Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 **Coordination Complex SH2 Domain Proteomimetics: An Alternative Approach to Disrupting Oncogenic Protein–Protein Interactions**

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Supporting Information

Chemistry: General Methods. Anhydrous solvents methanol, DMSO, CH₂Cl₂, THF and DMF were purchased from Sigma Aldrich and used directly from Sure-Seal bottles. Molecular sieves were activated by heating to 300 °C under vacuum overnight. All reactions were performed under an atmosphere of dry nitrogen in oven-dried glassware and were monitored for completeness by thin-layer chromatography (TLC) using silica gel (visualized by UV light, or developed by treatment with KMnO₄ stain or phosphomolybdic acid stain). ¹H and ¹³C NMR spectra were recorded on Bruker 400 MHz and a Varian 500 MHz spectrometers in either CDCl₃, CD₃OD or *d*₆-DMSO. Chemical shifts (δ) are reported in parts per million after calibration to residual isotopic solvent. Coupling constants (*J*) are reported in Hz. Before biological testing, inhibitors were subjected to further purification by reversed-phase HPLC (rpHPLC). Analysis and purification by rpHPLC were performed using either Atlantis Prep T3 10 µm C18 (2) 250 x 19 mm column run at 20 mL/min (preparative) or a Microsorb-MV 300 A C18 250 mm x 4.6 mm column run at 1 mL/min (analytical), using gradient mixtures of (A) water with 0.1% TFA and (B) 10:1 acetonitrile/water with 0.1% TFA. Ligand purity was confirmed by analytical rpHPLC using linear gradients from 100% A to 100% B, with changing solvent composition of either (I) 4.5% or (II) 1.5% per minute after an initial 2 minutes of 100% A.

Compound Characterization

Ligand of Complex 1 (L1) oil; δ_H (400 MHz, CDCl₃) 2.07 (s, 1H, O<u>H</u>), 2.63 (m, 4H, CHOH<u>CH₂</u>N), 3.86 (m, 8H, N<u>CH₂</u>Pyr), 3.97 (m, 1H, C<u>H</u>OH), 7.11 (t, *J* = 8 Hz, 4H, 4 CH (Ar)), 7.57 (t, *J* = 8 Hz, 4H, 4 CH

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 (Ar)), 8.49 (d, J = 6 Hz, 4H, 4 CH (Ar)); δ_{C} (400 Hz, CDCl₃) 58.91, 60.57, 67.12, 121.84, 122.95, 136.32, 148.69, 159.27; HRMS (ES+) calcd for C₂₇H₃₁N₆O [M+H] 455.2553. Found 455.2567; IR (KBr, cm⁻¹) 3378, 2923, 1640, 1614, 1585, 1474, 1435, 1273.

Proteomimetic 1: HRMS (ES+) calcd for $C_{27}H_{31}N_6OCu_2$ [M²⁺, Cu(II)] 290.0531. Found 290.0539; IR (KBr, cm⁻¹) 3498, 2959, 1612, 1574, 1484, 1448, 1259, 1172, 1032, 766, 641; mp = 115-120 °C; UV/Vis (D₂O, 10 mM HEPES, pH = 7.5) λ 253, 267.

Ligand of Complex 2 (L2) oil; $\delta_{\rm H}$ (400 MHz, CDCl₃) 3.68 (s, 2H, PhCH₂N-), 3.69 (s, 2H, PhCH₂N-), 3.80 (s, 4H, -NCH₂Pyr), 3.85 (s, 4H, -NCH₂Pyr), 3.92 (s, 6H, 2 –OCH₃), 7.14 (t, *J* = 6.1 Hz, 2H, 2 CH (Ar)), 7.27 (m, 3H, 3 CH (Ar)), 7.45 (s, 1H, CH (Ar)), 7.61 (m, 6H, 6 CH (Ar)), 8.20 (dd, *J* = 8.2 and 2.2 Hz, 2H, 2 CH (Ar)), 8.51 (d, *J* = 4.6 Hz, 2H, 2 CH (Ar)), 9.10 (d, *J* = 1.8 Hz, 2H, 2 CH (Ar)); $\delta_{\rm C}$ (400 MHz, CDCl₃) 52.2, 58.3, 58.6, 59.7, 59.8, 121.83, 122.1, 122.6, 124.4, 127.5, 127.8, 128.3, 129.2, 136.3, 137.4, 138.2, 139.0, 148.8, 150.2, 164.1, 165.6, 169.5; HRMS (ES+) calcd for C₃₆H₃₆N₆O₄ [M+H] 617.2870. Found 617.2841; IR (KBr, cm⁻¹) 3412, 2951, 2360, 1725, 1637, 1616, 1597, 1567, 1475, 1435. **Proteomimetic 2**: HRMS (ES+) calcd for C₃₆H₃₆N₆O₄Cu₂ [M²⁺, Cu(II)] 371.0689. Found 371.0701; IR (KBr, cm⁻¹) 3384, 2963, 1735, 1616, 1490, 1450, 1290, 1030, 963, 760, 637; mp = 110-125 °C; UV/Vis (D₂O, 10 mM HEPES, pH = 7.5) λ 223, 257, 278.

