

Encapsulated Binding Sites – Synthetically Simple Receptors for the Binding and Transport of HCl

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SUPPLEMENTARY INFORMATION

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1. Synthesis and Characterisation

(3,4-Bis-benzyloxy-phenyl)-acetic acid, an important building block in the synthesis of receptor 1 was synthesised according to literature methods.¹ Receptors **3**,² **7**,³ and **8**⁴ were also synthesised according to literature methods.

General synthesis method 1: To a solution of tris-(2-aminoethyl)-amine (1.02 mL, 6.9 mmol) in EtOAc, the appropriate acid anhydride (26.7 mmol) and pyridine (1.66 mL, 20.1 mmol) were added. The reaction was left to stir overnight and the solvent was removed under vacuum.

General synthesis method 2: To a solution of tris-(2-aminoethyl)-amine (1.02 mL, 6.9 mmol) and Et₃N (3.5 mL) in 100 ml of CH₂Cl₂ at 10°C was added the appropriate acid chloride (24 mmol) with stirring. The reaction was left to stir overnight and then poured into 200 mL of water. The aqueous layer was extracted with CH₂Cl₂ (2 x 100 mL). The organic layers were dried with MgSO₄, filtered and the solvent removed under vacuum.

Receptor 1. (3,4-Bis-benzyloxy-phenyl)-acetic acid (1.30 g, 3.7 mmol) was dissolved in CH₂Cl₂ (~30 mL). DCC (1.55 g, 7.5 mmol) and HOBT (1.01 g, 7.5 mmol) were added to the acid as a mixture of solids. The mixture was left to cool to 0°C for 10 min in an ice bath with a N₂ balloon. In a separate beaker, tris-(2-aminoethyl)-amine (0.19 mL, 1.25 mmol) and Et₃N (0.52 mL, 3.7 mmol) were dissolved in CH₂Cl₂ (~30 mL). This was added to the acid mixture and stirred for 3 d. The reaction mixture was filtered. The filtrate was evaporated to dryness under vacuum. The product was purified by column chromatography with a starting eluent of 99:1 CH₂Cl₂:MeOH, gradient elution was employed until a solvent ratio of 92:8 was reached. A small amount of Et₃N was added to the eluent. The product was obtained as a white solid in a 56% yield. m.p. 142°C. *R*_f 0.61 (9:1 CH₂Cl₂:MeOH). ¹H NMR (CDCl₃): δ 7.48-7.35 (12H, m, Ar H), 7.34-7.27 (18H, m, Ar H), 6.90 (3H, d, J = 1.8 Hz, Ar H), 6.83 (3H, d, J = 7.9 Hz, Ar H), 6.74 (3H, dd, J = 6.4, 1.8 Hz, Ar H), 6.32 (3H, t, J = 5.5 Hz, NH), 5.07 (6H, s, CH₂-O), 5.06 (6H, s, CH₂-O), 3.41 (6H, s, CH₂-C=O), 3.08 (6H, q, J = 4.9 Hz, NH-CH₂), 2.37 (6H, t, J = 5.5 Hz, CH₂). ¹³C NMR

(CDCl₃): δ 171.9 (C=O), 149.2 (Ar C-O), 148.3 (Ar C-O), 137.4 (Ar C-CH₂), 137.3 (Ar C-CH₂), 128.6 (Ar C), 128.5 (Ar C), 127.9 (Ar C), 127.5 (Ar C), 127.4 (Ar C), 122.4 (Ar C), 116.1 (Ar C), 115.4 (Ar C), 71.4 (CH₂-O), 71.2 (CH₂-O), 54.9 (CH₂-N), 43.1 (CH₂-C=O), 38.1 (NH-CH₂). ν_{\max} (solid): 3287m (N-H), 3063w (Ar C-H), 3032w (Ar C-H), 2932w (C-H), 2862w (C-H), 1639s (C=O), 1589m, 1543m, 1508s, 1454m, 1427m, 1381m, 1339m, 1258s, 1223s, 1157m, 1134s, 1080w, 1007s. ESIMS m/z : 1159 ([M+Na]⁺, 100%). HRESIMS: Calculated for C₇₂H₇₃N₄O₉, 1137.5372; found, 1137.5329.

