25 (hexane/AcOEt = 3/1); ¹H-NMR (400 MHz, CDCl₃) δ 0.97 (d, J = 6.9 Hz, 3H), 1.04 (d, J = 6.9 Hz, 3H), 2.20 (d sept, J = 8.2, 6.9 Hz, 1H), 3.22 (dd, J = 17.0, 5.0 Hz, 1H), 3.49 (dd, J = 17.0, 5.0 Hz, 1H), 4.33 (dddd, J = 9.2, 8.2, 5.0, 5.0 Hz, 1H), 7.18 (brd, J = 9.2 Hz, 1H), 7.40-7.53 (m, 6H), 7.59 (apparent t, J = 7.3 Hz, 1H), 7.76-7.80 (m, 2H), 7.95-7.99 (m, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 19.52, 19.86, 31.0, 39.3, 52.7, 126.9, 128.06, 128.51, 128.74, 131.3, 133.5, 134.7, 136.8, 166.9, 200.0; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 19/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 26.0$ min (R, minor), 34.0 min (S, major).

(E)-N-(4-Methyl-1-phenylpent-1-en-3-yl)benzamide

Colorless solid; R_f 0.38 (hexane/AcOEt = 3/1); ¹H-NMR (400 MHz, CDCl₃) δ 1.03 (d, J = 6.9 Hz, 6H), 2.02 (d sept, J = 6.9, 6.9 Hz, 1H), 4.72 (ddd, J = 8.7, 6.9, 6.9 Hz, 1H), 6.14 (brd, J = 8.7 Hz, 1H), 6.21 (dd, J = 15.8, 6.9 Hz, 1H), 6.60 (d, J = 15.8 Hz, 1H), 7.20-7.58 (m, 8H), 7.79-7.83 (m, 2H); HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 12/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 16.8$ min (minor), 21.6 min (major).



According to Typical Procedure B, the reaction of **1c** (81.6 mg, 0.25 mmol) and trichlorosilane (ca. 3 M CH₂Cl₂ solution, 2 equiv.) with (*S*)-BINAPO (16.4 mg, 10 mol%) in dry dichloromethane (1 mL) at rt for 9 h gave 4*H*-1,3-oxazine **2c** [58.5 mg, 75%, 53% ee (*S*)], keto amide **3c** [9.7 mg, 12%, 51% ee (*S*)] and an allylic amide (7.7 mg, 10%, 4% ee).

The sense of the enantioselectivity of 2c was found to be same as that of 3c by transforming 2c to 3c with hydrobromic acid in ethanol. The absolute configuration of 3c was unequivocally determined on comparison with the literature data.⁷ Thus, that of 2c was also determined.

(S)-2,4,6-Triphenyl-4*H*-1,3-oxazine (2c)

Colorless solid; mp 96-97 °C; TLC: R_f 0.63 (hexane/AcOEt = 3/1); $[\alpha]_{D}^{19}$ -33.5 (c 1.075, CHCl₃) for 53% ee (*S*); IR (KBr, cm⁻¹) 3088, 3061, 3028, 2862, 1687, 1647, 1495, 1450, 1319, 1240, 1119, 1070, 1024, 758, 696; ¹H-NMR (400 MHz, CDCl₃) δ 5.44 (d, *J* = 3.7 Hz, 1H), 5.65 (d, *J* = 3.7 Hz, 1H), 7.23-7.32 (m, 1H), 7.34-7.55 (m, 10H), 7.68-7.73 (m, 2H), 8.10-8.17 (m, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 56.6, 100.8, 124.2, 127.28, 127.36, 127.48, 128.23, 128.51, 128.70, 128.95, 131.1, 132.4, 132.8, 144.4, 146.3, 152.2; HRMS (FAB) calcd. for C₂₂H₁₈NO (M+H⁺) 312.1388, found 312.1383; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 29/1, flow rate 1.0 mL/min, UV detection at 254 nm): *t_R* = 10.5 min (*S*, major), 14.6 min (*R*, minor).

(S)-N-(3-Oxo-1,3-diphenylpropyl)benzamide (3c)

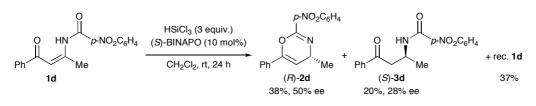
Spectroscopic data were consistent with the literature data.⁷

Colorless solid; TLC: R_f 0.15 (hexane/AcOEt = 3/1); ¹H-NMR (400 MHz, CDCl₃) δ 3.56 (dd, J = 17.0, 6.0 Hz, 1H), 3.86 (dd, J = 17.0, 4.6 Hz, 1H), 5.77 (ddd, J = 7.8, 6.0, 4.6 Hz, 1H), 7.20-7.60 (m, 11H), 7.60 (brd, J = 7.8 Hz, 1H), 7.83-7.87 (m, 2H), 7.91-7.95 (m, 2H); HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 4/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 21.5$ min (R, minor), 28.7 min (S, major).

N-(1,3-Diphenylallyl)benzamide

Spectroscopic data were consistent with the literature data.⁸

Colorless solid; TLC: R_f 0.39 (hexane/AcOEt = 3/1); ¹H-NMR (400 MHz, CDCl₃) δ 6.04 (apparent t, J = 6.7 Hz, 1H), 6.46 (dd, J = 16.0, 6.5 Hz, 1H), 6.53 (brd, J = 7.8 Hz, 1H), 6.62 (d, J = 16.0 Hz, 1H), 7.21-7.54 (m, 13H), 7.81-7.86 (m, 2H); HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 6/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 18.2$ min (minor), 20.2 min (major).



According to Typical Procedure B, the reaction of **1d** (78.1 mg, 0.25 mmol) and trichlorosilane (ca. 3 M CH₂Cl₂ solution, 2 equiv.) with (*S*)-BINAPO (16.6 mg, 10 mol%) in dry dichloromethane (2 mL; **1d** was partially dissolved at the beginning) at rt for 24 h gave 4H-1,3-oxazine **2d** [27.8 mg, 38%, 50% ee (*R*)] and keto amide **3d** [15.7 mg, 20%, 28% ee (*S*)]. **1d** was recovered in 37% (29.0 mg).

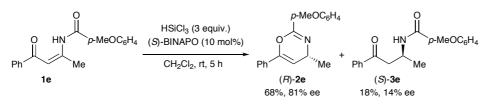
The sense of the enantioselectivity of 2d was found to be opposite to that of 3d by transforming 2d to 3d with hydrobromic acid in ethanol. Their absolute configurations were tentatively assigned by analogy.