Ligand of Complex 3 (L3) oil; $\delta_{\rm H}$ (400 MHz, CDCl₃) 3.64 (s, 4H, -NC<u>H</u>₂Pyr), 3.65 (s, 2H, PhC<u>H</u>₂N-), 3.67 (s, 2H, PhC<u>H</u>₂N-), 3.78 (s, 4H, 2 -NC<u>H</u>₂Pyr), 5.33 (s (br), 4H, 2 -NH₂), 6.48 (d, *J* = 8.3 Hz, 2H, 2 CH (Ar)), 6.85 (d, *J* = 7.6 Hz, 2H, 2 CH (Ar)), 7.12 (t, *J* = 6.1 Hz, 2H, 2 CH (Ar)), 7.25 (m, 3H, 3 CH (Ar)), 7.37 (t, *J* = 7.9 Hz, 2H, 2 CH (Ar)), 7.44 (s, 1H, CH (Ar)), 7.60 (m, 4H, 4 CH (Ar)), 8.49 (d, *J* = 4.5 Hz, 2H, 2 CH (Ar)); $\delta_{\rm C}$ (400 MHz, CDCl₃) 58.3, 58.5, 59.6, 59.9, 106.3, 112.2, 121.7, 122.6, 127.2, 127.4, 128.0, 128.9, 136.2, 138.0, 138.8, 139.2, 148.7, 157.7, 158.4, 159.8; HRMS (ES+) calcd for C₃₂H₃₄N₈ [M+H] 531.2979. Found 531.2953; IR (KBr, cm⁻¹) 3410, 2924, 2851, 2360, 1618, 1570, 1468. **Proteomimetic 3**: HRMS (ES+) calcd for C₃₂H₃₄N₈Cu₂ [M²⁺, Cu(II)] 327.0665. Found 327.0680; IR

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 (KBr, cm⁻¹) 3361, 2924, 1641, 1613, 1576, 1489, 1449, 1255, 1175, 1031, 803, 768, 640; mp = 95-114 °C; UV/Vis (D₂O, 10 mM HEPES, pH = 7.5) λ 237, 261, 302.

S1. Isothermal Titration Calorimetry Binding Experiments. Isothermal titration calorimetry (ITC) experiments were used to measure the binding of our metal complexes to various substrates, and were performed at 25 °C (298 K) using Microcal VP-ITC titration microcalorimeter. In order to minimize mixing heat effects caused by differences in solution composition, the substrates and receptor were both dissolved in freshly prepared HEPES buffer (50 mM, pH = 7.2) before each titration experiment. All solutions prior to experiments were degassed before being added to the calorimeter cell. The substrates, at a concentration of approximately 2.0 mM, were injected in 10µL increments into the reaction cell (cell volume 1.49 mL) containing complex at a concentration of *ca* 0.1 mM, until there occurred a saturation of binding sites. A 250 µL injection syringe with 310–400 rpm stirring was used to give a series of 10 µL injections at 3.5-minute intervals. Control experiments for heats of mixing and dilution were performed under identical conditions and used for data correction in subsequent analysis. Data acquisition and subsequent non-linear regression analysis were done in terms of a simple binding model using the Microcal ORIGIN software package.

ITC Experiments – representative traces for Proteomimetics 1-3



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S2. Fluorescence polarization (FP) binding assays.

