

**Supplementary Data for Manuscript:**

**“Discovery of dual fluorescent 1,8-NI dyes based on balanced Seesaw photophysical model”**

P. Nandhikonda, M.P. Begaye, Zhi Cao, Michael D. Heagy\*

Department of Chemistry, New Mexico Institute of Mining & Technology  
Socorro, NM 87801

(1) General experimental procedures along with characterization data .....	S1
(2) Appendix I: Photophysical properties of 4-substituted- <i>N</i> -(4'-phenyl)-1,8-naphthalimides .....	S6
(3) Appendix II: Copies of <sup>1</sup> H and <sup>13</sup> C NMR data for compounds <b>1-9</b> .....	S13

**General Experimental:** In a typical reaction, 1 mmol of the 4-substituted naphthalic anhydride was combined with 1.1 mmol of the 4-substituted aminoarene. The reactants were refluxed in approximately 3-4 mL pyridine for a period of 12 hr. Pyridine was removed in the fume hood via stream of air and the residue was filtered using a plug of silica gel with the appropriate solvent as found through TLC analysis. Recrystallization was further carried out using ethanol.

General procedure for preparation of 4-cyano-1,8-naphthalic anhydride:

4-bromo-1,8-naphthalic anhydride (3.1mmole) and tetraethyl ammonium cyanide (7.0mmole), CuCN (11.0 mmole), Pd<sub>2</sub>(dba)<sub>3</sub> (0.12mmole) (tris (dibenzylidene acetone) dipalladium), DPPF (0.48mmole) (1-Diphenylphosphino-1-(di-tert-butyl phosphino)-ferrocene were dissolved in 15 ml of anhydrous1,4-dioxane. The resulting mixture was refluxed in presence of argon gas for 3 hr. The reaction mixture was filtered over Celite-521 and the product was purified in 50:50 of ethyl acetate and hexane solution.

**4-Cyano-*N*-(4'-methoxyphenyl)-1,8-naphthalimide (1)** m.p. 180 °C; <sup>1</sup>H NMR DMSO-d6 δ, 8.61 (dd, *J* = 7.7 Hz, 2H), 8.37 (d, *J* = 8.1 Hz, 1H), 8.35 (d, 8.0 Hz, 1H) 8.15 (dd, *J*<sub>app</sub> = 7.5 Hz, 1H), 7.29 (d, *J* = 8.8 Hz, 2H), 7.07 (d, *J* = 8.5 Hz, 2H) 3.83 (s, 3H) <sup>13</sup>C DMSO-d6, δ 164.3, 159.5, 133.3, 132.3, 131.9, 131.7, 130.9, 130.6, 130.5, 129.8, 129.4, 128.7, 123.9, 114.8, 55.7.  
IR cm<sup>-1</sup> ν = 2361, 1707, 1513, 1364, 1238, 1187, 829, 789.

Anal. Calcd. for C<sub>20</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub>: C, 73.2; H, 3.66; N, 8.54. Found: C, 72.8; H, 3.74; N, 8.66

**4-Sulfo- *N*-(4'-methoxyphenyl)-1,8-naphthalimide (2)** m.p. 288 °C dec; <sup>1</sup>H NMR DMSO-d6, δ, 9.30 (d, *J* = 8.9 Hz, 1H), 8.47 (dd, *J* = 8.0 Hz, 2H), 8.26 (d, *J* = 6.9 Hz, 1H), 7.90 (dd, *J*<sub>app</sub> = 7.4 Hz, 1H), 7.34 (d, *J* = 14 Hz, 2H), 7.09 (d, *J* = 14 Hz, 2H). 3.83 (s, 3H)

<sup>13</sup>C NMR DMSO-d6 δ, 164.3, 164.1, 159.5, 134.9, 134.7, 131.1, 130.8, 130.7, 130.5, 129.1, 128.9, 127.2, 123.9, 123.2, 115.2, 114.8, 56.0.