(R)-4-Methyl-2-(4-nitrophenyl)-6-phenyl-4H-1,3-oxazine (2d)

Yellowish solid; mp 143-144 °C; TLC: R_f 0.61 (hexane/AcOEt = 3/1); $[\alpha]_{D}^{29}$ +5.8 (c 1.170, CHCl₃) for 61% ee (*R*); IR (KBr, cm⁻¹) 2974, 2929, 2848, 1689, 1649, 1601, 1522, 1340, 1244, 1146, 1093, 862, 760, 702; ¹H-NMR (400 MHz, CDCl₃) δ 1.47 (d, *J* = 6.9 Hz, 3H), 4.44 (dq, *J* = 6.9, 2.8 Hz, 1H), 5.52 (d, *J* = 2.8 Hz, 1H), 7.36-7.50 (m, 3H), 7.62-7.68 (m, 2H), 8.20-8.32 (m, 4H); ¹³C-NMR (100 MHz, CDCl₃) δ 25.0, 48.5, 102.2, 123.4, 124.1, 128.3, 128.6, 129.1, 132.5, 138.4, 146.7, 149.3, 150.3; HRMS (FAB) calcd. for C₁₇H₁₅N₂O₃ (M+H⁺) 295.1083, found 295.1073; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 29/1, flow rate 1.0 mL/min, UV detection at 254 nm): *t_R* = 8.7 min (*R*, major), 11.1 min (*S*, minor).

(S)-4-Nitro-N-(4-oxo-4-phenylbutan-2-yl)benzamide (3d)

Viscous oil; TLC: R_f 0.11 (hexane/AcOEt = 3/1); IR (neat, cm⁻¹) 3307, 3070, 2976, 1684, 1645, 1601, 1525, 1346, 1213, 1003, 870, 841, 754, 721, 690; ¹H-NMR (400 MHz, CDCl₃) δ 1.44 (d, J = 6.9 Hz, 3H), 3.25 (dd, J = 17.0, 5.5 Hz, 1H), 3.48 (dd, J = 17.0, 4.1 Hz, 1H), 4.71 (dddq, J = 7.8, 5.5, 4.1, 6.9 Hz, 1H), 7.37 (brd, J = 7.8 Hz, 1H), 7.50 (apparent t, J = 7.8 Hz, 2H), 7.62 (apparent t, J = 7.3 Hz, 1H), 7.95 (apparent d, J = 8.7 Hz, 2H), 7.98 (apparent d, J = 8.0 Hz, 2H), 8.28 (apparent d, J = 8.7 Hz, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 20.0, 42.5, 43.2, 123.7, 128.09, 128.13, 128.81, 133.8, 136.7, 140.1, 149.5, 164.5, 199.8; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 9/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 28.1$ min (R, minor), 33.3 min (S, major).



According to Typical Procedure B, the reaction of **1e** (73.8 mg, 0.25 mmol) and trichlorosilane (ca. 3 M CH₂Cl₂ solution, 2 equiv.) with (*S*)-BINAPO (16.4 mg, 10 mol%) in dry dichloromethane (1 mL) at rt for 5 h gave 4*H*-1,3-oxazine **2e** [47.2 mg, 68%, 81% ee (*R*)] and keto amide **3e** [13.7 mg, 18%, 14% ee (*S*)].

The sense of the enantioselectivity of **2e** was found to be opposite to that of **3e** by transforming **2e** to **3e** with hydrobromic acid in ethanol. The absolute configuration of **2e** was unequivocally determined as described later.

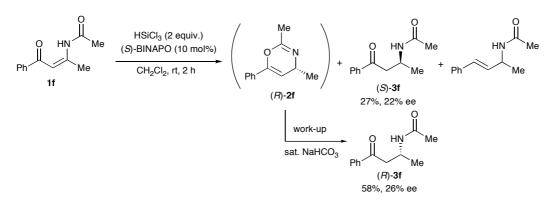
(*R*)-2-(4-Methoxyphenyl)-4-methyl-6-phenyl-4*H*-1,3-oxazine (2e)

Viscous oil; TLC: $R_f 0.40$ (hexane/AcOEt = 3/1); $[\alpha]_{0}^{30} + 5.3$ (c 1.295, CHCl₃) for 78% ee (*R*); IR (neat, cm⁻¹) 2966, 2926, 2837, 1689, 1608, 1512, 1250, 1169, 1146, 1068, 1030, 839, 760, 690; ¹H-NMR (400 MHz, CDCl₃) δ 1.43 (d, *J* = 6.9 Hz, 3H), 3.86 (s, 3H), 4.35 (dq, *J* = 6.9, 3.2 Hz, 1H), 5.51 (d, *J* = 3.2 Hz, 1H), 6.94-6.96 (m, 2H), 7.36-7.42 (m, 3H), 7.66-7.68 (m, 2H), 8.01-8.03 (m, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 25.1, 48.0, 55.3, 102.4, 113.5, 124.1, 125.1, 128.45, 128.67, 128.93, 133.1, 146.7, 151.8, 161.7; HRMS (FAB) calcd. for C₁₈H₁₈NO₂ (M+H⁺) 280.1338, found 280.1331; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 19/1, flow rate 1.0 mL/min, UV detection at 254 nm): *t_R* = 8.7 min (*R*, major), 19.8 min (*S*, minor).

(S)-4-Methoxy-N-(4-oxo-4-phenylbutan-2-yl)benzamide (3e)

Yellowish solid; mp 119-121 °C; TLC: R_f 0.06 (hexane/AcOEt = 3/1); IR (KBr, cm⁻¹) 3321, 3062, 2958, 1687, 1666, 1637, 1549, 1529, 756, 692; ¹H-NMR (400 MHz, CDCl₃) δ 1.40 (d, J = 6.9 Hz, 3H), 3.20 (dd, J = 16.8, 6.1 Hz, 1H), 3.47 (dd, J = 16.8, 4.0 Hz, 1H), 3.85 (s, 3H), 4.68 (dddq, J = 7.3, 6.1, 4.0, 6.9 Hz, 1H), 6.91 (apparent d, J = 8.9 Hz, 2H), 6.96 (brd, J = 7.3 Hz, 1H), 7.48 (apparent t, J = 7.3 Hz, 2H), 7.58 (apparent t, J = 7.3 Hz, 1H), 7.75 (apparent d, J = 8.9 Hz, 2H), 7.99 (apparent d, J = 7.4 Hz, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 20.1, 42.8, 43.3, 55.4, 113.7, 126.8, 128.1, 128.7 (2C), 133.5, 136.9, 162.1, 166.2, 199.7; HRMS (FAB) calcd. for C₁₈H₂₀NO₃ (M+H⁺) 298.1443, found 298.1470; HPLC (CHIRALPAK AD-H, 0.46 cm $\beta \times$ 25cmL, hexane/2-propanol = 19/1, flow rate 1.0 mL/min, UV detection at 254 nm): t_R = 77.8 min (*S*, major), 82.8 min (*R*, minor).