The fluorescene polarization (FP) measurements were performed on a custom-built multimodal confocal microscope described previously [Baoxu Liu et al, *J. Fluoresc*, A Photostable, pH-Invariant Fluorescein Derivative for Single-Molecule Microscopy]. In brief, a wavelength-tunable femtosecond laser (Tsunami HP, Spectra Physics, Mountain View, CA) is frequencydoubled through a β -BBO crystal (Newlight Photonics, Toronto, Canada) to produce a narrow excitation spectrum centered at 480 nm. The light is linearly polarized, passes through a 1.4 NA/100X plan-apochromat objective (Carl Zeiss Canada) and illuminates the sample at intensities of 50-100 W/cm². The emitted fluorescence is collected through the same objective and is spatially and spectrally filtered using a 50-µm pinhole and high-quality long-pass

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and interference filters (Semrock, Rochester, NY, USA) to remove out-of-focus fluorescence, Rayleigh and Raman scattering. Further, the fluorescence is divided into two components with orthogonal polarization, I_{vv} and I_{vh} , using a broadband polarizing cube beamsplitter (Newport, Irvine, CA). Each beam is then tightly focused onto a 50-µm diameter avalanche photodiode that features a low dark noise and high sensitivity (PD5CTC, Optoelectronic Components, Kirkland, QC, Canada). Each time a photon is detected the detector outputs an electric pulse and these pulses are counted by a PicoHarp300 module (PicoQuant Gmbh, Germany) to provide fluorescence intensity traces.

Prior to each set of experiments, a solution containing 65 nM of the dye Rhodamine110 in water was used to determine the correction factor G for the different detection sensitivity of the two channels. The fluorescence polarization p was then calculated according to the equation:

$$r = \frac{I_{vv} - G \cdot I_{vh}}{I_{vv} + 2 \cdot G \cdot I_{vh}} P = \frac{I_{vv} - G \cdot I_{vh}}{I_{vv} + G \cdot I_{vh}} \cdot 1000 \quad (\text{in mP units})$$

$$p = \frac{3 \cdot r}{2 + r} \cdot 1000 \text{ in mP}$$

Initially, the formation of the phosphopeptide-STAT3 complex was verified by FP; the overall titration curve and the value of $k_d = 142 \pm 42$ nM extracted from the fit concur with previous studies (Berg, Anal. Biochem, **2004**) Note that the free F*-peptide in solution exhibits an FP value of 15 ± 3 mP, much lower than the fitted saturation value of 226 ± 23 mP when virtually all phosphopeptide is bound to the protein.



(vii) Stat3-pYLPQTV-NH₂ binding measured by fluorescence polarization assay.

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 **S2.1. FP Binding Experiments Data**

All experiments were performed 20 minutes after mixing the starting solution containing 7.2 nM F*peptide and 292 nM STAT3 with various amounts of inhibitor stock solution (Figure **vii**). The concentration of inhibitor was varied between 0.1 μ M and 5 mM and all measurements were performed in a buffer (50 mM NaCl, 10 mM Hepes, pH 7.5, 2 mM dithiotreitol, w/w 1 mM EDTA). For statistical purposes, a series of five subsequent one-minute measurements were performed at each concentration of inhibitor.

S2.1. FP Binding Experiments Data



(x) Inhibitor 3

FP experiments with and without 1 mM EDTA in the buffer solution show that EDTA interferes in the interaction between the metal complex and the phosphopeptide. When EDTA is present, compound **2** has ca. 20% larger IC₅₀ value and a significantly steeper transition, which suggests an increase in

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 cooperativity. This suggests the chelation of copper atoms present in the inhibitor molecules by the EDTA molecules. We recorded similar EDTA mediated drops in affinity for each of the remaining inhibitors (data not shown). Purified Stat3 protein was purchased from SignalChem (CANADA) – 80-90 % purity.

Titration FP data was modelled by the equation:

$$p(X) = \frac{p_1 \cdot IC_{50} + p_2 \cdot X}{IC_{50} + X}$$

where *p* is the measured fluorescence polarization and *X* is the concentration of inhibitor. The free parameter is the half maximal inhibitory concentration (IC₅₀). The limiting values of the polarization (p_1 and p_2) were estimated from the plot and kept constant during the fit. The curve fitting was performed in Origin using the Levenberg-Marquardt algorithm and the reduced chi-square criterion for convergence. The inhibitor dissociation constant K_i was calculated from IC₅₀ according to the formula:

$$K_i = \frac{IC_{50}}{1 + \frac{[STAT3]}{K_d}}$$

where [STAT3] = 292 nM and $K_d = 100 \text{ nM}$.

