Supplementary Material

for

Pseudo-Allosteric Regulation of the Anion Binding Affinity of a

Macrocyclic Coordination Complex

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Experimental Section

General Methods and Instrument Details

All reactions were carried out under an inert atmosphere of nitrogen using standard Schlenk techniques or an inert atmosphere glove box unless otherwise noted. Diethyl ether, CH_2Cl_2 , and hexanes were purified by published methods.¹ All solvents were deoxygenated with N₂ bubbling prior to use. Deuterated solvents were purchased from Cambridge Isotope Laboratories Inc. and used as received. [Rh(nbd)₂]BF₄ and [RhCl(nbd)]₂ (nbd = norbornadiene) were purchased from Strem Chemicals and used as received. 4-(2-(Diphenylphosphino)ethylthio)phenylamine was synthesized according to literature methods.² All other chemicals were used as received from Aldrich Chemical Company. ¹H

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NMR and ¹³C{¹H} NMR spectra were recorded on a Varian Mercury 300 MHz FT-NMR spectrometer and referenced relative to residual proton resonances in CD₂Cl₂. ³¹P{¹H} NMR spectra were recorded on a Varian Mercury 300 MHz FT-NMR spectrometer at 121.53 MHz and referenced relative to an external 85% H₃PO₄ standard. All chemical shifts are reported in ppm. UV-vis spectra were recorded on a Varian Cary 50 Bio spectrophotometer in CH₂Cl₂. Electrospray ionization mass spectrometer, Elemental analyses were performed by Quantitative Technologies Inc. Whitehouse, NJ, USA.

Synthetic Methods

[**Rh**(**nbd**)₂]**B**(**C**₆**F**₅)₄. NBD (460.5 mg, 5.0 mmol) and LiB(C₆F₅)₄·Et₂O (380.0 mg, 0.5 mmol) in CH₂Cl₂ (2 mL) were added to a solution of [RhCl(nbd)]₂ (115.3 mg, 0.25 mmol) in CH₂Cl₂ (3 mL). The solution was stirred for 2 days before the resulting mixture was filtered through Celite. The crude product was purified by recrystallization twice from CH₂Cl₂, Et₂O, and hexanes (315.6 mg, 65%). ¹H NMR (CD₂Cl₂): δ 5.61 (s, olefinic, 4H), 4.30 (s, CH, 2H), 1.69 (s, CH₂, 2H). ESI-MS(*m*/*z*): [M-B(C₆F₅)₄]⁺ = 287.0, Calcd for [C₁₄H₁₆Rh]⁺ = 287.2. Anal. Calcd for C₃₈H₁₆BF₂₀Rh: C, 47.24; H, 1.67, Found: C, 47.12; H, 1.23.

N,N'-Bis{4-(2-diphenylphosphanylethylthio)phenyl}pyridine-2,6- dicarboxamide (5). To a CH₂Cl₂ (5 mL) solution of 2,6-pyridinedicarbonyl dichloride (102.0 mg, 0.5 mmol) and NEt₃ (0.17 mL, 1.2mmol) was added 4-(2-(diphenylphosphino)ethylthio)phenylamine (337.4 mg, 1.0 mmol) at 0 °C. The resulting solution was stirred for 2 days at room temperature before aqueous NaHCO₃ solution was added. The organic layer was extracted with CH₂Cl₂, washed with brine, and dried over MgSO₄. The solvent was removed under reduced pressure, and the resulting material was purified by silica gel chromatography (CH₂Cl₂:EtOAc = 40:1). Analytically pure product was obtained by recrystallization from

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CH₂Cl₂ and hexanes (302.4 mg, 75%). ¹H NMR (CD₂Cl₂): δ 9.53 (s, N-H, 2H), 8.44 (d, disubstituted pyridine, 2H, *J* = 8 Hz), 8.13 (t, disubstituted pyridine, 1H, *J* = 8 Hz), 7.70 (d, C₆*H*₄, 4H, *J* = 8 Hz), 7.40-7.28 (m, C₆*H*₄ and *PPh*₂, 24H), 2.96 (m, OC*H*₂, 4H), 2.37 (m, C*H*₂PPh₂, 4H). ¹³C{¹H} NMR (CD₂Cl₂): 161.6 (s), 149.4 (s), 140. 3 (s), 138.4 (d, *J*_{C-P} = 14 Hz), 136.3 (s), 133.2 (d, *J*_{C-P} = 19 Hz), 132.3 (s), 131.2 (s), 129.3 (s), 129.1 (d, *J*_{C-P} = 7 Hz), 126.1 (s), 121.2 (s), 31.3 (d, *J*_{C-P} = 22 Hz), 28.6 (d, *J*_{C-P} = 15 Hz). ³¹P{¹H} NMR (CD₂Cl₂): δ -16.3 (s). Anal. Calcd for C₄₇H₄₁N₃O₂P₂S₂: C, 70.04; H, 5.13; N, 5.21, Found: C, 69.69; H, 4.95; N, 5.16.