Supplementary Material (ESI) for Chemical Communications This journal is © The Royal Society of Chemistry 2009



To a solution of (*S*)-BINAPO (16.3 mg, 10 mol%) and **1f** (50.9 mg, 0.25 mmol) in dry dichloromethane (2 mL) was added dropwise trichlorosilane (ca. 3 M CH₂Cl₂ solution, 2 equiv.) at 0 °C. The reaction was stirred at rt for 2 h and quenched with water (3 mL) and dichloromethane (5 mL). The mixture was stirred for 1 h, filtered through a Celite pad with dichloromethane and extracted with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, evaporated, and purified by silica gel column chromatography (hexane/AcOEt = 1/1, AcOEt only, then CH₂Cl₂/EtOH = 10/1) to give keto amide **3f** [13.9 mg, 27%, 22% ee (*S*)]. On the other hand, the acidic aqueous layer (pH ca. 3) was basified with saturated aq. NaHCO₃ (pH ca. 9) and extracted with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, evaporated and evaporated to give pure keto amide **3f** [29.8 mg, 58%, 26% ee (*R*)].

The sense of the enantioselectivity of 3f extracted from the acidic aqueous layer was opposite to that of 3f extracted from the basic aqueous layer. The absolute configuration was assigned by analogy.

(R)-N-(4-Oxo-4-phenylbutan-2-yl)acetamide (3f)

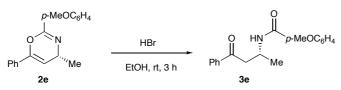
Yellowish solid; mp 74-76 °C; TLC: R_f 0.20 (hexane/AcOEt = 1/2); $[\alpha]_{D}^{20}$ +15.5 (c 0.995, CHCl₃) for 26% ee (*R*); ¹H-NMR (400 MHz, CDCl₃) δ 1.29 (d, *J* = 6.4 Hz, 3H), 1.96 (s, 3H), 3.09 (dd, *J* = 16.5, 6.4 Hz, 1H), 3.34 (brd, *J* = 16.5, 4.1 Hz, 1H), 4.47 (dddq, *J* = 7.1, 6.9, 6.4, 4.1 Hz, 1H), 6.32 (brd, *J* = 7.1 Hz, 1H), 7.47 (apparent t, *J* = 7.8 Hz, 2H), 7.58 (apparent t, *J* = 7.3 Hz, 1H), 7.96 (apparent d, *J* = 8.3 Hz, 2H), 7.25-7.97 (m, 5H); ¹³C-NMR (100 MHz, CDCl₃) δ 19.9, 23.4, 42.4, 43.4, 128.04, 128.65, 133.4, 136.7, 169.4, 199.2; HPLC (CHIRALPAK AS-H, 0.46 cmø × 25cmL, hexane/2-propanol = 29/1, flow rate 1.0 mL/min, UV detection at 254 nm): t_R = 13.6 min (*S*, minor), 19.2 min (*R*, major).

The formation of oxazine **2f** was confirmed by quenching the reaction mixture (another batch of the reaction) with deuterium oxide. After removal of the precipitates by filtration, the acidic deuterium oxide layer was washed with dichloromethane $(3\times)$ and checked directly by ¹H and ¹³C NMR analyses (for the spectra, see page S-21).

(R)-2-Methyl-4-methyl-6-phenyl-4H-1,3-oxazine (2f)

¹H-NMR (400 MHz, D₂O) δ 1.44 (d, *J* = 6.8 Hz, 3H), 2.40 (s, 3H), 4.34 (dq, *J* = 9.6, 6.8 Hz, 1H), 5.99 (d, *J* = 9.6 Hz, 1H), 7.21-7.28 (m, 3H), 7.31-7.37 (m, 2H); ¹³C-NMR (100 MHz, D₂O, MeOH was used as an internal standard) δ 19.0, 20.7, 44.5, 114.8, 125.6, 129.6, 130.6, 133.2, 149.9, 173.0.

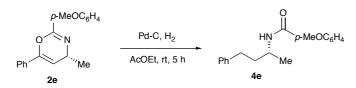
Transformations of 4H-1,3-Oxazine 2e



Hydrolysis: To a solution of **2e** [28.1 mg, 0.10 mmol, 81% ee (*R*)] in dry ethanol (1.0 mL) was added dropwise 47 % HBr (0.1 mL) at rt. The reaction was stirred at rt for 3 h and quenched with saturated aqueous NaHCO₃. The mixture was extracted with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, evaporated, and purified by silica gel column chromatography (hexane/AcOEt = 2/1) to give keto amide **3e** [23.3 mg, 78%, 81% ee (*R*)].

(R)-4-Methoxy-N-(4-oxo-4-phenylbutan-2-yl)benzamide (3e)

For physical data, see page S-11. $[\alpha]_{D}^{30}$ –12.8 (c 0.795, CHCl₃) for 81% ee (*R*); HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 19/1, flow rate 1.0 mL/min, UV detection at 254 nm): t_{R} = 71.5 min (*S*, minor), 75.0 min (*R*, major).

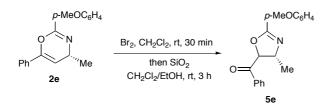


Reduction: A suspension of **2e** [26.8 mg, 0.096 mmol, 81% ee (*R*)] and 10% Pd/C (5.4 mg) in AcOEt (1.9 mL) was stirred at rt for 5 h under hydrogen atmosphere. The mixture was filtered through a Celite pad with AcOEt, evaporated, and purified by silica gel column chromatography (hexane/AcOEt = $3/1 \sim 2/1$) to give amide **4e** [24.5 mg, 90%, 81% ee (*R*)].

