Table S-1. Crystallographic information

	1	3	4	5
Empirical formula	C ₂₂ H ₃₄ Cl ₂ KO ₆ Ti	$C_{79}H_{167}K_2O_{12}Si_{16}Ti_2$	C ₆₃ H ₁₂₉ K ₂ O ₁₅ Si ₉ Ti	$C_{190}H_{349}K_6O_{36}Si_{24}Ti_3$
$M_{\rm w}$	552.39	1932.57	1505.57	4262.15
Temperature [K]	100(2)	100(2)	100(2)	100(2)
Size [mm]	0.38x0.30x0.22	0.32x0.25x0.18	0.46x0.33x0.20	0.37x0.30x0.20
Crystal system	monoclinic	triclinic	monoclinic	monoclinic
Space group	P2(1)/c	P-1	P2(1)/n	C2/c
a [Å]	11.410(2)	13.606(3)	11.572(2)	25.623(5)
b [Å]	11.789(2)	19.608(4)	34.572(7)	24.455(5)
c [Å]	19.048(4)	21.586(4)	21.520(4)	40.583(8)
α [°]	90	74.40(3)	90	90
β [°]	94.45(3)	79.95(3)	102.46(3)	106.02(3)
γ [°]	90	89.22(3)	90	90
V [Å ³]	2554(9)	5458(2)	8407(3)	24443(8)
Ζ	4	2	4	4
$\rho_{calc} [gcm^{-3}]$	1.436	1.176	1.190	1.158
Absorption coefficient [mm ⁻¹]	0.742	0.446	0.383	0.375
F(000)	1156	2086	3252	9172
θ range	1.79< 0 <25.00	1.25<0<25.00	1.13< 0 <26.38	1.04< 0 <24.00
Reflections collected/unique	18001/4495	19068/19068	65949/17177	80407/19181

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	1160
Data/restraints/parameters 4495/0/289 19068/0/1012 17177/0/829 19181/37/	
Goodness of fit on F^2 1.03 1.02 1.03 1.02	
Final R indices $[I>2\sigma(I)]$ R1=0.053, R1=0.075, R1=0.058, R1=0.123	,
wR2=0.107 wR2=0.160 wR2=0.120 wR2=0.29	9
R indices (all data) R1=0.075, R1=0.120, R1=0.096, R1=0.164	2,
wR2=0.118 wR2=0.178 wR2=0.134 wR2=0.33	3
Largest diff. Peak/hole $[e^{-7} Å^3]$ 1.02/-0.85 2.11/-0.42 1.03/-0.59 2.11/-1.19	

Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Center as supplementary publication no. CCDC-705716 (1), 705717 (3), 705719 (4), and 705718 (5). Copies of the data can be obtained free of charge at: http://www.ccdc.cam.ac.uk/products/csd/request/.

Compound 1 was crystallized from a mixture of pentane and toluene at low temperature (-60 to -70). The compound can be obtained in almost quantitative yield when Cp_2TiCl_2 is reacted with 0.5 equiv of the 18-rown-6 adduct of 1,4-di-potassio \neg 1,1,4,4-tetrakis \neg (trimethylsilyl) \neg tetramethyl \neg tetra \neg silane. The crown ether part of the structure of 1 contains some disorder.

The asymmetric unit of the structure of compound **3** contains two independent molecules of **3**. While in one of these molecules the potassium ion is located exactly centered above a Cp unit this is not true for the second molecule, where the potassium is located above one of the Cp carbons. In addition the asymmetric unit contains a poorly resolved benzene molecule which is distributed over two sites and can not be refined anisotropically.

The asymmetric unit of the structure of compound **4** contains an isolated anionic $[Cp_2Ti{Si(SiMe_3)_2SiMe_2}_2]$ fragment, a cationic $[Cp(18-crown-6\cdot K)_2]$ unit and in addition a 18-crown-6 crown ether molecule.

The crystal quality of compound **5** was poor. This is reflected by a value for R1 of 0.12 (at 48 degrees for 2 θ). The asymmetric unit contains two independent units of $[Cp_2Ti{Si(SiMe_3)(SiMe_2)_2}_2]$, three $[Cp(18\text{-}crown-6\cdot K)_2]$, a Cp^- , a benzene and a toluene molecule. The crown ethers are strongly disordered so that 37 restraints were used to treat this disorder. The fact that the measurement was carried out at 100K indicates that the nature of the disorder is statistical and not dynamic.