IR cm<sup>-1</sup> ν = 1658, 1514, 1243, 1262, 1196, 1070, 1032, 785, 754,

Anal. Calcd. for C<sub>19</sub>H<sub>13</sub>NSO<sub>6</sub>K • 2H<sub>2</sub>O: C, 50.0; H, 2.8; N, 3.06 Found: C, 49.7; H, 2.9; N, 3.10

**4-Chloro-N-(4'-methoxyphenyl)-1,8-naphthalimide (3)** m.p. 215 °C; <sup>1</sup>H DMSO-d6, δ 8.68 (d, *J* = 8.6 Hz, 1H), 8.42 (d, *J* = 7.1 Hz, 1H) 8.19 (d, *J* = 8.3 Hz, 1H), 7.69 (dd, *J*<sub>app</sub> = 8.3 Hz, 1H), 7.21 (d, *J* = 8.5 Hz, 2H), 7.03, (d, *J* = 8.5 Hz, 2H), 6.89 (d, *J* = 7.7 Hz, 1H), 3.83 (s, 3H)

<sup>13</sup>C DMSO-d6, δ 164.9, 163.9, 159.2, 153.3, 134.5, 131.6, 130.7, 129.7, 129.6, 124.5, 122.9, 121.2, 120.0, 114.5, 108.7, 108.5, 55.9.

IR cm<sup>-1</sup> ν = 1650, 1513, 1240, 1022, 818, 790

Anal. Calcd. for C<sub>19</sub>H<sub>12</sub>NO<sub>3</sub>: C, 75.4; H, 3.9; N, 4.64 Found: C, 75.3; H, 3.6; N, 4.78

**4-Cyano-N-(4'-aminophenyl)-1,8-naphthalimide (4)** m.p. 220 °C; <sup>1</sup>H NMR DMSO-d6 δ, 8.54 (d, *J* = 8.0 Hz, 2H), 8.31 (d, *J* = 7.7 Hz, 1H), 8.20 (d, 7.7 Hz, 1H) 7.99 (dd, *J*<sub>app</sub> = 8.0 Hz, 1H), 6.96 (d, *J* = 8.2 Hz, 2H), 6.65 (d, *J* = 8.5 Hz, 2H)

<sup>13</sup>CDMSO-d6, d, 163.9, 149.1, 132.2, 131.9, 131.5, 130.5, 130.1, 129.8, 129.8, 129.7, 129.5, 129.4, 129.3, 129.3, 124.2, 124.1, 123.3.

IR cm<sup>-1</sup> ν = 2359, 1709, 1654, 1514, 1374, 1237, 1168, 834, 783.

Anal. Calcd. for C<sub>19</sub>H<sub>11</sub>N<sub>3</sub>O<sub>2</sub>: C, 76.3; H, 3.68; N, 13.4: Found: C, 76.7; H, 3.74, N, 13.2.

**4-Sulfo-N-(4'-aminophenyl)-1,8-naphthalimide (5)** m.p. 270 °C;  $^1\text{H}$  DMSO-d6  $\delta$  9.38 (d,  $J = 8.6$  Hz, 1H) 8.51-8.45 (m, 2H), 8.27 (d,  $J = 7.4$  Hz, 1H), 7.90 (t,  $J = 7.4$ , 1H), 7.44 (d,  $J_{\text{app}} = 8.8$  Hz, 2H), 7.34 (d,  $J = 8.6$  Hz, 2H)  
 $^{13}\text{C}$  DMSO d-6  $\delta$  164.3, 163.9, 151.2, 135.9, 135.2, 134.9, 133.8, 131.0, 130.9, 129.2, 128.4, 127.3, 125.5, 123.6, 123.0, 122.1  
IR  $\text{cm}^{-1}$   $\nu = 1161, 1514, 1242, 1167, 1032, 751, 657$ .