(R)-4-Methoxy-N-(4-phenylbutan-2-yl)benzamide (4e)

Colorless solid; mp 119-121 °C; TLC: R_f 0.32 (hexane/AcOEt = 2/1); $[\alpha]_D^{21}$ -8.3 (c 0.915, CHCl₃) for 81% ee (*R*); IR (KBr, cm⁻¹) 3313, 2964, 2926, 2860, 1630, 1608, 1537, 1508, 1300, 1254, 1182, 1028, 843, 702, 700; ¹H-NMR (400 MHz, CDCl₃) δ 1.27 (d, *J* = 6.4 Hz, 3H), 1.89 (dt, *J* = 6.4, 7.8 Hz, 2H), 2.72 (t, *J* = 7.8 Hz, 2H), 3.84 (s, 3H), 4.27 (dtq, *J* = 7.8, 6.4, 6.4 Hz, 1H), 5.85 (brd, *J* = 7.8 Hz, 1H), 6.89 (apparent d, *J* = 8.7 Hz, 2H), 7.14-7.30 (m,

5H), 7.65 (apparent d, J = 8.7 Hz, 2H); ¹³C-NMR (100 MHz, CDCl₃) δ 21.1, 32.5, 38.6, 45.6, 55.4, 113.6, 125.9, 127.1, 128.32, 128.46, 128.54, 141.8, 162.0, 166.3; HRMS (FAB) calcd. for C₁₈H₂₂NO₂ (M+H⁺) 284.1651, found 284.1628; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 9/1, flow rate 1.0 mL/min, UV detection at 254 nm): $t_R = 19.5$ min (*R*, major), 23.1 min (*S*, minor).



Oxidation: To a solution of **2e** [31.7 mg, 0.11 mmol, 81% ee (*R*)] in dry dichloromethane (1.1 mL) was added dropwise bromine (0.76 M solution in dichloromethane, 0.3 mL) at rt. The reaction was stirred for 30 min and quenched with water. After being stirred for 5 min, the mixture was transferred to a separation funnel, basified with saturated aqueous NaHCO₃, and extracted with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂SO₄, filtered, and evaporated. The residue was then diluted with dichloromethane (2 mL) and ethanol (0.4 mL). Silica gel (ca. 300 mg) was added to the solution. The mixture was stirred for 3 h, evaporated, and charged directly on a silica gel column for chromatography (hexane/AcOEt = $3/1 \sim 1/1$ then CH₂Cl₂/EtOH = 20/1; **5e** was partially crystallized in the silica gel column. Therefore, elution with CH₂Cl₂/EtOH was needed) to give *trans*-**5e** (4.7 mg, 14%, 81% ee) and *cis*-**5e** (24.1 mg, 72%, 81% ee).

The relative configurations were assigned based on the coupling constants between H4 and H5 protons (*trans-***5e**: 6.4 Hz, *cis-***5e**: 10.5 Hz).

(4*R*,5*S*)-5-Benzoyl-2-(4-methoxyphenyl)-4-methyl-4,5-dihydrooxazole (*trans*-5e)

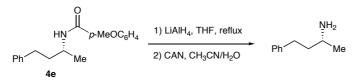
Viscous oil; TLC: R_f 0.31 (hexane/AcOEt = 2/1); IR (neat, cm⁻¹) 2970, 2933, 2839, 1695, 1653, 1608, 1514, 1257, 1173, 1028, 841, 692; ¹H-NMR (400 MHz, CDCl₃) δ 1.55 (d, J = 6.9 Hz, 3H), 3.86 (s, 3H), 4.50 (dq, J = 6.9, 6.4 Hz, 1H), 5.38 (d, J = 6.4 Hz, 1H), 6.93 (apparent d, J = 8.7 Hz, 2H), 7.53 (apparent t, J = 7.8 Hz, 2H), 7.65 (apparent t, J = 7.3 Hz, 1H), 7.93-8.02 (m, 4H); HRMS (FAB) calcd. for C₁₈H₁₈NO₃ (M+H⁺) 296.1287, found 296.1292; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 4/1, flow rate 0.4 mL/min, UV detection at 254 nm): t_R = 46.0 min (4*S*,5*R*, minor), 49.1 min (4*R*,5*S*, major).

(4*R*,5*R*)-5-Benzoyl-2-(4-methoxyphenyl)-4-methyl-4,5-dihydrooxazole (*cis*-5e)

Colorless solid; mp 127-129 °C; TLC: $R_f 0.19$ (hexane/AcOEt = 2/1); $[\alpha]_{D}^{22}$ +210.2 (c 0.245, CHCl₃) for 81% ee (4*R*,5*R*); IR (KBr, cm⁻¹) 2968, 1697, 1649, 1606, 1514, 1377, 1259, 1221, 1167, 1099, 1026, 960, 847, 694; ¹H-NMR (400 MHz, CDCl₃) δ 1.08 (d, *J* = 6.9 Hz, 3H), 3.87 (s, 3H), 4.85 (dq, *J* = 10.4, 6.9 Hz, 1H), 5.96 (d, *J* = 10.4 Hz, 1H), 6.93 (apparent d, *J* =

8.6 Hz, 2H), 7.53 (apparent t, J = 8.0 Hz, 2H), 7.64 (apparent t, J = 7.4 Hz, 1H), 7.93-8.02 (m, 4H); ¹³C-NMR (100 MHz, CDCl₃) δ 17.5, 55.4, 65.1, 83.2, 113.7, 119.4, 128.0, 129.0, 130.3, 133.9, 135.4, 162.4, 163.1, 194.5; HRMS (FAB) calcd. for C₁₈H₁₈NO₃ (M+H⁺) 296.1287, found 296.1298; HPLC (CHIRALPAK AD-H, 0.46 cmø × 25cmL, hexane/2-propanol = 1/1, flow rate 0.6 mL/min, UV detection at 254 nm): $t_R = 12.9$ min (4*S*,5*S*; minor), 48.2 min (4*R*,5*R*; major).

Determination of the Absolute Configuration of 2e



To a solution of **4e** (17.9 mg, 0.063 mmol, 81% ee) in dry THF (2.0 mL) was added LiAlH₄ (24.0 mg, 0.63 mmol) at rt. The reaction was refluxed for 16 h and, after cooling to rt, carefully quenched with water and 4M aq. NaOH. The mixture was extracted with dichloromethane (3×). The filtrate was dried over anhydrous Na₂CO₃, filtered, and evaporated to give crude (*R*)-*N*-(4-methoxybenzyl)-4-phenylbutan-2-amine. This crude amine was diluted with acetonitrile/water (2/1, 2 mL) and treated with CAN (173.1 mg, 0.32 mmol). The reaction was stirred at rt for 6 h and quenched with 1.0 M hydrochloric acid (5 mL). The mixture was filtered through a Celite pad with dichloromethane. The filtrate was washed with dichloromethane (3×). The combined organic layers were dried over anhydrous Na₂CO₃, filtered, and evaporated to afford (*R*)-4-phenylbutan-2-amine (4.3 mg, 46% over 2 steps).