Closed macrocyclic complex – $B(C_6F_5)_4$ (6a). To a CH₂Cl₂ (15 mL) solution of [Rh(nbd)₂]B(C₆F₅)₄ (96.6 mg, 0.10 mmol) was added drop wise compound **5** (80.6 mg, 0.10 mmol) in CH₂Cl₂ (15 mL) over 10 min. The resulting solution was stirred for 6 h before the solvent was evaporated. The product was recrystallized from CH₂Cl₂/hexanes (152.7 mg, 96%). ¹H NMR (CD₂Cl₂): δ 9.50 (s, N-H, 4H), 8.34 (d, disubstituted pyridine, 4H, J = 8 Hz), 8.05 (t, disubstituted pyridine, 2H, J = 8 Hz), 7.72 (d, C₆H₄, 8H, J = 8 Hz), 7.44-7.27 (m, PPh₂ and C₆H₄, 48H), 2.77 (m, OCH₂, 8H), 2.56 (m, CH₂PPh₂, 8H). ³¹P{¹H} NMR (CD₂Cl₂): δ 65.1 (d, J_{Rh-P} = 162 Hz). ESI-MS(m/z): [M-2B(C₆F₅)₄]²⁺ = 908.6, Calcd for [C₉₄H₈₂N₆O₄P₄Rh₂S₄]²⁺ = 908.1. Anal. Calcd for C₁₄₂H₈₂B₂F₄₀N₆O₄P₄Rh₂S₄: C, 53.70; H, 2.60; N, 2.65, Found: C, 53.66; H, 2.24; N, 2.55.

Closed macrocyclic complex – **BF**₄ (**6b**). To a CH₂Cl₂ (50 mL) solution of [Rh(nbd)₂]BF₄ (56.1 mg, 0.15 mmol) was added drop wise compound **5** (120.9 mg, 0.15 mmol) in CH₂Cl₂ (50 mL) over 10 min. The resulting solution was stirred for 6 h before the solvent was evaporated. The product was recrystallized from CH₂Cl₂/hexanes (147.7 mg, 99%). ¹H NMR (CD₂Cl₂): δ 9.92 (s, N-H, 4H), 8.28 (d, disubstituted pyridine, 4H, *J* = 8 Hz), 7.97 (d, C₆H₄, 8H, *J* = 8 Hz), 7.87 (t, disubstituted pyridine, 2H, *J* = 8 Hz), 7.47-7.29 (m, PPh₂ and C₆H₄, 48H), 2.74 (m, OCH₂, 8H), 2.57 (m, CH₂PPh₂, 8H). ³¹P{¹H} NMR (CD₂Cl₂): δ 65.3

(d, $J_{Rh-P} = 161$ Hz). ESI-MS(m/z): $[M-BF_4]^+ = 1904.4$, Calcd for $[C_{94}H_{82}BF_4N_6O_4P_4Rh_2S_4]^+ = 1904.2$. Anal. Calcd for $C_{94}H_{82}B_2F_8N_6O_4P_4Rh_2S_4\cdot CH_2Cl_2$: C, 54.96; H, 4.08; N, 4.05, Found: C, 54.94; H, 3.90; N, 4.12.