Anal. Calcd. for  $\text{C}_{18}\text{H}_{12}\text{N}_2\text{SO}_5$ : C, 58.7; H, 3.26; N, 7.60. Found: C, 58.3; H, 3.03, N, 7.59

**4-Chloro-N-(4'-aminophenyl)-1,8-naphthalimide (6)** m.p. 235 °C dec;  $^1\text{H}$  DMSO -d6,  $\delta$  8.62 (d,  $J = 8.6$  Hz, 1H), 8.57 (d,  $J = 8.0$  Hz, 1H), 8.42 (d,  $J = 8.0$  Hz, 1H), 8.10-8.00 (m, 2H), 6.99 (d,  $J = 8.5$  Hz, 2H), 6.66 (d,  $J = 8.6$  Hz, 2H), 5.25 (s, 2H)  
 $^{13}\text{C}$  DMSO-d6  $\delta$  164.1, 163.8, 149.2, 137.8, 132.1, 131.4, 130.5, 129.7, 129.2, 129.1, 129.0, 128.2, 124.1, 123.9, 122.7, 114.3.  
IR  $\text{cm}^{-1}$   $\nu = 1712, 1662, 1303, 1009, 771, 751$ .

Anal. Calcd. for  $\text{C}_{18}\text{H}_{11}\text{N}_2\text{O}_2\text{Cl}$ : C, 67.1; H, 3.42; N, 8.69 Found: C, 66.8; H, 3.38, N, 8.70

**4-Cyano-N-(4'-thiophenyl)-1,8-naphthalimide (7)** m.p. 200 °C;  $^1\text{H}$  NMR DMSO-d6  $\delta$ , 8.64 –8.56 (m, 4H), 8.13 (t,  $J = 7.5$  Hz, 1H), 7.76 (d, 7.7 Hz, 2H) 7.48 (d,  $J = 8.0$  Hz, 2H)  
 $^{13}\text{C}$  DMSO-d6,  $\delta$  163.6, 163.2, 146.3, 136.5, 134.4, 132.4, 131.3, 131.1, 130.7, 130.6, 129.8, 128.0, 127.9, 116.7, 114.7.  
IR  $\text{cm}^{-1}$   $\nu = 2360, 1677, 1648, 1561, 1345, 1228, 790, 738$ .

Anal. Calcd. for  $C_{19}H_{10}N_2SO_2$ : C, 69.1; H, 3.03; N, 8.48. Found: C, 68.9; H, 3.74; N, 8.42

**4-Sulfo-N-(4'-thiophenyl)-1,8-naphthalimide (8)** m.p. 230 °C dec;  $^1H$  DMSO-d6  $\delta$  9.31 (d,  $J = 8.5$  Hz, 1H), 8.49 (pseudo t,  $J = 7.8$  Hz, 2H), 8.25 (d,  $J = 7.4$  Hz, 1H) 7.91 (dd,  $J_{app}$  = 7.4 Hz, 1H), 7.75 (d,  $J = 8.5$  Hz, 2H), 7.53 (d,  $J = 8.5$  Hz, 2H)

$^{13}C$  DMSO-d6,  $\delta$  164.4, 164.1, 150.6, 134.9, 134.7, 131.1, 130.8, 129.1, 128.9, 128.3, 127.4, 125.8, 125.5, 123.9, 123.2, 115.2.

IR  $\text{cm}^{-1}$ ,  $\nu$  = 1661, 1371, 1240, 1192, 1068, 1042, 783, 753.

Anal. Calcd. for  $C_{18}H_{11}NS_2O_5 K \bullet 3H_2O$ : C, 45.2; H, 2.32; N, 4.84. Found: C, 45.2; H, 2.24; N, 4.76.

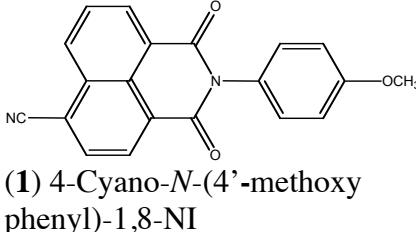
**4-Chloro-N-(4'-thiophenyl)-1,8-naphthalimide (9)** m.p. 226 °C dec;  $^1H$  DMSO-d6  $\delta$  8.64 (d,  $J = 8.2$  Hz, 1H), 8.42 (d,  $J = 7.4$  Hz, 1H), 8.18 (d,  $J = 7.9$  Hz, 1H), 7.67 (dd,  $J_{app}$  = 8.5 Hz, 1H), 7.46 (s, 2H), 7.18 (d,  $J = 7.5$  Hz, 2H), 7.03 (d,  $J = 9.1$  Hz, 2H), 6.87 (d,  $J = 8.6$  Hz, 1H).