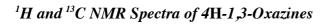
(R)-N-(4-Methoxybenzyl)-4-phenylbutan-2-amine

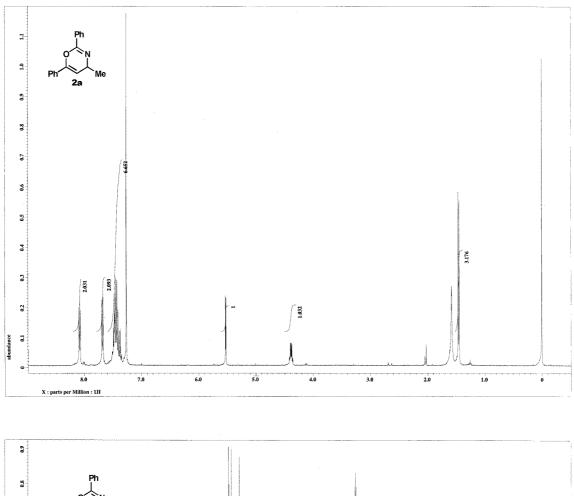
¹H-NMR (400 MHz, CDCl₃) δ 1.14 (d, *J* = 6.1 Hz, 3H), 1.48 (brd, 1H), 1.62-1.70 (m, 1H), 1.76-1.85 (m, 1H), 2.59-2.69 (m, 2H), 2.69-2.76 (m, 1H), 3.67 (d, *J* = 12.8 Hz, 1H), 3.76 (d, *J* = 12.8 Hz, 1H), 3.80 (s, 3H), 6.85 (apparent d, *J* = 8.6 Hz, 2H), 7.16-7.29 (m, 7H); ¹³C-NMR (100 MHz, CDCl₃) δ 20.3, 32.3, 38.7, 50.7, 51.9, 55.2, 113.7, 125.7, 128.31, 128.32, 129.3, 132.9, 142.4, 158.5.

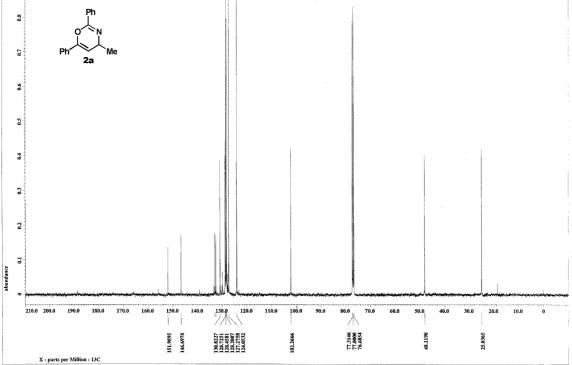
(R)-4-Phenylbutan-2-amine

Spectroscopic data are consistent with the literature data.⁹

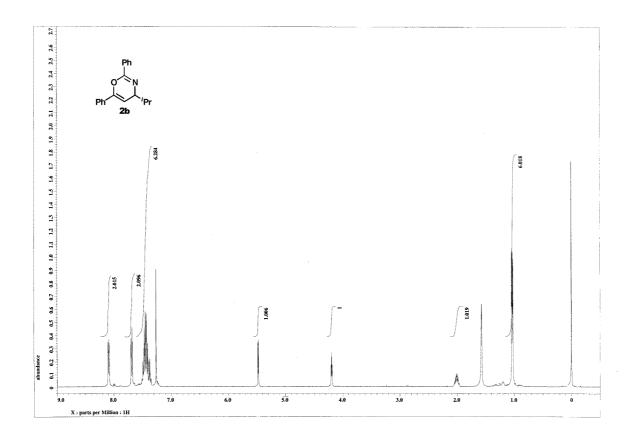
 $[\alpha]_{D}^{20}$ = 8.7 (c 0.215, CHCl₃) for 81% ee (*R*) [*lit*.⁹ [α]_{D}^{20} +6.4 (c 0.47, CHCl₃) for 98% ee (S)]; ¹H-NMR (400 MHz, CDCl₃) δ 1.11 (d, *J* = 6.1 Hz, 3H), 1.62 (brs, 2H, NH₂), 1.60-1.72 (m, 2H), 2.59-2.73 (m, 2H), 2.92 (sext, *J* = 6.1 Hz, 1H), 7.16-7.31 (m, 5H).

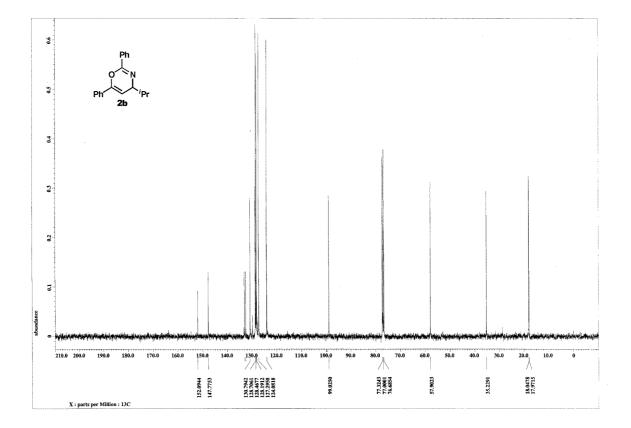




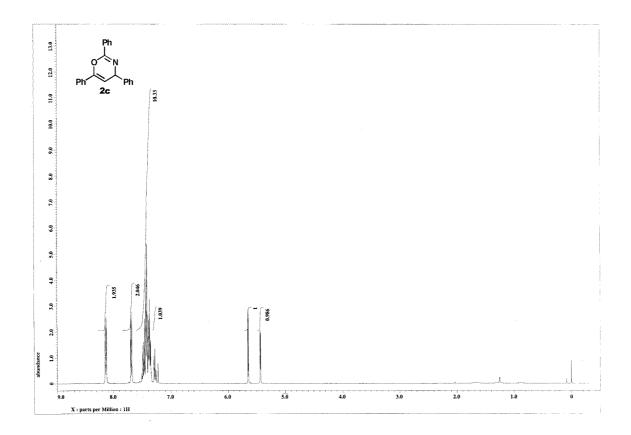


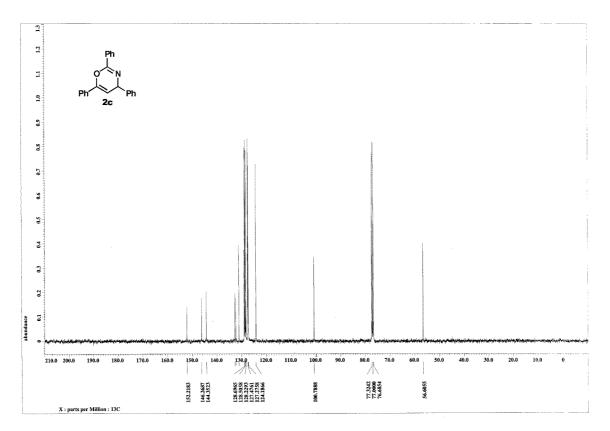
Supplementary Material (ESI) for Chemical Communications This journal is The Royal Society of Chemistry 2009