Closed macrocyclic complex – Cl (7). Compound 5 (40.3 mg, 0.050 mmol) in CH₂Cl₂ (5 mL) was added drop wise over 10 min to a CH₂Cl₂ (5 mL) solution of [RhCl(CO)₂]₂ (9.7 mg, 0.025 mmol). The resulting solution was stirred for 30 min before the solvent was evaporated. The product was recrystallized from CH₂Cl₂/hexanes (48.2 mg, 97%). ¹H NMR (CD₂Cl₂): δ 11.40 (s, N-H, 4H), 8.01 (m, disubstituted pyridine and C₆H₄, 12H), 7.70 (m, disubstituted pyridine, 2H, *J* = 8 Hz), 7.23 (m, PPh₂, 40H), 7.19 (d, C₆H₄, 8H, *J* = 8 Hz), 2.78 (m, OCH₂, 8H), 2.60 (m, CH₂PPh₂, 8H). ³¹P{¹H} NMR (CD₂Cl₂): δ 65.7 (d, *J*_{Rh-P} = 162 Hz). ESI-MS(*m*/*z*): [M-Cl]⁺ = 1851.3, Calcd for [C₉₄H₈₂ClN₆O₄P₄Rh₂S₄]⁺ = 1851.2. Anal. Calcd for C₉₄H₈₂Cl₂N₆O₄P₄Rh₂S₄·CH₂Cl₂: C, 57.82; H, 4.29; N, 4.26, Found: C, 58.18; H, 4.38; N, 4.30.

Open macrocyclic complex (8). To a CH₂Cl₂ (5 mL) solution of complex **6a** (63.5 mg, 0.02 mmol) was added ^{*i*}BuNC (19.9 mg, 0.24 mmol) in CH₂Cl₂ (5 mL). The resulting solution was stirred for 2 h before the volatiles were evaporated. The product was recrystallized from CH₂Cl₂/hexanes (63.1 mg, 99%). ¹H NMR (CD₂Cl₂): δ 9.61 (s, N-H, 4H), 8.38 (d, disubstituted pyridine, 4H, *J* = 8 Hz), 8.07 (t, disubstituted pyridine, 2H, *J* = 8 Hz), 7.65 (d, C₆H₄, 8H, *J* = 8 Hz), 7.42 (br, PPh₂, 40H), 7.16 (d, C₆H₄, 8H, *J* = 8 Hz), 2.89 (m, OCH₂, 8H), 2.57 (m, CH₂PPh₂, 8H). ³¹P{¹H} NMR (CD₂Cl₂): δ 18.2 (br). ESI-MS(*m*/*z*): [M-2B(C₆F₅)₄-2(^{*i*}BuNC)]²⁺ = 991.2, Calcd for [C₁₀₄H₁₀₀N₈O₄P₄Rh₂S₄]²⁺ = 991.2. Anal. Calcd for C₁₆₂H₁₁₈B₂F₄₀N₁₀O₄P₄Rh₂S₄: C, 55.46; H, 3.39 N, 3.99, Found: C, 55.54; H, 3.12; N, 3.90.³

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X-ray crystallography. Crystals of **6a** and **7** were mounted on fiber loops and cooled to 100 K. In both cases the centrosymmetric space group option was selected based on the results of the refinement. The structures were solved by direct methods, refined with anisotropic thermal parameters, and contained idealized hydrogen atoms, except for the fractional and disordered CH_2Cl_2 molecule in **6a** and **7**. All software is contained in the SMART, SAINT and SHELXTL software libraries of the Bruker XRD corporation. Disordered CH_2Cl_2 , the recrystallization solvent, was rendered using SQUEEZE. Found 623e/uc. Calculated for 16 CH_2Cl_2 , 672e/uc. Solvent is included in the formula and intensive properties.

Determination of the binding constant. The binding constants of **6a** and **8** for Cl⁻, respectively, were determined by UV-Vis spectroscopy in CH₂Cl₂ (303 nm for **6a** and 312 nm for **8**), employing the titration methodology (Figure S-1 and S-2). (^{*n*}Bu)₄NCl was used as the Cl⁻ source. The concentration of the complex was $5 \times 10^{-4} \text{ M}^{-1}$. The data were modeled with the nonlinear curve-fitting program Dynafit⁴ (Because the experiments were carried out by UV-vis spectroscopy and the binding constants were based upon small changes in absorbance, there could be significant error associated with these values. However, there is no doubt that the association constant of complex **8** is larger that for complex **6a**.).



Fig. S-1 Titration plot of **6a** with Cl⁻

Fig. S-2 Titration plot of 8 with Cl⁻

The Job plot for a determination of stoichiometry. Figure S-3 shows a Job plot which is based upon ¹H NMR chemical shift data for the amide N-H signal in CD_2Cl_2 for macrocycle 8 complexing Cl⁻ ([8] + [(^{*n*}Bu)₄NCl] = 10 mM).⁵



Fig. S-3 Job plot for complexation of 8 with Cl⁻

References

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