$^{13}C$  NMR DMSO-d6  $\delta$ , 164.0, 163.9, 137.4, 132.9, 131.6, 131.3, 131.1, 130.9, 130.5, 130.3, 129.6, 128.3, 128.2, 124.4, 124.1, 115.7

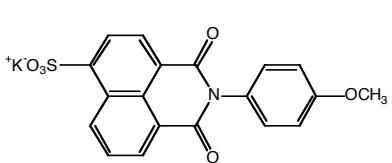
IR  $\text{cm}^{-1}$ ,  $\nu$  = 1656, 1583, 1512, 1365, 1238, 1173, 818, 789.

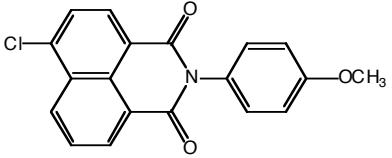
Anal. Calcd. for  $C_{18}H_{10}NSO_2Cl$ : C, 63.7; H, 3.11; N, 4.13 Found: C, 63.4; H, 3.03; N, 4.10.

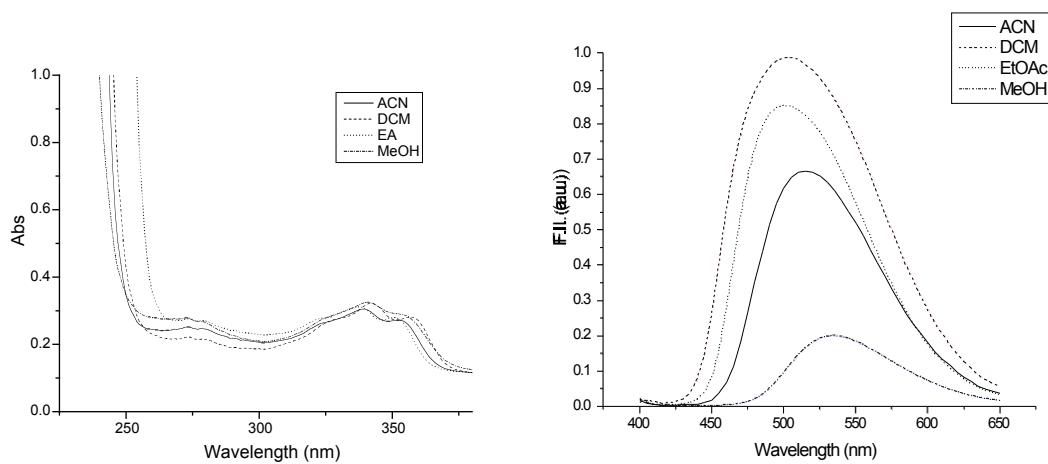
**Appendix I Photophysical properties of 4-substituted-*N*-(4'-phenyl)-1,8-naphthalimides**

 <b>(1)</b> 4-Cyano- <i>N</i> -(4'-methoxyphenyl)-1,8-NI	Absorbance $\lambda_{\max}$ , ( $\epsilon$ )	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity (a.u.)	Quantum Yield*
DCM $M = 1.0 \times 10^{-5}$	342(0.193) 410(0.126)	410	521 603	$1.8 \times 10^7$ $1.8 \times 10^7$	$4.6 \times 10^{-2}$ $4.6 \times 10^{-2}$
ACN $M = 1.0 \times 10^{-5}$	340(0.191) 410(0.126)	410	571	$1.8 \times 10^7$	$4.6 \times 10^{-2}$
EtOAc $M = 1.0 \times 10^{-5}$	338(0.200) 410(0.128)	410	527 601	$1.8 \times 10^7$ $1.8 \times 10^7$	$4.5 \times 10^{-2}$ $4.5 \times 10^{-2}$
MeOH $M = 1.0 \times 10^{-5}$	340(0.153) 410(0.120)	410	588	$1.6 \times 10^7$	$4.3 \times 10^{-2}$
Acetone $M = 1.0 \times 10^{-5}$	339(0.195) 410(0.125)	410	541 600	$1.8 \times 10^7$ $1.8 \times 10^7$	$4.6 \times 10^{-2}$ $4.6 \times 10^{-2}$