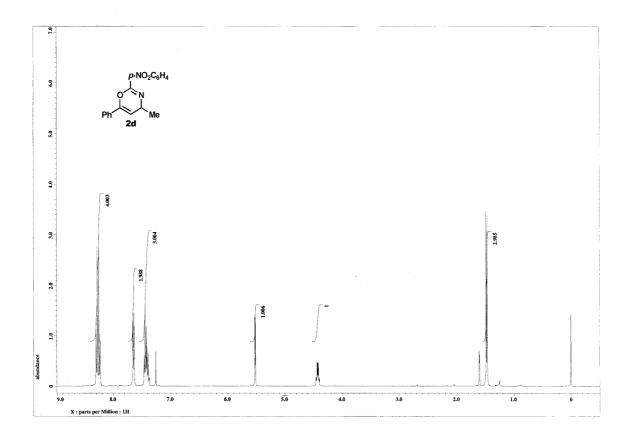


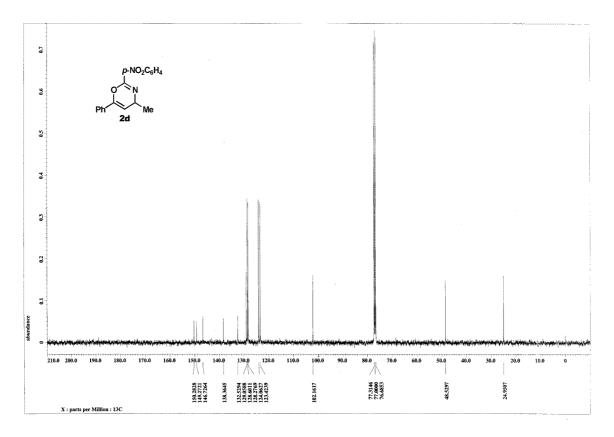
Supplementary Material (ESI) for Chemical Communications This journal is The Royal Society of Chemistry 2009

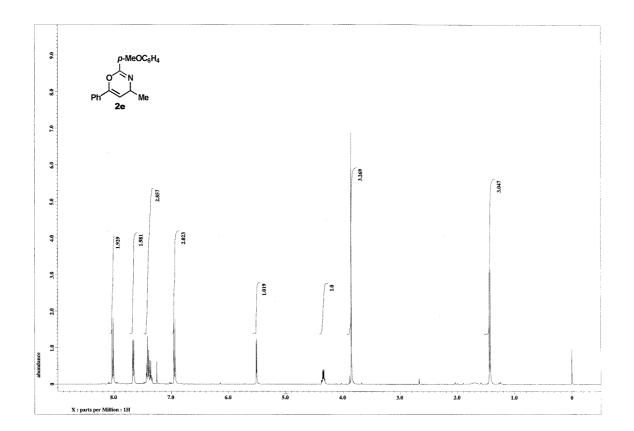


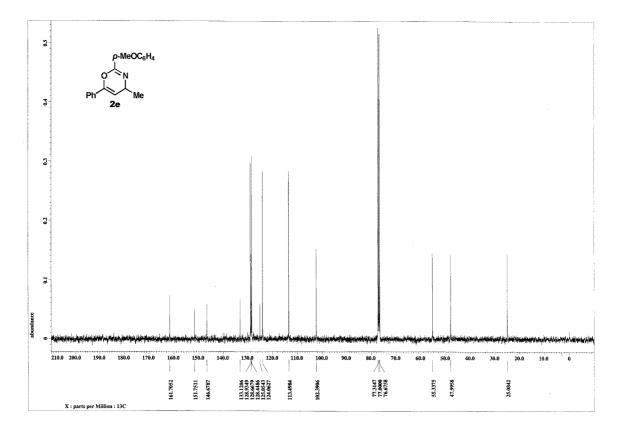


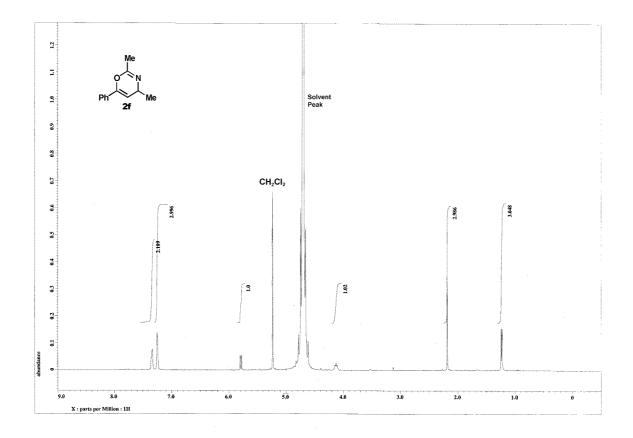
Supplementary Material (ESI) for Chemical Communications This journal is The Royal Society of Chemistry 2009

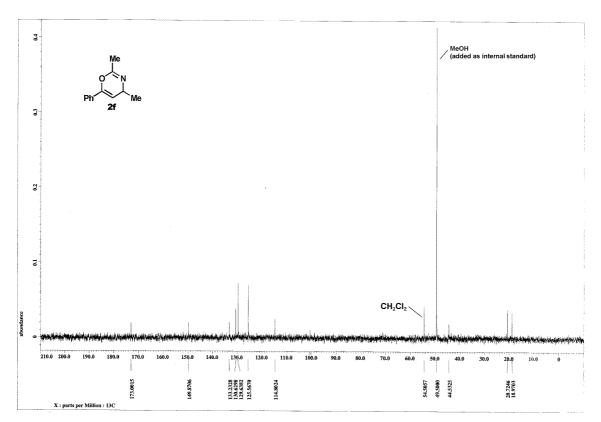




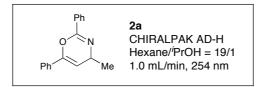








HPLC Traces of Optically Active Compounds



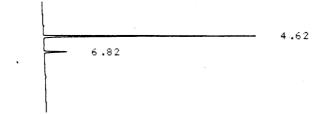
1. Racemic

CH. 1 C.S 2.50 ATT 8 OFFS 0 00/00/00 04:11

00/00/00 04:11 D-2500 TAG: METHOD: 4 CH: 1 0 CONC: AREA FILE: 1 CALC-METHOD: AREA% TABLE: RT AREA CONC ВC NO. 4.61 614007 49.998 88 1 614060 6.70 50.002 BB 2 TOTAL 1228067 100.000 PEAK REJ : 0

