

– Electronic Supplementary Material (ESI) –

Iron Catalyzed Oxidative Cyanation of Tertiary Amines

Wei Han and Armin R. Oftal*

*Department Chemie und Biochemie
Ludwig-Maximilians-Universität München
Butenandtstraße 5-13 (Haus F)
81377 München (Germany)
Fax: (+49) 89-2180-9977715
E-mail: oftal@lmu.de*

General. All reactions were carried out under an atmosphere of dry nitrogen. ^1H and ^{13}C NMR spectra of solutions in CDCl_3 were recorded on 200, 300, or 400 MHz NMR spectrometers. Chemical shifts were expressed in parts per million (ppm) downfield from tetramethylsilane and refer to the solvent signals (δ_{H} 7.24 and δ_{C} 77.0 ppm). Abbreviations for signal couplings are: s, singlet; d, doublet; t, triplet; m, multiplet. HRMS was performed on a Finnigan MAT 95Q mass spectrometer. Infrared spectra of neat substances were recorded on a Perkin-Elmer Spectrum BX II FT-IR spectrometer equipped with an ATR probe (diamond).

Materials. Commercially available tertiary amines were used as received. *N,N*-Dimethyl-*p*-anisidine,^{S1} 1-phenylpiperidine,^{S2} 2-phenyl-1,2,3,4-tetrahydroisoquinoline^{S2} and 2-(4-methoxyphenyl)-1,2,3,4-tetrahydroisoquinoline^{S2} were prepared according to literature procedures.

The following iron and copper salts were used: Iron(II) acetate (anhydrous, 97 %, Strem), iron(III) acetylacetone (99.9 %, Aldrich), iron(III) bromide (99 %, ABCR), iron(II) chloride (98 %, Aldrich), iron(III) chloride (anhydrous, 97 %, Grüssing), iron(III) chloride hexahydrate (99 %), iron(II) fluoride (anhydrous, 99 %, Strem), iron(II) gluconate hydrate (purum p.a., Fluka), iron(II) sulfate (99 %), copper(I) bromide (98 %, Acros), and copper(II) bromide (> 99%, Acros).

Trimethylsilyl cyanide (98 %, Acros) and *tert*.-butyl hydroperoxide (5.5 M solution in decane, purum, Aldrich) were purchased.

Typical Procedure for Iron-Catalyzed Cyanation of Tertiary Amines. Under an atmosphere of dry N_2 , a 25 mL Schlenk flask was charged with iron(II) chloride (10 mol-%, 13 mg). The tertiary amine (1.0 mmol), trimethylsilyl cyanide (2.0 mmol, 0.27 mL), and MeOH (2.0 mL) were added successively by syringe. To the mixture was added dropwise *tert*.-butyl hydroperoxide (2.5 mmol, 0.470 mL, 5.5 M solution in decane) over a period of 5 min. The mixture was stirred at room temperature for the indicated time. At the end of the reaction, the reaction mixture was poured into a saturated aqueous NaCl solution (20 mL) and extracted with CH_2Cl_2 (3×20 mL). The organic phases were combined, and the volatile components were evaporated in a rotary evaporator. The crude product was purified by column chromatography on silica gel (*n*-pentane/diethyl ether = 15:2, v/v).

(S1) J. A. Hodges and R. T. Raines, *Org. Lett.*, 2006, **8**, 4695-4697.

(S2) Z. P. Li, S. D. Bohle and C. J. Li, *Proc. Natl. Acad. Sci. USA*, 2006, **103**, 8928–8933.

[(4-Methoxyphenyl)-methyl-amino]acetonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 400 \text{ MHz})$ 2.90 (s, 3 H), 3.76 (s, 3 H), 4.06 (s, 2 H), 6.86 ppm (s, 4 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 39.9, 43.9, 55.5, 114.7, 115.4, 117.8, 142.1, 154.3 ppm; HRMS *m/z* (EI) 176.0944, C₁₀H₁₂N₂O requires 176.0950.

(Methyl-*p*-tolyl-amino)acetonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 2.29 (s, 3 H), 2.96 (s, 3 H), 4.12 (s, 2 H), 6.80 (d, *J* 8.5 Hz, 2 H), 7.12 ppm (d, *J* 8.5 Hz, 2 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 20.3, 39.4, 42.7, 115.3, 115.4, 129.8, 129.9, 145.6 ppm; HRMS *m/z* (EI) 160.0990, C₁₀H₁₂N₂ requires 160.1001.

