

Supplementary Material

Chirality and Magnetism of an Open-framework Cobalt Phosphite Containing Helical Channels from Achiral Materials†

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Synthesis: Typically, $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ (0.178g) was first dissolved in DMF (20mL) and then 4-diazabicyclo [2, 2, 2]octane (DABCO, 1.10g) was added with stirring. Finally, H_3PO_3 (0.6mL, 50wt %) was added to the above reaction mixture. A dark-blue gel was formed after stirring for 0.5h. The gel was sealed in a Teflon-lined stainless steel autoclave and heated under autogenous pressure at 60°C for 3 days. The resulting blue block crystals were isolated by filtration, washed with ethanol and dried in air. Inductively coupled plasma (ICP) analysis (Perkin–Elmer Optima 3300 DV ICP instrument) for **1** yielded Co 17.65, P 18.03% (calcd: Co 17.71, P 18.61%). Elemental analysis (%; Perkin–Elmer 2400 elemental analyzer) calcd: C 21.62, H 4.80, N 8.41; found: C 21.65, H 4.89, N 8.33).

Structure determination: Structural analysis of a single crystal (0.21×0.18×0.16 mm³) was performed on a Siemens SMART CCD diffractometer using graphite-monochromated MoK α radiation ($\lambda = 0.71073 \text{ \AA}$). Data processing was accomplished with the SAINT processing program. The structure was solved by direct methods and refined on F^2 by full-matrix least-squares using SHELXLT97. All Co, P, and O atoms were determined directly. The hydrogen atoms of the HPO_3 groups, C, and N atoms were subsequently found in the difference Fourier map. The hydrogen atoms in the DABCO molecules were placed geometrically. All non-hydrogen atoms

of the inorganic framework were refined anisotropically. Crystal data and details of the structure determination are given in Table S1.

Crystal data: $C_6N_2H_{16}O_6P_2Co$, $M_r = 333.08$, orthorhombic, space group $P2_12_12_1$ (No. 19), $a = 10.1040(4)$, $b = 10.8100(5)$, $c = 10.9930(5)$ Å, $V = 1200.70(9)$ Å³, $Z = 4$, $\mu = 1.713$ mm⁻¹, $\rho_{\text{calcd}} = 1.843$ gcm⁻³, 6673 reflections measured, 2368 unique ($R_{\text{int}} = 0.0180$). The final $wR(F^2)$ (all data) was 0.0608 and $R(F)$ ($I > 2\sigma(I)$) was 0.0209. The flack parameter was 0.034(14). CCDC-723093 contains the supplementary crystallographic data for **1**.

A crystal of the enantiomorph of **1** was selected for a second structure determination. The flack parameter was 0.031(15). In all other respects, the structures of the two enantiomorphs are identical, as expected. CCDC-723094 contains the supplementary crystallographic data for the enantiomorph of **1**. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

X-ray powder diffraction (XRD) data was collected on a Siemens D5005 diffractometer with CuK α radiation ($\lambda = 1.5418$ Å).

TG analysis was performed on Perkin-Elmer TGA 7 thermogravimetric analyzer in the air with a heating rate of 10 °Cmin⁻¹. TG analysis (Supporting Information Figure S9) of **1** shows a total weight loss of ca. 33.98 % between 180 and 750°C, which is attributed to the decomposition of the organic template molecules (calc. 34.23 %). XRD analysis of the residue left after heating to 750°C shows that the material had transformed into a dense $Co_2P_2O_7$ phase (JCPDS card No. 34-1378).

Infrared (IR) spectrum was recorded within the 400–4000 cm⁻¹ region on a Nicolet Impact 410 FTIR spectrometer using KBr pellets.

Diffuse reflectance spectra were collected on Perkin-Elmer Lambda20 within the range 300–800 nm.

The circular dichroism (CD) spectra were recorded on a JASCO J-810 spectropolarimeter with KBr pellets.

The magnetic measurements were carried out on crystalline samples with a MPMS-5 SQUID magnetometer.

Fig. S1 Thermal ellipsoids given at 50% probability, showing the atomic labeling scheme of **1**.

Fig. S2 A schematic view of the right-handed helical channels in which the diprotonated DABCO molecule reside.