\*Quantum yields relative to Quinine sulfate standard

	$\lambda_{\text{max}} (\text{Abs})$	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity A.U.	Quantum yield
<b>(2) 4-Sulfo -N (4-methoxy phenyl) NI</b>					
Dichloromethane $M = 1.0 \times 10^{-5}$	338 (0.233)	340	369 512	$2.62 \times 10^5$ $2.09 \times 10^5$	$3.6 \times 10^{-4}$ $2.8 \times 10^{-4}$
Ethyl Acetate $M = 1.0 \times 10^{-5}$	338 (0.246)	340	360 530	$2.10 \times 10^5$ $2.16 \times 10^5$	$2.7 \times 10^{-4}$ $2.8 \times 10^{-4}$
Acetonitrile $M = 1.0 \times 10^{-5}$	341 (0.242)	340	364 562	$2.92 \times 10^5$ $1.02 \times 10^5$	$3.8 \times 10^{-4}$ $1.3 \times 10^{-4}$
MeOH $M = 1.0 \times 10^{-5}$	339 (0.256)	340	368	$1.53 \times 10^5$	$1.9 \times 10^{-4}$

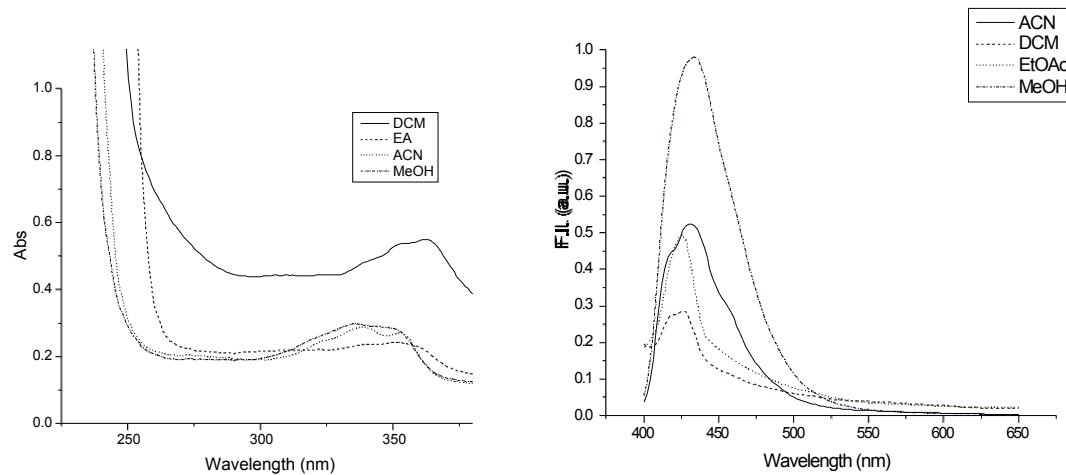
	$\lambda_{\text{max}} (\text{Abs})$	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity A.U.	Quantum yield
<b>(3) 4-Chloro-N-(4'-methoxyphenyl) NI</b>					
Dichloromethane $M = 1.0 \times 10^{-5}$	340 (0.305)	410	505	$7.35 \times 10^6$	$7.7 \times 10^{-3}$
Ethyl Acetate $M = 1.0 \times 10^{-5}$	342 (0.300)	410	494	$7.21 \times 10^6$	$7.6 \times 10^{-3}$
Acetonitrile $M = 1.0 \times 10^{-5}$	338 (0.312)	410	516	$6.21 \times 10^6$	$6.4 \times 10^{-3}$
Methanol $M = 1.0 \times 10^{-5}$	341 (0.324)	410	536	$2.87 \times 10^6$	$2.8 \times 10^{-3}$



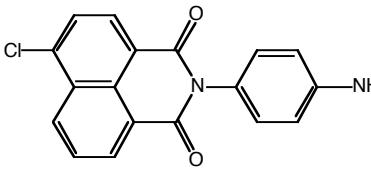
Supplementary Figure 1. Absorption and fluorescence spectra of (**3**) 4-Chloro-*N*-(4'-methoxyphenyl)-1,8-NI.