Receptor 2. 3,4-Dimethoxyphenyl acetic acid (1.00 g, 5.1 mmol) was dissolved in CH₂Cl₂ (~30 mL). DCC (2.10 g, 10.2 mmol) and HOBt (1.38 g, 10.2 mmol) were added to the acid as a mixture of solids. The mixture was left to cool to 0°C for 10 min in an ice bath with a N₂ balloon. In a separate beaker, tris-(2-aminoethyl)-amine (0.25 mL, 1.7 mmol) and Et₃N (0.71 mL, 5.1 mmol) were dissolved in CH₂Cl₂ (~30 mL). This was added to the acid mixture and stirred for 3 d. The reaction mixture was filtered. The filtrate was evaporated to dryness under vacuum. The product was purified by column chromatography (CH₂Cl₂:MeOH 98:2). A small amount of Et₃N was added to the eluent. The product was obtained as a white solid in a 42% yield. m.p. 123°C. R_f 0.53 (CH₂Cl₂:MeOH 9:1). ¹H NMR (DMSO-d₆): δ 7.86 (3H, t, J = 5.5 Hz, NH), 6.85-6.79 (6H, m, Ar H), 6.74 (3H, dd, J = 8.2, 1.2 Hz, Ar H), 3.69 (9H, s, OCH₃), 3.68 (9H, s, OCH₃), 3.30 (6H, s, CH₂-C=O), 3.04 (6H, q, J = 6.1 Hz, NHCH₂), 2.45 (6H, t, J = 6.7 Hz, CH₂N). ¹³C NMR (CDCl₃): δ 171.9 (C=O), 148.9 (Ar C-O), 148.0 (Ar C-O), 127.8 (Ar C), 121.3 (Ar C), 112.4 (Ar C), 111.3 (Ar C), 55.81 (O-CH₃), 55.78 (O-CH₃), 54.6 (CH₂-N), 42.9 (CH₂C=O), 37.9 (NH-CH₂). ν_{\max} (solid): 3283s (N-H), 2936m (C-H), 2835m (C-H), 1639s (C=O), 1593m, 1543m, 1512s, 1450m, 1420m, 1342m, 1261s, 1231s, 1138s, 1026s. ESIMS m/z : 703 ([M+Na]⁺, 100%). HRCIMS: Calculated for C₃₆H₄₉N₄O₉, 681.3500; found, 681.3509.

Receptor 4. Using general synthesis method 1 and propionic anhydride (3.44 mL) and purifying the product by washing with diethyl ether gave receptor **4** as a cream solid in a 21% yield. m.p. 80°C. ¹H NMR (CDCl₃): δ 6.50 (3H, bt, NH), 3.28 (6H, bq, NHCH₂), 2.54 (6H, bt, CH₂N), 2.24 (6H, q, J = 7.6 Hz, CH₂CO), 1.15 (9H, t, J = 7.3 Hz, CH₃). ¹³C NMR (CDCl₃): δ 174.8 (C=O), 54.6 (CH₂N), 37.8 (NHCH₂), 29.6

(COCH₂), 10.1 (CH₃). ν_{\max} (solid): 3294*m* (N-H), 2970*m* (C-H), 2936*m* (C-H), 2878*w* (C-H), 1643*s* (C=O), 1543*s*, 1454*m*, 1361*m*, 1319*w*, 1258*w*, 1234*m*, 1165*m*, 1049*m*, 999*m*. ESIMS *m/z*: 315 ([M+H]⁺, 100%). HRESIMS: Calculated for C₁₅H₃₁N₄O₃, 315.2391; found, 315.2396.

Receptor 5. Using general synthesis method 1 and butyric anhydride (4.3 mL) and purifying the product by washing with diethyl ether gave receptor **5** as a white solid in a 34% yield. m.p. 96°C. ¹H NMR (CDCl₃): δ 6.60 (3H, bt, NH), 3.28 (6H, bq, NHCH₂), 2.53 (6H, bt, CH₂N), 2.19 (6H, t, J = 7.3 Hz, CH₂CO), 1.67 (6H, sex, J = 7.6 Hz, CH₃CH₂), 0.94 (9H, t, J = 7.6 Hz, CH₃). ¹³C NMR (CDCl₃): δ 174.0 (C=O), 54.6 (CH₂N), 38.6 (NHCH₂), 37.7 (COCH₂), 19.3 (CH₂), 14.0 (CH₃). ν_{\max} (solid): 3294*m* (N-H), 2958*m* (C-H), 2932*m* (C-H), 2873*w* (C-H), 1649*s* (C=O), 1543*s*, 1423*m*, 1366*m*, 1319*m*, 1288*m*, 1250*m*, 1211*m*, 1165*m*, 1072*w*, 1034*w*, 995*w*. ESIMS *m/z*: 357 ([M+H]⁺, 100%). HRESIMS: Calculated for C₁₈H₃₇N₄O₃, 357.2860; found, 357.2860.