Fig. S3 The simulated and experimental X-ray diffraction patterns of compound **1**.

Fig. S4 Thermal analyses (TG and DTA) under air of **1**.

Fig. S5 X-ray diffraction pattern of **1** after heating at 750°C for 2h.

Fig. S6 IR spectrum of **1**.

Fig. S7 The UV-Vis diffuse reflectance spectrum for **1**.

Fig. S8 Circular dichroism spectra of compound **1** in KBr pellet.

Fig. S9 Plot of $\chi_M T$ versus T for **1** in an applied field of 1 kOe from 2 to 300 K; solid line represents the theoretical fit from Curie–Weiss law above 20 K.

Fig. S10 Magnetization versus H for **1** at 1.9 K. Inset: Hysteresis loop at 1.9 K for **1**.

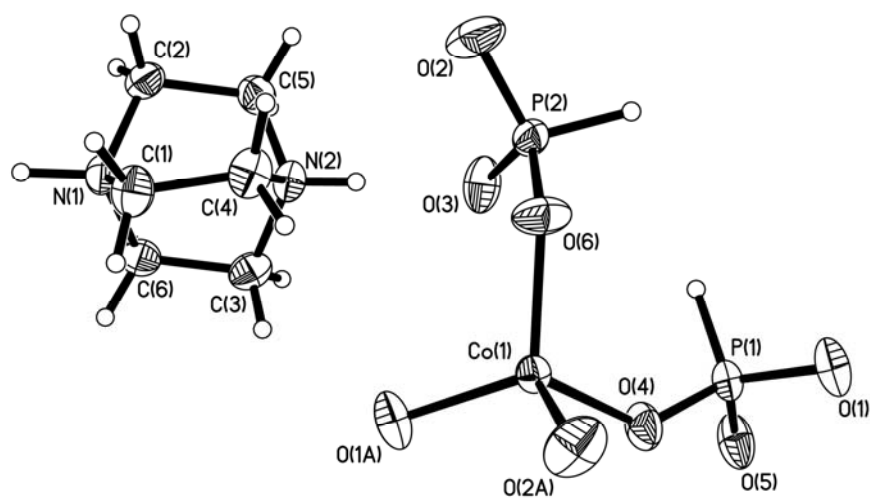


Fig. S1

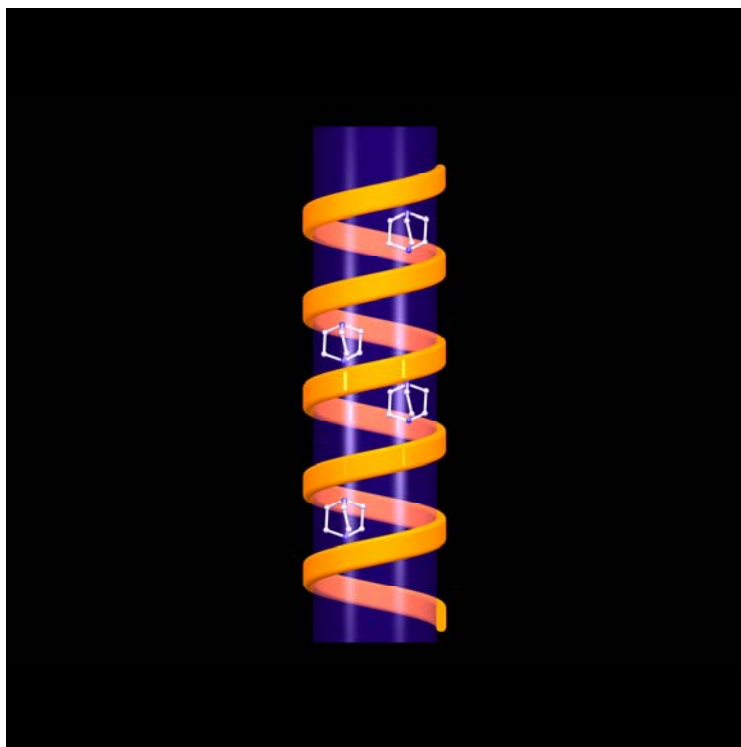


Fig. S2

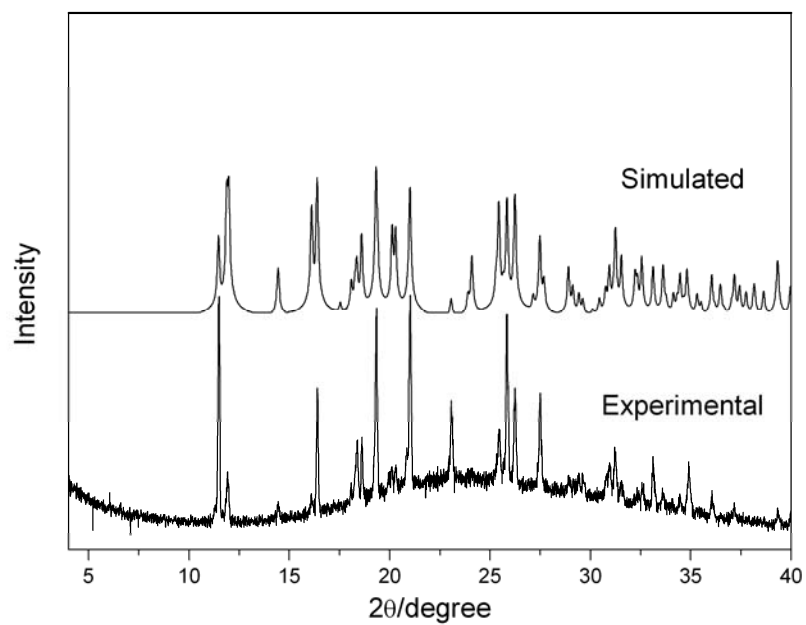


Fig. S3

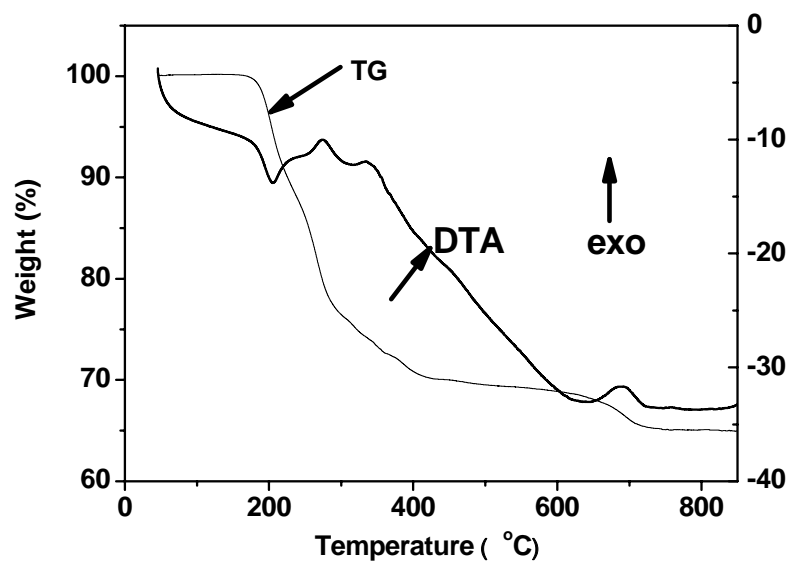


Fig. S4

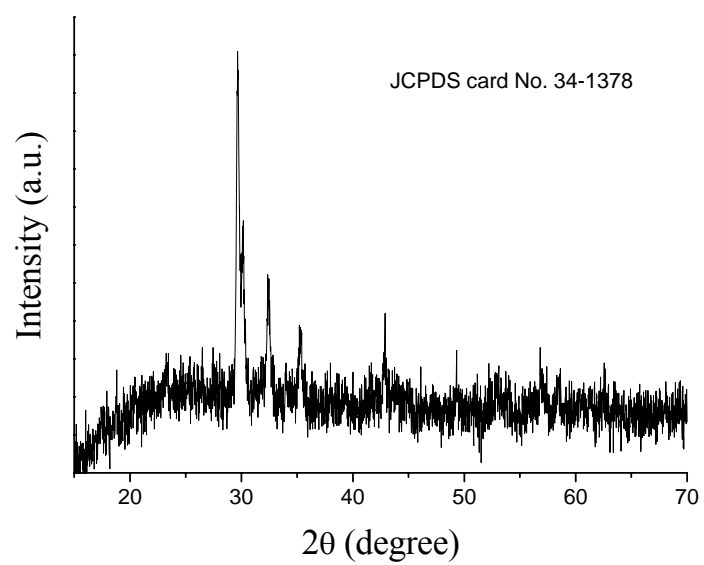


Fig. S5

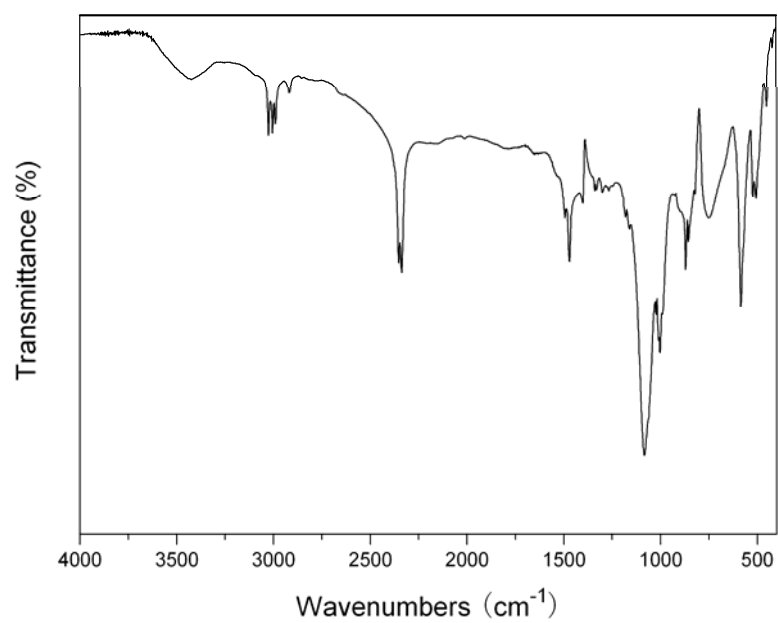


Fig. S6

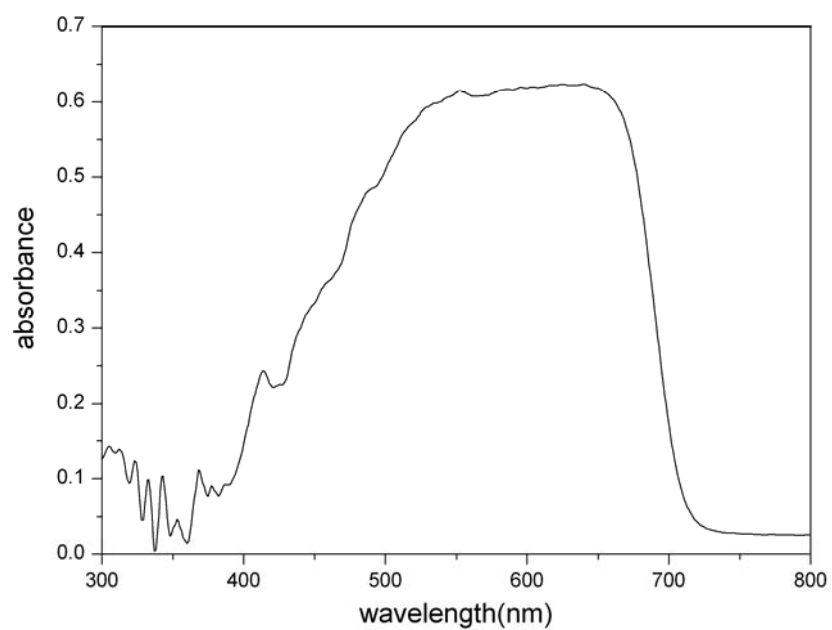


Fig. S7

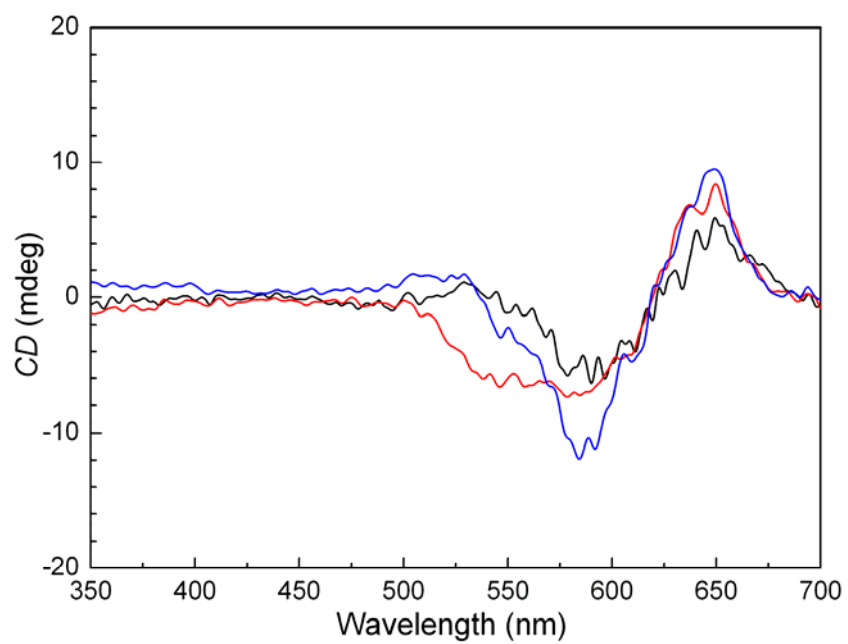


Fig. S8

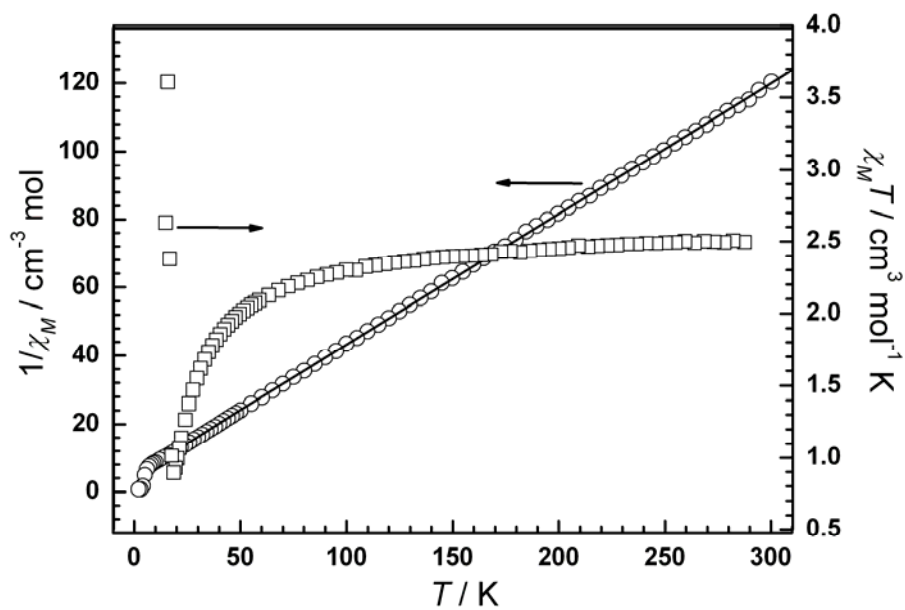


Fig. S9

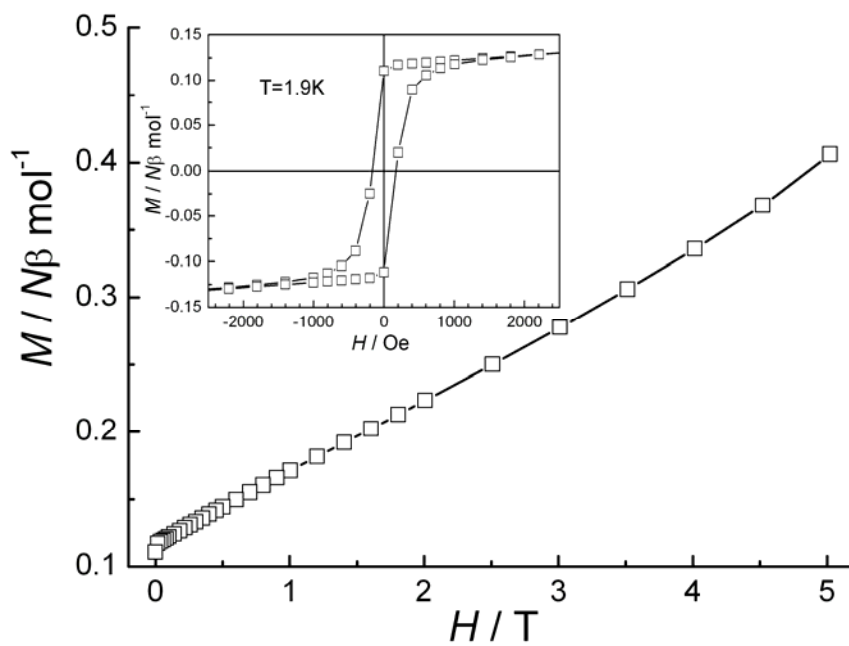


Fig. S10