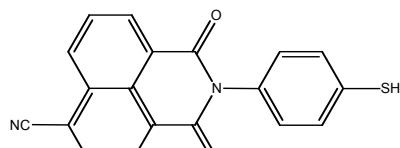
<chem>C#Cc1ccc2c(c1)C(=O)c3ccccc3N(c4ccc(N)cc4)C(=O)c2</chem> <b>(4)</b> 4-Cyano- <i>N</i> -(4'-amino phenyl)-1,8-NI	Absorbance	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity (a.u.)	Quantum yield
DCM $M = 1.0 \times 10^{-5}$	339(0.562) 420(0.134)	420	520 576	$5.4 \times 10^6$ $6.0 \times 10^6$	$1.3 \times 10^{-2}$ $1.4 \times 10^{-2}$
ACN $M = 1.0 \times 10^{-5}$	336(0.374) 420(0.126)	420	515 581	$2.0 \times 10^6$ $1.0 \times 10^7$	$5.1 \times 10^{-3}$ $2.5 \times 10^{-2}$
EtOAc $M = 1.0 \times 10^{-5}$	334(0.573) 420(0.141)	420	511 575	$1.9 \times 10^6$ $6.5 \times 10^6$	$4.3 \times 10^{-3}$ $1.5 \times 10^{-2}$
MeOH $M = 1.0 \times 10^{-5}$	337(0.562) 420(0.142)	420	522 579	$1.7 \times 10^6$ $4.0 \times 10^6$	$3.8 \times 10^{-3}$ $9.0 \times 10^{-3}$

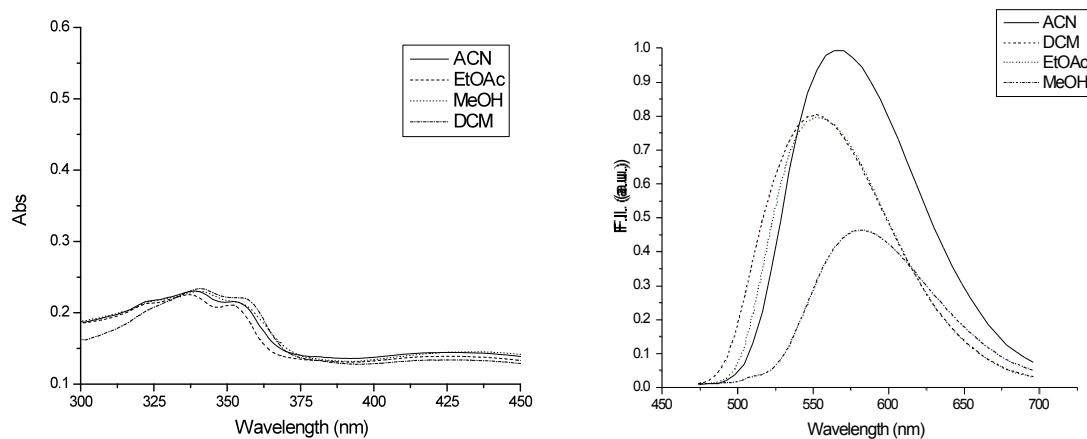
<chem>*[K+]O3S-C1=C(C=C2=C1C(=O)N(c3ccc(N)cc3)C(=O)c4ccc(*)cc4)C=C2</chem> <b>(5)</b> 4-Sulfo-N-(4-amino phenyl)-1,8-NI	$\lambda_{\text{max}}$ (Abs)	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity (a.u.)	Quantum yield
Dichloromethane $M = 1.0 \times 10^{-5}$	350 (0.230)	340	426	$9.18 \times 10^5$	$1.3 \times 10^{-3}$
Acetonitrile $M = 1.0 \times 10^{-5}$	338 (0.287)	340	430	$1.7 \times 10^6$	$1.9 \times 10^{-3}$
Ethyl Acetate $M = 1.0 \times 10^{-5}$	351 (0.243)	340	426	$1.6 \times 10^6$	$2.1 \times 10^{-3}$
Methanol $M = 1.0 \times 10^{-5}$	336 (0.297)	340	434	$3.1 \times 10^6$	$3.3 \times 10^{-3}$



Supplementary Figure 2. Absorbance and fluorescence spectra of **(5)** 4-Sulfo-N-(4'-aminophenyl)-1,8-NI.