S2.2. Interesting Note: Remarkably, the fluorescence of the fluorescein attached to the model peptide was gradually quenched upon increasing the inhibitor concentration in solution. When the inhibitor reached mM concentrations, the detected fluorescence intensity became ca. 10 times weaker than at low concentrations of inhibitor. The dilution of the peptide upon addition of inhibitor was less than 10%, far smaller than the observed drop in fluorescence intensity, whereas the adsorption of sample to the microscope coverslip was negligible during the time course of the experiments. Our data were acquired using the time-correlated single-photon counting (TCSPC) technique so they include fluorescence decay traces by default. As quenching is expected to influence the rate of fluorescence decay, the time-resolved

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curves were fitted to mono-exponential decays and the estimated fluorescence lifetime was plotted as a function of inhibitor concentration (figure **xi**, right). The titration curve closely resembles the FP data and it yields an IC₅₀ value that is virtually identical to the FP assay, 315 μ M vs. 324 μ M. We can conclude that the formation of the peptide-inhibitor complex promotes quenching of the fluorescein probe by bringing charged metal atoms in its close proximity. This finding is significant because it opens up exciting prospects for studying the inhibition potency of novel organo-metallic complexes using fluorescence lifetime, a method that is more sensitive and selective compared to fluorescence polarization.



(xi) *Left:* Fluorescence polarization measured upon titrating the F*-phosphopeptide (7.2 nM) with compound **3** in the presence of STAT3 protein (292 nM). *Right*: fluorescence lifetime titration.

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S3. Nuclear extract preparation and gel shift assays. Nuclear extract preparation from v-Srctransformed fibroblasts (NIH3T3/v-Src) or mouse fibroblasts overexpressing the human epidermal growth factor receptor (NIH3T3/hEGFR) and electrophoretic mobility shift assay (EMSA) were carried out as previously described (reference 5, 16). Nuclear extracts of equal total protein content were preincubated with different concentrations of compounds for 30 min at room temperature prior to incubation with the radiolabeled probe. The ³²P-labeled oligonucleotide probe used is hSIE (high affinity sisinducible element from the *c-fos* gene, m67 variant, 5'-AGCTTCATTTCCCGTAAATCCCTA) that binds STAT1 and STAT3. The EMSA trace for inhibitor 3 is shown below (not shown in manuscript)



(xii) EMSA analysis for NIH3T3/vSrc cells treated with increasing concentrations of compound 3



(xiii) Control EMSA analysis for NIH3T3/vSrc cells treated with increasing concentrations of 1,5-propyl-BDPA and Cu(OTf)_2 $\,$

S3.1 EMSA Whole Cell NIH3T3/vSrc Data. Cells were incubated with varying concentrations of inhibitors for 24 hr, and then the nuclear extract preparations were subjected to the Stat3–Stat3:DNA-binding assay *in vitro* using the hSIE probe followed by EMSA analysis. Encouragingly, our potent *in vitro* activity was mirrored in this whole cell assay, where agents induced significant levels of Stat3 dimer disruption. The three inhibitors were found to significantly reduce the levels of intracellular Stat3–Stat3 homodimer formation. A representative gel from the EMSA experiment for NIH3T3/vSrc fibroblasts treated with increasing concentrations of **1** is illustrated below in (xiii)

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(xiv) EMSA analysis of Stat3 DNA-binding activity in nuclear extracts prepared from NIH3T3/vSrc treated with compound 1 for inhibition of Stat3–Stat3 dimerization

S4. Whole cell viability studies in DU145, MDA468 and OCI-AML-2 cell lines (Table 3)

Human OCI-AML2 leukemia, DU145 prostate, and MDA468 breast cancer cells were seeded in 96 well plates. Cells were then treated with increasing concentrations of **1**, **2**, **3** and **4**. Seventy-hours after incubation, cell growth and viability was measured with the CellTiter96 aqueous nonradioactive (MTS) assay according to the manufacturer's instructions (Promega, Madison, WI) and as described previously (Simpson C and Schimmer, *Cancer Res.* **2009**). Data represent the mean $IC_{50} \pm SD$ (n = 5).