Table S1 Crystal data and details for the structure determination for **1**.

	Compound 1
Empirical formula	C ₆ H ₁₆ Co N ₂ O ₆ P ₂
Formula weight	333.08
Temperature (K)	293(2)
Wavelength (Å)	0.71069
Crystal system	<i>Orthorhombic</i>
Space group	<i>P2₁2₁2₁</i>
<i>a</i> (Å)	10.1040(4)
<i>b</i> (Å)	10.8100(5)
<i>c</i> (Å)	10.9930(5)
β (°)	90
Volume (Å ³)	1200.70(9)
<i>Z</i>	4
D _{calc} (g cm ⁻³)	1.843
Absorption coefficient (mm ⁻¹)	1.713
<i>F</i> (000)	684
Crystal size (mm)	0.21 x 0.18 x 0.16
θ range (°)	2.64 –26.07
Limiting indices	-12 ≤ <i>h</i> ≤ 12, -13 ≤ <i>k</i> ≤ 13, -8 ≤ <i>l</i> ≤ 13
Reflections collected / unique	6673 / 2368 [R(int) = 0.0180]
Refinement method	Full-matrix least-squares on <i>F</i> ²
Data / restraints / parameters	2368 / 2 / 168
Goodness-of-fit on <i>F</i> ²	1.084
Absolute structure parameter	0.034(14)
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0209, <i>wR</i> ₂ = 0.0603
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0221, <i>wR</i> ₂ = 0.0608
Largest diff. peak and hole (e Å ⁻³)	0.607 and -0.300

Table S2 Selected bond lengths [Å] and angles [°] for **1**.

Co(1)-O(1)#1	1.9581(17)	P(1)-O(1)	1.5128(18)
Co(1)-O(2)#2	1.9337(18)	P(1)-O(4)	1.5170(17)
Co(1)-O(4)	1.9538(16)	P(1)-O(5)	1.5189(17)
Co(1)-O(6)	1.9403(17)	P(2)-O(2)	1.5066(19)
P(1)-H(1)	1.360(10)	P(2)-O(3)	1.5160(19)
P(2)-H(2)	1.371(10)	P(2)-O(6)	1.5104(17)
O(1)#1-Co(1)-O(2)#2	111.88(9)	O(1)-P(1)-O(4)	111.96(10)
O(1)#1-Co(1)-O(6)	111.83(9)	O(1)-P(1)-O(5)	113.26(11)
O(1)#1-Co(1)-O(4)	110.15(7)	O(4)-P(1)-O(5)	110.08(10)
O(2)#2-Co(1)-O(4)	102.10(8)	O(2)-P(2)-O(3)	113.91(12)
O(2)#2-Co(1)-O(6)	109.77(8)	O(2)-P(2)-O(6)	108.85(11)
O(4)-Co(1)-O(6)	110.71(8)	O(3)-P(2)-O(6)	113.18(11)

Symmetry transformations used to generate equivalent atoms:

#1 $-x+1/2, -y, z-1/2$ #2 $-x+1, y+1/2, -z+1/2$