(Methyl-*m*-tolyl-amino)acetonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 2.31 (s, 3 H), 2.98 (s, 3 H), 4.25 (s, 2 H), 6.66-6.68 (m, 2 H), 6.74 (d, *J* 6.0 Hz, 1 H), 7.16-7.22 ppm (m, 1 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 21.8, 39.2, 42.3, 112.0, 115.5, 115.6, 121.1, 129.2, 139.2, 147.8 ppm; HRMS *m/z* (EI) 160.0990, C₁₀H₁₂N₂ requires 160.1001.

(Methyl-*o*-tolyl-amino)acetonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 2.29 (s, 3 H), 2.85 (s, 3 H), 3.84 (s, 2 H), 7.03-7.09 (m, 1 H), 7.17-7.22 ppm (m, 3 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 17.7, 41.0, 45.0, 115.6, 120.6, 124.9, 126.8, 131.3, 132.8, 148.4 ppm; HRMS *m/z* (EI) 160.0995, C₁₀H₁₂N₂ requires 160.1001.

(Methyl-phenyl-amino)acetonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 3.00 (s, 3 H), 4.16 (s, 2 H), 6.85-6.94 (m, 3 H), 7.28-7.33 ppm (m, 2 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 39.3, 42.3, 114.9, 115.4, 120.2, 129.4, 147.7 ppm; HRMS *m/z* (EI) 146.0841, C₉H₁₀N₂ requires 146.0844.

(S3) S. I. Murahashi, T. Nakae, H. Terai and N. Komiya, *J. Am. Chem. Soc.*, 2008, **130**, 11005–11012.

[(4-Bromophenyl)-methyl-amino]acetonitrile. The reaction mixture was heated to reflux. Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 2.98 (s, 3 H), 4.13 (s, 2 H), 6.72 (d, J 9.0 Hz, 2 H), 7.37-7.40 ppm (m, 2 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 39.3, 42.2, 112.6, 115.1, 116.4, 132.2, 146.7 ppm; HRMS m/z (EI) 223.9943, $\text{C}_9\text{H}_9\text{BrN}_2$ requires 223.9949.

[(2-Bromophenyl)-methyl-amino]acetonitrile

Viscous oil; ν_{max} (neat/ATR probe)/cm⁻¹ 2956, 2887, 2802, 2232 (C≡N), 1586, 1473, 1439, 1415, 1337, 1325, 1222, 1177, 1111, 1023, 987, 919, 871, 766, 752, 272, 722 and 653; $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 2.92 (s, 3 H), 4.07 (s, 2 H), 6.98-7.04 (m, 1 H), 7.25-7.35 (m, 2 H), 7.55-7.58 ppm (m, 1 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 40.6, 44.8, 115.1, 119.7, 122.8, 126.3, 128.5, 133.9, 147.3 ppm; HRMS m/z (EI) 223.9942, $\text{C}_9\text{H}_9\text{BrN}_2$ requires 223.9949; Anal. Calcd for $\text{C}_9\text{H}_9\text{BrN}_2$: C, 48.03; H, 4.03; N, 12.45. Found: C, 48.10; H, 3.70; N, 12.41.

[(4-Ethynylphenyl)-methyl-amino]acetonitrile. The reaction mixture was heated to reflux.

Viscous oil; ν_{max} (neat/ATR probe)/cm⁻¹ 3279 (≡C-H), 3042, 2916, 2823, 2240 (C≡N), 2101 (C≡C), 1671, 1605, 1512, 1360, 1332, 1246, 1181, 1113, 997, 924 and 819; $\delta_{\text{H}}(\text{CDCl}_3, 200 \text{ MHz})$ 2.99 (s, 1 H), 3.03 (s, 3 H), 4.18 (s, 2 H), 6.75 (d, J 9.0 Hz, 2 H), 7.42 ppm (d, J 9.0 Hz, 2 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 37.7, 40.5, 76.0, 84.7, 113.5, 113.9, 115.1, 133.5, 148.5 ppm; HRMS m/z (EI) 170.0839, $\text{C}_{11}\text{H}_{10}\text{N}_2$ requires 170.0844.