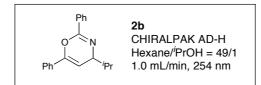
2. Optically active (74% ee)

CH. 1 C.S 2.50 ATT 6 OFFS 0 00/00/00 00:20

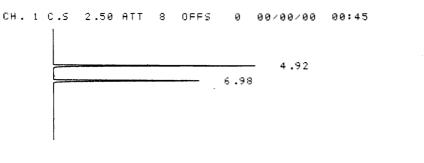


00/00/00 00:20 D-2500 METHOD: TAG: 2 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA NO. AREA CONC ВC RT 87.054 1 4.62 174135 8 B 2 6.82 25895 12.946 88 TOTAL 200030 100.000 РЕАК КЕЈ : 0

Supplementary Material (ESI) for Chemical Communications This journal is © The Royal Society of Chemistry 2009



1. Racemic



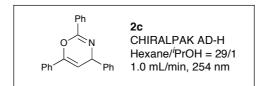
D-2500 00/00/00 00:45 METHOD: TAG: 2 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA CONC BC 50.312 BB 49.688 BB NO. RT AREA 4.92 694697 1 6.98 686082 2 TOTAL 1380779 100.000 РЕАК ВЕЈ : 0

2. Optically active (47% ee)

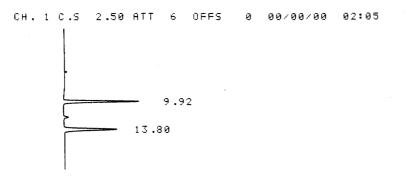
CH. 1 C.S 2.50 ATT 8 OFFS 0 00/00/00 01:14

00/00/00 01:14 D-2500 TAG: METHOD: 2 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA NO. RT AREA CONC BC 73.547 BB 26.453 BB 5.00 646183 1 232415 2 6.99 TOTAL 878598 100.000 PEAK REJ : 0

Supplementary Material (ESI) for Chemical Communications This journal is $\textcircled{\mbox{\scriptsize C}}$ The Royal Society of Chemistry 2009

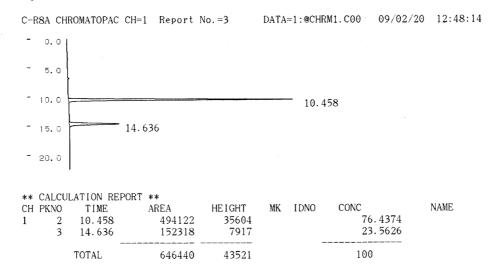


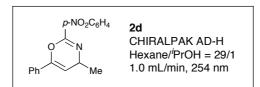
1. Racemic

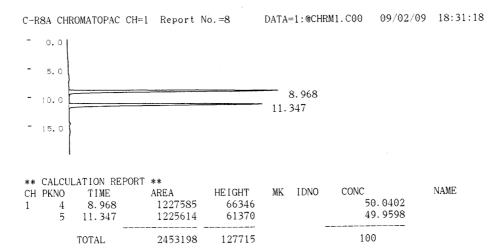


00/00/00 02:05 D-2500 TAG: METHOD: 3 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA NO. RT AREA CONC ВC 9.92 129911 49.917 BВ 1 130344 50.083 88 2 13.80 TOTAL 260255 100.000 РЕАК REJ : 0

2. Optically active [53% ee (S)]





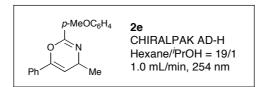


2. Optically active (50% ee)

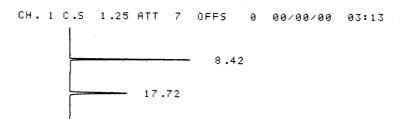
CH. 1 C.S 2.50 ATT 6 OFFS 0 00/00/00 01:04

00/00/00 01:04 D-2500 METHOD: TAG: 2 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA NO. RT AREA CONC ВC 8.74 219736 74.971 8 B 1 25.029 BB 2 11.08 73360 TOTAL 293096 100.000 РЕАК КЕЈ : 0

Supplementary Material (ESI) for Chemical Communications This journal is $\textcircled{\mbox{\scriptsize C}}$ The Royal Society of Chemistry 2009



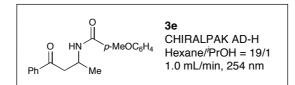
1. Racemic

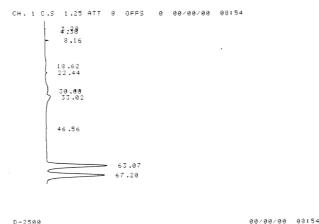


D-2500 00/00/00 03:13 METHOD: TAG: 3 CH: 1 FILE: 1 CALC-METHOD: AREA% TABLE: 0 CONC: AREA NO. RT AREA CONC ВC 8.42 1 344879 50.077 BB 2 17.72 343816 49.923 BB TOTAL 688695 100.000 PEAK REJ : 0

2. Optically active [81% ee (R)]

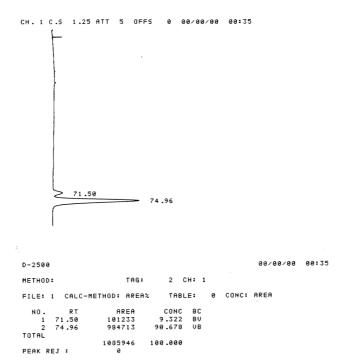
CH. 1 C.S 1.25 ATT 8 OFFS 0 00/00/00 00:44 8.74 19.75 00/00/00 00:44 D-2500 X METHOD: TAG: 2 CH: 1 0 CONC: AREA FILE: 1 CALC-METHOD: AREA% TABLE: AREA CONC BC NO. RT BB 8.74 1775558 90.353 1 2 19.75 189584 9.647 ΒB TOTAL 1965142 100.000 PEAK REJ : 0