S5. WST-1 viability assay procedures

Proliferating cells in 96-well plates were treated with inhibitors for 48 h for WST-1 assay analysis, according to manufacturer's (Roche) instructions.- <u>www.roche-applied-science.com</u>

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S6. Whole cell viability studies in healthy NIH3T3 cells

QuickTime™ and a decompressor are needed to see this picture.

Proteomimetic 2 viability data: NIH3T3 'healthy' cells treated with 2 versus breast cancer cells (MDA-

468) and acute myeloid leukemia cells (OCI-AML2) treated with 2. MDA-468 (Walker, S. R. Mol.

Cancer Res. 2009, 7, 966-976) and OCI-AML2 (Kube, D., Blood, 2001, 98, 762-770) cells are known to

have constitutively activated Stat3 activity. We noted significantly lower cytotoxic effects against healthy

NIH3T3 cells *cf.* cancer cells containing Stat3 activity.

S7. Whole cell viability studies of $Cu(II)(OTf)_2$ and Ligand 2 (L2)

Inhibitory assessment of Cu(II)(OTf)₂ on breast cancer cells (MDA-468), prostate cancer (DU145) and acute myeloid leukemia cells (OCI-AML2).

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It can be seen that there is no inhibitory affect due to the free metal at the concentrations at which our

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S8. Whole cell viability studies of Ligand 2 (L2)

Assessment of cytotoxicity of free ligand, L2 on breast cancer cells (MDA-468), prostate cancer (DU145)

and acute myeloid leukemia cells (OCI-AML2).



QuickTime™ and a decompressor are needed to see this picture.

The data show negligible cytotoxic effects upon cancer cell viablity by the free ligand L2. These data confirm that cytotoxicity is achieved only when the ligand is complexed with copper (II).

Supplementary Material (ESI) for Chemical Communications This journal is (c) The Royal Society of Chemistry 2009 **S9. Peptide Characterization**

Peptides used in FP and ITC experiients were purchased from CanPeptide (Canada) with purities > 97 %

by analytical HPLC analysis. Analytical HPLC traces and ESMS traces of the three peptides, Ac-

G(pY)LPQTV-NH₂, 5-FAM-G(pY)LPQTV-NH₂ and Ac-(pY)LKTK-NH₂ are shown below.

(i) Ac-(pY)LKTK-NH₂

CP08341 Sample ID Date Acquired 2/24/09 3:28:45 PM Sy stem Waters 600 3/19/09 9:44:02 AM Date Processed Multi-Solvent Delivery System Acq Method Set Standard Analysis Injection Processing Method LC PQ Processing Injection Volume 10.00 ul Solv ent A = 0.1%TFA/ACN Channel 2487Channel 1 B = 0.1%TFA/H2O 25.0 Minutes Run Time Flow 1ml/min 5-55% A in 25 min Column Vy dac Peptide & Protein Chemist/Book No P090312-08341 C-18, 4.6 x250mm Auto-Scaled Chromatogram 0.80 0.70 083 0.60 Ac-(pY)LKTK-NH2 0.50 ₽ 0.40 0.30 0.20 285:41-6 9.567 0.10 0.00 57 2.00 4.00 6.00 8.00 10.00 12.00 Minutes 14.00 16.00 18.00 20.00 22.00 24.00 Peak Results Nar RT Area Height % Area 9.063 11921855 756519 98.62 9.567 83533 7739 0.69 2 3 17,187 29603 5009 0.24 4 17.312 53528 7001 0.44 Original Filename: d:\voyager\data\canpept\090223\cp08341.ms This File # 1 : D:\VOYAGER\DATA\CANPEPT\090223\CP08341.MS Com 771.06 ^{{M-H}-} 5000 Ac-(pY)LKTK-NH2 4000 3000 2000 1000 500 600 700 800 1000 1100 1200 1300 900 Mass (m/z)

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(ii) Ac-G(pY)LPQTV-NH₂



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(iii) 5-FAM-G(pY)LPQTV-NH₂



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