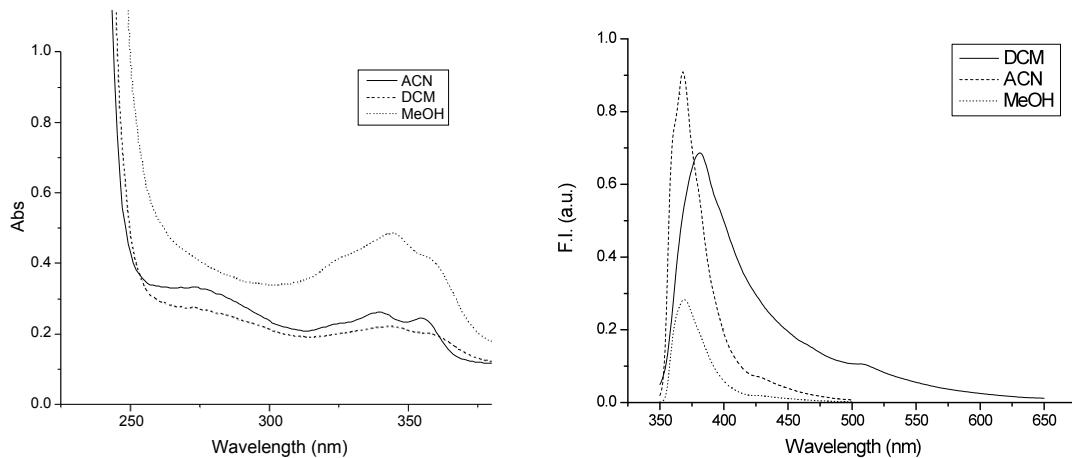
	$\lambda_{\text{max}} (\text{Abs})$	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity A.U.	Quantum yield
<b>(6)-4-Chloro-N-(4-amino phenyl)-1,8-NI</b>					
Dichloromethane $M = 1.0 \times 10^{-5}$	340 (0.346)	340	437 569	$2.75 \times 10^5$ $4.11 \times 10^6$	$2.5 \times 10^{-4}$ $3.8 \times 10^{-3}$
Acetonitrile $M = 1.0 \times 10^{-5}$	342 (0.240)	342	437 575	$2.07 \times 10^5$ $2.46 \times 10^5$	$2.8 \times 10^{-4}$ $3.3 \times 10^{-4}$
Ethyl Acetate $M = 1.0 \times 10^{-5}$	342 (0.217)	342	426 564	$1.24 \times 10^5$ $1.99 \times 10^5$	$1.8 \times 10^{-4}$ $2.9 \times 10^{-4}$
Methanol $M = 1.0 \times 10^{-5}$	342 (0.227)	342	434	$2.34 \times 10^5$ $9.55 \times 10^4$	$3.3 \times 10^{-4}$ $1.3 \times 10^{-4}$

	Absorbance	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F.Intensity (a.u.)	Quantum yield
<b>(7) 4-Cyano-N-(4'-thiophenyl)-1,8- NI</b>					
DCM $M = 1.0 \times 10^{-5}$	342(0.235) 410(0.140)	410	553	$17.3 \times 10^6$	$1.7 \times 10^{-2}$
ACN $M = 1.0 \times 10^{-5}$	340(0.274) 410(0.135)	410	567	$9.2 \times 10^6$	$2.2 \times 10^{-2}$
EtOAc $M = 1.0 \times 10^{-5}$	338(0.226) 410(0.136)	410	553	$7.4 \times 10^6$	$1.7 \times 10^{-2}$
MeOH $M = 1.0 \times 10^{-5}$	341(0.230) 410(0.138)	410	582	$4.3 \times 10^6$	$1.0 \times 10^{-2}$