[Methyl-(4-nitrophenyl)-amino]acetonitrile. The reaction mixture was heated to reflux.

Known compound; the NMR spectroscopic data agree with those given in lit.^{S4}. Solid, mp 112.5-113.4 °C (lit.,^{S4} 114 °C); $\delta_{\text{H}}(\text{CDCl}_3, 400 \text{ MHz})$ 3.17 (s, 3 H), 4.26 (s, 2 H), 6.79 (d, J 8.0 Hz, 2 H), 8.19 ppm (d, J 8.0 Hz, 2 H); $\delta_{\text{C}}(\text{CDCl}_3, 100 \text{ MHz})$ 39.3, 41.1, 112.3, 114.5, 126.0, 139.9, 152.0 ppm; HRMS m/z (EI) 191.0691, $\text{C}_9\text{H}_9\text{N}_3\text{O}_2$ requires 191.0695.

[Methyl-(3-nitrophenyl)-amino]acetonitrile

Solid, mp 99.3-99.5 °C (from Et₂O/pentane); ν_{max} (neat/ATR probe)/cm⁻¹ 3099, 3042, 2920, 2240 (C≡N), 1617, 1520, 1337, 1259, 1221, 1132, 1015, 881, 860, 789, 782, 737 and 667; $\delta_{\text{H}}(\text{CDCl}_3, 300 \text{ MHz})$ 3.11 (s, 3 H), 4.25 (s, 2 H), 7.09-7.13 (m, 1 H), 7.45 (t, J 8.3 Hz, 1 H),

(S4) M. Barzoukas, D. Josse, P. Fremaux, J. Zyss, J. F. Nicoud and J. O. Morley, *J. Opt. Soc. Am. B*, 1987, **4**, 977-986.

7.65-7.76 ppm (m, 2 H); δ_{C} (CDCl₃, 100 MHz) 39.3, 41.7, 108.7, 114.4, 114.7, 119.6, 130.3, 148.3, 149.3 ppm; HRMS *m/z* (EI) 191.0689, C₉H₉N₃O₂ requires 191.0695; Anal. Calcd for C₉H₉N₃O₂: C, 56.54; H, 4.74; N, 21.98. Found: C, 56.40; H, 4.73; N, 21.45.

2-Phenyl-1,2,3,4-tetrahydroisoquinoline-1-carbonitrile

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3}. δ_{H} (CDCl₃, 200 MHz) 2.91-3.24 (m, 2 H), 3.41-3.55 (m, 1 H), 3.71-3.82 (m, 1 H), 5.50 (s, 1 H), 6.97-7.10 (m, 3 H), 7.24-7.39 ppm (m, 6 H); δ_{C} (CDCl₃, 100 MHz) 28.1, 44.1, 53.4, 117.6, 117.7, 121.9, 126.8, 127.1, 128.8, 129.3, 129.5, 129.6, 134.1, 148.1 ppm; HRMS *m/z* (EI) 234.1146, C₁₆H₁₄N₂ requires 234.1157.

2-(4-Methoxyphenyl)-1,2,3,4-tetrahydroisoquinoline-1-carbonitrile

Solid; ν_{max} (neat/ATR probe)/cm⁻¹ 2996, 2933, 2837, 2221 (C≡N), 1647, 1510, 1465, 1454, 1259, 1245, 1206, 1179, 1030, 829 and 733; δ_{H} (CDCl₃, 400 MHz) 2.91-2.96 (m, 1 H), 3.12-3.21 (m, 1 H), 3.40-3.47 (m, 1 H), 3.55-3.59 (m, 1 H), 3.80 (s, 3 H), 5.36 (s, 1 H), 6.90-6.94 (m, 2 H), 7.07-7.11 (m, 2 H), 7.22-7.33 ppm (m, 4 H); δ_{C} (CDCl₃, 100 MHz) 28.7, 44.9, 55.5, 55.6, 114.8, 117.6, 121.0, 126.7, 127.1, 128.6, 129.4, 129.6, 134.3, 142.6, 155.7 ppm; HRMS *m/z* (EI) 264.1250, C₁₇H₁₆N₂O requires 264.1263; Anal. Calcd for C₁₇H₁₆N₂O: C, 77.25; H, 6.10; N, 10.60. Found: C, 76.88; H, 6.51; N, 10.43.