Receptor 6. Using general synthesis method 2 and valeroyl chloride (2.8 mL) and purifying the product by recrystallisation from ethyl acetate gave receptor **6** as a white solid in a 27% yield. m.p. 99°C. ¹H NMR (CDCl₃): δ 6.57 (3H, bt, NH), 3.28 (6H, bq, NHCH₂), 2.53 (6H, bt, CH₂N), 2.21 (6H, t, J = 7.6 Hz, CH₂CO), 1.61 (6H, quin, J = 7.6 Hz, CH₂), 1.34 (6H, sex, J = 7.6 Hz, CH₃CH₂), 0.91 (9H, t, J = 7.3 Hz, CH₃). ¹³C NMR (CDCl₃): δ 174.1 (C=O), 54.6 (CH₂N), 37.6 (NHCH₂), 36.3 (COCH₂), 28.0 (CH₂), 22.6 (CH₂), 13.9 (CH₃). ν_{\max} (solid): 3294*m* (N-H), 2955*m* (C-H), 2932*m* (C-H), 2866*m* (C-H), 2828*w* (C-H), 1639*s* (C=O), 1543*s*, 1454*m*, 1423*m*, 1381*m*, 1354*m*, 1319*w*, 1258*m*, 1231*m*, 1204*w*, 1165*m*, 1072*m*, 999*m*. ESIMS *m/z*: 399 ([M+H]⁺, 100%). HRESIMS: Calculated for C₂₁H₄₃N₄O₃, 399.3330; found, 399.3324.

2a Job Plot Data

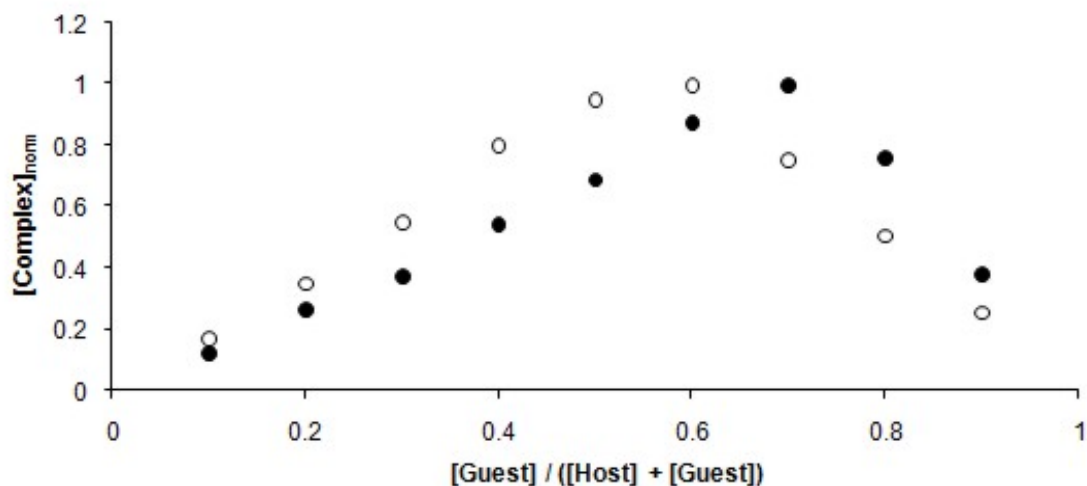
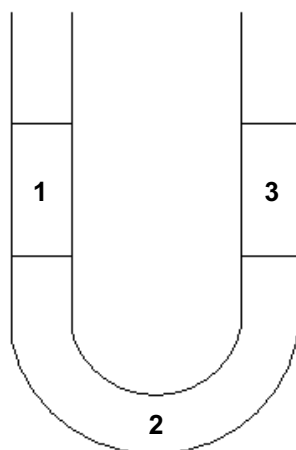


Figure S1. Job plot analysis for compound **2** on the addition of HCl (2.0 mM in diethyl ether) (black circles) or HBF_4 (54% wt. sol. in diethyl ether) (white circles) in $\text{CD}_3\text{CN}:\text{DMSO-d}_6$ (9:1), demonstrating different stoichiometries for the interaction of this receptor with the different acids.

2b U-Tube Method



All U-tube experiments were run in the same glassware with three separate solutions; 15.5 mL of 0.01 M transporter in dichloromethane (solution **2**), 6 mL of HCl (pH 0.9) (solution **1**) in deionised water and 6 mL of deionised water (solution **3**). The

organic phase was stirred using a custom made magnetic stirrer bar at a setting of 3.5 using a 728 Metrohm stirrer. The pH of the neutral phase was monitored using a 702 SM Titrino Metrohm potentiometer with a combined glass electrode taking readings every 10 minutes for 240 minutes, after an initial lag-time of 1 minute.

3. References

1. P. F. Kador, R. Venkatraman, D. F. Feller and D. D. Miller, *J. Med. Chem.*, 1977, **20**, 891-894.
2. K. E. Krakowiak, J. S. Bradshaw and R. M. Izatt, *J. Org. Chem.*, 1989, **55**, 3364-3368.
3. S. Valiyaveetil, J. F. L. Engbersen, W. Verboom and D. N. Reinhoudt, *Angew. Chem. Int. Ed.*, 1993, **32**, 900-901.
4. P. D. Beer, P. K. Hopkins and J. D. McKinney, *Chem. Commun.* 1999, 1253-1254.