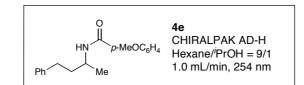


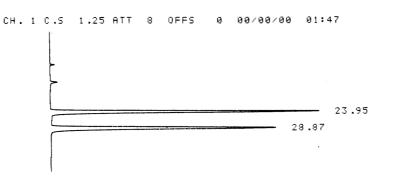


D-2500				00/00/00	08:54
METHOD:	TAG:	10 CH: 1			
FILE: 1 CALC-	METHOD: AREA%	TABLE:	0 CONC:	AREA	
NO. RT	AREA	CONC BC			
1 3.28	4667	0.045 BB			
2 4.38	1106	0.011 BB			
3 8.16	32105	0.310 88			
4 18.62	96053	0.929 88			
5 22.44	36463	0.353 88			
6 30.08	243532	2.354 BV			
7 33.02	547857	5.297 VB			
8 46.56	16521	0.160 BB			
9 63.07	4611172	44.579 BU			
10 67.20	4754272	45.963 UB			
TOTAL					
	10343748	100.000			
PEAK REJ :	0				

2. Optically active [81% ee (R)]







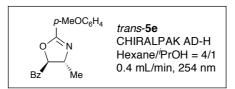
D-2500 00/00/00 01:47 METHOD: TAG: 3 CH: 1 0 CONC: AREA FILE: 1 CALC-METHOD: AREA% TABLE: NO. RT AREA CONC ВC 1 23.95 6179755 49.991 BВ 2 28.87 6182022 50.009 88 TOTAL 12361777 100.000 PEAK REJ : 0

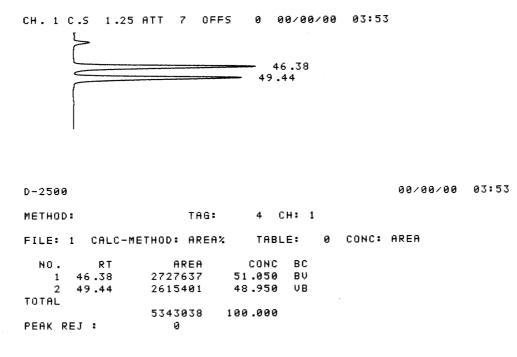
2. Optically active [81% ee (R)]

ž

CH. 1 C.S 1.25 ATT 7 OFFS 0 00/00/00 02:33

D-2500 00/00/00 02:33 METHOD: TAG: 3 CH: 1 FILE: 1 CALC-METHOD: AREA% 0 CONC: AREA TABLE: NO. RT AREA CONC BC 1 19.48 1731279 BB 90.561 2 23.10 180442 9.439 88 TOTAL 1911721 100.000 PEAK REJ : 0

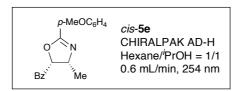




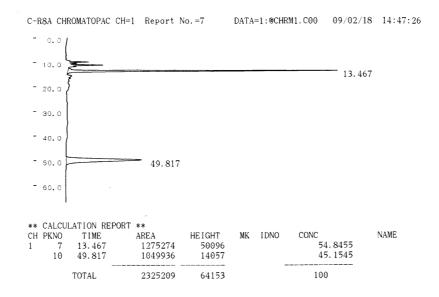
2. Optically active [81% ee (4R, 5S)]

CH. 1 C.S 1.25 ATT 7 OFFS 0 00/00/00 02:43 46.03 49.07 00/00/00 02:43 D-2500 METHOD: TAG: 3 CH: 1 FILE: 1 CALC-METHOD: AREA% 0 CONC: AREA TABLE: NO. RT AREA CONC ВC 46.03 404548 9.505 ΒV 1 2 49.07 3851608 90.495 VВ TOTAL 4256156 100.000 PEAK REJ : 0

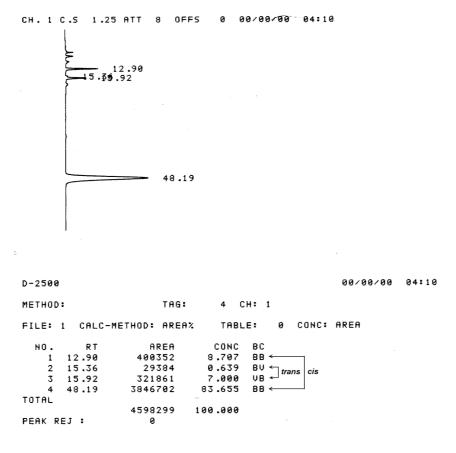
Supplementary Material (ESI) for Chemical Communications This journal is $\textcircled{\mbox{\scriptsize C}}$ The Royal Society of Chemistry 2009



1. Racemic



2. Optically active [81% ee (4R, 5R)]



References

- (1) S. Kotani, S. Hashimoto and M. Nakajima, *Tetrahedron* 2007, **63**, 3122.
- M. Nakajima, Y. Sasaki, M. Shiro and S. Hashimoto, *Tetrahedron: Asymmetry* 1997, 8, 341.
- (3) Y. K. Ramtohul and A. Chartrand, Org. Lett. 2007, 9, 1029.
- (4) Prepared from 4-methyl-2-butanone and methyl benzoate according to the literature procedure, see: T. Ishikawa, R. Kadoya, M. Arai, H. Takahashi, Y. Kaisi, T. Mizuta, K. Yoshikai, S. Saito, *J. Org. Chem.* 2001, **66**, 8000.
- (5) R. F. Klima, A. V. Jadhav, P. N. D. Singh, M. Chang, C. Vanos, J. Sankaranarayanan, M. Vu, N. Ibrahim, E. Ross, S. McCloskey, R. S. Murthy, J. A. Krause, B. S. Ault, and A. D. Gudmundsdóttir, *J. Org. Chem.* 2007, **72**, 6372.
- (6) Y.-D. Lin, J.-Q. Kao and C.-T. Chen, Org. Lett. 2007, 9, 5195.
- (7) M. Terada, K. Machioka and K. Sorimachi, Angew. Chem. Int. Ed. 2006, 45, 2254.
- (8) H.-H. Li, D.-J. Dong and S.-K. Tian, Eur. J. Org. Chem. 2008, 3623.
- (9) (a) K. Masutani, T. Minowa, Y. Hagiwara and T. Mukaiyama, *Bull. Chem. Soc. Jpn.* 2006, **79**, 1106. (b) J. González-Sabín, V. Gotor and F. Rebolledo, *Tetrahedron: Asymmetry* 2002, **13**, 1315.