1-Phenylpiperidine-2-carbonitrile from the reaction without methanol.

Known compound; the NMR spectroscopic data agree with those given in lit.^{S3,S5}. δ_{H} (CDCl₃, 400 MHz) 1.66-1.73 (m, 2 H), 1.84-1.87 (m, 2 H), 1.99-2.03 (m, 2 H), 3.00-3.06 (m, 1 H), 3.46 (d, *J* 12 Hz, 1 H), 4.63 (s, 1 H), 6.98-7.02 (m, 3 H), 7.25-7.34 ppm (m, 2 H); δ_{C} (CDCl₃, 100 MHz) 20.1, 25.0, 29.1, 46.4, 51.9, 117.1, 118.1, 122.0, 129.2, 149.6 ppm; HRMS *m/z* (EI) 186.1153, C₁₂H₁₄N₂ requires 186.1157.

Additional NMR signals of unidentified byproducts: δ_{H} (CDCl₃, 400 MHz) 1.3-1.5 (m), 5.93 (s), 6.80-6.82 (m); δ_{C} (CDCl₃, 100 MHz) 16.0, 21.6, 22.3, 69.4, 86.9, 113.1, 117.4, 121.9, 129.4, 146.5, 150.6 ppm. One of the byproducts is tentatively assigned to the dicyanation product 1-phenylpiperidine-2,6-dicarbonitrile^{S6} based on HRMS [*m/z* (EI) 211.1106,

(S5) E. Le Gall, J.-P. Hurvois, T. Renaud, C. Moinet, A. Tallec, P. Uriac, S. Sinbandhit and L. Toupet, *Liebigs Ann./Recueil*, 1997, 2089–2101.

(S6) K. Takahashi, T. Mikajiri, H. Kurita, K. Ogura and H. Iida, *J. Org. Chem.*, 1985, **50**, 4372–4375.

$C_{13}H_{13}N_3$ requires 211.1109]. Because of superimposition of resonances, however, an unambiguous assignment of the signals in the NMR spectra was not possible.

1-Phenylpyrrolidine-2-carbonitrile from the reaction without methanol.

Known compound; the NMR spectroscopic data agree with those given in lit.^{S7}. δ_H (CDCl₃, 200 MHz) 2.07-2.34 (m, 4 H), 3.26-3.38 (m, 2 H), 4.33-4.37 (m, 1 H), 6.58-6.63 (m, 2 H), 6.74 (t, *J* 7.4 Hz, 1 H), 7.16-7.25 ppm (m, 2 H); δ_C (CDCl₃, 100 MHz) 23.9, 31.5, 47.4, 49.0, 112.6, 118.1, 119.2, 129.4, 145.1 ppm; HRMS *m/z* (EI) 172.0991, C₁₁H₁₂N₂ requires 172.1000.

1-Phenylpyrrolidine-2,5-dicarbonitrile from the reaction in the presence of TMSCN (4.0 mmol). The Typical Procedure was followed to obtain the crude product which was purified by column chromatography (SiO₂, *n*-pentane/diethyl ether = 3:1, v/v).

Known compound (lit.^{S8}); ν_{max} (neat/ATR probe)/cm⁻¹ 3075, 3050, 3003, 2930, 2233, 1597, 1504, 1347, 1338, 1323, 1269, 1251, 1180, 1160, 966, 818, 745 and 693; δ_H (CDCl₃, 400 MHz) 2.56-2.63 (m, 4 H), 4.57 (t, *J* 2.8 Hz, 2 H), 6.78-6.81 (m, 2 H), 6.96-6.99 (m, 1 H), 7.35-7.39 ppm (m, 2 H); δ_C (CDCl₃, 100 MHz) 30.4, 48.9, 113.4, 117.9, 120.5, 129.9, 141.8 ppm. Attempts to elucidate the stereochemistry of the product were not made.

(S7) W. Liu, Y. Ma, Y. W. Yin and Y. F. Zhao, *Bull. Chem. Soc. Jpn.*, 2006, **79**, 577–579.

(S8) K. Takahashi, H. Saitoh, K. Ogura and H. Iida, *Heterocycles*, 1986, **24**, 2905–2910.





