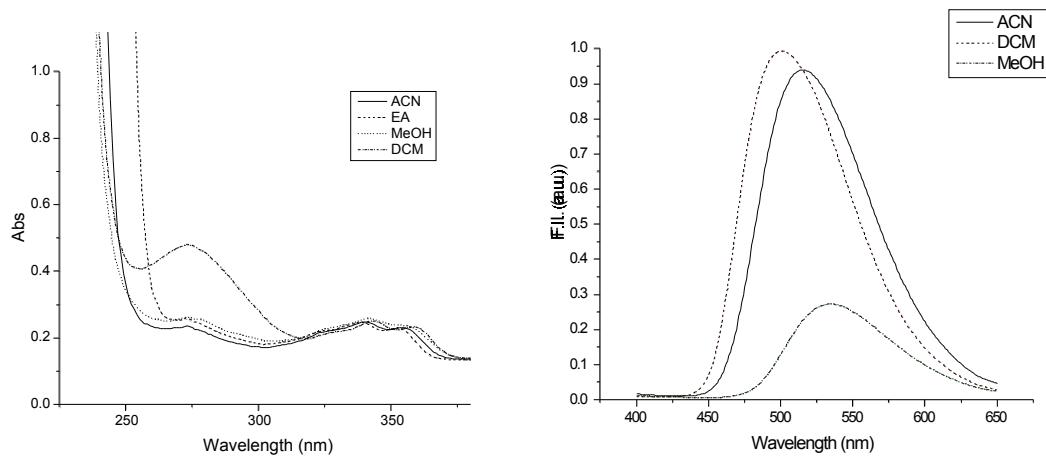
Supplementary Figure 3. Absorbance and fluorescence spectra of (**7**) 4-Cyano-*N*-(4'-thiophenyl)-1,8-NI.

<chem>*[K+]OS(=O)(=O)c1ccc2c(c1)C(=O)N(c3ccccc3S)C(=O)c2</chem> <b>(8)</b> 4-Sulfo- <i>N</i> -4-thiophenyl-1,8-NI	$\lambda_{\text{max}}$ (Abs)	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F. Intensity A.U.	Quantum yield
Dichloromethane M = $1.0 \times 10^{-5}$	344 (0.221)	340	388	$1.1 \times 10^6$	$1.6 \times 10^{-3}$
Acetonitrile M = $1.0 \times 10^{-5}$	340 (0.262)	340	386	$1.8 \times 10^6$	$2.2 \times 10^{-3}$
MeOH M = $1.0 \times 10^{-5}$	342 (0.245)	340	386	$1.8 \times 10^6$	$2.2 \times 10^{-3}$



Supplementary Figure 4. Absorbance and fluorescence spectra of (**8**) 4-Sulfo-*N*-(4'-thiophenyl)-1,8-NI.

<chem>O=C1C(=O)N(c2ccc(cc2)S)C(=O)c3cc(Cl)ccc31</chem> <b>(9)</b> 4-Chloro- <i>N</i> -(4'-thiophenyl)-1,8-NI	$\lambda_{\text{max}}$ (Abs)	$\lambda_{\text{ex}}$ (nm)	$\lambda_{\text{em}}$ (nm)	F. Intensity A.U.	Quantum yield
Methanol $M = 1.0 \times 10^{-5}$	343 (0.251)	343	538	$2.07 \times 10^6$	$2.6 \times 10^{-3}$
Acetonitrile $M = 1.0 \times 10^{-5}$	342 (0.247)	343	515	$7.01 \times 10^6$	$9.1 \times 10^{-3}$
Dichloromethane $M = 1.0 \times 10^{-5}$	343 (0.251)	343	500	$7.54 \times 10^6$	$9.6 \times 10^{-3}$



Supplementary Figure 5. Absorbance and fluorescence spectra of (**9**) 4-Chloro-*N*-(4'-thiophenyl)-1,8-NI.